# ABSTRACT <br> Synthesis of Vascular Disrupting Agent Payloads, Protease Specific Linkers, and their Corresponding Constructs 

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The discovery and development of small-molecule anticancer agents that are both highly efficacious and selectively targeted remains a formidable goal and challenge. Conventional chemotherapy regimens often employ the use of cytotoxic agents which fail to differentiate between normal and cancerous tissue and are thus limited by their high degree of toxicity to normal tissue. Antibody-drug conjugates (ADCs) represent an emerging class of biopharmaceutical therapeutics that deliver highly potent anticancer agents (payloads) selectively to tumors or components of the tumor microenvironment. ADCs are comprised of a drug (payload) which is joined through an appropriate linker to an antibody or engineered protein. Pairing both a high degree of selectivity through the antibody with a high degree of cytotoxicity through the payload, ADCs represent an emerging and promising class of chemotherapeutic agents. Two small-molecule payloads (KGP05 and KGP18, previously discovered in the Pinney Laboratory) function as both highly potent antiproliferative agents (cytotoxins) and as highly effective vascular disrupting agents (VDAs). Details regarding the synthesis of these payloads and the preparation of drug-linker constructs tethering these payloads to the protease cleavable dipeptide linker Val-Cit-PABOH, widely used in ADC research, are presented.

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# SYNTHESIS OF VASCULAR DISRUPTING AGENT PAYLOADS, PROTEASE SPECIFIC LINKERS, AND THEIR CORRESPONDING CONSTRUCTS 

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## ABBREVIATIONS

| ${ }^{\circ} \mathrm{C}$ | Degree Celsius |
| :---: | :---: |
| $\AA$ | Angstrom |
| ABC | ATP-Binding Cassette Transporter |
| AcBut | 4-(4-acetylphenoxy) butanoic acid |
| ACN | Acetonitrile |
| ADC | Antibody-Drug Conjugate |
| CA4 | Combretastatin A-4 |
| CA4P | Combretastatin A-4 Phosphate Salt |
| Cit | Citrulline |
| DIPEA | $N, N$-Diisopropylethylamine |
| DM1 | $N$-2'-Deacetyl-N2'-(3-mercapto-1-oxopropyl) maytansine |
| DMF | $N$, N -Dimethylformamide |
| DNA | Deoxyribonucleic Acid |
| ESI | Electrospray ionization |
| h | Hour |
| HPLC | High Pressure Liquid Chromatography |
| Hz | Hertz |
| KOtBu | Potassium tert-butoxide |
| SEM | Scanning Electron Microscopy |
| mAb | Monoclonal Antibody |
| Mc | Maleimidocaproyl |


| MCC | 4-[N-maleimidomethyl] cyclohexane-1-carboxylate |
| :--- | :--- |
| MDR | Multi-drug resistance |
| MHz | Megahertz |
| min | Minutes |
| mM | Millimole |
| MMAE | Monomethyl Auristatin E |
| NMR | Nuclear Magnetic Resonance Spectroscopy |
| OSu | O-Succinimide |
| PABOH | Para-aminobenzyl alcohol |
| PEG4Mal | 4-arm Polyethylene glycol-maleimide |
| PNP | $P$-Nitrophenol |
| SEM | Scanning Electron Microscopy |
| SMCC | Succinimidyl 4-(N-maleimidomethyl) cyclohexane-1-carboxylate |
| SPDB | $N$-succinimidyl-4-(2-pyridyldithio) butanoate |
| THF | Tetrahydrofuran |
| Val | Valine |
| VDA | Vascular Disrupting Agent |

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## DEDICATION

Honors thesis by Abigail Brinson Smith, dedicated
To,
My Loving Family,
Supportive Friends,
and
Encouraging Mentors

# CHAPTER ONE 

Background

## Cancer and Nature of Vasculature in Tumors

The National Cancer Institute defines cancer as a collection of related diseases that are connected by the general characteristic of a sub-set of cells that begin to continuously divide and spread into surrounding tissues. ${ }^{1}$ This is driven by mutations within a cell's DNA, prohibiting the natural growth and division checkpoints from being observed, leading to uncontrolled growth. ${ }^{1,2}$ This continuous division within a specific tissue eventually leads to the formation of a tumor from which malignant cells can break off of and invade other healthy tissues and form new growths around the body in an event known as metastasis. ${ }^{1}$ Without early detection and treatment, cancer can lead to severe health consequences, including death. Various methods have been sought to cure this disease including radiation, surgery, and chemotherapeutic drugs - the latter of the three being the focus of this thesis. ${ }^{2}$ These drugs serve to kill malignant cells; however, major problems with these drugs are specificity and cytotoxicity to cancer cells rather than healthy cells. ${ }^{2}$ Within anti-cancer biopharmaceutical drug research field is a branch that focuses on targeting the tumor's blood supply, or vasculature, as it embodies unique and targetable characteristics within a tumor. ${ }^{2}$

In healthy tissue, cells are provided nutrients through a complex and well-organized system of blood vessels and vasculature that deliver key metabolic components to tissue beds. ${ }^{3}$ These systems of vessels are maintained by pro-angiogenic processes and antiangiogenic processes that balance each other out in order to meet the demands of the
tissue. ${ }^{3}$ This process of developing new vessels from pre-existing ones is known as angiogenesis. ${ }^{3}$ However, once a cell becomes cancerous and begins the process of rapid division, it must quickly recruit additional vasculature to deliver more nutrients to meet the demands of the highly metabolic cancerous cells. ${ }^{3}$ The balance between pro-angiogenic and anti-angiogenic processes becomes severely skewed towards pro-angiogenesis without being deterred by the normal checkpoints. ${ }^{3}$ In doing so, the previously well-organized network of blood vessels is forced to haphazardly branch, forming a weak, brittle knot of vasculature within the developing tumor. ${ }^{3}$ The rapid rate at which these vessels grow contributes to their brittle nature, thus creating a distinct target for cancer therapeutic drugs. ${ }^{4}$ The most effective and correspondingly safest cancer therapeutic agents are able to target cancer cells while limiting deleterious effects to healthy cells. ${ }^{4}$


| 0 | 50 | 100 |
| :--- | ---: | ---: |
| $\mathrm{P}_{\mathrm{O}_{2}}$, | mmHg |  |

Figure 1: Panel A depicts a Scanning Electron Microscopy (SEM) of normal lung tissue. Panel B depicts an SEM of a human sigmoidal adenocarcinoma. The contrast between these two images reveals the obvious unorganized nature of tumor vasculature in comparison to that of normal tissue. Panels C and D display the differences in organization and distribution of $\mathrm{PO}_{2}$ of normal and cancerous tissue. These differences offer an ability for partial selectivity of therapeutic drugs. ${ }^{3}$ Reprinted from Cancer Treatment Reviews, Vol. 37, Dietmar W. Siemann, The unique characteristics of tumor vasculature and preclinical evidence for its selective disruption by Tumor-Vascular Disrupting Agents, Page No. 64, Copyright (2011), with permission from Elsevier.

Targeting the vascular network in cancer treatments comes in two forms. ${ }^{3}$ On one hand, drugs can target the process of the development of new vessel networks, otherwise known as the anti-angiogenic approach. ${ }^{4}$ Alternatively, drugs can target the already developed, but compromised blood vessel network in tumors, known as the antivasculature approach. ${ }^{4}$ These two approaches have also been used in conjunction with one another; the anti-angiogenesis drugs can prevent the tumor from growing larger and the anti-vasculature drugs can cut of blood supply to the interior of the tumor. ${ }^{4}$ In theory, together, these drugs can kill a tumor by starving it of the nutrients and oxygen it so heavily relies on. ${ }^{4}$ These unique microenvironment features allow for drugs to be developed that target these specific endothelial cells in tumors and lead to massive tissue necrosis. ${ }^{4}$


Figure 2: Diagram displaying the two different approaches to targeting tumor vasculature. ${ }^{3}$ Reprinted from Cancer Treatment Reviews, Vol. 37, Dietmar W. Siemann, The unique characteristics of tumor vasculature and preclinical evidence for its selective disruption by Tumor-Vascular Disrupting Agents, Page No. 65, Copyright (2011), with permission from Elsevier.

