

ABSTRACT

Kinetic Characterization of Thiosemicarbazones as Cysteine Protease Inhibitors and Their Potential Use as Therapeutic Agents against Metastatic Cancer and Chagas' Disease

Gustavo E. Chavarria Nolasco, Ph.D.

Mentor: Mary Lynn Trawick, Ph.D.

Human cathepsins L and K are cysteine proteases that play important roles in physiological and pathological processes, such as cancer metastasis, bone resorption and neurodegenerative diseases. This research has focused on the evaluation of synthetic thiosemicarbazones that could inhibit their proteolytic activities with the objective of preventing cancer metastasis. In addition, cruzain, a cathepsin L-like cysteine protease found in *Trypanosoma cruzi*, is a validated target for pivotal roles in the parasitic invasion in Chagas' disease, a condition that could be fatal if not treated. Currently, there is no effective treatment against the disease, which is rapidly extending to non-endemic areas in the United States and Europe. More than 150 synthetic thiosemicarbazones (obtained through a collaborative study) were tested against cathepsins L, K and cruzain. This work presents preliminary *in vitro* analysis of these compounds in order to characterize their potency and mode of inhibition. A number of potent inhibitors was found for each enzyme. A smaller subset of thiosemicarbazones were found to be selective. Results showed that compound **1**, one of the most potent inhibitors in this library is a slow binding, slowly reversible, competitive inhibitor of these targets.

Furthermore, **1** was able to delay and partially inhibit the activation of procathepsin K under acidic conditions. Similar results were found with 3-bromo-3'-hydroxybenzophenone thiosemicarbazone (**8**). Compounds **8** and **1** inhibited the *in vitro* type I collagenase activity of cathepsin L in a time-dependent manner and type IV collagenase activity of cathepsin K.

Analogs, **1**, **8**, **156**, **157**, and **168** were also used in cell studies. These compounds were able to delay cell migration and cell invasion in MDA-MB-231 cells, a type of carcinoma breast cancer cell line. It was determined that cell invasion and cell migration were inhibited in a concentration dependent manner.

Lastly, analog **17** was also found to be a slow reversible, slow binding, competitive inhibitor of cruzain. This compound was also able to inhibit the collagenase activity of recombinant cruzain when human type I collagen was used as a substrate.

Kinetic studies and molecular modeling indicate the best thiosemicarbazone inhibitors form a reversible covalent bond with each enzyme.

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by

Gustavo E. Chavarria, B.S.

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Patrick J. Farmer, Ph.D., Chairperson

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Approved by the Dissertation Committee

Mary Lynn Trawick, Ph.D., Chairperson

Sung-Kun Kim, Ph.D.

Bryan F. Shaw, Ph.D.

Kevin G. Pinney, Ph.D.

Jaime L. Diaz-Granados, Ph.D.

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J. Larry Lyon, Ph.D., Dean

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LIST OF ABBREVIATIONS

[E]	Enzyme concentration
[E _T]	Total enzyme concentration
[I]	Inhibitor concentration
[S]	Substrate concentration
ΔG°	Gibbs Free Energy
°C	Degrees Celsius
3-D	3-Dimensional
A	Alanine
AA	Amino acid
ABL	V-ABL Abelson Murine Leukemia Viral Oncogene Homolog 1
ABR	Assay Buffer for Reversibility Studies (Cathepsin L)
Abz	3-Aminobenzoic Acid
ACI	Type I Collagen in Acetic Acid
ACIV	Type IV Collagen in Acetic Acid
AD	Anno Domini
ADT	Androgen Deprivation Therapy
AKR	Assay Buffer for Reversibility Studies (Cathepsin K)
Ala	Alanine
AMC	7-Amino-4-Methylcoumarin
Arg	Arginine

Asn	Asparagine
Asp	Aspartate
ATCC	American Type Culture Collection
ATG	Start codon
AZ	Arizona
BC	Before Christ
BCL	Apoptosis regulator proteins
BCR-ACL	Tyrosine-Kinase Inhibitors
bFGF	Basic Fibroblast Growth Factor
Bid	Interacting-Domain Death Agonist
BRCA	Breast Cancer Type 1 Susceptibility Protein,
BT474	Human Breast Carcinoma Cell Line
BZ	Benznidazole
C	Cysteine
C-	Carboxy Terminal
C=O	Carbonyl, Carbonyl Bond
C4-S	Chondroitin 4-Sulfate
C4-S	Chondroitin 4-Sulfate
cABL	Oncogene
CATH	Protein Structure Classification
Cbz	Carboxybenzyl
CD20	B-Lymphocyte Antigen
CD33	Siglec-3

Cdc42	Cell Division Control Protein 42
CK	Cathepsin K
CKR	Cathepsin L Stock Solution for Reversibility Studies (Cathepsin K)
CL	Cathepsin L
CLC	Cathepsin L Stock Solution for Collagenase Studies
CLIP	Class II-Associated Invariant Chain Peptide
CLR	Cathepsin L Stock Solution for Reversibility Studies (Cathepsin L)
C-N	Carbon-Nitrogen Angle
COPD	Chronic Obstructive Pulmonary Disease
CRES	Cystatin-Related Epididymal Spermatogenic
CRT	Conformal Radiation Therapy
CTC	Stop Codon
CTX-I	C-Telopeptide Of Type I Collagen
Cys	Cysteine
CZ	Cruzain
D	Aspartate
DKK-1	Dickkopf-Related Protein 1
DMEM	Dulbeco's Modified Eagle's Medium
DMSO	Dimethylsulfoxide
DNA	Deoxyribonucleic Acid
DTT	Dithiothreitol
E	Glutamate
E	Enzyme

E-64	Irreversible Inhibitor Of Cysteine Proteases
EC	Enzyme Classification
ECM	Extracellular Matrix
EDTA	Ethylenediaminetetraacetic acid
EGF	Epidermal Growth Factor
EGFR	Epidermal Growth Factor Receptor
ELISA	Enzyme-Linked Immunosorbent Assay
Erk	Extracellular-Signal-Regulated Kinases
ES	Michaelis-Menten Complex
ET-1	Endothelin
F	Phenylalanine
FAK	Focal Adhesion Kinase-1
FBS	Fetal Bovine Serum
FDA	United States Food And Drug Administration
G	Glycine
g	Gram
g	Gram
G1-S	Post-Mitotic Phase To Synthesis Phase
GEF	Guanine Nucleotide Exchange Factors
Gln	Glutamine
Glu	Glutamate
Gly	Glycine
H	Histidine

h	Hour
HER2/neu	Human Epidermal Growth Factor Receptor 2
His	Histidine
HIV	Human Immunodeficiency Virus
HPV	Human Papillomavirus
HUVEC	Human Umbilical Vein Endothelial Cells
I	Isoleucine
I	Inhibitor
IC ₅₀	Half Maximal Inhibitory Concentration
ID	Identification
IFA	Indirect Immunofluorescen Assay
IGF-1	Insulin-Like Growth Factor 1
IL-1 α	Interleukin 1 Alpha
Ile	Isoleucine
IMRT	Intensity-Modulated Radiation Therapy
ING	Inhibitor Of Growth 1 Gene
IORT	Intraoperative Radiation Therapy
JNK	C-Jun N-Terminal Kinases
K	Lysine
kDa	Kilodalton
K_I	Inhibition Constant
K_I^{app}	Apparent Inhibition Constant
KIT	Kit

K_M	Michaelis-Menten Constant
k_{obs}	Rate Constant For Conversion From The Initial Velocity Phase To The Steady State Velocity Phase.
k_{off}	Rate Of Dissociation
k_{on}	Rate Of Association
l	Liter
L-	Left Domain
LDL	Low-Density Lipoprotein
Leu	Leucine
LHRH	Luteinizing Hormone-Releasing Hormone
Lys	Lysine
M	Methionine
M	Molar
MAPK	Mitogen-Activated Protein Kinases
MES	2-(N-morpholino)ethanesulfonic acid
Met	Hepatocyte Growth Factor Receptor
Met	Methionine
MHC II	Major Histocompatibility Complex Ii
min	Minute
mM	Millimolar
MMP	Metalloprotease
MOPS	3-(N-morpholino)propanesulfonic acid
mRNA	Messenger Ribonucleic Acid

MW	Molecular Weight
MW	Molecular Weight
N	Asparagine
N-	Amino
NaCl	Sodium Chloride
NADPH	Nicotinamide Adenine Dinucleotide Phosphate (Reduced)
NaOAc	Sodium Acetate Buffer
NaOAc	Sodium Acetate Buffer
NaOH	Sodium Hydroxide
NaOH	Sodium Hydroxide
NC1D	Noncollagen Domain
NF- κ B	Nuclear Factor Kappa-Light-Chain-Enhancer Of Activated B Cells
NGF	Nerve Growth Factor
NH	Amino Group
NIH3T3	Mouse Embryonic Fibroblast Cell Line
nM	Nanomolar
NO ₂	Nitrogen Dioxide
Nx	Nifurtimox
o-But	O-Butyl
OPG	Osteoprotegerin
P	Proline
p38	P38 MAPK
p53	Tumor Protein 53

PAGE	Polyacrylamide gel electrophoresis
PCR	Polymerase Chain Reaction
PDB	Protein Data Bank
PDGF	Platelet-Derived Growth Factor
PDGFRA	Alpha-Type Platelet-Derived Growth Factor Receptor
pH	Measure Of Hydrogen Ion Concentration
Phe	Phenylalanine
PI3K	Phosphatidylinositol 3-Kinases
PIM	Proto-Oncogene Serine/Threonine-Protein Kinase Pim-1
pM	Picomolar
Pro	Proline
Q	Glutamine
R	Arginine
R	Radical
R-	Right Domain
r^2	Coefficient Of Determination
Rac	A G-Protein
RANK	Receptor Activator Of Nuclear Factor K B
RANKL	Receptor Activator Of Nuclear Factor Kappa-B Ligand
Ras	A G-Protein
Rho	A G-Protein
RTK	Receptor Tyrosine Kinases
S	Serine

S	Substrate
s	Second
SA-C	Solution A with DMSO (SA-C)
SA-I	Solution A with Inhibitor in DMS
SAR	Structure-Activity Relationship
SCCA	Squamous Cell Carcinoma Antigen
SCOP	Structural Classification Of Proteins
SDS	Sodium Dodecyl Sulfate
SE	Standard Error
sec	Second
Ser	Serine
Src	Proto-Oncogene Tyrosine-Protein Kinase
SRP	Signal Recognition Particle
<i>T. cruzi</i>	Trypanosoma Cruzi
t-But	Tertiary-Butyl
TGF- β	Transforming Growth Factor Beta
Thr	Threonine
TIC	Type I Collagen
TIC	Type I Collagen
TIVC	Type IV Collagen
TIVC	Type IV Collagen
TM	Trademark
TNF	Tumor Necrosis Factors

TNF- α	Tumor Necrosis Factor-Alpha
Trp	Tryptophan
TUB	Tubulin
Tyr	Tyrosine
U87MG	Human Glioblastoma-Astrocytoma, Epithelial-Like Cell Line
uPA	Urokinase-Type Plasminogen Activator
USA	United States of America
V	Valine
v	Velocity
Val	Valine
VEGF	Vascular Endothelial Growth Factor
VEGFR	Vascular Endothelial Growth Factor Receptor
V_{MAX}	Maximum Velocity
v_{max}	Catalytic Rate Of An Enzyme Without The Presence Of An Inhibitor (Lowest Concentration, IC ₅₀ Determination)
v_{min}	Catalytic Rate Of An Enzyme With The Presence Of An Inhibitor (Highest Concentration, IC ₅₀ Determination)
v_o	Initial Velocity
v_s	Steady-State Velocity
W	Tryptophan
WNT	Wingless Gene
X	Nonspecific Amino Acid
X	Inhibitor Concentration

Y	Tyrosine
Y	Fractional Activity
Z-FR-AMC	N-Carbobenzoxy-L-Phenylalanyl-Larginine Amide, Hydrochloride
α	Alpha
β	Beta
γ	Gamma
μ	Micro
μM	Micromolar
μm	Micrometer

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DEDICATION

To

My parents

German Chavarria Umanzor

and

Lidia Nolasco

This is the product of your love and dedication

CHAPTER ONE

Introduction

Statement of Purpose and Significance

Science has always been fascinated by the extraordinary ability of enzymes to catalyze numerous reactions in living organisms. Their catalytic activity has been a major challenge in drug discovery due to their vital role in pathological diseases and conditions. Therefore, enzymes are often the main target in drug discovery. Examples of enzymes implicated in pathological conditions include the kinases, proteins that have a pivotal role in cancer.¹ To date, there are cases where treatments have been accepted for the treatment of specific conditions. For example, the Food and Drug Administrations have approved several medications for the treatment of the Human Immunodeficiency Acquired (HIV) that directed toward enzymatic targets. The list of enzymes include: nucleoside reverse transcriptase, HIV integrase and proteases.²

Proteases comprise one of the most studied groups due to their implications in diseases. The cysteine cathepsins are proteases and have become a 'hot' target in research due to their participation in numerous processes. Two members of this family, cathepsins L and K have been extensively characterized and studied. Pioneer cathepsin K research in drug discovery has lead to the discovery of selective inhibitors of this protease that are currently in clinical trials as potential anti-osteoporosis agents. To date, cathepsin L inhibitors are not currently in clinical trials.

Cathepsins L and K play important roles in bone metastasis and cathepsin L overexpression is associated with a number of metastatic cancers. The main cause of

death in patients with cancer is its metastatic spread. Overexpression of cathepsins L and K promote the degradation of the components of the extracellular matrix in cancer microenvironments.

The main objective of this research project was to examine a library of small molecules as inhibitors of individual enzymes, cathepsin L and cathepsin K, for potential use in targeted therapies as anti-metastatic agents. The ideal chemotherapeutic agent should meet several criteria, including: potency, selectivity, solubility, and be non-toxic in patients. It should produce only minor side effects, and exhibit stability when administered with long shelf life (chemically stable), etc. One of the major disadvantages when using non-specific chemotherapy as a cancer treatment is the considerable list of negative side effects (i.e. vomiting, hair loss, skin rash). In targeted therapy, the inhibition of one specific enzyme or protein has the potential to limit these negative side effects.

The discovery of cathepsins L and K inhibitors involve several considerations. One the first steps in the search of these compounds was the creation of a library of synthetic compounds.³⁻⁷ These compounds were synthesized by members of Dr. Kevin G. Pinney laboratory. Thiosemicarbazone analogs were first applied to the cathepsin L-like enzyme cruzain^{8,9}, and then to cathepsin L as enzyme inhibitors.¹⁰⁻¹³ Assay optimization is a key step in drug discovery. Discrepancies in results might lead to misinterpretations and false positives. Thus, the necessity of solid and reliable assays was mandatory. The analysis of their structure-activity relationships SARs was extremely helpful in order to achieve these goals. SARs helped to optimize pre-clinical studies. They also helped to identify possible selectivities and lead to the design of more

analogues that could be potent inhibitors. The characterization of these compounds, including the kinetic mechanism of action is important for the design and study of possible candidates that can function as effective drugs in the treatment of cancer metastasis. This involved numerous evaluations. Fluorometric based assays were utilized to study various assay parameters in inhibitory activities, reversibility studies, determination of K_I and the mechanism and mode of inhibition.

We also explored the application of thiosemicarbazones as potential inhibitors of the activation of procathepsin K which occurs under acidic conditions. The inhibition of the activation of procathepsin K was explored by using one of the lead thiosemicarbazone inhibitors of cathepsin K. Experiments using human type IV collagen, a natural substrate of cathepsin K, were carried out in order to determine if thiosemicarbazones were capable of inhibit the outstanding collagenase activity of cathepsin K using molecular biology techniques and fluorescent protein stains.

Results indicated that a number of thiosemicarbazones are lead compounds as cathepsin L inhibitors. We also explored the inhibition of the type I collagenase activity of cathepsin L by using a good inhibitor of this protease that was found in the library of inhibitors.

Cancer metastasis is characterized by enhanced cell invasion and cell migration. Two dimensional cell based assays were used to determine the inhibitory ability of these compounds within the extracellular matrices of a metastatic breast cancer cell line.

Another cysteine protease, cruzain, is the most potent hydrolase found in *Trypanosoma cruzi*, the responsible of Chagas' disease. Chagas' disease, once upon considered as a disease affecting countries in development, is rapidly spreading

throughout developed countries. The parasitic infection is now considered a threat in some European countries and the United States. Cruzain is mainly involved in cell host invasion and organ invasion due to its proteolytic capability of degrading surrounding tissues.

On the other hand, the treatment of Chagas' disease, a condition that has become chronic in millions of people in South America, is a major challenge. There is no effective treatment for the eradication of the disease and efforts in the search for better alternatives are non-existing. The disease was considered a neglected condition, but this is rapidly changing due to the increasing number of patients in the United States and Europe. We evaluated many analogs in the library of thiosemicarbazones against cruzain in the search for therapeutic agents against Chagas' disease.

These studies were part of an ongoing collaboration between the Trawick and Pinney group (Baylor University). Synthesis of synthetic thiosemicarbazones and biological evaluations were carried out by members of Dr. Kevin G. Pinney group and Dr. Mary Lynn Trawick group respectively.

General Overview of Cancer

Defining cancer can be quite difficult. There is not a unique definition due to the complexity of the disease. Many factors can affect the specific definition of cancer, such as the origin, organ affected as well as others. Thus, a combination of these concepts will help to understand the basics of the disease. Cancer is a generic name that numerous diseases share in their pathology: growth of malignant cells that invade healthy tissues and organs. Thus, tumors are the collection of malignant cells. Additionally, a malignant tumor has the ability to invade healthy organs and tissues.^{14,15} Perhaps, the complexity of

cancer is the biggest obstacle when finding appropriate treatments. Several factors promote the development and spread of this disease in normal tissues; thus a specific route for cancer treatment seems a long term objective. The combination of chemotherapy, radiotherapy, surgery and targeted therapies increases survival rates in cancer patients. Nowadays, chemotherapy is one of the most used therapies in the treatment of cancer.

Types of Cancer and Organ Distribution

As previously stated, cancer is a generic term that comprises more than 100 diseases with similar characteristics. However, a general form to identify them is by defining the organ(s) that are mainly affected. In general, the most common cancers distribute within oral cavities, endocrine, digestive, respiratory genital, urinary, bones, heart, skin, breast, eye, and brain systems. Also, lymphoma, myeloma, and leukemia are also other types of cancer.¹⁶ However, organs can be affected by more than one type of cancer. For example, there are several different types of brain cancers affecting children (gliomas, astrocytomas, brain stem, ependymoma, medulloblastomas and other).¹⁷

History of Cancer

Cancer has been prevalent for centuries. Nevertheless, early civilizations did not understand the causes and origins of this disease. The oldest written description of cancer was originated in 3000 B.C. in the Egyptian civilization. However, the term cancer was used for the first time by Hippocrates in 460 B.C. The Greek physician used the terms “carcinos” and “carcinoma” when referring to ulcer-based tumors. Modern terminologies, “cancer” and “oncos”, were first used by Celsus (28-50 B.C.) and Galen

(130-200 A.D.), respectively. “Cancer” and “oncos” are the Greek words for crab and swelling.^{18,19}

Statistics of Cancer

Cancer and its variations combined represent the second leading cause of death in the United States.²⁰ More than 550,000 patients died in 2010 of cancer.²¹ It is also important to observe the economical impact of cancer in society. Cancer accounted for more than 115 billion dollars in lost productivity when other indirect costs, such as care giving and job wages, are included, the costs are much higher. The total cost of cancer was calculated to be \$232 billion in 2000. Projections estimate this number can increase up to \$308 billion in 2020.²²

Lung and bronchus cancers account for 14% of new cases in the United States in 2012 (468610 cases).²¹ However, breast and prostate cancer are the leading types of cancer for the same period in women and men respectively. Prostate cancer represents an alarming thirty percent of new cases within the male population. In the female population, thirty percent of new cases of breast cancer account for in the United States.

Other leading types of cancer found in the male population are: colon, melanoma, leukemia, and pancreas. Similarly, new cases of cancer in females are distributed among uterus, thyroid, melanoma and colon. Figure 1 shows the ten leading cancer types only in the United States.²¹

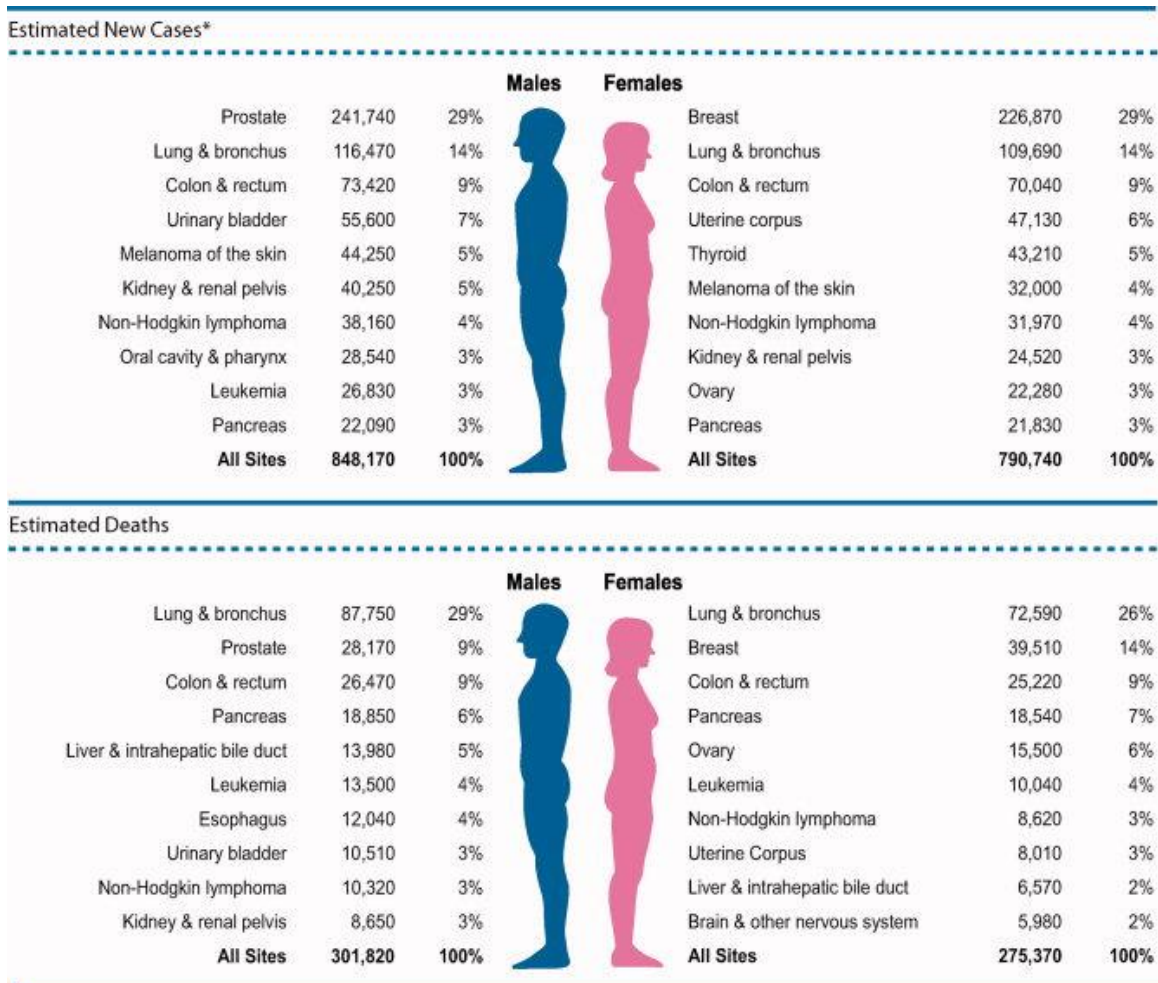


Figure 1. Ten Leading Cancer Types for the Estimated New Cancer Cases and Deaths by Sex, United States, 2012 (directly reproduced from Siegel , page 13)

Causes of Cancer

Advances in science and technology have opened the opportunity to study in depth some of the cancer that attacks our modern society. Research also has demonstrated that cancer can be triggered by internal and external factors that influence the development of malignant tumors.