## Natural Products and Tubulin Inhibition

The biological effects associated with inhibition of tubulin polymerization include both antiproliferative (cytotoxic) behavior as well as destruction of tumor-associated vasculature mediated by morphology (shape) changes in the long, thin endothelial cells causing them to become round, thereby affecting the integrity of the blood vessel walls by introducing leaky holes in the vessel network. ${ }^{3}$ This subsequently, leads to reduced red blood cell flow which leads to red blood cell stacking which leads to high blood viscosity. ${ }^{3}$ All of these factors result in the insufficient delivery of nutrients to the core of the tumor. ${ }^{3}$ Once the nutrient supply has been cut off to the center of the growth, tumor necrosis ensues. ${ }^{3}$


Figure 3: This image depicts the effects of vascular disruption. ${ }^{5}$ Reprinted by permission from Springer Nature: Nature Reviews Cancer, Disrupting tumour blood vessels, Gillian M. Tozer et al, Copyright (2005).

In order to induce this conformational change, it was discovered that the natural product, colchicine was a strong inhibitor of tubulin polymerization and was initially investigated for potential use in cancer therapies. However, the use of colchicine had undesired effects that ultimately diminished its utility as a chemotherapeutic agent; however, hope remained as the colchicine binding site on tubulin could still be targeted through synthesis of colchicine derivatives and analogues. Some adverse effects of colchicine include gastrointestinal irritation at low doses, while at higher doses more serious implications such as hepatocellular damage, CNS toxicity, respiratory depression, and cardiovascular collapse may occur making it too dangerous to be seriously considered as a chemotherapeutic. ${ }^{6}$ However, in smaller doses colchicine is currently approved to medically treat acute gout and recurrent pericarditis. ${ }^{6}$

The discovery and development of new small-molecule inhibitors of tubulin polymerization that interact at the colchicine binding site offers the potential promise for
improved therapeutic efficacy and selectivity while minimizing the unintended side-effects inherent to colchicine. ${ }^{6}$ The natural products combretastatin A-1 (CA1) and combretastatin A-4 (CA4) were isolated from the African bush-willow tree, Combretum caffrum, by Pettit and co-workers. ${ }^{7}$ These molecules where discovered to be potent inhibitors of tubulin polymerization, like colchicine, but with potentially fewer and less serious adverse effects, thus offering better promise as chemotherapeutic agents in comparison to colchicine (Figure 6). ${ }^{7}$ From structure activity relationship (SAR) studies of the combretastatins, it was found that optimum inhibition of tubulin polymerization occurred when the following features were present: (1) the cis $(Z)$ geometry about the alkene bridge, (2) the $4-5 \AA$ distance between the aryl moieties, and (3) the trimethoxy phenyl group. ${ }^{8}$ The payloads synthesized by The Pinney Group, KGP05 and KGP18, contain these three important structural features (Figure 4). ${ }^{9-15}$ These VDAs bind to the colchicine binding site on the $\beta$ tubulin subunit of the $\alpha \beta$-tubulin heterodimer leading to tubulin inhibition and ultimately the disassociation of the endothelial cells lining the microvessels associated with the tumor leading to internal tumor necrosis. ${ }^{3,11,12}$ The Pinney Group has found high degrees of cytotoxicity against cancer cell lines in their design and development of structurally diverse benzosuberene-based molecules along with corresponding analogues and derivatives. ${ }^{12-15}$


Figure 4: This image depicts the similarities between the tubulin inhibitors colchicine, CA1, CA4, and the synthetic payloads KGP05, KGP18, KGP156, and OXi6196.

The drug-linker constructs described in this thesis are intended to offer a high degree of selectivity for the tumor endothelial cells and a high degree of cytotoxicity through tubulin inhibition. Building on previous synthetic protocols established by the Pinney Group, the synthesis of these small-molecule therapeutic agents was modified to enhance overall yield and decrease production of undesired side products. ${ }^{12,13,16,17}$ The linker Mc-Val-Cit-PABOH was synthesized and subsequently covalently tethered to the tubulin polymerization inhibitor, KGP05, and the resultant drug-linker construct is currently undergoing biological evaluation for its potential utility as a component in various ADCs and related conjugates. ${ }^{18}$

## Antibody-Drug Conjugates and Payloads

ADCs consist of three basic components: the monoclonal antibody, the linker, and the drug payload. ${ }^{18-21}$ Each piece of the ADC brings about an advantage to the overall compound in either increasing its selectivity or its cytotoxicity. ${ }^{7}$

VDAs are highly cytotoxic, but alone are only effective if they can impart selectivity to the endothelial cells lining the microvessels feeding tumors. ${ }^{18}$ Thus, it is important when constructing an ADC to evaluate whether the mechanism of drug release is selective to conditions unique to the target cell or tumor microenvironment. ${ }^{18}$ The tumor vasculature can be targeted through direct apoptotic effects or by inhibiting tubulin polymerization in order to alter endothelial cell shape and consequently preventing the vessel network from delivering nutrients and oxygen throughout the tumor. ${ }^{3}$ In other words, disrupting the endothelial cell's shape, will impart severe consequences on its capacity to carry out its intended function. ${ }^{3}$ The drugs discussed herein have been synthesized with the intention of altering the tubulin polymerization of tumor endothelial cells and have shown to be successful in this way. ${ }^{9,10}$ The payloads KGP05 and KGP18 both inhibit the ability of tubulin to function properly. ${ }^{9,10}$ This benzosuberene analog (KGP18) and dihydronaphthalene analog (KGP05) have been shown in biological studies to possess strong cytotoxicity (low nM to pM range) due to their ability to inhibit tubulin polymerization in human cancer cell lines. ${ }^{22}$

A major obstacle faced in the past with VDAs is multidrug resistance (MDR) facilitated by ATP-binding cassette (ABC) transporters in the cell membrane which function to pump out any foreign material present in the cell, severely decreasing the potency and efficiency of these VDAs; however, to circumvent this issue, there have been
studies looking into inhibiting these ABC transporters to further increase a drug's intracellular effect. ${ }^{23}$ Moving forward, when testing the cytotoxicity of the VDAs synthesized herein, it would be interesting to compare their effect alone versus in the presence of an ABC transporter inhibitor. This problem was faced with the FDA approved, Mylotarg ${ }^{\mathrm{TM}}$, and is discussed in greater detail later on.

Another major obstacle with tumor VDAs, is the problem of the viable rim. ${ }^{5}$ The tumor cells on the outer edge of the solid mass are able to remain viable despite the necrosis of the internal aspect of the tumor (Figure 5). ${ }^{5}$ The periphery is less sensitive to vascular disruption which has been contributed to cells ability to extract nutrients and oxygen from the surrounding tissues. ${ }^{5}$ However, studies have shown the problem is more complex revealing that the vasculature itself at the rim of the tumor is less affected than at the center. ${ }^{5}$ This could be explained by the differences in interstitial pressure and the vascular architecture. ${ }^{5}$ Interstitial pressure steadily rises moving from the rim to the center of a tumor and as a result, tumor microvessel permeability increases and becomes susceptible to losing vascular integrity. ${ }^{5,24}$ Further, with respect to the architecture, there are more small caliber vessels present at the center than at the rim of the tumor, which display a much higher sensitivity to the increase in interstitial pressure than the larger vessels present at the rim. ${ }^{5,25}$ This problem is circumvented by pairing VDA drugs with antibodies as the tumor periphery is the only tumor region that is easily accessible by high-molecular-weight therapeutics such as antibodies, making the combination of VDAs with these new approaches very attractive with ADC therapeutics. ${ }^{5,26}$


Figure 5: This image depicts the viable (V) cells at the rim of a tumor versus the necrotic $(\mathrm{N})$ cells at the center of the tumor in a hematoxylin and eosin staining of paraffin embedded tissue of a mouse mammary carcinoma. ${ }^{5}$ Reprinted by permission from Springer Nature: Nature Reviews Cancer, Disrupting tumour blood vessels, Gillian M. Tozer et al, Copyright (2005).

Additionally, proper function of the ADCs relies on a stable linker. ${ }^{20}$ The linker is structurally responsible for the connection between the antibody and payload and functionally responsible for directing the subsequent release of the payload once the compound has reached the targeted site through either lysosomal processing of the antibody or a cleavable linker. ${ }^{3,20}$ Its specific purpose is to keep the VDA drug (payload) in connection to the monoclonal antibody in order to minimize or prevent non-specific release. ${ }^{18}$ There are cleavable linkers and there are non-cleavable linkers - this classification is assigned based on the mechanism of drug release (Figure 6). ${ }^{18}$ Noncleavable linkers rely on lysosomal degradation in order for drug release to occur, while cleavable linkers are sensitive to subtle changes in physiological conditions that can result in the release of the drug. ${ }^{18}$ Further, cleavable linkers can be classified as either chemically labile or enzymatically labile. ${ }^{18}$ Chemically labile linkers can be cleaved in acidic
environments or by an abundance of glutathione which reduces disulfide based linkers, while enzymatically labile linkers are cleaved by proteases that are highly expressed in cancer cells. ${ }^{18}$ The linker synthesized herein was designed for selective proteolytic cleavage by cathepsin B. ${ }^{4,18,30}$

The first FDA approved ADC was Mylotarg ${ }^{\circledR}$ in 2000 that utilizes a non-peptide based linker; it was withdrawn from the market due to clinical failures, but has since been re-approved. ${ }^{21,27}$ Mylotarg ${ }^{\circledR}$ clinical testing was expedited by the FDA and granted accelerated approval for patients with acute myeloid leukemia (AML) aged 60 years and older who were not eligible for cytotoxic chemotherapy. ${ }^{28}$ Once more testing was complete, it was observed that as a result of Mylotarg®, patients suffered severe hepatotoxicity in the form of sinusoidal obstructive syndrome (SOS) or hepatic venoocclusive disease (VOD); Mylotarg ${ }^{\circledR}$ offered no survival benefit in comparison to standard chemotherapeutic treatments and was complicated by severe adverse effects, prompting its subsequent removal from the market. ${ }^{28}$ One proposed reason for this was that the cancer cells exhibited an overabundance P-glycoprotein (a type of ABC transporter) expressed in tumor cells which mediated drug-resistance. This is due to the fact that when the conjugate is processed intracellularly, hydrophobic cytotoxic metabolites are generated which are prime substrates for P-glycoprotein and are subsequently pumped out of the target cell. ${ }^{28}$ In order to circumvent this issue of cell-mediated drug resistance, Kovtun and colleagues have described a maleimidyl-based hydrophilic linker, PEG4Mal (Figure 6), that appears to be a poor substrate for P-glycoprotein, thereby enabling the cytotoxic drug to remain inside the cell. ${ }^{28,29}$ Enabling the cytotoxic drug to remain inside the cell has a double benefit as it enables the drug to impart its cytotoxic effects intracellularly, and it reduces the
severity of adverse effects as the amount of free circulating drug is minimized. ${ }^{28}$ Since its removal from the market, Mylotarg ${ }^{\circledR}$ has been reapproved by the FDA as it was observed that Mylotarg ${ }^{\circledR}$ showed significantly less adverse effects for patients with acute promyelocytic leukemia (APL), a subtype of AML, due to the significantly lower presence of P-glycoprotein in these cancer cells. ${ }^{28}$

Especially prevalent is the cathepsin B cleavable Mc-Val-Cit-PABOH linker utilized in many pre-clinical ADC candidates, as well as the FDA approved ADC Adcetris ${ }_{\circledR}$ (brentuximab vedotin) in 2011. ${ }^{21}$ An alternative route for the synthesis of the cathepsin B cleavable Mc-Val-Cit-PABOH linker is reported herein that involved six steps from $l$-Citrulline. ${ }^{18}$ Importantly, this route avoided undesirable epimerization and proceeded with improved overall yield. Utilizing this methodology, a drug-linker construct incorporating a potent small-molecule inhibitor of tubulin polymerization (referred to as KGP05), was synthesized as a representative example. This linker was designed to be a cathepsin B cleavable dipeptide moiety for lysosomal release of the attached prodrug; however, biological studies suggest that it can also be cleaved extracellularly. ${ }^{18,30,31}$ For example, it was observed that antigens such as CD20, CD21, CD22, which are known to be poorly internalized worked well in mouse cancer models with an anti-lymphoma ADC products. ${ }^{32,33}$ This indicates that the ADC can maintain its cytotoxic effects even when the payload is unleashed outside of the cell. Because of this, using antibodies specific to fibrin or to collagen IV (extracellular matrix components) have been proposed as an attractive tumor targeting strategy known as the Cancer stromal targeting (CAST) approach. ${ }^{32,34,35} \mathrm{It}$ has been frequently observed in solid tumors that the blood coagulates at a higher than normal frequency which then subsequently creates a targetable fibrin deposition. ${ }^{32,34,35}$

This is important because in vivo studies have shown that antibodies have a slow rate of diffusion to the center of a solid tumor; however, if ADCs can target features of the outer rim tumor extracellular space and release their drug payloads here, they can diffuse and subsequently internalize into the innermost tumor cells, endothelial cells and in tumorresident leukocytes at a faster rate..$^{32,34,35}$

A: Cleavable Linkers




Linker 3
Linker 4


B: Non-Cleavable Linkers


Linker 5

Linker 6

Figure 6: This image displays the wide variety and diversity in linker structures that can be used in ADCs. Panel A presents cleavable linkers: Linker 1 is a lysosomal protease sensitive Val-Cit dipeptide linker; Linker 2 is a Glutathione sensitive SPDB linker; Linker 3 is an acid sensitive AcBut linker; and Linker 4 is a $\beta$-Glucuronidase sensitive linker. Panel B presents non-cleavable linkers: Linker 5 is a SMCC linker; and Linker 6 is a PEG4Mal linker. ${ }^{36}$

Finally, the major source of drug specificity arises from the practice of tethering a monoclonal antibody ( mAb ) to the linker in order to deliver the highly cytotoxic payload with a high degree of specificity to antigens that are over-expressed in cancer cells. ${ }^{20}$

Determining which mAb will provide the highest degree of specificity is a whole area of research in itself. ${ }^{37}$ The resulting physiological and morphological differences in tumors give rise to the expression of certain types of antigens that wouldn't normally be available for targeting. ${ }^{37}$ It is complicated by the fact that every cancer is different, and every cancer presents different types of antigens on its surface and in different concentrations. ${ }^{37}$ For example, even within the realm of breast cancer there are many different antigens that could be expressed depending on the specific type of cancer; specifically the ADC, Kadcyla ${ }^{\mathrm{TM}}$, targets the antigen HER2 which is often overexpressed in breast cancer., ${ }^{3,38}$

The overall mechanism of the ADC begins when the antibody attaches to the specific antigen expressed on the surface of the cancerous cell. ${ }^{20}$ This can be followed either by receptor mediated endocytosis or extracellular release of the payload into the tumor microenvironment (Figure 7). ${ }^{3,20}$ The mechanism of action depends on the antigen targeted and the type of linker constructed. ${ }^{20}$ The proteolytic cleavage can occur on the surface of the cells after the antibody attaches to the antigen, or it can occur within the cell during lysosomal degradation. ${ }^{4,20}$


Figure 7: The image depicts the ADC mechanism of action in the delivery of a payload to a target cell.