Oncogenes

In general, an oncogene is a gene that promotes the development of cancer (tumorigenesis). Numerous genes have the potential of becoming oncogenes and extensive research has been done to study the role of oncogenes in the development of cancer.²³ The most common oncogenes found in human cancer belong to the Rat Sarcoma family (RAS). The family is a well known series of related proteins that participate in cellular signaling, replication stress and activation of DNA damage responses.²⁴ The Proto-Oncogene Proteins c-pim-1 family (PIM) constitutes a family of related proteins, serine/threonine kinases, which are associated with the prognosis of solid tumors, especially prostate cancer. However, they are considered as proto-oncogenes.²⁵ The Hepatocyte Growth Factor Receptor Met, a tyrosine kinase oncogene has been strongly related to the progress of basal-like breast carcinoma.²⁶ Other oncogenes have been identified and several reviews show the importance of these targets. The Human Epidermal Growth Factor Receptor 2 HER2/neu is a proto-oncogene; however, its overexpression in normal cells can catalyze the activity of this protein as a oncogene.²⁷ Other proteins that have been found to have oncogene-properties are BCL-ABL (Proto-Oncogene Proteins fused with Abelson proto-oncogene), Tyrosine-Protein Kinase (KIT), and Platelet-Derived Growth Factor Receptor, α -polypeptide (PDGFRA).²⁸⁻³⁰

Tumor-Suppressor Genes

In contrast tumor-suppressor genes have the ability to arrest the development of tumors in human bodies.³¹ Thus, tumor-suppressor genes are also known as anti-oncogenes. Tumor-suppressor gene research is more challenging due to the difficulty to

restore the anti-oncogene activity of these targets. Some of these genes include p53, a much studied tumor-suppressor gene with a high activity against genome mutation, and ING genes, tumor-suppressor genes in the formation of head and neck carcinomas.^{32,33} Another example of a tumor-suppressor gene is breast cancer type 1 susceptibility protein BCRA.³⁴

Viruses as Causes of Cancer

Viruses have become one the biggest centers of attention for scientists for their ability to induce cancer in humans. The discovery of viruses as cancer-promoter agents one hundred years ago opened an undiscovered world that viruses that could attack healthy cells. Hepatitis B and C viruses can promote liver cancer.³⁵ Another herpes virus, the Epstein-Barr virus has been linked with non-Hodgkin lymphomas and respiratory system cancer.³⁶ Patients with human immunodeficiency virus (HIV) have higher risks to develop Kaposi sarcoma and non-Hodgkin lymphomas among others.³⁷ Perhaps, the best known case is related to the human papilloma virus (HPV).³⁸ This virus is the reason for a high number of female patients with genital cancers including cervix, vulva, vagina, and anus. Furthermore, the carcinogen action of HPV has been extended in the male population (penis cancer) as well.³⁹ New studies have related oral cancer (tongue, tonsil, and throat) with this virus.⁴⁰

Non-Biological Carcinogens

Oncogenes, tumor-suppressor genes, and Human papillomavirus (HPV) are considered biological carcinogens due to their expression (or sometimes overexpression) or transmission among mammalian organisms. Nevertheless, the existence of hundreds

of carcinogens is well known and studied, and they account for a significant number of new cases of cancer every year. Physical, chemical, and external biological agents are classified as carcinogens and human exposure to these agents is now restricted. The most common carcinogens found on daily basis are tobacco, sunlight, and coal-tars. Cancer is also promoted by exposure to other agents or material such as arsenic, coal production, secondhand smoke, sunbeds, X-radiation, and asbestos.⁴¹⁻⁴⁴

Cancer Treatments

Medicinal research has achieved important goals in the fight against cancer. Death rates have decreased and cancer patients have higher chances of overcoming this terrible disease. Generally, cancer treatments can be divided into five major categories: Surgery, hormone therapy, radiation, immunotherapy, targeted therapy, and chemotherapy.

Surgery

Overall, surgery is one the most common techniques in cancer treatment. The main objective of surgical operations is the removal of the tumor along with the blood vessel network that has been formed around the tumor and the lymph nodes. This technique has been greatly evolved from early days when radical surgeries were carried out (like complete mastectomies with women affected with breast cancer) to modern techniques where less invasive surgeries are performed. Endoscopic surgeries are efficient tools when it comes to surgeries. This technique utilizes modern technology (exploratory surgery) to remove tumors in the colon, esophagus, and breast. Other

modern methods that are classified as surgeries are cryosurgery (by using liquid nitrogen), and laser surgery.⁴⁵⁻⁴⁸

Hormone Therapy

Cancer is often strongly hormone dependent, especially in breast (estrogen) and prostate (testosterone) cancers. The discovery of drugs such as Tamoxifen, has shown that patients with breast cancer have higher probabilities of overcoming their conditions.⁴⁹ Similarly, male patients with prostate cancer have been treated with Androgen Deprivation Therapies (ADT).⁵⁰ Briefly, testosterone is one of the main responsible in abnormal cell growth in the prostate. Aromatase inhibitors (LHRH or Luteinizing Hormone-Releasing Hormone) are also considered as great promises for the treatment of cancer in male and female patients.⁵¹

Radiation

Another technique that is greatly utilized for the treatment of cancer is radiation, also known as radiotherapy. Today's technology offers radiotherapy devices with excellent precision when treating a tumor without affecting neighboring organs or tissues. Conformal radiation therapy (CRT), intensity-modulated radiation therapy (IMRT), stereotactic radiosurgery, stereotactic radiation therapy, and intraoperative radiation therapy (IORT) are also other used techniques that belong to this category.⁵²⁻⁵⁵

Immunotherapy

Immunotherapy is the biological analogy of chemotherapy (also discussed in this chapter). It is also known as biological response modifier therapy, biologic therapy, or biotherapy. The idea is simple: researchers are looking for natural components found in

the human body that control the growth of abnormal cells.⁵⁶ Examples of these agents are interferons, interleukins, or cytokines.⁵⁷⁻⁶⁰ The production of monoclonal antibodies that could target cancer cells has been extensively promoted. Currently, Baviximab, is in clinical trials. The antibody was designed against the membrane phospholipid phosphatidylserine and has been utilized in the treatment of lung cancer.⁶¹ Rituximab (a monoclonal antibody against CD20, a protein highly expressed in B cells) and trastuzumab (a monoclonal antibody against HER2/neu, a proto-oncogene) are FDA approved immunotherapy agents that are currently used in the treatment of lymphoma and breast cancer, respectively.⁶²⁻⁶⁴ Furthermore, prostate cancer patients now have the option to get treated with Sipuleucel-T, a vaccine for hormone-refractory prostate cancer.⁶⁵

Targeted Therapies

The latest discoveries in cancer treatment can be compiled under this category. The objective of these agents is to control the growth, division, and life-cycle of cancer cells. Currently, there are three sub-categories for targeted therapies. Growth signal inhibitors, angiogenesis inhibitors, and apoptosis-inducing drugs. Growth signal inhibitors are compounds that specifically target growth factors and other proteins, such as receptors, that are closely involved in cell signaling. Gefitinib, imatinib, cetuximab, dasatinib, and nilotinib are some of these compounds that showed good efficacy when treating patients with lung cancer, kidney cancer, and glioblastomas.^{66,67} Angiogenesis inhibitors are aiming to target the generation of blood vessels that are formed around tumors. Angiogenesis is the biological process that generates new networks of blood vessels. Bevacizumab, an angiogenesis inhibitor, has been extensively used for the

treatment of numerous cancers, including kidney, lung, and advanced colorectal.⁶⁸

Apoptosis-inducing drugs are specifically designed to promote programmed cell death in cancer cells. Apoptosis, which is the process of programmed cell death, is another biological process for target-based therapies.⁶⁹ Table 1 summarizes some of these anti-cancer agents and their targets.⁷⁰

Table 1. Selected Anti-Cancer Agents that are Used as Targeted Therapeutic Agents⁷⁰

Hematologic Malignancies		Solid Tumors	
Molecular Target	Anti-Cancer Agent	Molecular Target	Anti-Cancer Agent
CD20	Rituximab ⁹⁰ Y-Ibritumomab ¹³¹ I-Tositumomab	HER2/neu	Lapatinib Trastuzumab
CD33	Gemtuzumab ozogamicin	EGFR	Erlotinib Gefitinib Cetuximab Panitumumab
CD52	Alemtuzumab	VEGFR	Sorafenib Sunitinib
BCR-ACL	Imatinib Dasatinib	VEGF	Bevacizumab

Chemotherapy

One of the most important therapies in the fight against cancer is chemotherapy. This therapy relies on the power of chemical agents in the eradication of cancer cells by altering their DNA or affecting important steps in cell growth and replication. Aminopterin, for example, is a potent inhibitor in DNA replication and is used for the treatment of leukemia.⁷¹ Methotrexate, is also used for the treatment of choriocarcinoma (a form of uterus cancer).⁷² The list of chemotherapeutic agents is extensive and several reviews offer more complete information about each one of them. However, undesired

effects accompany these treatments. Some side-effects include nausea, vomiting, and hair loss. Thus, researchers are always looking for new chemotherapeutic agents that overcome these problems. Chemotherapeutic agents are usually divided into seven categories. Table 2 shows the classification of chemotherapeutic agents, their targets, and some examples for each type of compounds.⁷³

Cancer Metastasis

It is well known that the main reason for death within patients with any kind of cancer is the metastasis of the tumor. Thus, it is important to understand the biochemistry and cell biology of metastasis in order to identify potential targets with new and more powerful agents. Metastasis is the process when a cancer tumor (primary) spreads and invades other organs or tissues (metastasis). However, the metastatic tumor conserves identical features to the primary tumor, that is, overexpression or underexpression of proteins, morphology, and similar chromosomal composition. Once the tumor undergoes the metastatic process, survival rates decrease dramatically and treatment becomes inefficient.⁷⁴ Metastasis can occur in almost every case of cancer. Lungs are the most common target organ for metastasis. Several cancers including breast, colon, kidney, melanoma, ovary, pancreas, prostate, rectum, stomach, thyroid, and uterus metastasize to this organ. Lung cancer accounts for the highest number of new cases and deaths among cancer patients. Other metastatic organs are liver, bones, peritoneum, and adrenal glands.^{69,75}

Table 2. Classification of Chemotherapeutic Agents

Type	Mechanism of action	Examples	References
Alkylating agents	Alters DNA structure	Cyclophosphamide Busulfan Temozolomide Cisplatin	76-78
Antimetabolites	Inhibits DNA replication	Cladribine Flouxuridine Fludarabine Methotrexate	77,79
Anti-tumor antibiotics	Inhibits enzyme involved with DNA replication	Daunorubicin Doxorubicin Epirubicin Mitomycin-C	80,81
Topoisomerase inhibitors	Inhibits topoisomerases	Topoisomerase I inhibitors Topotecan Irinotecan Topoisomerase II inhibitors Etoposide Teniposide	82,83
Mitotic inhibitors (Tubulin-binding agents)	Inhibit cell mitosis or enzymes related to cell reproduction	Paclitaxel Docetaxel Vinblastine Vinorelbine	84
Corticosteroids	Hormone-type therapy	Methylprednisolone Dexamethasone	85-87
Others		Bortezomib (L-asparaginase inhibitor)	88

The Metastatic Mechanism

Several reviews have been written about the metastatic cycle. One of them is presented by Geiger and Peeper.⁸⁹ The complexity of this mechanism can be summarized into eight steps. Invasion and migration (steps one through three) will be further discussed in chapter two.

1. Tumor cells experience an epithelial-mesenchymal transition.
2. Tumor cells degrade the extracellular matrix (ECM) by using proteolytic enzymes.
3. Tumor invades surrounding tissues or organs.
4. Tumor intravasation into a new or pre-existing network of blood vessels
5. Tumor cells are transported through the blood vessel system.
6. Tumor cells undergo extravasation at the final destination (metastasis site)
7. Tumor cells could undergo a lag phase (asymptomatic phase)
8. Tumor cells start the formation of a new tumor in the metastatic site. This process requires ECM remodeling and angiogenesis.
9. Cells surrounding the acidic microenvironment undergo anoikis or programmed cell death.

Proteolytic Enzymes (Proteases)

One of the key steps in cancer metastasis is the role of proteolytic enzymes in the migration and invasion of cancer cells. Thus, the importance of proteases as anti-cancer targets has been observed by several researchers.⁹⁰

Classification of Proteases

A general classification divides proteases by their active site. Usually, proteases are divided into four categories: serine proteases, aspartate proteases, cysteine proteases, and metalloproteases.⁹¹ Table 3 summarizes the main features and differences for each sub classification of these proteases.

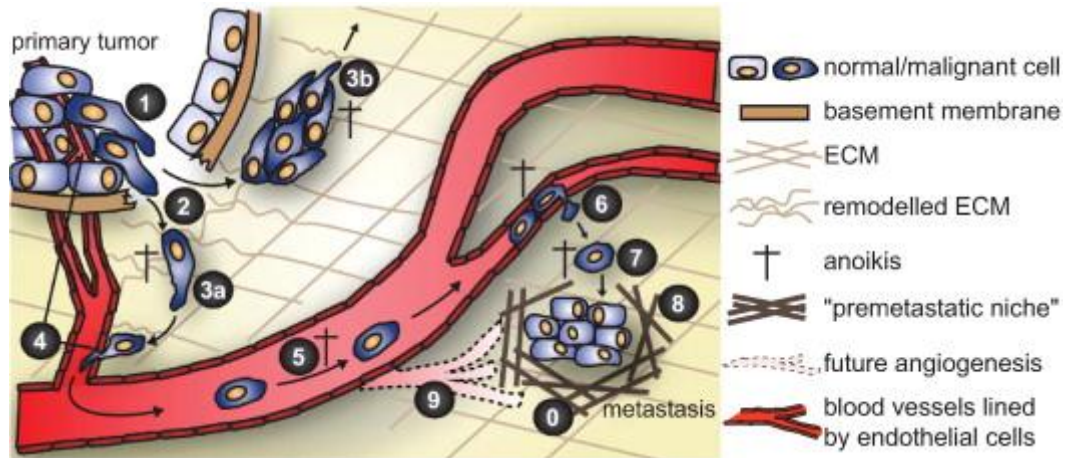


Figure 2. The Metastatic Process. (directly reproduced from Geiger, page 294)⁸⁹

The Cathepsins (Cysteine Proteases)

Currently, there are fifteen well characterized cathepsins found in nature.⁹² Interestingly, they do not belong to the same classification as serine, aspartate, or cysteine proteases. Two cathepsins are classified as serine proteases (cathepsins A and G);^{93,94} two cathepsins are aspartyl proteases (cathepsins D and E),^{95,96} and the rest of them are classified as cysteine proteases (cathepsins B, C, F, H, K, L, O, S, V, W, and X).⁹² The cathepsins are considered as peptidases; that is, they have the ability of cleave peptidic bonds. Expression of the cathepsins is mainly found in the lysosomal, cellular organelles responsible of processing cellular debris.⁹⁷ Cathepsin activity is usually maximal under

acidic conditions. Cysteine proteases are also classified by their specific target. Cathepsins L, S, K, V, and F are endopeptidases, while cathepsin C has exopeptidase and aminopeptidase activities; cathepsin X is an exopeptidase; and cathepsin H is an aminopeptidase and exopeptidase. Cathepsin B has both exopeptidase and endopeptidase activities. The present work will focus on two cathepsin cysteine proteases, cathepsin K, and cathepsin L, and cruzain, a cathepsin L-like enzyme. Cruzain, a parasitic protease found in *Trypanosoma cruzi* will be discussed in detail in Chapter 4.

Table 3. Classification of Hydrolases

Protease	Main feature/mechanism of action	Examples	References
Serine/Threonine proteases	Cleave peptidic bonds by using a serine as a nucleophilic residue in the active site	Trypsin Elastase Chymotrypsin	98,99
Aspartate proteases	Two highly conserved aspartate residues are found in the active site (acid-base mechanism)	Pepsins Renins Cathepsins	100
Cysteine proteases	A cysteine residue is the key amino acid in the catalytic activity of the enzyme	Cathepsins Papain Caspases	92,101
Metalloproteases	Their active site contains a metal, usually zinc or cobalt	MMP1 MMP8 MM14	102

Classification of the Cathepsins

Proteins can be classified according to their structural characteristics. The Structural Classification of Proteins (SCOP) and the CATH protein structure classification are usually well accepted classification systems for macromolecules. Table 4 shows how cysteine proteases are classified based on their structural features. The

SCOP system classifies cysteine proteases as alpha + beta proteins. The fold is named cysteine proteases, and the superfamily is called cysteine proteases. Finally, they all belong to the papain family. A similar classification can be found under the CATH system.^{103,104}

Table 4. SCOP Classification of Cysteine Proteases

SCOP	Name
Classification	Alpha + beta
Fold	Cysteine proteinases
Superfamily	Cysteine proteinases
Family	Papain

Distribution of Cysteine Cathepsins

Seven cysteine proteases (cathepsins B, L, H, O, F, X, and C) are ubiquitously distributed in mammals. Liver, for example, expresses five of these proteins (cathepsins F and C are not expressed). Other organs that express these cysteine proteases are thyroid gland, kidneys, spleen, and placenta. In contrast, cathepsins S, K, V, and W are expressed in specific organs or cells. For example, cathepsin K is usually expressed in osteoclasts and macrophages; cathepsin S is expressed in the heart and spleen. Similar cases can be seen for cathepsin V and W that are expressed in cornea and lymph nodes, respectively.¹⁰⁵ Table 5 shows cysteine proteases that are found ubiquitously in mammals organs.

Table 5. Expression of Cysteine Proteases that are Distributed Ubiquitously
(Adapted from Berdowska)

Cathepsin	Liver	Thyroid Gland	Kidney	Spleen	Placenta	Ovary	Lung
B	+	+	+	+	-	-	-
L	+	+	+	-	-	-	-
H	+	-	+	+	-	-	-
O	+	-	-	+	+	+	-
X	+	-	+	-	+	-	+
C	-	-	+	+	+	-	+

Biological Roles of Cysteine Proteases

Proteases play an important role in biological processes. Specifically, cysteine proteases are quite involved in the degradation and processing of important macromolecules involved in several cycles.^{105,106} Table 6 summarizes some of the most important biological processes where cathepsins are primarily involved. The most important biological functions are related to protein processing, immunity, bone remodeling, and differentiation of keratinocytes, reproduction processes, and apoptosis.

Additionally, cathepsin H has been linked with the atherogenesis processes, and the transformation of LDL and protein C macromolecules.¹⁰⁷ Similarly, cathepsin X is linked to immune system processes and inflammation.¹⁰⁸ Cathepsin V is able to degrade plasminogen, releasing subproducts that are angiostatin-like proteins.¹⁰⁹ It has been demonstrated that cathepsins L, S, and K might participate in obesity and weight loss in animal studies. The importance of cysteine protease in pathological roles will be discussed separately.¹¹⁰

Table 6. Selected Biological Roles of Cysteine Proteases.

Function	Specific activity	Cathepsin	References
Protein activation	Proteolysis of thyroglobulin	B, L, K	111–113
	Activation of β -galactosidase, renin and trypsin	B	
	Activation of granzymes A and B	C	
	Activation of SP-C in lung	H	
Mediation of antigen presentation	Generation of antigenic peptides	B, L, S	114,115
	Mediation of MHC II antigen	S, L	
	Release of CLIP by Ii processing	S, L, F, V	
Bone remodeling	Bone remodeling (resorption and formation)	K	116
Hair follicle cycle	Anomalies in hair follicle cycle and skin morphology	L	117
	Terminal differentiation of keratinocytes		
Reproduction cycles	Spermatogenesis	L	118
	Oogenesis and embryogenesis		
	Embryo partition		
Apoptosis	Granzymes A and B (implicated in apoptosis)	C	119
	TNF- α induced apoptosis	B	
	Activation of procaspase 3	L	

Mechanism of Action of Cysteine Proteases

There are three key active site residues: Cys25 (L), His159 (R), and Asn175 (R). The former two residues form a thiolate-imidazolium ion that is stabilized by Asn175 via hydrogen bonds. The difference between serine and cysteine proteases is cysteine residues are present in an ionized form. The thiolate moiety attacks the carbonyl amide carbon of the substrate to form a tetrahedral intermediate. Then, the complex converted into an acyl enzyme, releases the C-terminal portion of the substrate. A molecule of

water hydrolyzes the acyl-enzyme intermediate by forming a second tetrahedral complex. Lastly, the complex, enzyme-intermediate, releases the intact enzyme and the N-terminal portion of the substrate. (Figure 3).¹⁰⁶

General Structure of Cysteine Proteases

Cysteine proteases are expressed with three well defined portions. They all contain a signal peptide, a propeptide, and the mature domain. The signal peptides are short polypeptides having between 15 (cathepsin K) and 24 residues (cathepsin C). The propeptides are medium size polypeptide regions that range between 38 (cathepsin X) and 251 residues (cathepsin F).¹²⁰

Finally, the mature domain varies in length depending on each protein. Cathepsin F contains the shortest amino acid sequence –ironically, this protease’s pro-region is bigger than the mature form- and cathepsin B is the biggest protein with 260 amino acid residues.¹⁰⁶ The catalytic active site is composed of three amino residues: Cys25 (L), His159 (R), and Asn175 (R) (papain numbering). Cys25, which acts as nucleophile during peptide hydrolysis, is surrounded by a conserved polypeptide made of 9 residues: Cys-Gly-Ser-Cys-Trp-Ala-Phe-Ser. The second residue, His159, is embedded in conserved hexapeptide that has the sequence: Gly-**His**-X-X-X-Gly, where X could be any aliphatic hydrophobic residue (Val, Ile, or Leu). Lastly, Asn175, the third amino acid, composes a tripeptide with the form **Asn**-Ser-Trp. Both Cys25 and His159 interact with two polar and one hydrophobic residue (Gln19, Gly68, Trp183), the N-terminus Pro2, and some cysteine residues as well.

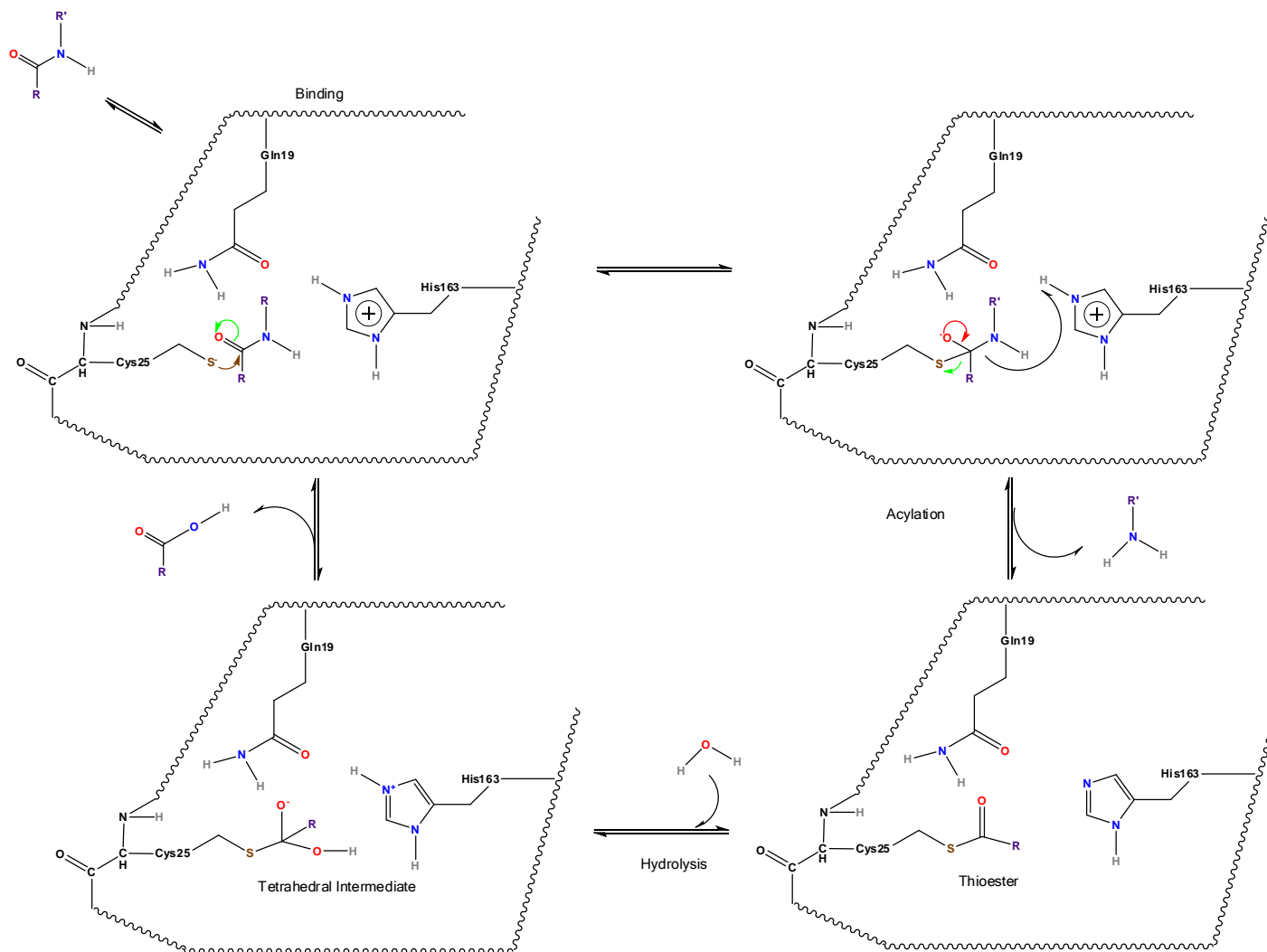


Figure 3. Mechanism of Action of Cysteine Proteases

Deeper analysis of sequence alignment reveals that cysteine proteases can also be divided into three cathepsin-like: cathepsin B-like, cathepsin L-like, and cathepsin F-like.