Currently, the FDA has approved seven ADCs, two of which employ smallmolecule inhibitors of tubulin polymerization as their payloads. ${ }^{37,39}$ The seven that are currently approved include: Ado-trastuzumab emtansine (Kadcyla ${ }^{\mathrm{TM}}$ ), Brentuximab vedotin (Adcetris ${ }^{\text {TM }}$ ), Inotuzumab ozogamicin (Besponsa ${ }^{\text {TM }}$ ), Gemtuzumab ozogamicin (Mylotarg ${ }^{\mathrm{TM}}$ ), Polatuzumab vedotin-piiq (Polivy ${ }^{\mathrm{TM}}$ ), Enfortumab vedotin (Padcev ${ }^{\mathrm{TM}}$ ), and Trastuzumab deruxtecan (Enhertu ${ }^{\mathrm{TM}}$ ). ${ }^{39}$ Three of the seven FDA approved ADCs, Brentuximab vedotin (Adcetris ${ }^{\text {TM }}$ ), Polatuzumab vedotin-piiq (Polivy ${ }^{\text {TM }}$ ), and Enfortumab vedotin ( $\operatorname{Padcev}^{\mathrm{TM}}$ ), have similar chemical structures and all target microtubule polymerization through a Val-Cit Cathepsin B cleavable linker and MMAE payload tethered to a different antibody ( mAb ) based on type of cancer cell targeted. ${ }^{40,41}$ Inotuzumab ozogamicin (Besponsa ${ }^{\text {TM }}$ ) and Gemtuzumab ozogamicin (Mylotarg ${ }^{\text {TM }}$ ) both feature payloads that function to cause double stranded breaks in DNA in target cells. ${ }^{42,43}$

Finally, Trastuzumab deruxtecan (Enhertu ${ }^{\mathrm{TM}}$ ) which features a payload that functions as a topoisomerase I inhibitor. ${ }^{44}$


Figure 8: Depicted is the structure of the FDA approved ADC, Ado-trastuzumab emtansine (Kadcyla ${ }^{\mathrm{TM}}$ ) featuring a Tratuzumab antibody, MCC linker, and the DM1 payload which serves to inhibit tubulin polyermization. ${ }^{45}$

The four that employ small-molecule inhibitors of tubulin polymerization as their payloads are Brentuximab vedotin (Adcetris ${ }^{\circledR}$ ) for Hodgkin's and anaplastic large cell lymphomas (Figure 9), ado-trastuzumab emtansine (Kadcyla ${ }^{\circledR}$ ) for the treatment of HER2positive breast cancer (Figure 8), Enfortumab vedotin (Padcev ${ }^{\mathrm{TM}}$ ) for urothelial cancer, and Polatuzumab vedotin-piiq (Polivy ${ }^{\text {TM }}$ ) for relapsed or refractory diffuse large B-cell lymphoma. ${ }^{37}$ The drug, Adcetris ${ }^{\circledR}$ tethers a mAb targeting CD30 antigens to a cathepsin B protease cleavable linker which is then attached to a mono-methyl auristatin E (MMAE) payload (Figure 9). ${ }^{46}$ Adcetris ${ }^{\circledR}$ is a strong model for future drugs because the linker that
is employed in this drug is the dipeptide, Valine-Citrulline, which is selectively cleaved by the protease cathepsin $\mathrm{B} .{ }^{46}$ Cathepsin B is a good target due to its high abundancy specifically in the tumor microenvironment. ${ }^{46}$ Once cleavage has occurred by cathepsin B, the spacer (para-amino benzyl alcohol moiety) then releases the payload at the targeted site through an elimination reaction. ${ }^{46}$ The spacer is a moiety that is self-immolative and undergoes a spontaneous 1,6-elimination resulting in release of the prodrug payload within the cell. ${ }^{18,47}$ There have been many studies conducted on the Mc-Val-Cit-PABOH linker in the ADC model of drug delivery; however, the work reported herein is novel in that it attaches unique and highly potent small-molecule VDAs to this promising linker.


Figure 9: Adcetris ${ }^{\circledR}$ mechanism of drug release.

## CHAPTER TWO

Materials and Methods

## General Information

Tetrahydrofuran (THF), dichloromethane (DCM), dimethylformamide (DMF), and methanol (MeOH) were used in anhydrous forms. All of the reactions were carried out under an atmosphere of nitrogen in oven-dried glassware with magnetic stirring unless otherwise specified. Analytical thin layer chromatography was performed on 0.25 mm thick, silica gel 60-F254 plates. Visualization was accomplished with UV light or aqueous potassium permanganate solution staining followed by air heating. Purification of reaction products was carried out on a Biotage Isolera flash purification system using silica gel (200-400 mesh, $60 \AA$ ).
${ }^{1} \mathrm{H}$ NMR spectroscopic data were recorded on a Bruker Ascend $600(600 \mathrm{MHz})$ spectrometer and are reported in ppm using solvent as internal standard (e.g. DMSO-d6 at $2.50 \mathrm{ppm})$. Data are reported as: $\mathrm{b}=$ broad, $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{p}=$ pentet, $\mathrm{m}=$ multiplet; coupling constant(s) are expressed in $\mathrm{Hz} .{ }^{13} \mathrm{C}$ NMR spectroscopic data were recorded on a Bruker Ascend $600(150 \mathrm{MHz})$ spectrometer. Chemical shifts are reported in ppm with solvent resonance employed as the internal standard (e.g. DMSO-d6 at 39.5 ppm ). High-resolution mass spectra were obtained under positive ESI using a Thermo Scientific LTQ Orbitrap. All the reagents were purchased from common suppliers.

HPLC data were recorded on an Agilent 1200 HPLC system with a diode-array detector $(\lambda=190-400 \mathrm{~nm})$, a Zorbax XDB-C18 HPLC column ( $4.6 \mathrm{~mm} \times 150 \mathrm{~mm}, 5 \mu \mathrm{~m}$ ), and a Zorbax reliance cartridge guard column and are reported in percent purity. The
method used to obtain this data was through a $30-90 \mathrm{ACN}$ gradient with a flowrate of 1 mL per minute (Table 1).

| $30-90 \mathrm{ACN}$ | Percent Water | Percent ACN |
| :--- | ---: | ---: |
| 0 min | 70 | 30 |
| 25 min | 10 | 90 |
| 30 min | 10 | 90 |

Table 1: 30-90 ACN method used in obtaining HPLC data.

## Synthesis of Mc-Val-Cit-PABOH Linker

Synthesis of Fmoc-Citrulline (Compound 1) ${ }^{18}$


A solution of $l$-citrulline $(1.500 \mathrm{~g}, 8.562 \mathrm{mmol}, 1.100 \mathrm{eq})$ in water $(0.2 \mathrm{M})$ was treated with sodium bicarbonate $(1.439 \mathrm{~g}, 17.13 \mathrm{mmol}, 2.201 \mathrm{eq})$. In a separate container, a solution of $\mathrm{Fmoc}-\mathrm{Cl}(2.013 \mathrm{~g}, 7.784 \mathrm{mmol}, 1.000 \mathrm{eq})$ in THF ( 0.2 M ) was prepared. Following stirring for 30 minutes at room temperature, the $\mathrm{Fmoc}-\mathrm{Cl}$ solution was added to the reaction. The reaction was allowed to stir for 24 hours at room temperature before removing the THF under reduced pressure. The aqueous layer was extracted (3X) with ethyl acetate and the organic layer was discarded. The aqueous layer was then treated with $\mathrm{HCl}(2 \mathrm{M})$ until a white precipitate was observed. The organic layer was then extracted (3X) following the addition of PrOH [ $10 \%$ by volume] - EtOAc. The organic phase was dried over sodium sulfate. The solvent was removed under reduced pressure. The remaining white solid was then sonicated and triturated with diethyl ether, which was then decanted off. This step was repeated twice to obtain a white solid which was dried under reduced pressure to obtain an analytically pure sample of Fmoc-Cit ( $3.165 \mathrm{~g}, 7.963 \mathrm{mmol}$, $93 \%$ yield). No further purification was required. ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 12.56$
$(\mathrm{s}, 1 \mathrm{H}), 7.89(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.73(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.67(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.42(\mathrm{t}$, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.28(\mathrm{~m}, 2 \mathrm{H}), 5.94$ (brs, $J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.38$ (brs, 2H), $4.31-4.26$ (m, 2H), $4.23(\mathrm{q}, J=8.0,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.93(\mathrm{~m}, J=9.5,8.0,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.95(\mathrm{brt}, J=6.6$ $\mathrm{Hz}, 2 \mathrm{H}), 1.70(\mathrm{~m}, 1 \mathrm{H}), 1.57(\mathrm{~m}, J=14.1,9.8,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.49-1.32(\mathrm{~m}, 2 \mathrm{H}), 1.03(\mathrm{~d}, J$ $=6.1 \mathrm{~Hz}, 1 \mathrm{H})$.

$$
\text { Synthesis of Fmoc-Cit-PABOH (Compound 2) }{ }^{18}
$$



A solution of Fmoc-Citrulline ( $1.037 \mathrm{~g}, 2.620 \mathrm{mmol}, 1.000 \mathrm{eq})$ and 4-aminobenzyl alcohol $(0.964 \mathrm{~g}, 7.86 \mathrm{mmol}, 3.00 \mathrm{eq})$ in DMF $(0.2 \mathrm{M})$ was treated with DIPEA $(0.456 \mathrm{~mL}$, $2.62 \mathrm{mmol}, 1.00 \mathrm{eq})$. Following stirring at room temperature for 15 minutes, HATU was added to the reaction. The solution was stirred for 48 hours in the dark at room temperature. DMF was removed under reduced pressure and the crude product was purified by flash column chromatography using $2-10 \% \mathrm{MeOH}-\mathrm{DCM}$ as solvent. Product was isolated as a white solid. ( $0.443 \mathrm{~g}, 0.882 \mathrm{mmol}, 39 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 9.99$ (s, $1 \mathrm{H}), 7.89(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.75(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.56(\mathrm{~d}, J$ $=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.44-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.24(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.00$ (brt,
$J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.43(\mathrm{~s}, 2 \mathrm{H}), 5.10(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.43(\mathrm{~d}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.29-4.12$ $(\mathrm{m}, 4 \mathrm{H}), 3.07-2.99(\mathrm{~m}, 1 \mathrm{H}), 2.98-2.89(\mathrm{~m}, 1 \mathrm{H}), 1.72-1.63(\mathrm{~m}, 2 \mathrm{H}), 1.62-1.54(\mathrm{~m}$, $2 \mathrm{H}), 1.50-1.42(\mathrm{~m}, 1 \mathrm{H}), 1.42-1.33(\mathrm{~m}, 1 \mathrm{H})$.

Synthesis of Fmoc-Val-Cit-PABOH (Compound 3) ${ }^{18}$


A solution of Fmoc-Cit-PABOH ( $0.594 \mathrm{~g}, 1.180 \mathrm{mmol}, 1.00 \mathrm{eq}$ ) in DMF ( 0.2 M ) was treated with trimethylamine ( $3.31 \mathrm{~mL}, 23.8 \mathrm{mmol}, 20.1 \mathrm{eq}$ ). The reaction was stirred for 24 hours at room temperature. DMF and excess trimethylamine were removed under reduced pressure. Trace trimethylamine was removed by co-evaporation with DMF. The resulting residue was dissolved in DMF ( 0.1 M ) and to the solution, Fmoc-Val-OSu (0.568 $\mathrm{g}, 1.30 \mathrm{mmol}, 1.10 \mathrm{eq})$ was added. The reaction was stirred for 20 hours at room temperature. DMF was removed under reduced pressure and the resulting residue was purified by flash column chromatography using 3-12 \% MeOH-DCM as solvent. Product was obtained as a white solid. ( $0.462 \mathrm{~g}, 0.767 \mathrm{mmol}, 65 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , DMSO- $d_{6}$ ) $\delta 9.98(\mathrm{~s}, 1 \mathrm{H}), 8.11(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.89(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.74(\mathrm{dd}, J=$ $10.2,7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.54(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.47-7.38(\mathrm{~m}, 3 \mathrm{H}), 7.32(\mathrm{dd}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H})$,
$7.23(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 5.97(\mathrm{t}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.41(\mathrm{~s}, 2 \mathrm{H}), 5.10(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H})$, $4.45-4.39(\mathrm{~m}, 3 \mathrm{H}), 4.34-4.19(\mathrm{~m}, 3 \mathrm{H}), 3.93(\mathrm{dd}, J=9.0,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.06-2.97(\mathrm{~m}$, $1 \mathrm{H}), 2.97-2.89(\mathrm{~m}, 1 \mathrm{H}), 2.03-1.92(\mathrm{~m}, 1 \mathrm{H}), 1.75-1.64(\mathrm{~m}, 1 \mathrm{H}), 1.63-1.53(\mathrm{~m}, 1 \mathrm{H})$, $1.43(\mathrm{q}, J=7.4,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.40-1.31(\mathrm{~m}, 1 \mathrm{H}), 0.87(\mathrm{dd}, J=14.0,6.7 \mathrm{~Hz}, 6 \mathrm{H})$.

## Synthesis of Mc-Val-Cit-PABOH ${ }^{18}$



A solution of Fmoc-Val-Cit-PABOH $(0.725 \mathrm{~g}, 1.20 \mathrm{mmol}, 1.00 \mathrm{eq})$ in DMF $(0.2$ M) was treated with trimethylamine ( $3.36 \mathrm{~mL}, 24.0 \mathrm{mmol}, 20.0 \mathrm{eq}$ ). The reaction stirred for 24 hours at room temperature. DMF and excess trimethylamine was removed under reduced pressure. Trace trimethylamine was removed by co-evaporation with DMF. The resulting residue was dissolved in $\mathrm{DMF}(0.1 \mathrm{M})$ and to the solution, $\mathrm{Mc}-\mathrm{OSu}(0.423 \mathrm{~g}, 1.37$ $\mathrm{mmol}, 1.14 \mathrm{eq}$ ) was added. The reaction was stirred for 20 hours at room temperature. DMF was then removed under reduced pressure and the resulting residue was purified by flash column chromatography using 3-12\% MeOH-DCM as solvent. Product was obtained as a white solid. $\left(0.415 \mathrm{~g}, 0.724 \mathrm{mmol}, 60 \%\right.$ yield). ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 9.92$ (s, 1H), 8.06 (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.82(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{~d}, 2 \mathrm{H}), 7.22(\mathrm{~d}, 2 \mathrm{H}), 7.00$ $(\mathrm{s}, 2 \mathrm{H}), 6.01(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.42(\mathrm{~s}, 2 \mathrm{H}), 5.09(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.42(\mathrm{~d}, J=5.1 \mathrm{~Hz}$,

2H), $4.40-4.30(\mathrm{~m}, 1 \mathrm{H}), 4.18(\mathrm{~d}, J=8.6,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.05-2.97(\mathrm{~m}, 1 \mathrm{H}), 2.97-2.89$ $(\mathrm{m}, 1 \mathrm{H}), 2.22-2.05(\mathrm{~m}, 2 \mathrm{H}), 2.00-1.89(\mathrm{~m}, 1 \mathrm{H}), 1.75-1.63(\mathrm{~m}, 1 \mathrm{H}), 1.63-1.53(\mathrm{~m}$, $1 \mathrm{H}), 1.53-1.39(\mathrm{~m}, 5 \mathrm{H}), 1.39-1.30(\mathrm{~m}, 1 \mathrm{H}), 1.21-1.09(\mathrm{~m}, 3 \mathrm{H}), 0.83(\mathrm{dd}, 6 \mathrm{H})$. HPLC: 90\% pure.