For example, cathepsins S, K, V, and L are cathepsin L-like. Furthermore, Karrer's work found specific characteristics for each subclass. Cathepsin L-like proteases share the ERF/WNIN motif. The research group found cathepsins, with the exception of cathepsin B, have proregions made with 107 amino acids containing a highly conserved hexapeptide (ERFNIN) using cloning techniques.¹²¹ Similarly, cathepsin F-like subfamily contains the ERFNAQ/A motif. Finally, cathepsins C, O, and X cannot be classified as a cathepsin-like subclass.

Crystal Structures of Cysteine Proteases

Crystal structures of nine members of human cathepsins have been elucidated. The list includes cathepsins, B, L, K, H, X, V, C, S, and F. Crystal structures for cathepsins O, W, and X have not yet been resolved.^{92,122–130}

Cysteine proteases are formed by two domains: left (L) and right (R). The L-domain's main characteristic is formed by three α -helices. The longest helix is called the central helix, and is a polypeptide of more than 30 residues long. The R-domain could be described as a β -barrel forming a coil-type motif. The barrel is enclosed by an α -helix at the bottom of the macromolecule. One of the components of the catalytic triad, His159, is located at the top of the barrel. Overall, both domains open up, forming what is called the active-site cleft. This is the part of the protease where Cys25 and His159 are located. Cys25 comes from the N-terminus of the L-domain while His159 comes from the C-terminus of the R-domain.

The active site surface is made of residues coming from four loops, two from each domain. The L-domain loops are shorter and disulfide-bond connected. The R-domain loops, on the other hand, are larger in length and are placed at the top of the β -barrel motif.

However, endopeptidases and exopeptidases have subtle differences in their crystal structures that help their substrate affinity. First, the active-site in endopeptidases is located between both domains. Nevertheless, exopeptidases have fewer binding sites due to the fact that more loops are occupying space of the active-site. Figure 4 shows General representation of cysteine proteases. Three active residues (Cys25, His159 and Asn175) are shown in ball and stick form. The structure is colored according to the following: α -Helices are red, β -sheets are cyan, turns are green, and coils are white. (PDB ID: 1ME3).¹³¹

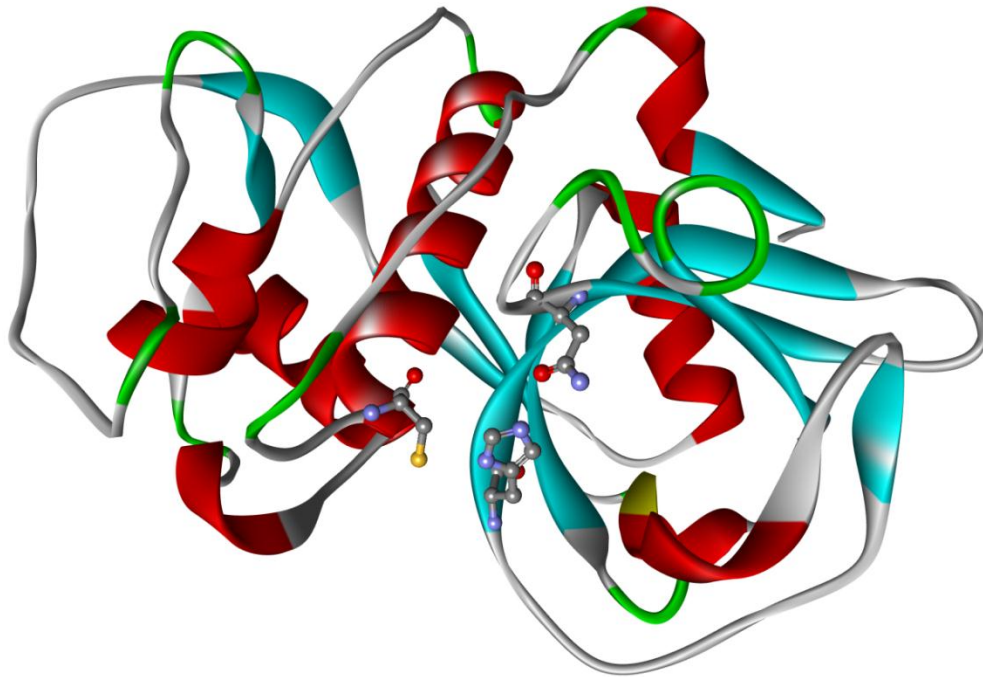


Figure 4. General Representation of Cysteine Proteases

Relationship between Cysteine Proteases and their Substrates: The Active Site

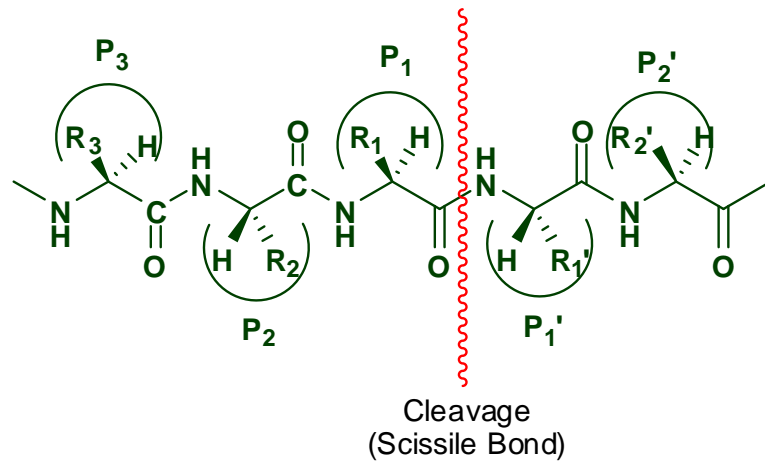
Standard nomenclature for cysteine proteases was established by Schechter and Berger forty-five years ago. They used papain to determine binding sites between the hydrolase and its substrates. This nomenclature is still used and accepted among cysteine proteases. The structures of the cathepsin family form several different subsites: S_n , S_{n+1} , ..., S_n' and S_{n+1}' . Nonprimed subsites refer to the N-terminal while primed nomenclature refers to the C-terminal of the substrate. Figure 5 is a two-dimensional representation of papain binding sites. In general, five substrate binding sites are distributed around the catalytic active site. S_1 , S_3 , and S_2' are located at the left of the active site, while S_2 and S_1' can be found at the right side of the active site. S_2 , S_1 , and S_1' are substrate binding sites but S_2 is the only one that forms a pocket. The other substrate-binding sites are just considered areas. Four loops, two from each domain, are involved in the substrate-binding process: Gln19-Cys25, Arg59-Tyr67 (L-domain), and Leu134-His159, Asn175-Ser205 (R-domain). A disulfide bond made between Cys22 and Cys63 is also found.⁹² Figure 5 shows the Schechter-Berger notation for binding sites in proteases^{132,133} and the schematic presentation of substrate binding sites for cysteine proteases is shown in a view along the active site cleft.⁹⁷

Characteristics of Procathepsins

Cysteine proteases are expressed as preproenzymes. The peptide signal, with a length average of approximately eighteen amino acids, binds to the signal recognition particle (SRP).

The propeptide region is a polypeptide that varies in length and differs from every cathepsin (38-251 amino acids). The importance of the propeptide can be summarized in three different functions.¹²⁰

A



B

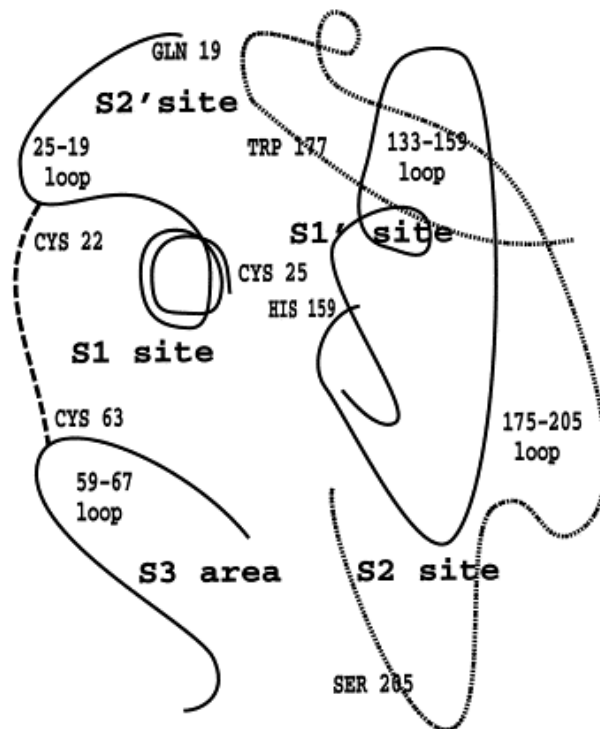


Figure 5. **A.** Schechter-Berger Notation for Binding Sites in Proteases.(adapted from Fersht).^{132,133} **B.** Schematic Presentation of Substrate Binding Sites for Cysteine Proteases. (directly reproduced from reference Turk, page 100)

1. Protein fold. Some of these proregions function as a moiety for protein folding of the mature form.
2. Transporter. The proregion can help to transport the mature form of the enzyme to the lysosomes by using specific mannose-6-phosphate receptor pathway.¹³⁴
3. Inhibitors. The proregion acts a potent reversible inhibitor of the enzyme and undesired activation of the cathepsins is prevented.¹³⁵

Similarities of Proenzymes

Four proenzymes of the cathepsin family have been elucidated (procathepsins B, L, K, and X).¹³⁶⁻¹³⁹ All of them showed that their proregions bind to the mature forms of the proteases blocking the active-site binding site preventing substrate-enzyme interaction. Hydrophobic interactions, salt bridges, and hydrogen bonds are the most common types of interaction between proregions and their respective mature forms. Interestingly, procathepsin X shows a disulfide bridge between Cys25 and one cysteine residue coming from the proregion (Cys10).¹³⁹

Mechanisms of Proenzyme Activation

Activation of cysteine proteases is simply cleaving the proregion that is connected to the mature form. In general, there are two forms by which proenzymes can be activated:

1. Enzyme-catalyzed reaction. In this mechanism, other proteins are involved in the activation of cysteine proteases. That is the case of cathepsin D, an

aspartyl protease. Cathepsins C and X do require other cathepsins (L and S) for their respective activation.¹⁴⁰⁻¹⁴²

2. pH dependent. Some cysteine proteases can undergo an autocatalytic process under acidic conditions. The mechanisms involve several other mechanisms that are uni- and bimolecular. First, the procathepsin cleaves its propeptide under acid conditions. Then, a newly mature proenzyme intervenes and helps the activation process of inactive macromolecules, generating a chain reaction that helps the activation process.¹⁴³

Similarities of Propeptides Amino Acid Sequences

Cathepsin propeptides show little identity. This can be seen in the discrepancies found in lengths. As previously stated, cathepsin X propeptide is a 38-residue polypeptide, while cathepsin F's proregion is 6 times as big as its cathepsin X analog (251 amino acid polypeptide). However, procathepsin L-like proregions average 100 residues and contain two conserved motifs: ERF/WNIN and GNFD.¹²¹

Natural Substrates of Cysteine Proteases

Cysteine proteases are non-specific proteases with high promiscuity. This behavior explains the wide distribution and expression of these proteases among several tissues and organs. The extracellular matrix (ECM) is composed of numerous proteins that are degraded by cysteine proteases. Important components of the ECM are proteoglycan, collagen, elastin, fibronectin, laminin, osteocalcin, and osteonectin. Table 7 shows ECM components that have been shown to be susceptible to degradation by selected cysteine proteases.¹⁴⁴

Table 7. Degradation of ECM Components by Cysteine Proteases (adapted from Parks)

Cat	Proteoglycan	Collagen	Elastin	Fibronectin	Laminin	Osteocalcin	Osteonectin
B	+	+	-	+	+	+	+
S	-	+	+	+	-	+	-
V	-	-	+	-	-	-	-
H	-	-	-	-	-	+	

Inhibitors of Cysteine Proteases

Cysteine proteases are tightly regulated by other macromolecules found in mammalian organisms. This section will present a summary of the most important natural inhibitors found in nature. The three most important classes of cysteine protease inhibitors are cystatins, thyrpains, and serpins.¹⁴⁵⁻¹⁴⁸ However, it is important to add that cathepsins' proregions are also potent inhibitors of the mature forms. Other macromolecules that show inhibition against cathepsins are: CRES (cystatin-related epididymal spermatogenic protein), testatin, cystatin-T, sialostatin L2, and sialostatin.⁹² Specific inhibitors for cathepsins L and K will be discussed in chapters two and three.

Synthetic Inhibitors of Cysteine Proteases

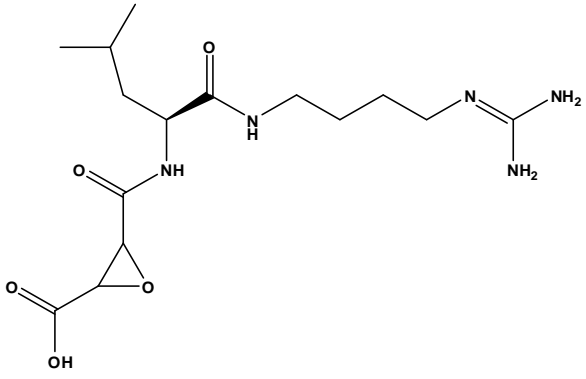
The importance of cathepsins in pathological processes, such as cancer, has led to numerous investigations with the aim to synthesize potent, yet selective, cathepsin inhibitors. The volume of generic and specific inhibitors for the cathepsins is quite large, and previous generations of inhibitors have been investigated.¹⁴⁹⁻¹⁵²

One of the most potent inhibitors of cysteine proteases is E-64, a natural compound isolated from *Aspergillus japonicas*. Chemically, E-64 is an epoxide that inhibits irreversibly numerous cysteine proteases including cathepsins B, L, K, and S.

Table 8 depicts the chemical structure of E-64 and selected IC₅₀ values against selected cathepsins. E-64 became the building block for analogs containing epoxides as their main moiety.¹⁵³ Other groups synthesized CA030 and CA074 as cathepsin B inhibitors.¹⁵⁴ CA030 is a peptidyl ethyl ester derivative of E-64, while CA074 is a methylated analog of E-64. Similar compounds have been developed for cathepsin S as well.¹⁵⁵

Leupeptin, an aldehyde, is also a potent cysteine protease inhibitor. The disadvantage of this natural product is it is not specific. Leupeptin inhibits serine and cysteine proteases and therefore its use in cysteine proteases research is limited.¹⁵⁶ Nevertheless, other functional groups have also demonstrated generic and specific activity against cathepsins. The list includes: diacyl bis-hydrazides, diamino pyrrolidinone, aldehydes, cyclic ketones, nitriles, epoxysuccinyl derivatives, vinyl sulfones, diazomethanes, azepanone-based compounds, ally sulfones, and β-lactams.¹⁰⁶ Synthetic inhibitors for cathepsins L and K will be reviewed in chapters two and three.

Table 8. Chemical Structure of E-64 and Selected IC₅₀ Values against Cathepsins.¹⁵⁶

	Cathepsin	IC ₅₀ (nM)
	L	2.5
	S	4.1
	K	1.4

Thiosemicarbazones as Potential Cancer Chemotherapeutic Agents

Thiosemicarbazones, derivatives of semicarbazones, are a well known type of synthetic compounds. They have received special attention due to their inhibitory activity of biological processes. Research has been conducted to test their antibacterial, antiviral, antifungal and antineoplastic effects.¹⁵⁷ However recent studies have confirmed that thiosemicarbazones have the ability to inhibit human and parasitic cysteine proteases.⁸⁻¹³

Importance of Cysteine Proteases in Medicine

A general overview has been described throughout about this chapter about biological and chemical properties of human cysteine proteases. However, cysteine proteases are also found in other organisms, such as bacteria, protozoa, and plants (papain, the main cysteine protease, is a potent hydrolase found in papaya, a tropical fruit). Table 9 summarizes some of the roles of cysteine proteases in human pathological conditions.¹⁰⁵

Table 9. Cysteine Proteases and Their Participation in Human-Related Diseases

Disease	Description	Cathepsin	References
<i>Osteoporosis</i>	Bone-related disease. Characterized by excessive fragility in human bones. Osteoclasts, bone cells that are responsible for bone resorption, express high concentrations of cathepsin K. Cathepsin K is a powerful hydrolase that can catalyze several types of collagen and other proteins. Deficiency of cathepsin K produces a pycnodysostosis.	K	158

Table 9. Cysteine Proteases and Their Participation in Human-Related Diseases
(Continued)

Disease	Description	Cathepsin	References
<i>Rheumatoid arthritis</i>	This is an inflammatory-related disease that is characterized by excessive degradation of type II collagen, one of the cartilage's main components. Overexpression of cathepsins L, B, S, and K contributes to this condition.	K, L, B, S	159,160
<i>Inflammatory and immune diseases</i>	Myasthenia gravis, inflammatory bowel disease, and asthma are classified under this category. <i>In vivo</i> research has concluded that selective inhibition of cathepsin S might be a potential therapeutical target in the combat against inflammatory diseases. Furthermore, overexpression of cathepsin V, a cathepsin-L like protease, has been found in patients with myasthenia gravis, an autoimmune disease.	S, V	161,162
<i>Atherosclerosis</i>	This condition is characterized by a reduction of blood flow in arteries. Cathepsins S and K are strongly linked to the degradation of elastin. Low levels of elastin reduce the elasticity of arteries and veins	S, K	163,164
<i>Periodontitis</i>	Strong evidence relates cathepsins L and B to this condition which is characterized by the inflammation of gums and destruction of connective tissues supporting the teeth.	L, B	165

Table 9. Cysteine Proteases and Their Participation in Human-Related Diseases
(Continued)

Disease	Description	Cathepsin	References
<i>Respiratory system-related diseases</i>	Overexpression of numerous cathepsins (B, S, L, and H) catalyzes a chronic obstructive pulmonary disease (COPD). COPD is characterized by a chronic bronchitis and emphysema.	L, B, S	166,167
<i>Pancreatitis</i>	<i>In vivo</i> research with mice demonstrated that cathepsin B prematurely activates trypsinogen, one of the most potent serine hydrolases found in the pancreas. A disruption in the trypsinogen activation cycles produces acinar cell necrosis.	B	168
<i>Psoriasis</i>	Expression of procathepsins L and B have been found in patients with this condition. Cathepsin S is also overexpressed in human keratinocytes.	L, B	169–171
<i>Papillon-Lefevre and Haim-Munk syndromes</i>	Both syndromes are gene-related diseases. Characteristics of these conditions are teeth loss and hyperkeratosis of specific epithelial tissues. Research has demonstrated a cathepsin C mutation gene is responsible for these syndromes.	C	172
<i>Other diseases</i>	Patients with Gaucher disease overexpress cathepsins B, K, and S. Cathepsin H is strongly linked with patients with asthma. Finally, Sjogren’s syndrome, a rare condition characterized by xerostomia, is cathepsin B-dependent.	B, K, S, H	173–175

General Introduction of Enzyme Kinetics and Enzyme Inhibition

Enzymes are proteins that catalyze biochemical reactions in organisms. They utilize compounds found in nature and convert them into different products that can be used for other purposes.

Enzymatic reactions are extremely complex; however, models have been proposed to explain these processes. Nevertheless, the Michaelis-Menten model is a good approximation for many of these biochemical systems.¹⁷⁶

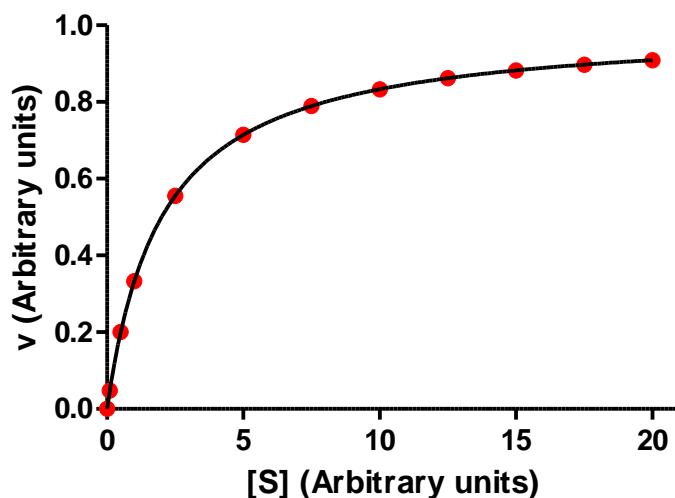
The model establishes a hyperbolic relationship between substrate concentration and enzymatic activity. Scheme 1 describes chemical reactions between an enzyme and its substrate.

First, a reversible reaction occurs between both species (E= enzyme; S= substrate), creating an intermediate, the Michael-Menten complex (ES). The second step of the process is a very slow irreversible process. The Michaelis-Menten complex is broken down into the final product and the release of the enzyme.