## Synthesis of Linker-Payload Construct

Synthesis of Mc-Val-Cit-PABC-KGP05 ${ }^{18}$


A solution of KGP05 ( $0.200 \mathrm{~g}, 0.586 \mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{DCM}(0.1 \mathrm{M})-\mathrm{NaHCO}_{3}$ Sat $[1: 1]$ was treated with triphosgene $(0.521 \mathrm{~g}, 1.78 \mathrm{mmol}, 3.00 \mathrm{eq})$. The reaction was stirred for 2 hours at room temperature to generate the desired isocyanate. The resulting residue was purified by flash column chromatography using $3-12 \% \mathrm{MeOH}-\mathrm{DCM}$ as solvent. The isocyanate was taken to the next step without further purification. A solution of Mc-Val-Cit-PABOH ( $0.205 \mathrm{~g}, 0.358 \mathrm{mmol}, 1.00 \mathrm{eq})$ and the isocyanate compound $(0.197 \mathrm{~g}, 0.537$ mmol, 1.50 eq ) in DMF ( 0.2 M ) was treated with dibutyltin dilaurate $(0.42 \mathrm{~mL}, 0.72 \mathrm{mmol}$, 2.0 eq). The reaction was stirred for 3 days at room temperature. DMF was removed under
reduced pressure and the resulting residue was purified by flash column chromatography using 3-12\% MeOH-DCM as solvent. The product was obtained as a white solid. ( 0.028 g , $0.029 \mathrm{mmol}, 5 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( 600 MHz, DMSO- $d_{6}$ ) $\delta 10.00(\mathrm{~s}, 1 \mathrm{H}), 8.66(\mathrm{~s}, 1 \mathrm{H}), 8.08$ $(\mathrm{d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.61(\mathrm{~s}, 2 \mathrm{H}), 7.35(\mathrm{~s}, 2 \mathrm{H}), 7.00(\mathrm{~s}, 2 \mathrm{H}), 6.91$ $-6.85(\mathrm{~m}, 1 \mathrm{H}), 6.81(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{~s}, 2 \mathrm{H}), 5.97(\mathrm{dd}, J=11.8,5.8 \mathrm{~Hz}, 2 \mathrm{H}), 5.41$ (s, 2H), $5.03(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.38(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{dd}, J=8.6,6.7 \mathrm{~Hz}, 1 \mathrm{H})$, $3.75(\mathrm{~s}, 5 \mathrm{H}), 3.73(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 3 \mathrm{H}), 3.69(\mathrm{~s}, 3 \mathrm{H}), 3.36(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.02(\mathrm{dd}, J=$ 13.4, $6.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.94(\mathrm{p}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.62(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.24(\mathrm{~s}, 2 \mathrm{H}), 2.18(\mathrm{td}$, $J=7.3,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.11(\mathrm{dt}, J=17.0,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.96(\mathrm{td}, J=13.5,11.6,4.9 \mathrm{~Hz}, 2 \mathrm{H})$, $1.69(\mathrm{~s}, 1 \mathrm{H}), 1.63-1.56(\mathrm{~m}, 1 \mathrm{H}), 1.48(\mathrm{q}, J=8.6 \mathrm{~Hz}, 6 \mathrm{H}), 1.35(\mathrm{~s}, 2 \mathrm{H}), 1.18(\mathrm{dd}, J=11.0$, $4.5 \mathrm{~Hz}, 2 \mathrm{H}), 0.85(\mathrm{dd}, J=6.8,3.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.82(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}(151 \mathrm{MHz}$, DMSO) $\delta 172.72,171.75,171.53,171.06,159.33,154.88,153.15,139.39,137.13,136.42$, $134.91,132.21,127.89,125.33,124.78,119.36,108.83,106.09,60.50,58.03,56.32$, $55.39,53.57,37.48,35.41,30.89,29.77,28.24,26.26,25.39,24.97,22.89,22.57,19.73$, 18.67. HPLC: $84 \%$ pure.

## CHAPTER THREE

## Results and Discussion

Over the course of this study, the synthesis of Mc-Val-Cit-PABOH was performed with varying conditions to enhance the overall yield. Following the synthesis of the linker, the payload, KGP05, was attached with overall success.


Scheme 1: Synthesis of Mc-Val-Cit-PABOH.

Figure 1 outlines the steps taken to prepare the peptide linker Mc-Val-Cit-PABOH. The first step resulted in successful addition of the protecting group Fmoc to the $l$-Citrulline amino acid resulting in compound $\mathbf{1}$ which was then subjected to a HATU coupling reaction to attach the para-amino benzyl alcohol to the C terminus of the amino acid. Next, Fmoc-Cit-PABOH, compound 2, was treated with base to remove the Fmoc group. Now this position was open to accept the Fmoc protected valine amino acid in the next step of the reaction which resulted in compound $\mathbf{3}$. This reaction had to be repeated as there was excess tri-ethyl amine (base) when reacting with Fmoc-Val which resulted in the undesired deprotection of the Valine amino acid. Compound $\mathbf{3}$ was then treated with base to deprotect and remove the Fmoc group from the Valine. Finally, the residue was reacted with McOSu to yield the desired Mc-Val-Cit-PABOH compound. Each reaction in this synthesis was purified by flash column chromatography before moving to the next step. ${ }^{1} \mathrm{H}$ NMR analyses were conducted to properly characterize and confirm the reaction products. Also, an HPLC analysis was conducted on the final linker compound which was found to have 90\% purity.

Following the synthesis of the dipeptide linker, the payload KGP05 was synthesized with the intent of attaching it to the linker in a later reaction. Scheme 2 outlines the synthesis of KGP05 which has been reported previously by the Pinney group. ${ }^{13}$ To start off, compound 4, 6-methoxy-1-tetralone, was nitrated to form two constitutional isomers, with one isomer being the desired product, compound 5, called 5-nitro-6-methoxy-1tetralone. ${ }^{13}$ KGP05 was then obtained by reacting 5-bromo-1,2,3-trimethoxybenzene with n-butyllithium, followed by the addition of compound 5 dissolved in THF to form compound $6 .{ }^{13}$ The reaction of compound $\mathbf{6}$ with Zn in the presence of acetic acid resulted
in reduction of the nitro group, and following a subsequent condensation reaction, the desired product, KGP05, was obtained with a $72 \%$ yield. ${ }^{13}$ Each reaction in this synthesis was purified by flash column chromatography before moving to the next step. ${ }^{1} \mathrm{H}$ NMR analyses were conducted to properly characterize and confirm the reaction products.


Scheme 2: Synthesis of KGP05. ${ }^{13}$

Upon successful synthesis of the linker (Mc-Val-Cit-PABOH) and payload (KGP05), the drug payload KGP05 was attached to yield a unique drug-linker construct ready to be evaluated for its specificity and cytotoxicity to human cancer cell lines by the Trawick Group. Scheme 3 displays the synthesis of Mc-Val-Cit-PABC-KGP05. Here, the KGP05 payload was reacted with triphosgene and sodium bicarbonate acting as the base in order to produce the isocyanate, compound 7, through a carbonylation reaction. ${ }^{48}$ Compound 7 was then mixed with the linker, Mc-Val-Cit-PABOH, with dibutyltin dilaurate, a catalyst. In this step, the isocyante of the payload reacts with the alcohol moiety
of the linker to produce the desired final drug-linker construct product, Mc-Val-Cit-PABCKGP05. Each reaction in this synthesis was purified by flash column chromatography before moving to the next step. ${ }^{1} \mathrm{H}$ NMR analyses were conducted to properly characterize and confirm the reaction products. Also, ${ }^{13} \mathrm{C}$ NMR and HPLC analyses were conducted on the final drug-linker construct which was found to have $84 \%$ purity.


Scheme 3: Synthesis of KGP05 drug linker construct, Mc-Val-Cit-PABC-KGP05.

Next, the drug payload KGP18 was attempted to be attached to the linker through the process shown in Scheme 5. Upon analysis, it was determined that the desired product was not produced, but an undesired double addition of the KGP18-PNP on both sides of the linker occurred during the reaction. Upon analyzing this product by ESI mass spectrometry it was obvious that KGP18-PNP has reacted with not only the benzylic alcohol of the spacer but also with the free amine of the valine moiety of the dipeptide.

Scheme 4 and 5 display the synthetic process attempted to create KGP18 drug linker construct and the undesired product produced.


Scheme 4: Synthesis of KGP18-PNP.


Scheme 5: Attempted synthesis of the KGP18 drug linker construct.

Scheme 6 outlines the next approach attempted in this synthesis to produce the desired KGP18 drug-linker construct. This proposed method used triethyl amine as the base rather than DIPEA in an effort to have more selectivity of KGP18-PNP addition onto the benzylic alcohol of the spacer rather than the free amine of the valine moiety of the dipeptide linker which becomes deprotected in the presence of base. While the desired KGP18 drug-linker construct was produced as confirmed by mass spectroscopy, an undesired side-product was also produced. Upon several purifications attempts by liquid column chromatography and prep TLC, the two compounds could not be separated from each other most likely due to their similarity in polarity.


Scheme 6: Attempted synthesis of KGP18 drug-linker construct.