Scheme 1. Chemical Reactions Between Enzyme and Substrate

This simple model is the basis for the derivation of the Michaelis-Menten equation (Equation 1.1), where K_M : Michaelis-Menten constant, S: substrate concentration, V_{max} : maximum enzymatic activity, and v : velocity. The Michaelis-Menten constant, K_M , represents the amount of substrate that is necessary to achieve half maximal enzyme's activity. The value also includes the dissociation constant. The lower the value, the tighter the interaction is between both species (substrate and enzyme). A graphical representation of simulated data is depicted in Figure 6.



$$v = \frac{v_{MAX} [S]}{K_M + [S]} \quad (1.1)$$

Figure 6. Michaelis-Menten Equation. Simulated Data Give Calculated K_M and V_{max} Values of 2 and 1 Arbitrary Units, Respectively

Enzyme Inhibition. Basic Concepts

An enzyme inhibitor is a natural or synthetic compound that has the ability to reduce the catalytic activity of an enzyme. In general, inhibitors can mimic the structure of the intermediates that can be formed during the biochemical reactions (Enzyme-Substrate). Several examples of transition-state inhibitors can be found in literature and some of them have clinical applications. Hence, the importance of understanding the mechanism of action and the reaction optimal conditions are required to study the mode of inhibition. It is important for the design and study of possible candidates to function as inhibitors. Assay optimization is perhaps the most critical step in drug discovery and development. Parameters such as temperature, pH, chelating metals in solution, buffer components, and viscosity can determine the success or failure of any enzymatic assay.¹⁷⁷

Several inhibitors work in a reversible fashion. Reversible inhibition can occur when the enzymatic activity can be recovered through the time of reaction. On the contrary, irreversible inhibitors interact with the target enzyme by inactivating it, and the

protein is no longer capable of catalyzing a reaction. Extensive research aims at designing reversible and irreversible inhibitors. Irreversible inhibitors are the best candidates for infectious diseases (bacterial, viral, fungal, etc.). However, this is not applicable when the target enzyme is found in the human body and participates in biological processes. Inactivation of a human protein can be an undesired consequence of a medical treatment, and patients could be in danger of suffering subsequent conditions. For example, irreversible inactivation of human cathepsin K, a potential target in osteoporosis, could lead to osteopetrosis. Symptoms of osteopetrosis include harder and denser bones, which lead to bone fracture.¹⁷⁸

Reversible inhibitors can be classified based on their mechanism of action. In general, there are four cases: competitive, non-competitive, uncompetitive, and mixed reversible inhibition. Viagra, nevirapine, and camptothecin are classical examples of these modes of inhibitions.^{179–181} Thus, inhibition mode studies are crucial and play a major role in drug discovery and development. Quantitative comparisons can be utilized to select the best candidates for future pre-clinical and clinical trials. Kinetics parameters such as the inhibition constant (K_I), IC_{50} value and binding Gibbs free energy (ΔG^0) of that occur when the inhibitor binds to the target protein, can quantify the structural changes of the system (enzyme, inhibitor, or both) that undergo during the reaction. An important feature in inhibitor design is its selectivity. It is not uncommon to find a compound that inhibits a family of proteins with similar characteristics (usually structure or mechanism-based). Therefore, inhibition assays are performed with different enzymes in order to investigate a possible selectivity or a generic application. *Trans-*

Epoxy succinyl-L-leucylamido(4-guanidino)butane, or better known as E-64, is a potent irreversible cysteine protease inhibitor widely used in research.¹⁸²

The discovery, optimization, and utilization of various compounds that can be potential inhibitor agents are the main focus of our research. The creation of a library and the analysis of their structure-activity relationship (SAR) can be extremely helpful in order to achieve these goals. SARs can help to optimize future and advanced pre-clinical and *in vivo* studies. They also help to identify possible selectivity and lead to the design of more analogs that could be potent inhibitors.¹⁸³⁻¹⁸⁵

Modes of Inhibition

An inhibitor is considered as competitive when it binds to the catalytic active site of the enzyme and competes with the substrate for that site. Noncompetitive inhibition occurs when the compound binds to an established enzyme-substrate inhibitor or the free enzyme. Uncompetitive inhibition is usually found when the compound binds to the enzyme-substrate complex. Finally, mixed inhibition occurs when the inhibitor binds in more than one fashion.

Table 10. Classical Examples of Reversible Inhibitors

Mode of Inhibition	Compound	Medical Condition	Enzyme Target
Competitive	Viagra (Sildenafil Citrate)	Erectile Dysfunction	Phosphodiesterase Type-5 Monophosphate
Non-competitive	Nevirapine (nucleoside)	HIV	Reverse Transcriptase
Uncompetitive	Camptothecin (Quinoline Alkaloid)	Cancer	Topoisomerase

The Definition of IC_{50} and K_I and Their Importance in Inhibition Studies

There are two important concepts in enzyme inhibition that are widely used when testing potential inhibitors with several targets: IC_{50} and K_I .

The IC_{50} value or half maximum inhibitory concentration is the concentration of a specific compound (inhibitor) that is required to inhibit the enzymatic activity of a target by fifty percent. Thus, IC_{50} values are concentration dependent and equation 1.2 describes the mathematical relationship between both variables. Y represents the inhibited activity (compared to control), X is $\log([\text{inhibitor}])$ in M , v_{\min} and v_{\max} represent the velocities of the enzyme when it is pre-incubated with the highest and lowest inhibitor concentrations, respectively. Finally, the Hill slope value is the slope of the sigmoidal curve. The data fit into a sigmoidal curve and can be seen in Figure 7.

The physiological significance of this parameter resides in the search of compounds that at relatively lower concentrations can reduce the activity of a specific target (enzyme).

The Hill slope usually ranges between -1 and 1. Unity is considered an ideal inhibitor behavior; however, there are cases when this value is not close to unity, meaning there are other factors affecting the behavior of the compound.

The determination of IC_{50} for any enzyme is method dependent. This parameter is susceptible to laboratory conditions such as temperature, buffer (type, concentration), pH, enzyme concentration, preincubation time, solvent effect, substrate inhibition, etc. Thus, the importance of establishing a validated, consistent protocol is vital for comparison and reproducibility purposes. This is also a determinant factor when comparing IC_{50} values for other systems.¹⁸⁶

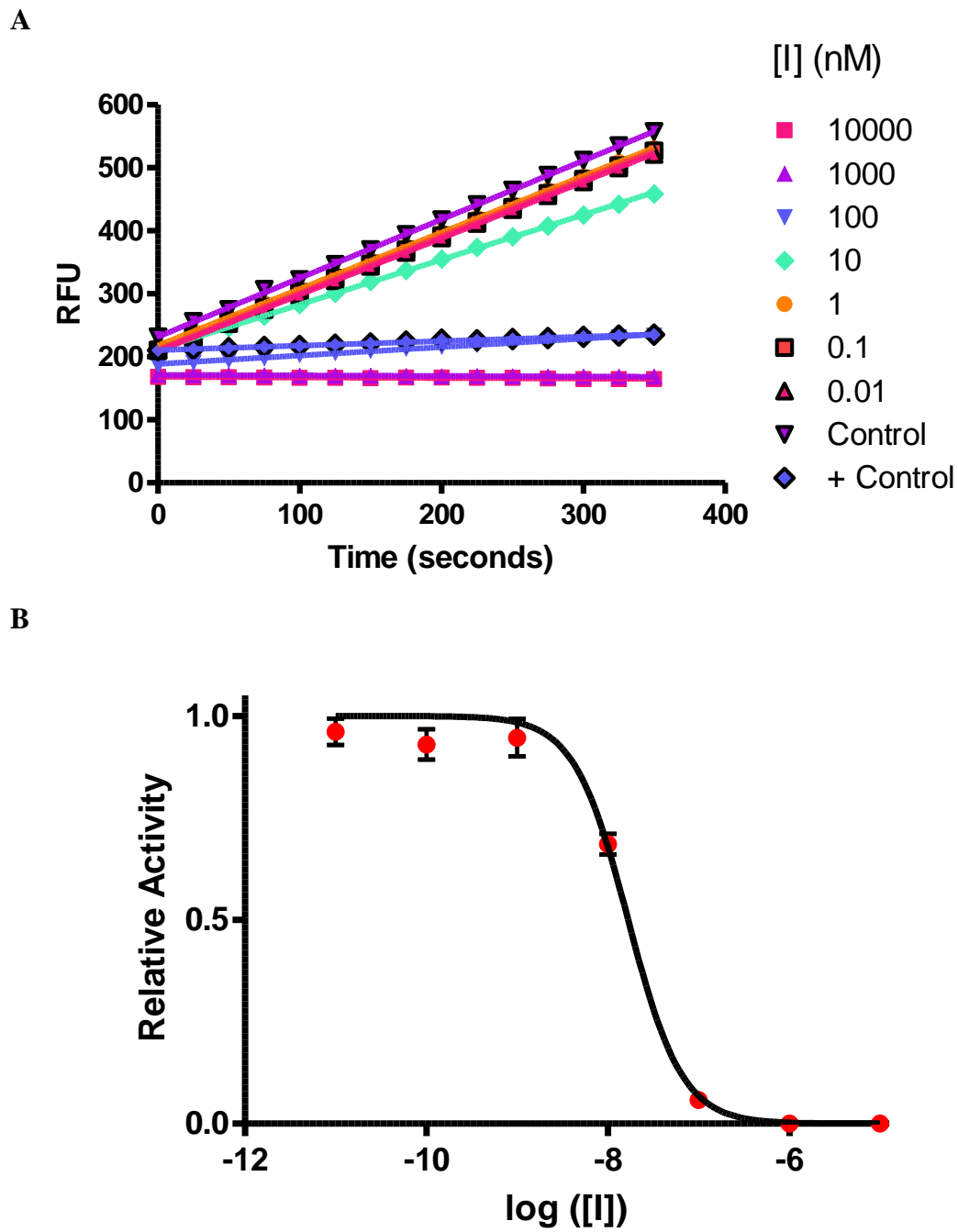


Figure 7. **A.** Inhibition of Cathepsin K Activity by an Inhibitor ($0 \leq [I] \leq 10 \mu\text{M}$). **B.** Calculation of IC_{50} values. Real Data were Fitted into Equation 1.2, v_{\min} and v_{\max} were Given Constraints Set to 0 and 1, Respectively. Results were: IC_{50} : 16.7 nM, Hillslope: -1.14 and r^2 : 0.9863

The inhibition constant (K_I , Equation 1.3) is another concept that correlates the effect of any inhibitor with a given enzyme (the lower the value, the better the inhibitory activity of any compound). In general, K_I is the ratio of k_{off} and k_{on} and, the rate constants for the inhibition offset and onset of both species (enzyme and inhibitor), respectively.

$$Y = \frac{v_{MIN} + (v_{MAX} - v_{MIN})}{1 + 10^{(\log(IC_{50}-X)*Hillslope)}} \quad (1.2)$$



There are several approaches to calculate this parameter. Equation 2.3 offers the simplest form to calculate K_I , assuming both k_{on} and k_{off} are already known.

Unfortunately, the determination of these rate constants is not always a simple process, and several authors have come up with alternative models aiming the determination of the inhibition constant. It is also important to understand the limitation of each approach in order to guarantee the most exact value for K_I . The exact calculation of K_I can be done by using the transient approach. Data sets are fitted into a system of differential equations (usually five or six equations). However, this approach requires the mode of inhibition being established or previously known; and a complex mathematical program.

A simple, yet well accepted approach was determined by Cheng and Prusoff in 1973 when they derived a system of equations that correlate IC_{50} values and K_I values.¹⁸⁷

Three equations describe these relationships when a reversible inhibitor works as a competitive (Equation 1.4), noncompetitive (Equation 1.5), and uncompetitive inhibitor (Equation 1.6). The simplicity of these equations attracts the majority of researchers.

However, the mode of inhibition needs to be established when choosing the appropriate equation.

$$IC_{50} = K_I \left(1 + \frac{[S]}{K_M} \right) \quad (1.4)$$

$$IC_{50} = \frac{[S] + K_M}{\frac{K_M}{K_I} + \frac{[S]}{\infty K_I}} \quad (1.5)$$

$$IC_{50} = \infty K_I \left(1 + \frac{K_M}{[S]} \right) \quad (1.6)$$

Slow Binding and Tight Inhibition. Determination of Mode of Inhibition

A compound is considered to be a slow binding inhibitor when the inhibition process occurs as a response of time. This means the potential inhibitor binds to an enzyme at longer times and slowly releases throughout the course of the reaction. A visual form to determine if the compound acts as a slow binding inhibitor occurs when comparing treated and nontreated reaction progress curves. A non-linear behavior is the key characteristic of this form of inhibition. Several examples of slow binding inhibitors can be found in literature. Figure 8 shows an example of reaction progress curves of an enzyme when it is inhibited by a slow binding inhibitor.

The treated progress curve can be fitted into equation 1.7 by fixing the concentrations of substrate, inhibitor, and enzyme. The release of product is dependent of the time reaction. Fitting the data into this equation solves for three different parameters (v_o , v_s , and k_{obs}); the rates v_o and v_s are the initial and steady-state velocities, t is the time, and k_{obs} the apparent first-order rate constant for the steady state that is

formed between all the species present in the reaction (enzyme, inhibitor, enzyme-inhibitor complex, and substrate).¹⁸⁸

$$P = v_s t \frac{(v_o - v_s)}{k_{obs}} (1 - e^{-k_{obs}t}) \quad (1.7)$$

The vast majority of inhibitors reported in literature have been found to work as competitive inhibitors. Nevertheless, a considerable number of compounds are noncompetitive or uncompetitive inhibitors. The mode of inhibition can be determined by using mathematical equations such as Michaelis-Menten and Lineweaver-Burk and using experimental data.^{176,189}

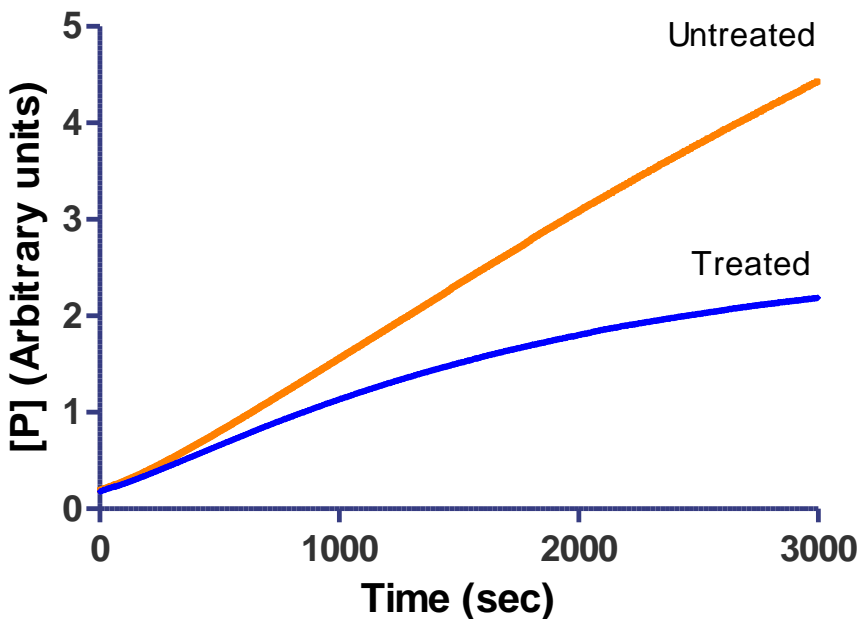


Figure 8 Typical Reaction Progress Curves of Untreated (orange) and Treated Sample with a Slow Binding Inhibitor (blue)

Alternatively, the mode of inhibition slow-binding inhibitors can be determined by using equation 1.7. The k_{obs} values are substrate dependent as seen in equations 1.8,

1.9 and 1.10. Calculated rate constants can be plotted versus substrate concentration if the inhibitor concentration is constant (Figure 9).¹⁷⁷

For competitive inhibitors, rate constants (k_{obs}) decrease hyperbolically by increasing substrate concentration (See equation 1.8). In the case of uncompetitive inhibition (equation 1.9), rate constants are substrate concentration independent. Finally, the rate constants for non-competitive inhibitors will increase hyperbolically with higher substrate concentrations (equation 1.10).

$$k_{obs} = \frac{k}{1 + \frac{[S]}{K_M}} \quad (1.8)$$

$$k_{obs} = k \quad (1.9)$$

$$k_{obs} = \frac{k}{1 + \frac{K_M}{[S]}} \quad (1.10)$$

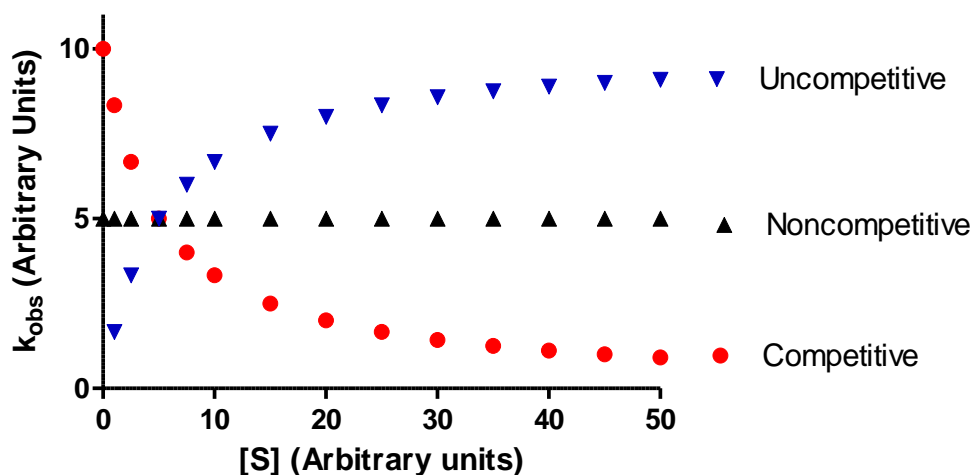


Figure 9. Determination of Mode of Inhibition for Slow-Binding Inhibitors. See Equations 1.7-1.10. Values are in Arbitrary Units. K_M : 5; k : 10 (for competitive and uncompetitive inhibition; k : 5 (for noncompetitive inhibition)

Relationship between Enzyme Concentration and K_I : The Morrison's Equation

Perhaps, the best mathematical method to evaluate inhibition constants for slow, tight-binding inhibitors is the Morrison's quadratic equation (equation 1.11). The equation expresses the K_I value in terms of the enzyme species concentrations. Thus, the inhibition constant is dependent of free enzyme and enzyme bound to the inhibitor. Mathematically, the Morrison's mathematical model is a quadratic equation; that is, it has two plausible solutions. Obviously, only one solution is physically possible because $K_I > 0$. Nevertheless, Morrison's equations solves the apparent inhibition constant K_I^{app} . Equation 1.11 solves the true inhibition constant K_I by solving Equation 1.12 with a relationship between K_I^{app} , K_I , $[S]$, and K_M .

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2 [E]_T} \quad (1.11)$$

$$K_I^{app} = K_I(1 + \frac{[S]}{K_M}) \quad (1.12)$$

Equation 1.11 presents the relation between K_I^{app} , K_I , $[S]$, and K_M and inhibitor. K_I^{app} , K_I , $[S]$, and K_M Inhibited (v_i) and uninhibited (v_o) can be experimentally determined. Michaelis-Menten, and substrate concentration should be constants. Thus, the apparent inhibition constant can be then determined by solving the dependence of the fractional activity (i.e. v_i/v_o) versus inhibitor concentration. However, other considerations, such as optimization and limitation of the model have also been studied.¹⁹⁰

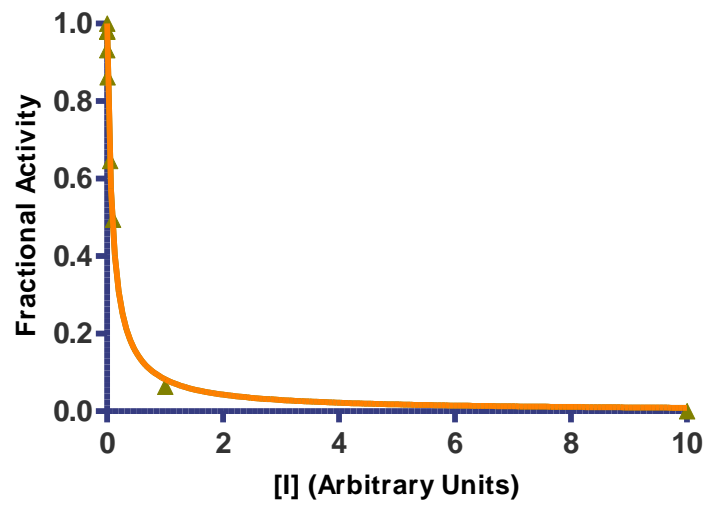


Figure 10. Graphical Representation of the Morrison's Equation. Real data was Used to Calculate K_I^{app} Conditions (in μM): [S]:50; K_M : 1; $[E_T]$: 0.001; v_o : 1; r^2 : 0.9949

CHAPTER TWO

Evaluation of Thiosemicarbazones as Cathepsin L Inhibitors

Nomenclature, Classification and Historical Background

Human cathepsin L (EC 3.4.22.15) is a hydrolase and belongs to the sub-class of peptidases, enzymes that act on peptidic bonds.¹⁹¹ Cathepsin L is also classified as an endopeptidase, that is, a hydrolase that only acts on peptidic bonds located within amino acid sequences.¹⁹²

The official name of the protein is cathepsin L (CL). CL is also known as cathepsin L1.¹⁹³ One of the first literature references about cathepsin L is from 1977 when its purification from rat liver, characterization and catalytic activity were reported.¹⁹⁴

Localization and Biological Roles of Cathepsin L

Cathepsin L is a ubiquitous hydrolase found in mammals. Cloning experiments revealed that cathepsin L mRNA is expressed among several cell lines in similar quantities. Human cathepsin L was found to be encoded in the chromosome 9q21-22.¹⁹⁵

The extraordinary catalytic activity of cathepsin L is quite fascinating because of its participation in numerous yet diverse biological processes. Cathepsin L is mostly considered a lysosomal protease and its distribution in human tissues and organs is quite extensive. Skin, liver, thyroid glands, and kidneys are organs where considerable concentrations of this protease have been detected.¹⁹⁶⁻¹⁹⁹ Animal studies showed considerable amounts of cathepsin L in intervertebral disc tissue, Type A (macrophages)

and Type B (fibroblast) cells that form the synovial lining layer.^{200,201} However, recent studies have shown that cathepsin L is also expressed in other cellular organelles. Active cathepsin L was detected in the nucleus of NIH3T3 cells during G1-S transition cell phase, controlling cell cycle progression.²⁰² Similarly, Sullivan and coworkers detected cathepsin L activity in colorectal cancer cell nuclei.²⁰³ Cathepsin L is also found as a secreted protease in the secretory vesicles where it participates in the activation of proenkephalin confirming that cathepsin L is also selectively secreted under specific conditions.²⁰⁴

Serum, cytoplasmic and nuclear proteins, are also susceptible to be degraded by this protease.^{203,205,206} Furthermore, the versatility of cathepsin L has been found in multiple processes: bone resorption, sperm maturation, intestinal neoplasia, and processing of neuropeptides into neurotransmitters and some hormones.^{118,207,208}

Cathepsin L hydrolyzes thyroglobulin in the thyroid glands, releasing two important subproducts (thyroxine and thyronine). Animal studies also revealed this protease is highly involved in the processing of peptide epitopes in thymic epithelial cells. It has also been implicated in other immunogenic processes. For example, interferon- γ enhances the activity of cathepsin L in human monocytes.^{209,210}

Genetically manipulated mice (cathepsin L expression was suppressed) showed these animals had problems with hair loss, epidermal hyperplasia, acanthosis, and hyperkeratinosis. Other studies support the role of cathepsin L in the differentiation of keratinocytes and its importance in hair pigmentation and the biology of hair cells.

Reproduction cycles (spermatogenesis) also require the activity of this protease. For example, Sertoli cells in the male reproductive system express considerable amounts

of cathepsin L. Cathepsin L-deficient mice showed dysfunctional seminiferous tubules and reduced spermatogenesis levels.^{209–213}

Female reproductive systems require the proteolytic activity of cathepsin L. Cathepsin L has been linked to oocyte generation (oogenesis). Other studies confirmed higher concentrations of cathepsin L mRNA in placenta, specifically during implantation and before labor. Mice studies also confirmed higher amounts of cathepsin L in the visceral yolk sack.^{214,215}

Finally, cathepsin L plays a crucial role in angiogenesis. The proteolytic degradation of the components of the extracellular matrix (ECM) is increased by the activity of cathepsin L.²¹⁶

Natural Substrates of Cathepsin L

Cathepsin L is one of the most potent cysteine cathepsins. Its ubiquitous distribution explains its importance. However, there is little information regarding the potency of this hydrolase in physiological conditions. Research has shown its *in vitro* efficacy in recent reports.