## CHAPTER FOUR

## Conclusions and Future Studies

The design and synthesis of new drug-linker constructs is vital for the successful preparation of a variety of ADCs. This study evaluates the synthesis of a peptide linker, VDA drug payloads, and their constructs. Future biological studies will be conducted on these drug-linker constructs to evaluate their cytotoxicity, inhibition of tubulin polymerization, and inhibition of the colchicine binding site. If these biological results show high cytotoxicity, cleavability, and specificity, then they could be paired to antibodies specific to antigens expressed in the tumor microenvironment to produce a unique ADC .

In future studies, an important way to increase the efficiency of these drug-linker constructs could be to add a moiety that increases the construct's water solubility. One strategy for increasing water solubility could be using amino acids in the linker like glycine and serine because these possess shorter hydrocarbon and polar side chains, are reported to be more readily water soluble and more likely to be cleaved due to their structural simplicity. ${ }^{13,49}$ Another strategy for increasing water solubility of the drug-linker construct could be by adding a hydrochloride salt moiety. ${ }^{13,50}$ There have been studies demonstrating that these salt payloads still have acute cytotoxicity and ability to inhibit tubulin polymerization. ${ }^{24,27}$ Further, an ADC synthesized to deliver glucocorticoids has shown success in using a phosphate bridge attached to the aliphatic alcohol of the linker to increase plasma stability while also maintaining rapid release of payload in the lysosomal environment. ${ }^{52}$ This method could be applied to oncological drug-linker constructs in the future. While increasing water solubility would increase the stability of an ADC in the
blood, adding another step in the synthetic pathway of the drug-linker construct could lead to a decrease in overall yield or the synthesis of a bulky molecule with low stability. These are problems future scientists in the Pinney lab and beyond will have to tackle in the coming years.

The goal to synthesize pure KGP18 drug-linker construct was unsuccessful in this study, but has been successfully achieved by the Pinney Group in the past. In the attempted synthetic approach, the desired product along with an undesired side product were obtained and these two molecules were unable to be separated from each other during the course of these studies. In the future, the hope is to apply the knowledge gained from this study in order to devise and test an alternate synthetic route which minimizes or inhibits the double addition of the KGP18 prodrug to the linker or the production of the inseparable sideproduct. These studies focused on the synthesis of the VDA payloads, KGP05 and KGP18; the linker, Mc-Val-Cit-PABOH; and their unique drug-linker constructs as a cleavable tubulin inhibitor anti-cancer therapeutic. While there were obstacles faced along the way, these reported methods proved to be efficient and provided good yields for the final druglinker construct.

APPENDIX

Appendix A: NMR Spectra



ABS-I-23_2nd.10.fid


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Appendix B: HPLC Data

```
Data File C:\Chem32\1\Data\Jake\ABS-I-75.D
Sample Name: ABS-I-75
\begin{tabular}{|c|c|}
\hline Acq. Operator & : SYSTEM \\
\hline Sample Operator & SYSTEM \\
\hline Acq. Instrument & : 1200 HPLC Location : 1 \\
\hline Injection Date & : 7/13/2019 4:57:13 PM \\
\hline & Inj Volume : No inj \\
\hline Acq. Method & : C:\Chem32\1\Methods\GRAD 2 50-90 ACN.M \\
\hline Last changed & : 4/30/2014 1:53:57 AM by ERICAP \\
\hline Analysis Method & C: \Chem32\1\Methods \MASTERMETHOD2.M \\
\hline Last changed & : 12/2/2015 12:37:42 PM by Eric Lin \\
\hline Sample Info & : MC-Val-Cit-PABOH (ABS-I-75) \\
\hline
\end{tabular}
```




DAD1 E, Sig=280,16 Ref=off (JakelABS-I-75.D)


DAD1 G, Sig=300,16 Ref=off (JakelABS-I-75.D)


## Data File C: \Chem32\1\Data\Jake\ABS-I-75.D <br> Sample Name: ABS-I-75 <br> 

Area Percent Report

| Area Percent Report |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Sorted By | : | Signal |  |  |  |
| Multiplier | : | 1.0000 |  |  |  |
| Dilution | : | 1.0000 |  |  |  |
| Sample Amount: |  |  | 1.00000 | [ng/ul] | (not used |

Use Multiplier \& Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=off

| Peak \# | RetTime [min] | Type | Width [min] | $\begin{gathered} \text { Area } \\ {[\mathrm{mAU} \mathrm{~s}]} \end{gathered}$ | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.467 | BB | 0.0916 | 6455.05371 | 996.19574 | 96.1627 |
| 2 | 3.048 | BV | 0.1429 | 13.72619 | 1.32496 | 0.2045 |
| 3 | 3.235 | VV | 0.1412 | 110.01940 | 11.33673 | 1.6390 |
| 4 | 3.838 | VB | 0.1633 | 12.89438 | 1.11212 | 0.1921 |
| 5 | 4.525 | BB | 0.1454 | 13.41488 | 1.33347 | 0.1998 |
| 6 | 7.260 | BB | 0.1903 | 79.12875 | 6.06225 | 1.1788 |
| 7 | 10.078 | BB | 0.2678 | 28.40033 | 1.53519 | 0.4231 |

Totals :
6712.637641018 .90046

Signal 2: DAD1 C, Sig=210,8 Ref=off

| Peak \# | RetTime [min] | Type | Width [min] | $\begin{gathered} \text { Area } \\ {[\mathrm{mAU} * \mathrm{~s}]} \end{gathered}$ | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.467 | BV | 0.1030 | 1.16916e4 | 1683.49255 | 95.1412 |
| 2 | 2.290 | VB | 0.1827 | 65.13793 | 4.91289 | 0.5301 |
| 3 | 2.914 | BV | 0.0968 | 20.14769 | 3.22897 | 0.1640 |
| 4 | 3.041 | VV | 0.0976 | 20.67145 | 3.10874 | 0.1682 |
| 5 | 3.235 | VB | 0.1393 | 264.61975 | 27.73666 | 2.1534 |
| 6 | 3.839 | BB | 0.1446 | 34.34644 | 3.43550 | 0.2795 |
| 7 | 4.525 | BB | 0.1416 | 55.25455 | 5.67673 | 0.4496 |
| 8 | 7.265 | BB | 0.1998 | 15.66669 | 1.14405 | 0.1275 |
| 9 | 10.076 | BB | 0.2804 | 78.81969 | 4.09758 | 0.6414 |
| 10 | 24.423 | BB | 0.3049 | 42.41915 | 2.02081 | 0.3452 |


| Data File C:\Chem32\1\Data\Jake\ABS-I-75.D <br> Sample Name: ABS-I-75 |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| Totals : |  | 1.22886e4 | 1738.85447 |  |
| Signal 3: DAD1 $\mathrm{E}, \mathrm{Sig}=280,16$ Ref=off |  |  |  |  |
| $\begin{aligned} & \text { Peak RetTime Type } \\ & \# \quad[\mathrm{~min}] \end{aligned}$ | Width <br> [min] | $\begin{gathered} \text { Area } \\ {\left[\mathrm{mAU}{ }^{*} \mathrm{~s}\right]} \end{gathered}$ | Height [mAU] | $\begin{gathered} \text { Area } \\ \text { \% } \end{gathered}$ |
| 1 1.467 BV | 0.0885 | 796.15808 | 128.10965 | 83.0364 |
| 21.692 VB | 0.1511 | 81.38235 | 7.02850 | 8.4879 |
| $3 \quad 3.235$ BB | 0.1358 | 44.12374 | 4.77422 | 4.6019 |
| 4 4.524 BB | 0.1394 | 16.31908 | 1.70852 | 1.7020 |
| $5 \quad 10.078$ BB | 0.2622 | 20.82310 | 1.15565 | 2.1718 |
| Totals: 9588.80636142 .77654 |  |  |  |  |

Signal 4: DAD1 G, Sig=300,16 Ref=off

| Peak <br> \# | RetTime [min] |  | Width [min] | $\begin{gathered} \text { Area } \\ {\left[m A U^{*} \mathrm{~s}\right]} \end{gathered}$ | Height <br> [mAU] | $\begin{gathered} \text { Area } \\ \% \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.468 |  | 0.0834 | 262.36572 | 45.40654 | 73.9776 |
| 2 | 1.698 | VB | 0.0980 | 31.98127 | 4.66147 | 9.0176 |
| 3 | 3.235 | BV | 0.1197 | 14.69276 | 1.86862 | 4.1428 |
| 4 | 3.837 | BB | 0.1365 | 12.86430 | 1.38264 | 3.6273 |
| 5 | 4.525 | BB | 0.1422 | 32.75137 | 3.40603 | 9.2347 |
| Totals |  |  |  | 354.65543 | 56.72530 |  |