In vitro studies have proved extracellular matrix components, are optimal cathepsin L natural substrates under physiological conditions. Cathepsin L is capable of degrading proteoglycans, aggrecan, elastin, fibronectin, laminin, and osteocalcin.^{217–221}

Collagen deserves special attention regarding cathepsin L activity. The collagenase activity has been well demonstrated by using several types of this macromolecule. Cathepsin L collagenase activity has been demonstrated by using types II (cartilage, vitreous humor), IX (cartilage), XI (cartilage), I (skin, tendon, bone), and IV (basal lamina).^{222–224}

Catalytic Activity of Cathepsin L and Substrate Specificity

Cathepsin L is considered one of the most potent cysteine proteases found in mammalian organisms. As previously stated, the protease is able to process a wide range of proteins, from small peptides to high molecular weight proteins, such as type I collagen (Molecular Weight: ~ 300 kDa). Recent reports have characterized the substrate specificity of cathepsin L. The protease cleaves Arg-Ser peptidic bonds when synthetic polypeptides were used to test the hydrolase activity specificity. Substitution of the serine residue by a hydrophobic residue (Phe) reduces the hydrolytic activity of cathepsin L.

It was also demonstrated that cathepsin L is able to cleave more than one peptidic bond. The protease was able to hydrolyze both Phe-Ser and Ser-Ser peptidic bonds in the synthetic substrate Abz-KLFSSKQEDDNp, while Abz-KLLSSKQ-ED-DNp was hydrolyzed at two different sites (Leu-Ser and Ser-Ser). Basic amino acids residues at the P1 position (Arg and Lys) are the preferred targets for cathepsin L for peptidic cleavage. In contrast, substrates with acidic residues (Glu, Asp) and polar residues (Asn, Thr, Asn) at the P2 position are poor cathepsin L substrates. Proline-containing peptides are the worst substrates for the protease.¹⁰⁹

Other studies confirmed the preference of cathepsin L for substrates containing hydrophobic amino acids at the P2 position, positive-charge (Lys and Arg) at the P1 position and broader specificities at P3 and P4 positions.²²⁵

Structure of Cathepsin L

Cathepsin L contains 217 residues per monomer with a molecular weight between 25 and 29 kDa.¹⁹³ Cathepsin L contains one heavy chain (MW: 20-24 kDa; 175 residues,

Residues 1-175) and one light chain (MW: 5 kDa; 42 residues, Residues 176-215).^{226,227}

The amino acid sequence of the heavy chain is shown in Figure 11. Peptide sequences also confirmed the high sequence identity of this enzyme when it is compared to papain up to 41%.^{123,228}

1	APRSVDWREK	GYVTPVKNQG	QCGSCWAFSA	TGALEGQMFR	KTGRLISLSE
51	QNLVDCSGPQ	GNEGCNGL	DYAFQYVQDN	GGLDSEESYP	YEATEESCKY
101	NPKYSVANDT	GFVDIPKQEK	ALMKAVATVG	PISVAIDAGH	ESFLFYKEGI
151	YFEPDCSSED	MDHGVLVVGY	GFESTESDNN	KYWLVKNSWG	EEWGMGGYVK
201	MAKDRRNHCG	IASAASYPTV			

Figure 11. Amino acid Sequence of Cathepsin L. Legend: Blue, Heavy Chain; Red, Light Chain; Catalytic Residues, Green.^{229,230}

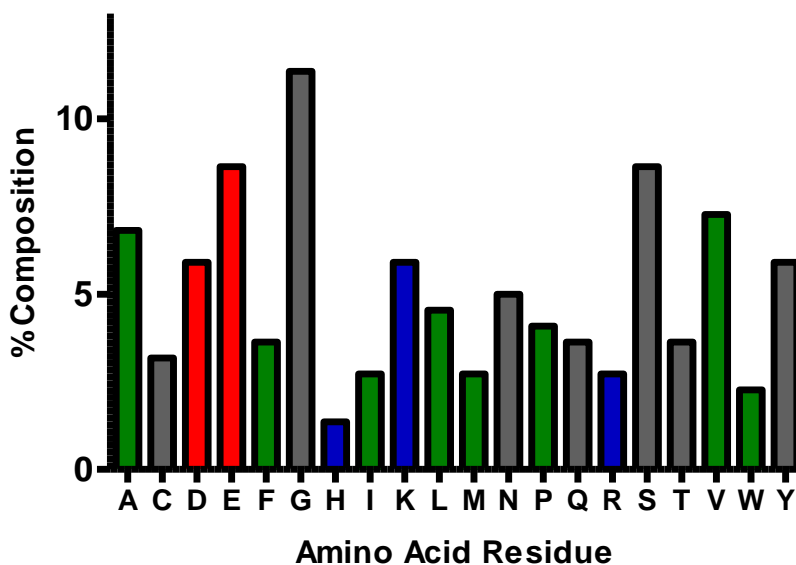


Figure 12. Composition of Human Cathepsin L. Legend= Blue: Asic Residues; Red: Acidic Residues; Green: Hydrophobic (Non Polar) Residues; Gray: Neutral, Polar Residues

The three most abundant residues are glycine, glutamate and serine (25, 19, and 19 residues, respectively). On the other hand, the least abundant residues are histidine (three residues) and tryptophan (five residues). Hydrophobic residues account for almost

one third of the total composition of cathepsin L. Approximately, 41 percent of the composition is entirely made of polar, uncharged residues.

Table 11. Amino Acid Residue Distribution of Cathepsin L

AA	#	%	AA	#	%	AA	#	%
Ala	15	6.82	Ile	6	2.73	Arg	6	2.73
Cys	7	3.18	Lys	13	5.91	Ser	19	8.64
Asp	13	5.91	Leu	10	4.55	Thr	8	3.64
Glu	19	8.64	Met	6	2.73	Val	16	7.27
Phe	8	3.64	Asn	11	5.00	Trp	5	2.27
Gly	25	11.36	Pro	9	4.09	Tyr	13	5.91
His	3	1.36	Gln	8	3.64	Total	220	100.0

The secondary structure reveals that ten α -helices constitute thirty percent of the total amino acid sequence. However, only one fifth of the sequence forms beta sheets distributed in fifteen strands.^{229,230}

Crystal Structure of Cathepsin L

One of the first reports of the analysis of cathepsin L by X-ray crystal structures revealed important characteristics of the structure.¹²³ The S' regions of cathepsin L are described as shallow depressions and show numerous similarities when compared to the S' regions of papain. The depression is limited by a flat bottom that involves interaction of the Trp189 side chain, Gln21, Gly20, Asn18, Glu192, Trp193 and Leu144. The oxyanion hole of cathepsin L is made by three residue side chains (Gln16, Trp189, and His163) and the Cys25 backbone. The S_2 pocket of cathepsin L is remarkably hydrophobic, a characteristic that is found among several cysteine proteases. The bottom S_2 pocket of cathepsin L consists of Met70, Ala135, Leu69, Met161 and Gly164, which are hydrophobic. Detailed inspection of that S_2 site reveals that Asp162, Met161, Asp160,

and Ala214 build one of the walls that encapsulate the S₂ pocket. Furthermore, this wall is the reason the S₂ subsite is a narrow pocket. Met70 is the responsible for the S₂ pocket shallow characteristic. The S₃ subsite of the pocket can be found on one of the sides of the left domain of the cleft. The S₃ residues Gly67 and Gly68 were found to be located in the middle of the former subsite. They both form an ample substructure that is surrounded by Asn66, Glu63, Leu69 and the carbonyl oxygen of Gly61. Figure 13 shows the Crystal Structure of Human Cathepsin L. Three active residues (Cys25, His159, and Asn175) are shown in ball and stick form. The structure is colored according to the following: α -Helices are red, β -sheets are cyan, turns are green, and coils are white (PDB ID: 1ICF).²²⁸

General Considerations of Procathepsin L

Cysteine proteases are often produced in mammalian organism as precursors, also known as procathepsins. Cathepsin L is expressed as a proenzyme in eukaryotic organisms. The preproform of cathepsin L, named preprocathepsin L (MW ~ 39 kDa) is inactive and consists of three major polypeptide regions: the signal peptide, consisting of 17 residues; the proregion made of 96 amino acid residues; and the mature form (217 amino acid residues).^{137,231} The proregion is located between the signal sequence and the N-terminus of the mature form. The proregion has regulatory activities. It controls the activation and activity of cathepsin L, so it acts a natural peptidic inhibitor of cathepsin L. However, it has also been suggested that both activation and processing of cathepsin L are two related, yet different processes.^{232,233} The proenzyme is found to be more stable than the mature form at higher pH ranges, because the proregion helps to stabilize the macromolecule in basic environments.¹⁴² The crystal structure of procathepsin L has also

been resolved using cloning techniques. The proregion is located over the mature portion of cathepsin L.¹³⁷

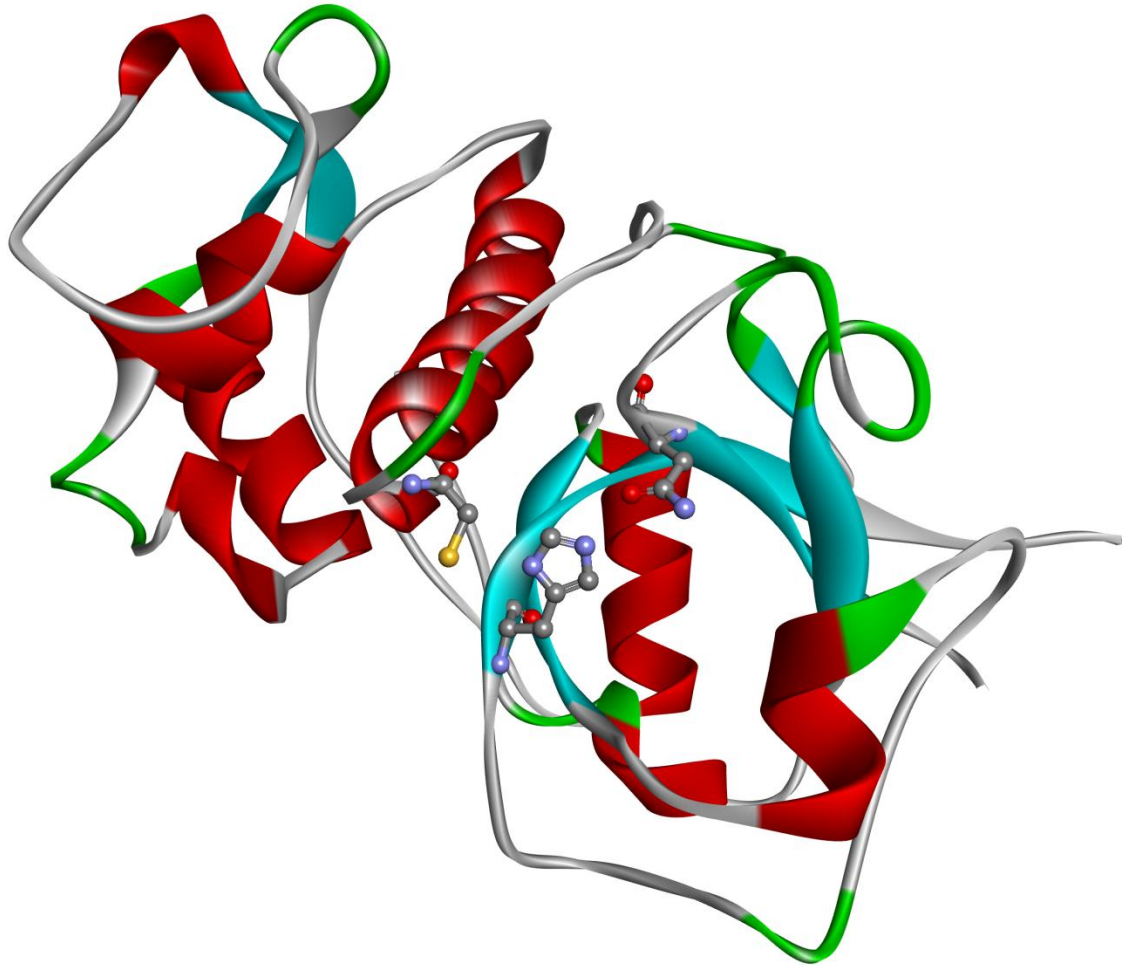


Figure 13. Crystal Structure of Human Cathepsin L (PDB ID: 1ICF)

In general, the proregion is a 75 residue globular-shaped segment and an elongated 21 residue segment. There are two major interactions sites between the proregion and cathepsin L. The first one is found by the side of the catalytic active site, and the second interaction site is with one of the connecting loops of the mature form. These findings explain the inhibitory function of the proregion against the mature form.

A hydrophobic residue (Phe71) and a small peptide region (Met75-Glu79) are the specific connecting points between the proregion and cathepsin L active site by forming numerous hydrogen bonds. The hydrophobic globular segment contains three α -helices and corresponding connecting loops. Furthermore, activation of procathepsin L can be a pH dependent mechanism. Acidic conditions promote the activation of the protease by an auto activation process. Activation of procathepsin L generates two major macromolecules: the proregion and mature cathepsin L with molecular weights of 10 and 29 kDa respectively. Further investigation has revealed the proregion of the procathepsin L can be cleaved at two different sites.²³⁴ Figure 14 shows the crystal structure of human procathepsin L. Three active residues (Cys25, His159 and Asn175) are shown in ball and stick form. The structure is colored according to the following: α -Helices are red, β -sheets are cyan, turns are green, and coils are white, and the propeptide is dark green (96 amino acid residues). PDB ID: 1CS8¹³⁷

Inhibitors of Cathepsin L

As proteases found in mammalian organisms, cathepsin L has natural inhibitors that regulate its catalytic activity and degradation. Also, numerous synthetic inhibitors have been developed in recent years to inactivate its ability due to its importance in medicine. To date, none of the possible inhibitors developed have been used in clinical trials.

Natural Inhibitors of Cathepsin L

One of the first cathepsin L natural inhibitors that was discovered is α -macroglobulin.²³⁵ This protein is extremely large (MW: 725 kDa) and due to its size, it is able to encapsulate cathepsin L within its structure, inactivating the latter one. The

cystatins are a well known class of protein inhibitors of cathepsin L.²³⁶ These natural inhibitors might act as cysteine protease regulators due to the high catalytic activity of the cathepsins. Generally, the superfamily of cystatins is divided into three major groups: stefins, cystatins, and kininogens.²³⁷ The stefins are found in epithelial cells and numerous cells and tissues.²³⁸ An interesting characteristic of this subfamily is their lack of disulfide bonds and carbohydrates.²³⁹ The inhibitory potency of stefins can be seen in reported values for stefins, which are in the picomolar range. The second family of cysteine protease inhibitors is the cystatins.

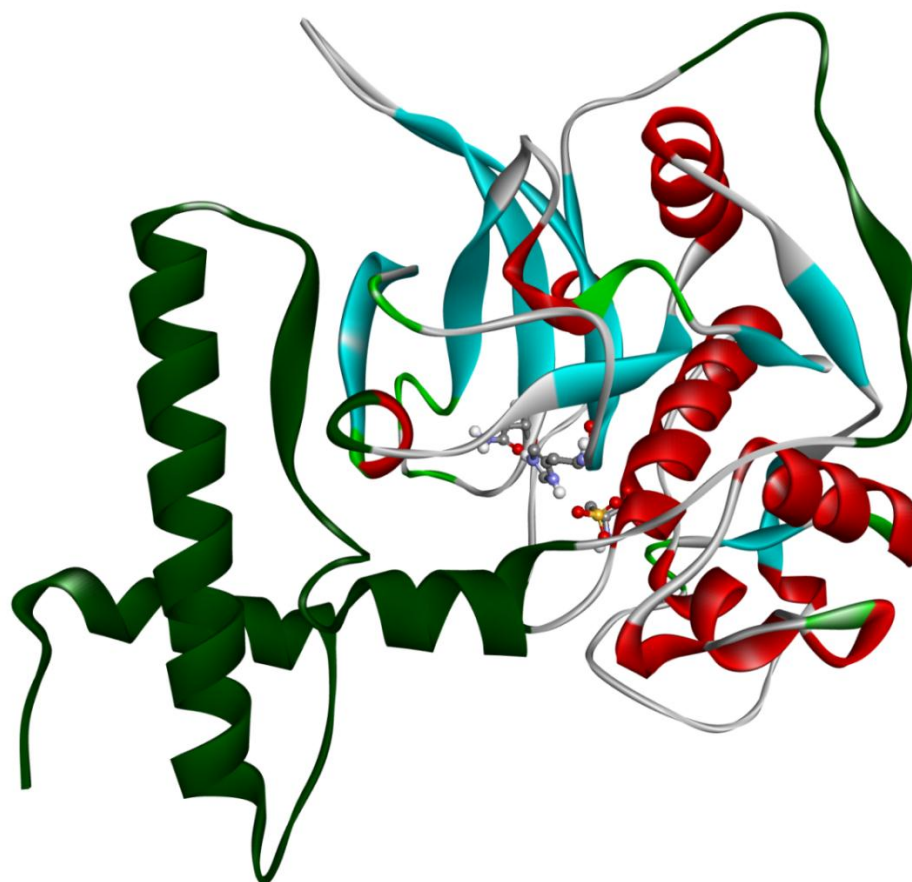


Figure 14. Crystal Structure of Human Procathepsin L (PDB ID: 1CS8)

The cystatins have similar amino acid sequences to the egg-white chicken protein found more than 30 years ago.¹⁴⁶ The characteristic feature of these proteins is the presence of two disulfide bond in close proximity to the carboxyl-terminus. Cystatins can be found in several fluids of mammalian organisms in a wide range of concentrations.²⁴⁰ Cystatin C is one of the most abundant proteins found in seminal plasma and cerebrospinal fluid, and traces have been found in plasma, saliva, and urine. Lastly, the third family of inhibitors of proteases that belong to the cystatin superfamily is the kininogen family. These proteins have regulatory activities and function as precursors of kinins, an important class of proteins that act on vasodilation and contraction of smooth muscles tissues processes.¹⁴⁸

However, other studies have confirmed there are other two different types of macromolecules that act as natural cathepsin L inhibitors. First, the thyropins, specifically the MHC class II-associated p41 invariant chain (Ii) is a potent cathepsin L inhibitor;²²⁸ and serpins, that were originally known to inhibit serine proteases. Hurpin, a member of the serpin family, is also a known inhibitor of cathepsin L.²⁴¹ Additionally, the squamous cell carcinoma antigen (SCCA) shows inhibitory activity by inhibiting cathepsin L.

Synthetic Inhibitors of Cathepsin L

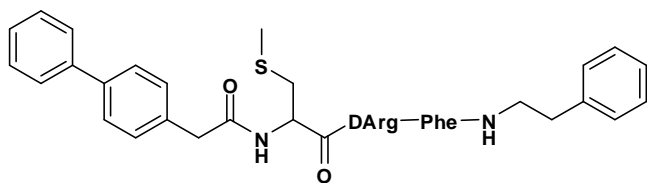
The importance of cathepsin L in physiological processes and pathological conditions has lead to a significant number of research projects that focused on potent, yet selective cathepsin L inhibitors. Many of these inhibitors have small peptides as a key moiety of their structures. Furthermore, many non-peptidic inhibitors, which are usually smaller, have better bioavailability. The large list of peptidic inhibitors includes

derivatives of: ene diones, epoxysuccinyl, nitrile, aldehydes, cyanopyrrolidines, cyclic heptapeptides, glyoxal and the cathepsin V propeptide. The list of nonpeptidic inhibitors of cathepsin L includes: Hydroperoxides, succinyl epoxides, diazomethylketones, thiocarbazates, chalcones, acyl hydrazones, thiosemicarbazones and unsaturated amides.^{106,242} A sample of peptidic and non-peptidic inhibitors reported within the last ten years can be seen in Figure 15. The list of inhibitors includes oxocarbazate, nitriles, thiocarbazates, aldehydes and azepanones.

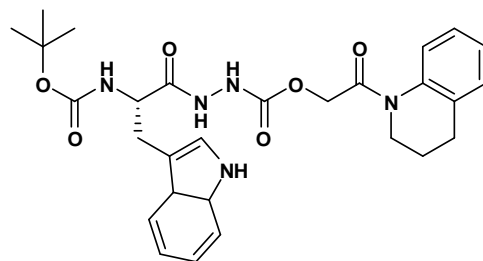
Crystal Structures of Reversible and Irreversible Inhibitors of Cathepsin L

Shenoy and Sivaraman recently reported the crystal structures of two inhibitors of cathepsin L. They obtained the crystal structures of cathepsin L complexed with a keto-aldehyde based and a diazomethylketone derivative inhibitors that are commercially available.²³⁰

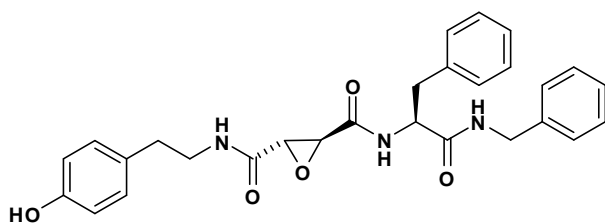
The aldehyde derivative (*Z*-Phe-Tyr-(*O*-But)-COCHO), α -keto- β -aldehyde) has shown to be a slow, tight-binding inhibitor of cathepsin L with a calculated K_I value of 0.6 nM.²⁵⁰ They found the reversible inhibitor binds to the active residue Cys25 via a transient covalent bond. Overall, the aldehyde forms a tetrahedral thiosemiacetal complex with cathepsin L. One of the key features of its inhibitory potency is due to the α -ketone oxygen which is in front of cathepsin L's oxyanion hole. Cys25 is covalently modified by the aldehyde group of the analog forming a thiosemiacetal moiety. Lastly, the S1 pocket is occupied by the Tyr-(*O*-But) moiety. The rest of the inhibitor (phenyl, and carboxybenzoxy groups) occupies the S2 and S3 pockets of the cysteine protease.



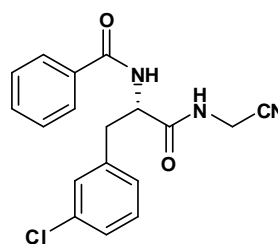
Biphenylacetyl Peptide (K_I : 19 nM)²⁴³



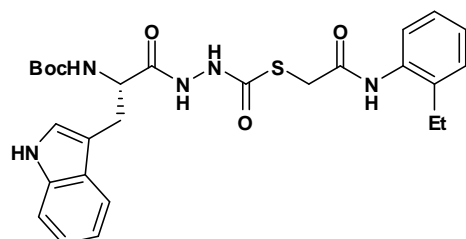
Oxocarbazate (IC_{50} : 6.9 nM)²⁴⁴



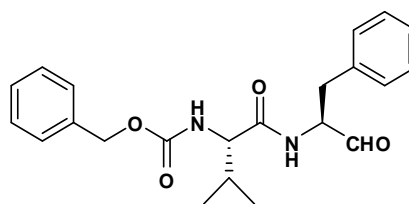
Epoxysuccinyl Peptide (CAA0225)
(IC_{50} : 6.9 nM)²⁴⁵



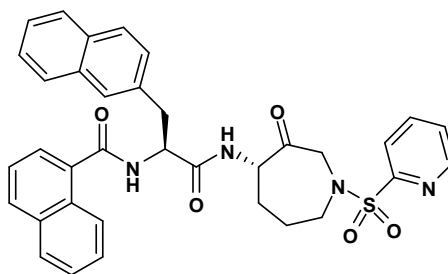
Dipeptidyl Nitrile (pIC_{50} : 6.6)²⁴⁶



Thiocarbazate (IC_{50} : 56 nM)²⁴⁷



Dipeptide Aldehyde (MDL21870)
(IC_{50} : 2.5 nM)²⁴⁸



Azepanone (K_I : 0.43 nM)²⁴⁹

Figure 15. Selected Cathepsin L Non-Thiosemicarbazone Inhibitors Reported Between 2003 and 2012

The crystal structure also showed there were six hydrogen bonds between the inhibitor and the active site of the enzyme. Two of these hydrogen bonds involve His163 and Cys25, two of the active site residues.

The diazomethylketone Z-FY(*t*-Bu)-DMK is an irreversible inhibitor of cysteine proteases in *in vitro* and *in vivo* studies. The investigators found the irreversible inhibitor binds to the active residue Cys25 via covalent bond. The diazomethane methylene moiety makes a covalent bond with the thiol group of Cys25. The S1, S2 and S3 pockets are occupied by the Tyr(*t*-Bu), phenylalanine phenyl group and the carboxybenzyl groups of the inhibitor. The crystal structure also showed there were five hydrogen bonds between the inhibitor and the active site of the enzyme. Two of these hydrogen bonds are made between the inhibitor and the Gln19 side chain and Cys25 backbone. Additionally, the carbonyl oxygen of the synthetic compound was found to be facing the oxyanion hole of the enzyme.

Role of Cathepsin L in Cancer

The overexpression and abnormal activity of cathepsin L is also been evidenced in pathological conditions, as previously described in chapter one. Furthermore, proteolytic enzymes have become one of the field targets in the search for new and potent anticancer agents due to their importance in metastatic processes like cell invasion and migration.²⁵¹ Extensive research has demonstrated the participation of cathepsin L in malignant tumors due to its role in apoptosis, angiogenesis, cell invasion and cell migration.²⁵² Invasion and migration will be discussed separately.

Apoptosis

One of the particular characteristics of cancer cells is their ability to extend their life time by altered programmed cell death processes (apoptosis). In normal cells, apoptosis is an active, but highly regulated process. One of the proposed mechanisms of apoptosis reduction relates to cathepsin L and its ability to cleave Bid, a protein with apoptotic characteristics. Cleavage of Bid will liberate cytochrome C from the mitochondria which also induces apoptosis.²⁵³ Interestingly, animal and *in vitro* studies using cancer cell lines give somewhat conflicting results with respect to the role of cathepsin L in apoptosis.

Cell culture with human transfected cells shows that cathepsin L has apoptotic functions.²⁵⁴ Animal studies with cathepsin L knockout mice, indicate the lack of cathepsin L reduces cell proliferation and tumor growth, yet contributes to the apoptotic processes.²⁵⁵

Angiogenesis

Angiogenesis, or the process of formation of new blood vessels is vital for tumor growth.²⁵⁶ Endostatin, a degradation product of type XVIII collagen, has angiogenic properties. Cathepsin L has excellent collagenase activity as discussed previously. Research has shown that endostatin can be produced by degrading type XVIII collagen.²⁵⁷ However, research has not provided specific evidence that supports the regulation of endostatin by cathepsin L.

Research has also investigated the role of cathepsin L in angiogenesis in animal models; however, their results are quite contradictory. Angiogenesis was reduced in mice treated with non specific cysteine protease inhibitors.²⁵⁸ In contrast, animal studies

revealed that the angiogenesis process was not stopped in cathepsin L knockout mice.²⁵⁵

Additionally, Urbich and coworkers reported overexpression of cathepsin L promoted the invasive characteristics of mature endothelial stem cells and neovascularization in mice was also found.²⁵⁹

Cathepsin L and Cancer in the Male Population

Research has found cathepsin L is overexpressed in six of the most common types of cancers found in the United States. The list includes: prostate, bladder, colorectal, lung, melanoma and kidney cancers which constitute 70% of the new cases in 2012.^{203,260–264} Specifically, cathepsin L is highly expressed in male urine, prostate cancer cell culture and male cancer tissues (colorectal, lung, melanoma, and kidney). Interestingly, the protease is downregulated in prostate cancer tissue.¹⁰⁵

Cathepsin L and Cancer in the Female Population

Similarly, upregulation of cathepsin L has been found in seven of the most common types of cancer. The list includes: breast, ovarian, colorectal, lung, melanoma, kidney and uterus cancers which constitute an alarming 88% percent of the new cases in 2012. Overexpression of cathepsin L is found in female cancer tissues (breast, colorectal, lung, and melanoma); female serum (ovarian and uterus) and cell culture (breast). Downregulation was also found in female cancer tissue (kidney and uterus).^{265–267}

Cathepsin L and Other Types of Cancer

The importance of cathepsin L in other cancers has also been reviewed and investigated. Nasopharyngeal, gastric, brain, head, neck cancer tissues also expressed elevated concentrations of this protease.^{105,268–270}

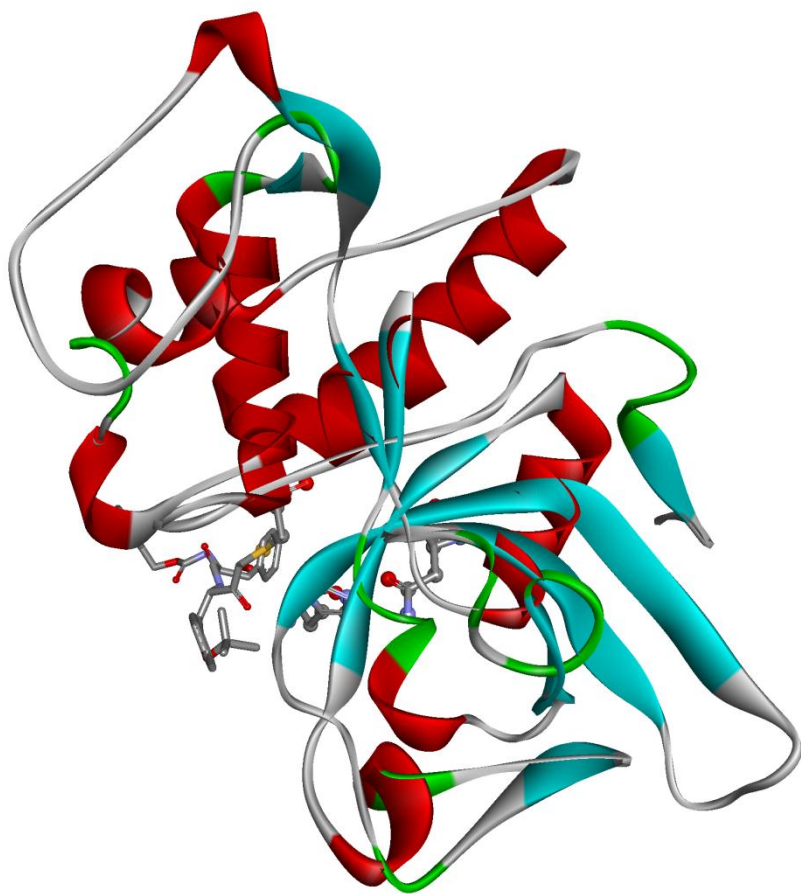
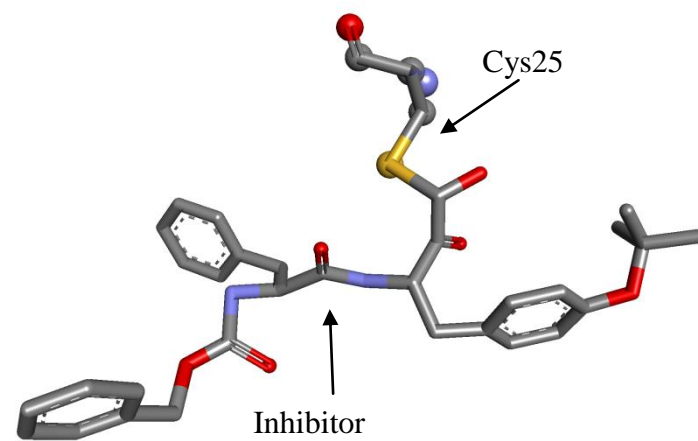
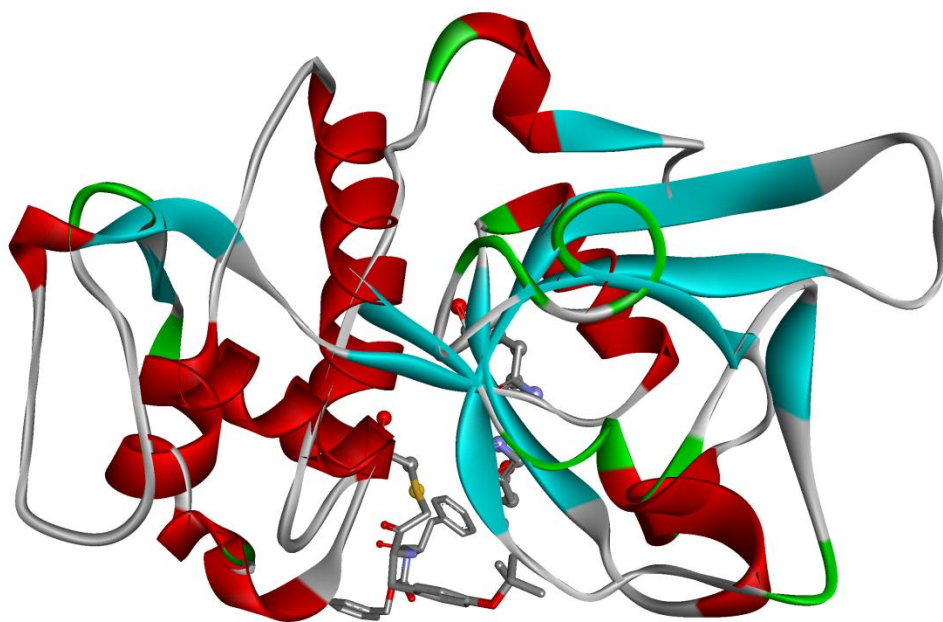
A**B**

Figure 16. Crystal Structure of Human Cathepsin L with Z-Phe-Tyr(OBut)-COCHO, a Reversible, Slow, Tight Binding Inhibitor. **A.** Overall Structure. **B.** Covalent Modification of Cys25 by the Inhibitor (PDB ID: 3OF8)

A



B

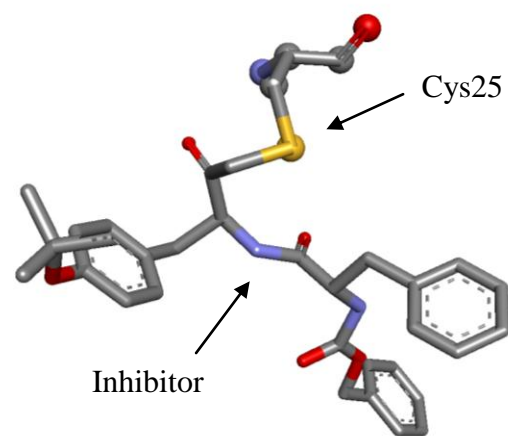


Figure 17. Crystal Structure of Human Cathepsin L with Z-F-Y(t-Bu)-DMK, an Irreversible Inhibitor. A. Overall Structure. B. Covalent Modification of Cys25 by the Inhibitor (PDB ID: 3OF9)

Cathepsin L and Breast Cancer

More than 225,000 new cases of breast cancer in women are predicted to be detected in 2012. This deadly disease is the second cause of death among female cancer patients. Almost 40,000 women are expected to die in the same year. Thus, it is important to find new chemotherapeutic targets that can arrest or delay this cancer, yet with minimal side effects. As previously stated, cathepsin L is upregulated in breast cancer tissue and cell lines.

Investigators have linked the invasiveness of breast cancer with cathepsin L activity.²⁷¹⁻²⁷³ Therefore, *in vitro* assays for cell invasion and cell migration are performed usually with Matrigel™. This synthetic matrix has also used for other cell culture studies including angiogenesis, 3D cell culture, neuroscientific studies, metabolic and toxicological studies.²⁷⁴⁻²⁷⁶ The composition of this proprietary matrix includes laminectin (56%), type IV collagen (31%) and entactin.²⁷⁷ These three proteins are also important components of the extracellular matrix in physiological conditions.²⁷⁸⁻²⁸⁰ Matrigel™ also contains several growth factors such as bFGF, EGF, IGF-1, PDGF, NGF, and TGF-β. Two- and three-dimensional arrays have been developed to investigate the invasiveness and motility of cancer cells in *in vitro* studies.^{281,282}

Two-Dimensional Experiments

The main objective of two-dimensional studies is to mimic three-dimensional environments that tumor cells encounter when they are attempting to invade surrounding tissues. Perhaps, this approach is one of the most used in cancer research when testing novel compounds that delay cell invasion.

Typically, *in vitro* invasion assays are carried out using Boyden chambers. The chambers consist of two parts: a polycarbonate membrane. Special features of the layer are its porosity and average diameters of its pores, which range between 1 and 8 μm . The second part of the chamber is the thick layer of a basement membrane component (i.e. collagen) or a combination of them (i.e. MatrigelTM). This layer is also present as a porous material. A similar format can be found for migration assays. The main difference from migration assays is Boyden chambers do not contain a basement membrane (i.e. MatrigelTM).²⁸³ Cells are placed in the inner chamber (a top chamber incubated for an extended period of time, usually 24 hours). Cells with invasive characteristics tend to invade through they layer of basement membrane. Cells that pass the membrane(s) can be determined by colorimetric or fluorometric methods. A chemoattractant is also added in the bottom chamber in order to attract cells. A graphical representation of the modified Boyden chamber for invasion and migration assays can be seen in Figure 18.²⁸⁴

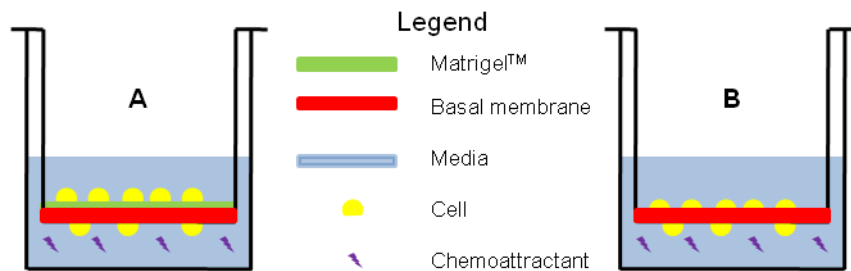


Figure 18. Graphical Representation of Boyden Chambers for Cell (A) Invasion; and (B) Migration Assays

There are also attempts to correlate cathepsin L mRNA and protein expression as a potential biomarker for the detection of breast cancer. This idea seems to be

impractical due to discrepancies in similar studies that have or have not found a direct relationship between cathepsin L and breast cancer samples.^{285,286}

Cell Invasion and Migration. Role of Proteases in ECM Degradation

As previously stated, the metastatic cascade can be summarized in nine general steps. The mechanism is not simple due to the numerous biochemical and physiological processes that need to be carried out in order for one cancer cell to metastasize from the primary tumor. The cell then must migrate, invade and be deposited to its final destination (metastatic place)

Numerous proteases (MMPs, cysteine and serine proteases) are required for cell invasiveness. Proteolytic reactions degrade surrounding extracellular matrix and permit cancer cells to migrate from the primary site and invade surrounding organs or tissues.

The degradation of the ECM occurs via two mechanisms:

1. Proteolytic degradation caused by individual cell-secreted proteases from cells.
2. Surface-related proteolytic degradation caused by membrane-bound proteases.²⁸³

Examples of diffuse degradation proteases are MMPs and some cathepsins. MMP-2 and MMP-14 are examples of surface-related proteases. Finally, MMPs and cathepsins are also found in this category.

Cell Invasion and Migration. Cell Signaling between MMPs and Cathepsins

Cancer cells are not the only cells that contribute to their invasive/motility characteristics.²⁸⁷ Stromal cells, and overall, the tumor microenvironment, which enhances protease activity are included. Table 12 summarizes the proteolytic contribution of numerous tissue connective cells as part of the tumor microenvironment. The regulatory activity of all of them (excluding endothelial cells) include mutual

activation between proteases found in both types of cells (stromal and tumor); also, inhibition of proteases found in tumor cells by more than one stromal protease.

Macrophages, neutrophils, fibroblasts, and endothelial cells express more than four cathepsins. For example, macrophages produce six out of the seven cathepsins expressed in this type of stromal cell. Fibroblasts express six different metalloproteases. Table 13 also describes the distribution of other important proteases such as urokinase-type plasminogen activators (uPAs) and plasmin in cancer stromal cells. The distribution of this enzyme, along with other proteins is more selective. For example, pericytes only express MMP9 and monocytes express only cathepsin B and MMP9.

Table 12. Enzyme Distribution of MMPs and Cathepsins within Cancer Stromal Cells. (Adapted from Mason, page 230²⁸⁷)

Protease Cell	Metalloprotease (MMP-)						Cathepsin						
	2	3	9	11	13	14	B	C	D	G	H	L	S
Monocyte	-	-	+	-	-	-	+	-	-	-	-	-	-
Pericytes	-	-	+	-	-	-	-	-	-	-	-	-	-
Endothelial	+	+	-	-	-	+	+	-	+	-	-	+	+
Fibroblast	+	+	+	+	+	+	+	+	+	-	-	+	-
Lymphocytes	-	+	+	-	-	-	-	+	+	-	+	-	-
Neutrophils	-	-	+	-	-	-	+	+	+	+	+	-	-
Macrophages	+	-	+	-	-	+	+	+	+	-	+	+	+
Mast	+	-	+	-	-	-	-	+	+	+	-	-	-
Mesenchymal stem	+	-	-	-	-	-	+	-	+	-	-	-	-

Table 13. Enzyme Distribution of Other Proteases within Cancer Stromal Cells. (Adapted from Mason, page 230²⁸⁷)

Cell	Protease					
	Plasmin	tPA	uPA	Elastase	Granzyme B	MMP4
Endothelial	-	-	+	-	-	-
Fibroblast	-	-	+	-	-	-
Lymphocytes	-	-	-	-	+	-
Neutrophils	-	-	-	+	-	-
Macrophages	+	+	+	-	-	-
Mast	-	-	-	+	+	+

The proteolytic network is quite complex, and extensive research has done a great deal to explain possible pathways that link these different types of proteases. For example, two cathepsins (B and C) are responsible for the activation of other proteases. Cathepsin C regulates the activity of Granzyme B, Cathepsin G, and Elastase.²⁸⁸ On the other hand, cathepsin B is one of the major key regulators in proteolytic cell signaling. It regulates two metalloproteases: MMP3, and MMP2.²⁸⁹ Cathepsin B also regulates (and is regulated by) cathepsin D and uPA.²⁹⁰ Finally, tPA regulates this cysteine protease.²⁹¹

Additionally, cathepsin L is regulated by cathepsin D (a cathepsin B regulated protease) and regulates two more cathepsins (C and X).^{292,293} As seen in these examples, the extensive network of proteases and its members is closely linked. Proteases do not work independently; however they play proteolytic and regulatory functions among other important features. Figure 19 depicts the interaction between the members of the cathepsin family. Cathepsin L regulates two cysteine proteases (cathepsins C and X) and is regulated by cathepsin D, an aspartic protease (legend: blue: cysteine protease; red: aspartic protease; green: serine protease; adapted from Mason, page 231).²⁸⁷

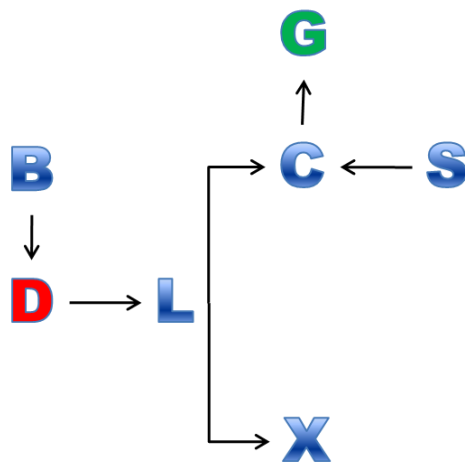


Figure 19. The Cathepsin Network

Cell Signaling of Cell Invasion and Migration

It has been demonstrated that both cysteine proteases and metalloproteases interact in a symbiotic fashion. However, these regulatory activities are also the result of numerous activities divided in signal transduction pathways that are specialized or related for specific functions. Cell migration and cell invasion are regulated by three important, signal transduction pathways: FAK, Integrin, and MAPK pathways.²⁹⁴⁻²⁹⁶

The regulation and activation of focal adhesion kinase (FAK) has taken a pivotal role in research due to its role in cell migration and cell invasion. Overexpression of FAK has been linked to highly invasive cancers such as squamous cell carcinomas and lung adenocarcinomas. Additionally, accumulation of FAK has been found in invadopodia, cell regions rich in integrins and MMPs.²⁹⁷

Integrins, are receptors that mediate interactions between cells and their microenvironment. In the case of cancer cells, integrins promote tumor metastasis via extracellular matrix interaction. In general, integrins partially regulate MMPs and uPA protease activities. Integrins and FAK have shown to promote cell motility in cancer cells. However, there are cases when integrins are the controllers of this characteristic.²⁹⁸

The mitogen-activated protein kinase family comprises multiple proteins that have direct effect on cell migration and invasion. Examples include p38, Erk, and JNK. The Jun N-terminus kinase promotes cell migration via phosphorylation of microtubule-associated proteins. Another member of the family, p38 regulates protein kinase 2/3, which is directly implicated in cell migration. In summary, MAPKs have proven to regulate cell migration by more than one signal transduction pathways.²⁹⁹

The basic migration mechanisms are actin and myosin II dependent. First, cell migration occurs via actin filament assembly.²⁹⁹ Alternatively, myosin II performs ‘contractions’ in the actomyosin networks, so cells can migrate through healthy tissue. These two mechanisms are the result of the downstream product of activated (or unactivated) Rac, Rho, and Cdc42.

Perhaps, a major force of cell migration and cell invasion is the integrins.³⁰⁰ These heterodimeric proteins make connections between tumor cells and ECM, so the former one can adhere to the matrix. The connections are via adaptor proteins, such as talin, vinculin, and focal adhesion kinase (FAK). Another function of the integrins is to send intercellular and extracellular signals as well. This is the moment when MMPs and cathepsins perform their hydrolytic interaction: the invasion of ECM and the implicit migration throughout the matrix. On the other hand, the cadherins are the responsibility of cancer cell invasion behavior. The list includes E-cadherin, N-cadherin, and other proteins such as connexins and immunoglobulins. The complexity of this process can be seen in the expression of the cadherins. They both play selfregulatory activities. For instance, E-cadherin is the anti-invasion and N-cadherin gets overexpressed when the invasion process occurs.

Another class of cell migration promoter proteins is made of chemokines. The importance of these G-protein-coupled receptors lies in their ability to promote tumor invasion by using tumor macrophages and lymphocytes, important stromal cells rich in other metalloproteases and cathepsins (See Table 12).

Kinases can also promote cell invasion and migration. On one hand, migration is promoted by PI3K and Erk, two kinases that activate RTK, a third kinase. Invasion is

also promoted by the complex formed with FAK and Src. The role of FAK in cell migration and invasion is vital for tumor cells. First, the FAK-p190RohGEF complex promotes the activation of Rho, formation of actin fibers in the cytoplasm and stabilizes new adhesion points.

Bone Metastasis

As previously discussed in chapter one, bone is the preferred metastatic site for breast and prostate cancers.³⁰¹ Bone metastasis can be divided into two forms. Osteoclastic metastasis occurs in male patients with prostate cancer, while osteoblastic metastasis is mainly found in breast cancer patients.^{302,303} Bone metastasis is the result of what is known as the vicious cycle.³⁰⁴ Tumor cells express RANKL, a ligand that binds to RANK and osteoprotegerin, a specific RANKL receptor. The ratio RANKL/OPG determines the osteoclast cycle which begins with bone resorption and growth factors. Growth factors act as tumor promoters.³⁰⁵ Currently, there are compounds in early or late clinical trials. For example, imatinib is in preclinical trials. It has been proven to reduce osteolytic metastasis. Bisphosphonates have been also investigated. Other compounds include atrasentan, and denosumab in preclinical and clinical studies involving prostate cancer.³⁰⁶⁻³⁰⁹

Cathepsin L in Bone Metastasis, Cell Invasion and Cell Migration

The number of research studies relating cathepsin L with bone metastasis is increasing and supports the theory of cathepsin L as one of the major macromolecules contributing to bone metastasis. Therefore, it is not surprising cathepsin L levels in breast cancer tissues and cell lines are higher than in normal tissues.³¹⁰ This

overexpression is the downstream result of some proinflammatory cytokines in the bone resorption processes (IL-1 α , Il-6, TNF- α).^{311,312} Specific cathepsin L inhibitors designed by Katunuma and coworkers (CLIK-148) were able to inhibit bone resorption and hypercalcemia in mice.³¹³ Also, *in vitro* studies osteosarcoma cells treated with cathepsin L inhibitors showed reduced levels of cathepsin L mRNA.³¹⁴ Additionally, cathepsin L is a powerful hydrolase with collagenase; elastase among other extracellular matrix components can be degraded as previously described. In depth studies have also concluded that cathepsin L participates indirectly in the bone remodeling process by activating key players in this process (MMPs, uPA and heparanase).³¹⁵ Cathepsin L might also be the responsible for the activity of E-cadherin, a cell-adhesion molecule.²⁵⁵

In general, other studies using HUVEC, U87MG (glioblastoma), and neuroblastoma cell lines also confirm the importance of cathepsin L in bone metastasis.³¹⁶⁻³¹⁸ Cathepsin L has also shown to promote cell invasion and cell migration. Reduced cell adhesion and an increment of ECM degradation are consequences of overexpression of cathepsin L in cancer cells.²⁹⁰ One of proposed mechanisms of the participation of cathepsin L in cell invasion and cell migration is the ability of the protease in cleaving E-cadherin. Lower levels of this cell-adhesion protein is downregulated in highly invasive cells.³¹⁹ The pathways for cell signaling and cathepsin L activity in cell invasion and cell migration are largely unexplored. The possibility of inhibition of cell invasion and/or cell migration via inhibition of cathepsin L has become a reference point for ongoing investigations using specific synthetic inhibitors designed for cathepsin L.