Signal 5: DAD1 H, Sig=320,16 Ref=off

| Peak \# | RetTime [min] |  | Width <br> [min] | $\begin{gathered} \text { Area } \\ {\left[\mathrm{mAU}{ }^{*} \mathrm{~s}\right]} \end{gathered}$ | Height <br> [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.468 |  | 0.0832 | 192.66597 | 33.43027 | 72.1637 |
| 2 | 1.703 |  | 0.1373 | 20.76944 | 2.06575 | 7.7793 |
| 3 | 3.836 |  | 0.1280 | 13.97581 | 1.59794 | 5.2347 |
| 4 | 4.525 | BB | 0.1414 | 39.57329 | 4.14356 | 14.8223 |

[^0]

Data File C: \Chem32\1\Data\Jake\ABS-I-99.D
Sample Name: ABS-I-99



Signal 1: DAD1 A, Sig=254,4 Ref=off

| Peak \# | RetTime [min] | Type | Width <br> [min] | $\begin{gathered} \text { Area } \\ {[\mathrm{mAU} \mathrm{~s}]} \end{gathered}$ | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.495 |  | 0.1131 | 54.22970 | 6.38362 | 2.3489 |
| 2 | 1.665 |  | 0.0888 | 17.06932 | 2.73506 | 0.7393 |
| 3 | 1.766 | VB | 0.0647 | 7.00096 | 1.61192 | 0.3032 |
| 4 | 1.944 | BV | 0.1724 | 21.39836 | 1.77838 | 0.9269 |
| 5 | 2.185 | VB | 0.1032 | 13.52821 | 1.89746 | 0.5860 |
| 6 | 2.607 | BV | 0.1135 | 16.43190 | 2.14274 | 0.7117 |
| 7 | 2.805 | VV | 0.1244 | 88.95795 | 10.54220 | 3.8532 |
| 8 | 3.059 | VB | 0.1265 | 19.16593 | 2.22500 | 0.8302 |
| 9 | 3.401 | BV | 0.1343 | 60.09003 | 6.72206 | 2.6028 |
| 10 | 3.669 | VB | 0.1409 | 1937.25708 | 203.76689 | 83.9110 |
| 11 | 5.032 | BV | 0.1805 | 11.81313 | 1.05618 | 0.5117 |
| 12 | 5.386 | VB | 0.1417 | 12.22289 | 1.32429 | 0.5294 |
| 13 | 6.070 | BB | 0.2242 | 28.76205 | 2.06884 | 1.2458 |
| 14 | 9.561 | BB | 0.2618 | 20.77636 | 1. 18959 | 0.8999 |
| Total | Is : |  |  | 2308.70389 | 245.44426 |  |

Signal 2: DAD1 C, Sig=210,8 Ref=off

| Peak \# | RetTime [min] | Type | Width <br> [min] | Area [mAU*s] | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.483 | BV | 0.1296 | 161.31943 | 18.90460 | 3.2182 |
| 2 | 1.664 | VV | 0.1334 | 115.15079 | 11.64610 | 2.2972 |
| 3 | 1.954 | VV | 0.1738 | 115.66602 | 9.38527 | 2.3075 |
| 4 | 2.192 | VB | 0.1016 | 79.51984 | 11.09081 | 1.5864 |

Data File C:\Chem32\1\Data\Jake\ABS-I-99.D
Sample Name: ABS-I-99

| Peak \# | RetTime [min] |  | $\begin{aligned} & \text { Width } \\ & {[\mathrm{min}]} \end{aligned}$ | $\begin{gathered} \text { Area } \\ {[\mathrm{mAU} * \mathrm{~s}]} \end{gathered}$ | Height [mAU] | $\begin{gathered} \text { Area } \\ \% \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 5 | 2.606 |  | 0.1133 | 31.54139 | 4.12077 | 0.6292 |
| 6 | 2.806 | VV | 0.1222 | 190.00183 | 23.03302 | 3.7904 |
| 7 | 3.059 | VB | 0.1316 | 49.92138 | 5.62284 | 0.9959 |
| 8 | 3.397 | BV | 0.1362 | 110.18508 | 12.33825 | 2.1981 |
| 9 | 3.669 | VB | 0.1406 | 3911.42358 | 412.57855 | 78.0300 |
| 10 | 5.024 | BV | 0.2483 | 42.98986 | 2.52891 | 0.8576 |
| 11 | 5.387 | VV | 0.1604 | 41.95073 | 3.87322 | 0.8369 |
| 12 | 6.075 |  | 0.2580 | 89.54916 | 5.32926 | 1.7864 |
| 13 | 9.558 | BB | 0.2623 | 56.30305 | 3.18405 | 1.1232 |
| 14 | 10.229 | BB | 0.2053 | 17.19465 | 1.25928 | 0.3430 |
| Total |  |  |  | 5012.71680 | 524.89493 |  |

Signal 3: DAD1 E, Sig=280,16 Ref=off

| Peak \# | RetTime [min] |  | Width <br> [min] | $\begin{gathered} \text { Area } \\ {\left[\mathrm{mAU}{ }^{*} \mathrm{~s}\right]} \end{gathered}$ | Height <br> [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.499 | BV | 0.1286 | 10.58568 | 1.25229 | 1.1527 |
| 2 | 1.661 | VV | 0.0985 | 36.74335 | 5.32257 | 4.0011 |
| 3 | 2.052 | VV | 0.1664 | 18.03585 | 1.49839 | 1.9640 |
| 4 | 2.194 | VB | 0.0995 | 21.44307 | 3.07095 | 2.3350 |
| 5 | 2.805 | VB | 0.1071 | 24.55502 | 3.53312 | 2.6739 |
| 6 | 3.399 | BV | 0.1386 | 19.37074 | 2.11923 | 2.1094 |
| 7 | 3.669 | VB | 0.1405 | 771.93689 | 81.47997 | 84.0591 |
| 8 | 6.080 | BB | 0.1978 | 15.65542 | 1.15759 | 1.7048 |
| Total |  |  |  | 918.32603 | 99.43412 |  |

Signal 4: DAD1 G, Sig=300,16 Ref=off

| Peak \# | RetTime [min] |  | Width <br> [min] | $\begin{gathered} \text { Area } \\ {\left[\mathrm{mAU} U^{*} \mathrm{~s}\right]} \end{gathered}$ | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.661 | BB | 0.0938 | 35.94694 | 5.53011 | 10.2711 |
| 2 | 2.050 | BV | 0.1183 | 9.72066 | 1.22862 | 2.7775 |
| 3 | 2.194 | VB | 0.0947 | 20.68534 | 3.14590 | 5.9104 |
| 4 | 2.804 | BB | 0.1052 | 7.19315 | 1.06024 | 2.0553 |
| 5 | 3.392 | BV | 0.1388 | 9.41847 | 1.02932 | 2.6911 |
| 6 | 3.669 | VB | 0.1405 | 267.01852 | 28.18267 | 76.2947 |
| Total |  |  |  | 349.98308 | 40.17687 |  |

Data File C:\Chem32\1\Data\Jake\ABS-I-99.D
Sample Name: ABS-I-99

Signal 5: DAD1 H, Sig=320,16 Ref=off

| Peak \# | RetTime [min] |  | Width <br> [min] | $\begin{gathered} \text { Area } \\ {\left[m A U^{*}\right. \text { s] }} \end{gathered}$ | Height [mAU] | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1.662 |  | 0.0778 | 6.56233 | 1.27988 | 10.3584 |
| 2 | 2.195 |  | 0.0963 | 12.99295 | 1.93521 | 20.5088 |
| 3 | 3.669 | BB | 0.1366 | 43.79768 | 4.79417 | 69.1328 |
| Total |  |  |  | 63.35296 | 8.00926 |  |

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