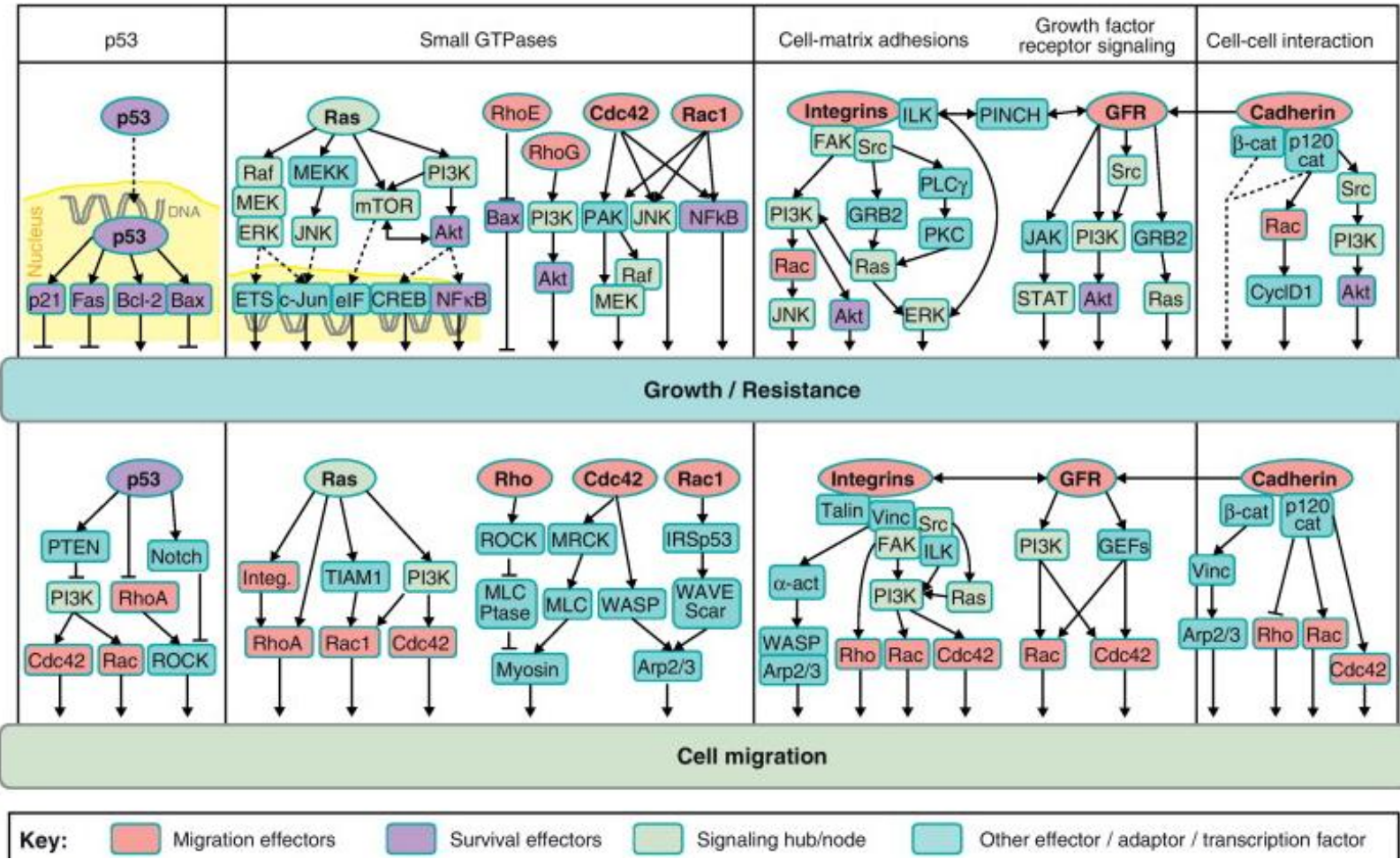


Figure 20. Signaling Pathways Controlling Tumor Cell Growth, Survival and Invasion. (reproduced from Alexander, page 17)²⁹⁹

Material and Methods for the Biological Evaluation of Thiosemicarbazones Derivatives as Inhibitors of Cathepsin L

Materials and Equipment

Ultra pure water was generated using a Barnstead Nanopure Diamond water purification system. This reagent will be called water for future references. Sodium acetate (anhydrous), ethylenediaminetetraacetic acid (EDTA), and dithiothreitol (DTT) were obtained from EMD. Recombinant, cathepsin L from human liver, 30 % Brij 35, and sodium dodecyl sulfate (SDS) were obtained from Sigma. Pure (99.9 %) Dimethyl sulfoxide (DMSO) and pure 7-Amino-4-methylcoumarin (AMC) were products from Acros and Anaspec. Z-Phe-Arg-AMC (7-Amino-4-methylcoumarin, N-CBZ-L-phenylalanyl-L-arginine amide, hydrochloride, Z-FR-AMC) was purchased from Bachem. Human type I collagen and gentamicin were acquired from Calbiochem. Simply Blue™ and electrophoresis reagents, including 10% Bis-Tris SDS-PAGE gels, antioxidant reagent, reducing agent, protein standard marker, SYPRO®, and loading buffer 4X were all products from Invitrogen. 3-(N-morpholino)propanesulfonic acid (MOPS) and Tris Base were purchased from BDH and Fisher Scientific. Matrigel™ invasion and migration assay kit were purchased from BD Biosciences. MDA-MB-231 cells were obtained from ATCC. DMEM media, FBS, and DIFF QUIK were products from, Cellgro, Gibco, and Imeb, respectively. Thiosemicarbazone derivatives were synthesized by members of Dr. Kevin G. Pinney' laboratory. Black 96 well flat bottom microplates (3991) and 0.2 µm filters were obtained from Corning. Ethanol was a product of Pharco. Single and twelve-channel pipeters as well pipettes tips were Eppendorf products. Range of pipettes used varied between 0.5 and 5000 µl. Finally, the

pH meter (Mettler Toledo), Fluoroskan Ascent FL (Thermo), XCell SureLock® Minutesi (Invitrogen), GE Typhoon 9400 FL, and Zeiss Axiovert 40 CFL inverted microscope were the instruments to calibrate the pH of the prepared solutions, fluorometric experiments, the separation of proteins, fluorescent imaging, and manual cell counting.

Preparation of Buffers and Stock Solutions

Preparation of 1 M sodium acetate buffer, pH 5.5. This buffer was prepared as a stock solution for further dilutions. It will be referred to as a 1 M buffer stock solution. One liter of this stock was prepared by weighing 69.65 g (0.85 mol) of sodium acetate anhydrous (NaOAc, MW: 82.03 g/mol) and 8.64 g (0.15 mol) of acetic acid glacial (MW: 60.05 g/mol; 1.049 g/ml) in a clean volumetric flask. The contents were dissolved with pure water. Glacial acetic acid or sodium hydroxide 5 M were used to adjust the pH of the buffer. The solution was stored at 4 °C .

Preparation of 5M NaOH. Fifty milliliters were prepared by weighing 10 g (0.25 mol) of NaOH pellets (NaOH, MW: 40 g/mol) in a clean volumetric flask. The contents were dissolved with pure water.

Preparation of 400 mM sodium acetate buffer, pH 5.5. One liter of this solution was prepared by diluting 400 ml of 1 M buffer stock solution to a final volume of one liter. Glacial acetic acid or sodium hydroxide 5 M (MW: 40 g/mol) were used to adjust the pH of the buffer. This solution was stored at 4 °C.

Preparation of 40 mM EDTA. A stock concentration of this solution was prepared by dissolving 14.89 mg (40 µmol) of EDTA (MW: 372.29 g/mol) in one

milliliter of water. Fifty milliliters of this solution (744.5 mg; 2 mmol) was prepared for multiple experiments. The solution of 40 mM EDTA was stored at 4 °C.

Preparation of 80 mM DTT. A freshly made stock concentration of this solution was prepared by dissolving 14 mg (80 µmol) of DTT (MW: 175 g/mol) in one milliliter of sodium acetate 400 mM, pH 5.5. The 80 mM DTT solutions were not stored for future experiments.

Preparation of 4 mM AMC. This stock solution was prepared by dissolving 0.91 mg (4 µmol) of AMC (MW: 229.2 g/mol) in one milliliter of pure DMSO. This solution was stable for several months when stored at low temperatures (-20 °C).

Preparation of 29.94 mM Z-FR-AMC. A sample of 19.43 mg (29.93 µmol) of Z-FR-AMC (MW: 649.15 g/mol) were dissolved in one milliliter of pure DMSO. This solution was stable for several months when stored at low temperatures (-20 °C).

Recombinant cathepsin L from human liver. Recombinant, human cathepsin L was purchased from Sigma. Concentration of the protease varied based on lot number. A typical cathepsin L concentration is 0.285 µg/µl (9.83 µM; MW: 29 kDa).

Preparation of human type I collagens solution from human skin. Type I collagen (0.4 mg) was dissolved in 1 ml 50 mM acetic acid (pH 2.5). Solution was stirred for 24 hours at room temperature.

Thiosemicarbazone derivatives stocks solutions. More than 150 compounds synthesized in the laboratory of Dr. Kevin G. Pinney (Baylor University) were used to

make stock solutions (20 mM) in pure DMSO.^{3,6,7,10-12} Usually, one milligram was weighed to give a final volume that varied between 100 and 200 μl of these stock solutions.

Preparation of inhibitor dilutions. Each inhibitor was serially diluted in pure DMSO to provide seven different concentrations (named A-G) that were ten-fold diluted. The concentrations of these dilutions varied between 2 mM (solution A) and 2 nM (solution G). Each dilution was diluted ten-fold in a solution that contained 35% DMSO in water (named 1-8). Concentrations of the aqueous solutions varied between 200 μi and 200 pM. An eighth solution was made by using solution C (20 μM) to prepare an intermediate solution (1 μM in 35% DMSO). Solutions made in pure DMSO were stored for two weeks at (-80 °C). Aqueous solutions were made prior to the experiment and kept at (4 °C). Figure 21. Serial Dilution Flowchart represents a graphical procedure of the preparation of the stock and inhibitor dilutions.

Preparation of cathepsin L assay buffer. Cathepsin L assay buffer (will be referred as assay buffer for future reference) contained 1.8 mM EDTA, 5.4 mM DTT, 0.02% Brij 35 and 120 mM NaOAc pH 5.5. Ten milliliters of assay buffer were made by adding 450 μl of 40 mM EDTA, 675 μl of 80 mM DTT, 6 μl of Brij 35, 3 ml of 400 mM NaOAc pH 5.5, and 5.87 ml of water in a plastic 50 ml conical tube. Table 14 summarizes a typical calculation sheet to make different volumes of assay buffer. The solution was stable and used for a maximum of 24 h.

Preparation of 12% DMSO solution. An aqueous solution of DMSO was prepared by dilution of 120 μl of DMSO with 8800 μl of water to bring a total of 1 ml.

Preparation of 35% DMSO solution. An aqueous solution of DMSO was prepared by dilution of 350 μl of DMSO with 650 μl of water to bring a total of 1 ml.

Preparation of 50% DMSO solution. An aqueous solution of DMSO was prepared by dilution of 500 μl of DMSO with 500 μl of water to bring a total of 1 ml.

Preparation of cathepsin L stock solution. A stock solution of cathepsin L (CL) contained 1 mM EDTA, 3 mM DTT, 0.01% Brij 35, 10 nM cathepsin L and 400 mM NaOAc pH 5.5. One milliliter of cathepsin L stock solution was made by adding 25 μl of 40 mM EDTA, 37.5 μl of 80 mM DTT, 0.33 μl of Brij 35, 1.02 μl of 0.285 $\mu\text{g}/\mu\text{l}$ cathepsin L, and 936 μl of 400 mM NaOAc, pH 5.5, in disposable glass test tubes. Table 15 summarizes a typical calculation sheet to make different volumes of cathepsin L stock solution. The solution was made prior to every kinetic or inhibition experiment.

Table 14. Preparation Table for Cathepsin L Assay Buffer

Assay Buffer (ml)	40 mM EDTA (μl)	80 mM DTT (μl)	Brij 35 (μl)	400 mM NaOAc (μl)	Water (μl)
5	225	338	3	1500	2935
10	450	675	6	3000	5870
20	900	1350	12	6000	11740
40	1800	2700	24	12000	23477

Table 15. Preparation Table for Cathepsin L Stock Solution

CL (ml)	40 mM EDTA (μl)	80 mM DTT (μl)	Brij 35 (μl)	400 mM NaOAc (μl)	Cathepsin L (μl)
1	25	37.5	0.33	936	1.02
2	50	75	0.67	1872	2.04
3	75	112.5	1	2808	3.06

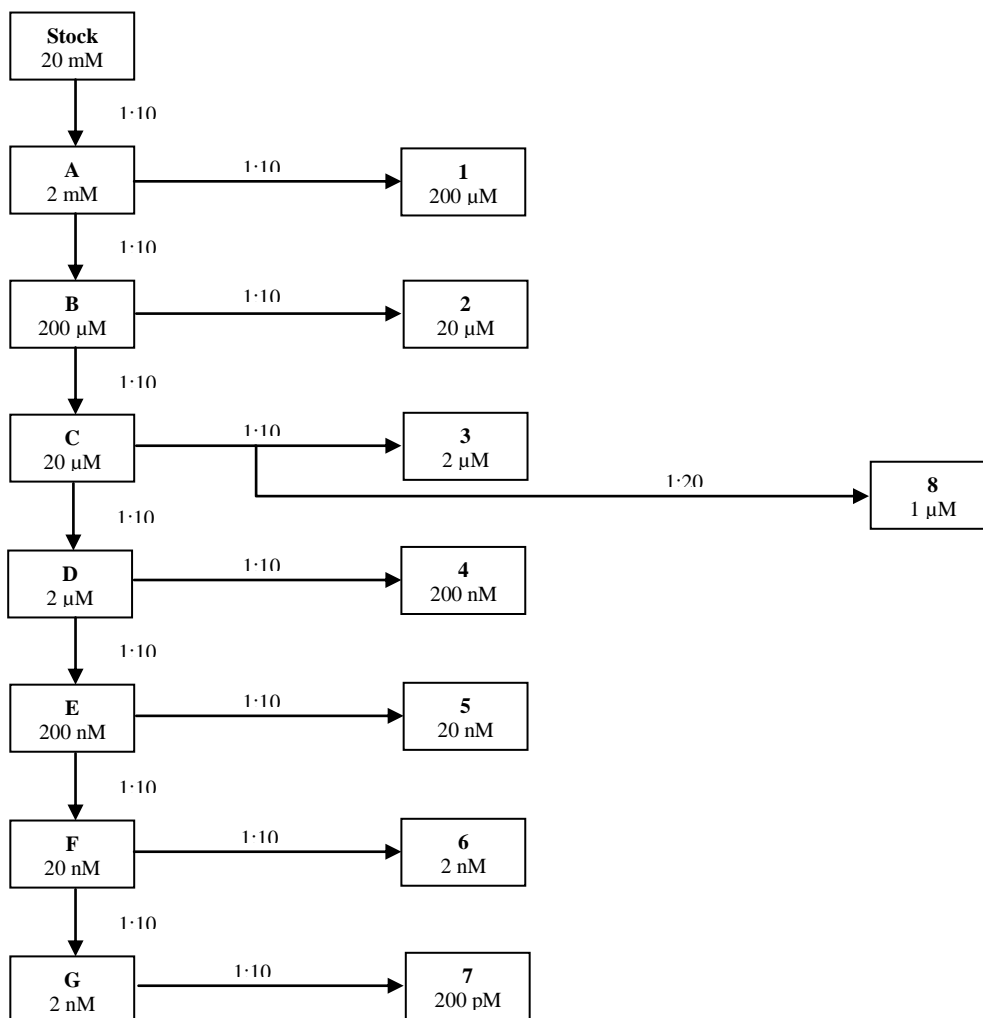


Figure 21. Serial Dilution Flowchart

Preparation of Z-FR-AMC stock solution. A stock solution of Z-FR-AMC (500 μM) in 2.5% DMSO was prepared by adding 20 μl of 29.94 mM Z-FR-AMC, 10 μl of DMSO and 1170 μl of water in order to have a final volume of 1.2 ml. The solution was made in a dark plastic 15 ml conical tube. Table 16 shows representative calculations to prepare various amounts of the substrate solution. The stock solution was stable and used for a maximum of 24 hours.

Table 16. Preparation Table for 500 μ M Z-FR-AMC for Cathepsin L Assays

500 μ M Z-FR-AMC (ml)	29.94 mM Z-FR-AMC (μ l)	DMSO (μ l)	Water (μ l)
1	16.7	8.3	975
2.5	41.8	20.8	2440
5	83.5	41.5	4875
10	167.0	83.0	9750

Assay buffer for reversibility studies (ABR). Three milliliters of ABR were prepared by mixing 1200 μ l of 250 mM NaOAc pH 5.5, 1.01 μ l of Brij 30, 112.5 μ l of 80 mM DTT, 75 μ l of 40 mM EDTA, 55 μ l of DMSO, 5 μ l of 29.94 mM Z-FR-AMC and 1552 μ l of water. ASR was prepared prior to the experiment in a 15 ml plastic conical tube kept in the dark. Final conditions of ASR are: 100 mM NaOAc pH 5.5, 1 mM EDTA, 80 mM EDTA, 0.01% Brij 30, 50.5 μ M Z-FR-AMC, and 2% DMSO. Table 17 summarizes these conditions.

Table 17. Preparation Table for Cathepsin L Assay Buffer for Reversibility Studies

Volume (ml)	40 mM EDTA (μ l)	80 mM DTT (μ l)	Brij 35 (μ l)	250 mM NaOAc (μ l)	29.94 mM Z-FR- AMC (μ l)	DMSO (μ l)	Water (μ l)
3	25	37.5	0.33	936	1.02	55	1200

Preparation of cathepsin L stock solution for reversibility studies (CLR). One hundred forty-four microlites of CLR were prepared by mixing 115.2 μ l of 250 mM NaOAc pH 5.5, 0.10 μ l of Brij 30, 10.8 μ l of 80 mM DTT, 7.2 μ l of 40 mM EDTA, 2.93 μ l of 0.285 μ g/ μ l cathepsin L, and 7.8 μ l of water in a glass test tube. Final conditions of CLR are: 200 mM NaOAc pH 5.5, 2 mM EDTA, 6 mM EDTA, 0.02% Brij 30, and 100 nM cathepsin L. Table 18 summarizes the preparation tables for CLR.

Table 18. Preparation Table for Cathepsin L Stock Solution for Reversibility Studies

	Volume (ml)	40 mM EDTA (μ l)	80 mM DTT (μ l)	Brij 35 (μ l)	250 mM NaOAc (μ l)	Cathepsin L (μ l)	Water (μ l)
CLR	0.14	7.2	10.8	0.1	115.2	2.93	7.8

Preparation of cathepsin L stock solution for collagenase studies (CLC).

Seventy-two microliters of CLC were prepared by mixing 31.2 μ l of 1 M NaOAc pH 5.5, 0.10 μ l of Brij 30, 11.7 μ l of 80 mM DTT, 7.8 μ l of 40 mM EDTA, 8.7 μ l of 0.285 μ g/ μ l cathepsin L, and 12.5 μ l of water in a glass test tube. Conditions of CLC are: 433.3 mM NaOAc pH 5.5, 4.3 mM EDTA, 13 mM EDTA, 0.04% Brij 30, and 1192 nM cathepsin L. Table 19 summarizes preparation tables for CLC.

Dulbecco's modified Eagle's medium (DMEM) supplemented media. A sterile bottle of DMEM media is supplemented with 50 ml of 10X FBS and 830 μ l of a 3 mg/ml gentamicin stock solution. Supplemented DMEM media conditions are: 10% FBS and 50 ng/ml gentamicin. The media will be called as supplemented DMEM medium in future references.

Table 19. Preparation Table for Cathepsin L Assay Buffer for Collagenase Inhibition Studies

CLC	Total Volume	40 mM EDTA	80 mM DTT	Brij 35	1 M NaOAc	CL	Water
Volume (μ l)	72	7.8	11.7	0.1	31.2	8.7	12.5

Experimental Section

Kinetic Cathepsin L Assay

Kinetic analysis of cathepsin L was carried out by using a Thermo Fluoroskan microplate reader with 3691 96-well black microplates.⁵ The total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 50 μ l of water, 10 of 35% DMSO solution, 20 μ l cathepsin L stock solution, and 20 μ l of Z-FR-AMC stock solution. A mixture containing assay buffer, 35% DMSO, was preincubated at 25 °C during 5 minutes using 96-well black microplates. The catalytic activity of cathepsin L was monitored by adding a concentration of Z-FR-AMC to every reaction was carried out in triplicates. The production of AMC was monitored for 5 minutes at 25 °C using excitation and emission reference filters of 355 and 460 nm respectively. Readings were taken every 10 seconds for five minutes. Reactions were carried out in triplicate. The final concentrations of the kinetic assay were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM EDTA, 0.01% Brij 35, and 1 nM cathepsin L. Final concentrations of Z-FR-AMC varied between 0.2 and 14 μ M.

Preliminary Inhibition Studies

Thiosemicarbazone analogs (provided by Dr. Kevin G. Pinney's laboratory, Baylor University)^{3,6,7,10-12} were prescreened to determine if they have inhibitory activity against cathepsin L. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 50 μ l of water, 10 μ l of 35% DMSO or 10 μ l of dilution "1" (See Figure 21; final concentration: 10 μ M), 20 μ l cathepsin L stock solutions, and 20 μ l of Z-FR-AMC stock solution. The enzyme-inhibitor mixtures (180 μ l) assay buffer, 35%

DMSO or inhibitor, and cathepsin L was preincubated at 25 °C during 5 minutes using 96-well black microplates. Reactions were started by adding 20 µl of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five minutes. Final reactions were started by adding 20 µl of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five minutes. The final concentrations of the preliminary inhibitory studies were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM EDTA, 0.01% Brij 35, 1 nM cathepsin L, 10 µM of the screened inhibitor and 50 µM of Z-FR-AMC. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Final concentrations of the inhibition cathepsin L assay were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM EDTA, 0.01% Brij 35, 1 nM cathepsin L and 50 µM of Z-FR-AMC. Final concentrations of the inhibitors varied between 10 µM and 10 pM. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Compounds that did not inhibit cathepsin L activity more than 50% at a final concentration of 10 µM (i.e. $v_i/v_o \leq 0.5$) were considered as 'inactive' compounds and a general IC₅₀ value greater than 10000 nM was assumed. Compounds that inhibited cathepsin L activity more than 50% were further considered for cathepsin L inhibitory studies and an exact IC₅₀ value was determined.

Cathepsin L Inhibition Assay

Analysis of cathepsin L and its inhibitors was carried out by using a modified protocol of the kinetic cathepsin L assay, previously described. Total volume of the reaction was 200 µl. Every well contained 100 µl of assay buffer, 50 µl of water, 10 µl of 35% DMSO or 10 µl of inhibitor dilutions, 20 µl cathepsin L stock solutions, and 20 µl of Z-FR-AMC stock solution. A 180 µl mixture containing assay buffer, 35% DMSO

or inhibitor, and cathepsin L was preincubated at 25 °C during 5 minutes using 96-well black microplates. Reactions were started by adding 20 µl of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five minutes. The final concentrations of the cathepsin L inhibition assay were: 100 mM NaOAc, pH 5.5, 1 mM EDTA, 3 mM EDTA, 0.01% Brij 35, 1 nM cathepsin L and 50 µM of Z-FR-AMC. Final concentrations of the inhibitors varied between 10 µM and 10 pM. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate.

Tables Table 20 and Table 21 summarize final volumes and final conditions for the kinetic and cathepsin L inhibition assays.

Construction of AMC Calibration Curve

The cathepsin L kinetic assay was slightly modified to construct the 7-amino-4-methylcoumarin calibration curve. Final volume was 200 µl. The assay consisted of 120 µl assay buffer, 10 µl of water, 10 µl of stock solution and 20 µl of AMC dilutions made in 20% DMSO. Serial dilution was used to prepare twelve AMC samples that were 10 fold final concentrations.

Table 20. Preparation Table for Kinetic, Cathepsin L Inhibition Assays and Construction of AMC Calibration Curves

Reagent	Kinetic assay (µl)	Inhibitory assay (µl)	AMC Curve (µl)
Assay Buffer	100	100	100
Water	50	50	60
Control	10	0	0
Inhibitor	0	10	0
CL	20	20	20
Z-FR-AMC	20	20	20
Total (µl)	200	200	200

Table 21. Final Conditions for Kinetic Cathepsin L and Inhibition Cathepsin L assays

NaOAc (mM)	EDTA (mM)	DTT (mM)	Brij 35 (%)	DMSO (%)	Cathepsin L (nM)	Z-FR- AMC (μ M)
100	1	3	0.01	2	1	0.1- 50

A 4 mM AMC in DMSO was used as a primary stock solution. Table 22 shows calculations for the preparation of these dilutions. Final concentration varied between 15 μ M and 1 nM of AMC in solution. Final conditions were 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM DTT, 0.01 % Brij, and 2 % DMSO. Assay buffer, water, cathepsin L stock solution (with no inhibitor) and AMC dilutions were added to the 96-well black microplates, incubated at 25 °C for 5 minutes. Samples were monitored 25 °C for 800 seconds at excitation and emission wavelengths of 355 nm and 460 nm respectively. Linear regression for data fitting was carried out using GraphPad 5.0, where the independent and dependent variables are time (in seconds) and relative fluorescence units. The Y- intercept (dependent variable) was plotted vs AMC concentration (in μ M).

Table 22. Preparation Table for the Construction of AMC Calibration Curve

#	Final Concentration		Add from	AMC dilutions in 20% DMSO			
	AMC (μ M)	DMSO (%)		Volume (μ l)	DMSO (μ L)	Water (μ l)	DMSO (%)
0	25	2	Stock	63	138	800	20
1	15	2	Stock	38	163	800	20
2	10	2	Stock	25	175	800	20
3	5	2	Stock	13	188	800	20
4	1	2	# 1	67	187	747	20
5	0.75	2	# 1	50	190	760	20
6	0.5	2	# 1	33	193	773	20
7	0.3	2	# 3	60	188	752	20
8	0.1	2	# 3	20	196	784	20
9	0.050	2	# 6	100	180	720	20
10	0.010	2	# 6	20	196	784	20
11	0.005	2	# 6	10	198	792	20
12	0.001	2	# 6	2	200	798	20

FC: Final concentration during assay

Effect of Solvent Concentration on Cathepsin L Inhibition Assays

Preincubation studies with compounds **1**, **136**, and **168** were carried out by a slight modification of the cathepsin L inhibition assay. Different sets of mixtures containing assay buffer, inhibitor and cathepsin L were preincubated at various DMSO concentrations between 0.7 and 3% (final concentration). Reactions were taken every 25 seconds for five minutes and carried out in triplicate.

Effect of Inhibitor Concentration on Cathepsin L Progress Curves

Final concentrations, conditions, and volumes are similar to those previously described in the cathepsin L inhibition assay. Assay buffer, inhibitors (final concentrations varied between 100 nM and 20 μ M) and Z-FR-AMC (final concentration: 5 μ M) were added to the 96-well black plates (volume of the substrate-inhibitor mixture: 180 μ l). Then, 20 μ l of cathepsin L stock solution were added immediately without preincubation time. Readings were taken every 3 seconds for fifty minutes.

Effect of Preincubation Studies on Cathepsin L Inhibition Assays

Preincubation studies with compounds **1** and **8** were carried out by a slight modification of the cathepsin L inhibition assay. Different sets of mixtures containing assay buffer, inhibitor and cathepsin L were preincubated at various periods of times between 0 and 240 minutes. Reactions were taken every 25 seconds for five minutes and carried out in triplicate.

Determination of K_i^{app} using Morrison's Quadratic Equation

Data that was obtained in the effect of preincubation studies for compounds **1** and **8** was further analyzed. The possibility of these compounds to be tight binding inhibitors

was examined by fitting the data by a nonlinear regression using Morrison's quadratic equation.

Cathepsin L Reversibility Studies

Three milliliters of assay buffer for reversibility studies (BSR) were prepared as previously described. Twenty five microliters of cathepsin L (100X: 100 nM) assay buffer for reversibility studies (CLR) were pre-incubated with an equal amount of a concentrated solution of the inhibitor (100X: 0.1 IC₅₀) at 25 °C between one and four hours. Then, two microliters of the enzyme-inhibitor mixture were rapidly mixed with 198 µl of ASR in order to start the reaction. Total reaction volume was 200 µl. Readings were taken every twenty five seconds for four hours. Final concentrations are similar as described previously. Final conditions were: 100 mM NaOAc, pH 5.5, 1 mM EDTA, 3 mM EDTA, 0.01% Brij 35, 1 nM cathepsin L and 50 µM of Z-FR-AMC. Table 21 describes required volumes for this experiment.

Table 23. Preparation Table for Cathepsin L Reversibility Assay

Reagent	Reversibility studies (µl)
Assay Buffer	198
Inhibitor/CL	2
Total	200

Effect of Substrate Concentration (Z-FR-AMC) on IC₅₀ Values

The effect of [Z-FR-AMC] was studied with compounds **1** and **8**. Minor modifications of the cathepsin L inhibition assay were carried out. Different sets of mixtures containing assay, inhibitor and cathepsin L were preincubated at specific preincubation times (5 or 240 minutes). Reactions were initiated by the addition of

different concentrations of Z-FR-AMC. Reading were taken every 25 seconds for five minutes and carried out in triplicate.

Effect of Substrate Concentration (Z-FR-AMC) on Cathepsin L Progress Curves

Final concentrations, conditions, and volumes are similar to those previously described for the cathepsin L inhibition assay. Assay buffer, one inhibitor concentration (Final concentration for **8**: 5 μ M) and Z-FR-AMC (final concentrations varied between 0.5 and 20 μ M) were added to the 96-well black plates (volume of the substrate-inhibitor mixture: 180 μ l). Then, 20 μ l of cathepsin L stock solution was added immediately without preincubating. Readings were taken every 3 seconds for fifty minutes.

Inhibition of Cathepsin L Collagenase Activity by Thiosemicarbazone Derivatives

A solution of human type I collagen ([ACI]: 0.4 mg/ml) and cathepsin L stock solution (CLI) were prepared as previously described. The inhibitor stock solution was prepared by mixing 1.3 μ l of 20 mM stock solution of the inhibitor in DMSO, 9.1 μ l of DMSO and 89.6 μ l of water. Conditions of this stock solution are: 10.4% DMSO and 260 μ M of the inhibitor. The experiment was started by preincubating 3 μ l of ACI and 2.5 μ l of the inhibitor stock solution in a microcentrifuge tube at 37 °C for 2 hours. A 1X staining solution was prepared by diluting 10 μ l of 5000X SYPRO® red protein gel staining dye with 49990 μ l of 7.5% acetic acid. The cathepsin L-inhibitor mixtures were preincubated for 2 hours. Similarly, another group of control samples (without the inhibitor) were also set up. Then, reactions were initiated by adding 7.5 μ l of type I collagen and monitored between 0 and 20 hours at 37 °C. Final conditions of the reactions were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM DTT, 0.01% Brij 30, 275

nM cathepsin L, 2% DMSO, 50 μ M of the tested inhibitor, and 0.01 mg/ml type I collagen. Reactions were stopped by adding 2 μ l of LDS NuPAGE® sample buffer, heated at 90 degrees for ten minutes and immediately stored at -80°C. Inactivated samples (15 μ l) were loaded onto 4-12 % NuPAGE® Bis-Tris gels. Electrophoresis was carried out by using MOPS running buffer at 200 V for 50 minutes. Gels were stained in 50 ml of 1X SYPRO® staining solution at room temperature between one and two hours followed by a destaining process using water (1X) and 7.5% acetic acid (1X). Finally, gel imaging was performed with a GE Typhoon 9400 FL with excitation and emission wavelengths of 550 and 630 nm, respectively. Another experiment was carried out with similar conditions but cathepsin L, inhibitor and type I collagen were added at the same time with no preincubation time. Table 24 describes conditions for inhibition of collagenase activity studies.

Table 24. Preparation Table for Cathepsin L Collagenase Activity Assay

Reagent	Control (μ l)	Inhibited (μ l)	Control (μ l)	Inhibited (μ l)
CLC	3.0	3.0	3.0	3.0
Control (Water)	2.5	0	2.5	0
Inhibitor	0	2.5	0	2.5
TIC	7.5	7.5	7.5	7.5
Preincubation time (h)		2		0
Total	13	13	13	13

Cell Culture Experiments. MDA-MB-231 Cell Subculture Maintenance

One milliliter of passage 2, MDA-MB-231 cells that were kept under cryogenic temperatures, was thawed at 0 °C. Then, cells were cultured with 12 ml supplemented DMEM medium in T75 cm² with flasks with canted necks in incubators at 37 °C and 5%

CO₂/air. Supplemented DMEM medium was changed every two days with 12 ml of fresh DMEM medium. Cell subculture was carried out once cells were 80% confluent.

Cell Subculture (passaging) Protocol

A new generation of MDA-MB-231 cells or passaging was carried out by removing DMEM medium from confluent cells. Immediately, 4 ml of 0.25% (w/l) trypsin with EDTA were added to the cells to induce cells to detach. Flasks were incubated with trypsin in the incubator for 4 minutes. After incubation time, cells were observed under the microscope to confirm if they were completely rounded and detached from the flask surface. Eight milliliters of supplemented DMEM media 1X were added to the flasks in order to neutralize the proteolytic activity of trypsin. Trypsinized cells were used to generate a new generation of cells. Four milliliters of cells were transferred to a new T75 cm² flask with canted neck that contained 8 ml of fresh DMEM media. New passages were cultured and maintained as previously described. Also, trypsinized cells were frozen to maintain and generate new passages for future experiments.

Cell Freezing Protocol

Trypsinized cells were periodically frozen to keep and maintain an inventory for future experiments. Stock solutions were centrifuged at 10000 rpm for 5 minutes to form a cell pellet. Suspended media was removed and replaced with DMEM medium that is supplemented with 5% DMSO. A cell counter was used to determine cell concentration in the new stock solutions. Cell concentration varied between one and 5 million cells/ml. These cell stock solutions were transferred to cryogenic vials. Cells were kept under cryogenic conditions (liquid nitrogen).

Cell Invasion Assays

Invasion assays were carried out using BD Bioscience Matrigel™ invasion kit assay. Independent MDA-MB-231 passages were cultured and passaged as described. Three independent experiments were performed with passages 13, 13 and 13 cells. Cells were trypsinized when they were 80% confluent. However, trypsin neutralization was done with DMEM with no FBS. Cell density was determined with a Beckman Coulter Z-Coulter cell counter. The three stock solutions were diluted with DMEM with no FBS in order to reach a 200,000 cell/ml concentration in each case. Five thiosemicarbazone inhibitors with high cathepsin L selectivity were selected to test its efficacy by delaying cell invasion. Additionally, E-64, a known generic cysteine protease inhibitor was used as a positive control. Stock solutions (20 mM in DMSO) of the compounds and E-64 were prepared by dissolving 1 mg of each sample according to the calculations listed in Table 25.

Table 25. Preparation Table for 20 mM of Selected TSCs for Invasion Assays

Compound	Molecular Weight (g/mol)	Sample (mg)	Sample (μ mol)	Volume (μ l)
1	413.13	1	2.42	121.0
8	350.23	1	2.86	142.8
156	359.44	1	2.78	139.1
157	395.42	1	2.53	126.4
168	474.32	1	2.11	105.4
E-64	357.41	1	2.80	139.9

Then, a 1:10 fold dilution in DMSO ([A]: 2 mM) of each compound was used to prepare two stock aqueous dilutions using DMEM without FBS ([Stock solutions]: 50 and 20 μ M). DMSO was added to each of these dilutions in order to reach a concentration of 4%. Table 26 shows the preparation of these solutions.

Table 26. Preparation Table for 50 and 20 μ M for Selected TSC Inhibitors

Concentration (μ M)	A (μ l)	DMSO (μ l)	Media (μ l)	[DMSO] (%)
50	25	15	960	4
20	10	30	960	4
Control	0	40	960	4

Simultaneously, several BD BioCoat Matrigel® invasion chambers, which contain an 8 micron pore size PET membrane with a layer of Matrigel basement, were prepared for the experiment. Both sides of every matrix were rehydrated using 500 μ l DMEM media with no FBS for two hours in a 5% CO₂ environment at 37 °C. Media were removed and inserts were ready to be used to start the invasion assay. The experiment was initiated by adding 750 μ l of DMEM supplemented with 10% FBS (which functions as a chemoattractant) and gentamicin to a 24-well microplate (i.e. lower chamber). Then, the inserts were carefully placed on top of the wells containing the chemoattractant, avoiding air bubbles between the bottom part of the insert and medium. Two hundred fifty microliters of the cell stock solutions (200,000 cells/ml) and 250 μ l of 50 and 20 μ M of each compound to be tested were added to every insert. Final conditions for treated cells were: 2% DMSO, 50,000 cells and 25 or 10 μ M of the compounds. Final conditions for untreated (controls) cells are: 2% DMSO and 50,000 cells. The 24-well plates containing the invasion studies chambers were placed in an incubator with a 5% CO₂ environment for 24 hours at 37 °C.

Reactions were terminated by removing medium from the inserts. Each insert was cleaned twice with cotton swabs to remove cells located at the top side of the membrane. Membranes were stained with a Diff-Quik staining kit (IMEB, Inc.). Cells were fixed with 100% methanol for a minimum time of two minutes. Then, membranes

were rinsed with deionized water and consecutive staining solutions of azure and xanthenes dyes. Samples were placed in each solution for a minimum of two minutes and rinsed with water in between. Finally, samples were air-dried for a minimum for two hours in a biological safety cabinet.

Membranes were removed using sterile scalpels and placed on glass slides. Each sample was observed under a Zeiss Axiovert 40 CFL inverted microscope to perform manual cell counting. Ten fields were chosen and observed under a 40 X objective. Eight fields were located at the periphery of the circular sample and numbered clockwise. Two more fields were counted at the middle of each sample, as depicted in figure 18. Experiments were performed in triplicate. The inhibition of the invasiveness of MDA-MB-231 cells was measured by using the formula: treated cells that invaded the Matrigel™ layer/untreated cells that invaded the Matrigel™ layer.

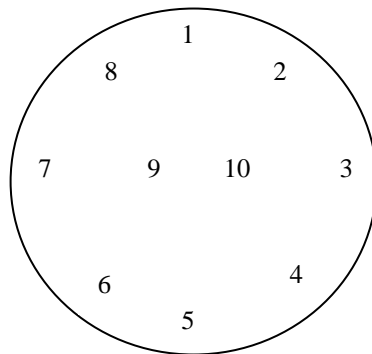


Figure 22. Invasion and Migration Assay Sample Fields

Cell Migration Assays

The experimental procedure for the determination of the ability of MDA-MB-231 to migrate is similar to that described for the cell invasion assays. However, the inserts with an 8 μm did not contain the Matrigel™ layer. Both, invasion and migration assays

were performed simultaneously and used the same MDA-MB-231 cell stock solutions. The inhibition of the motility of MDA-MB-231 cells was measured by using the formula: treated cells that migrated to the 8 μm membrane/untreated cells that migrated to the 8 μm layer.

Results and Discussion

Through a continuing collaboration with Dr. Kevin G. Pinney and his group, our results have indicated that some thiosemicarbazone derivatives are lead compounds as cathepsin L inhibitors.^{3,6,7,10-12} Thiosemicarbazones were first discovered as inhibitors of cruzain, a cathepsin L-like enzyme in a high-through put screening process.⁸ Synthesis of a small library of thiosemicarbazones led to the discovery of a number of thiosemicarbazone derivatives as inhibitors of cruzain. This library was tested toward cathepsin L as well showing promising results. The library of compounds required a complete screening to verify the potency of novel analogs. The Structure-Relationship Activity (SAR) was characterized after completing *in vitro* testing against cathepsin L (IC_{50} values).

The characterization of compounds that could be used as potential therapeutic agents involved numerous evaluations. Understanding the kinetic effect of these inhibitors helped to identify their mechanism of action. Fluorometric based assays were utilized to study various assay parameters in inhibitory activities, determination of K_i , and reversibility of thiosemicarbazone inhibitors. Two dimensional cell based assays (colorimetric cell invasion and cell migration assays) were used to determine the inhibitory ability of these compounds within the extracellular matrices of metastatic breast cancer cell lines.

To understand interactions between inhibitors and cathepsin L, molecular modeling of thiosemicarbazones was employed. These studies will also aid in the understanding of developing inhibitory moieties that could be explored in combination with existing inhibitors.

Cathepsin L studies were carried out using a 96-well microplate fluorometric based assay. The major advantage of this type of assay is the duration of the assay without compromising the reproducibility of the results. Microplate-based assays are commonly used in drug discovery; furthermore, high throughput screening is also available where hundreds or thousands of samples can be evaluated simultaneously. In general, cysteine proteases, including cathepsins L, K, and cruzain, have the highest catalytic activity under acidic conditions. Thus, NaOAc was selected due to its stability, effective and optimal pH range between 3.7 and 5.6.^{5,320} EDTA functions as a chelator and prevents any interaction of possible metal ion traces found in aqueous solution such as mercury with the target protein. DTT is a powerful reducing agent. It is frequently used in enzyme kinetics because it has the ability to protect thiol groups that are essential in cysteine proteases. It can also reduce disulfide bonds by undergoing self oxidation processes by disulfide exchange. Brij 35 is a nonionic detergent that is used for its ability to prevent possible protein aggregation and adherence to containers. DMSO is a universal solvent used to solubilize reagents with low aqueous solubility such as the substrate and thiosemicarbazone derivatives. Z-FR-AMC is a fluorogenic substrate that has been used with serine and cysteine proteases. The lists of enzymes that can cleave this substrate include serine proteases, cathepsins, kallikrein, and plasmin.^{321,322} The basis of the fluorescent assay of cysteine proteases is their ability to release AMC from

Z-FR-AMC, a nonfluorescent substrate. AMC is a compound that fluoresces when it is excited at 354 nm (excitation: 354 nm; emission: 442 nm).³²³ Final conditions for the cathepsin L fluorometric assay were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 3 mM DTT, 0.01% Brij 35, 0.7 - 2% DMSO, 1 nM cathepsin L, and 0.5 - 50 μ M Z-FR-AMC.

Assay Optimization. Effect of DMSO on Cathepsin L Inhibition Assays

Research related to cathepsin L has increased significantly over time. Numerous experimental conditions are described and all of them are customized based on the main objectives of the research. An important factor to consider is the effect of DMSO during enzymatic catalysis. Reports of cathepsin L assays range from 0.7% to 10% final concentration.^{324,325} Thus, it was necessary to analyze the effect of DMSO concentration in our samples. Experiments were conducted using analogs **1**, and **156**, to study the effect of DMSO concentration (See Table 27)

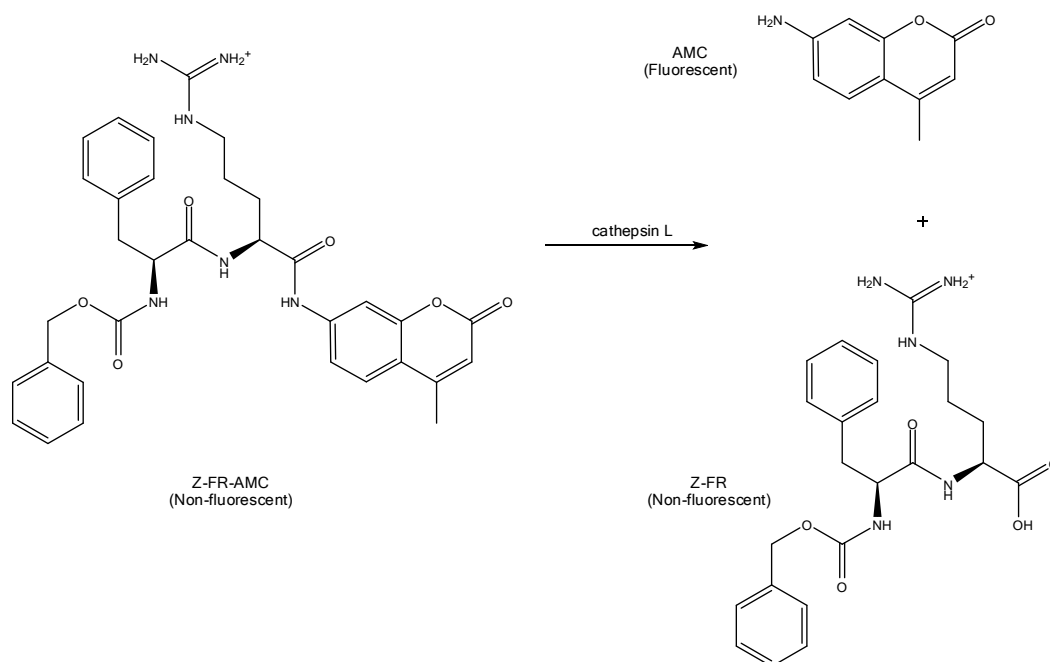
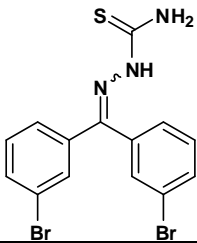
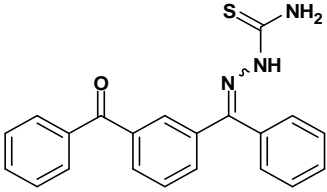


Figure 23. Hydrolysis of the Z-FR-AMC using cathepsin L

Table 27. Effect of DMSO Concentration on Inhibition Studies

DMSO (%)		
	1	156
	IC ₅₀ (nM)	
0.7	70	ND
2	16.7	9.85
3	6.64	7.7

[Z-FR-AMC]: 50 μM; preincubation time: 5 min

The obtained results showed the effect of DMSO on IC₅₀ values of cathepsin L inhibitors. Solubility of the lead compounds (**1** and **156**) was increased upon addition of DMSO and their inhibitory activity was enhanced accordingly. The efficiency of **1** is 6 times better when comparing its IC₅₀ at 0.7 and 2%, an overall of 2.85 fold increment of DMSO in the enzymatic assay. (IC₅₀= 70 and 11.8 nM respectively). On the other hand, the effect of the solvent is not as evident when comparing IC₅₀ values for **156**. (IC₅₀= 9.85 and 7.7 nM). It was also found the stability of the enzyme activity was also increased with higher concentrations of DMSO.

Construction of AMC Calibrations Curves

Fluorometric based assays were used to study the potency, mechanism and mode of inhibition of thiosemicarbazones as cathepsin L inhibitors. The principle of the assay relies on the ability of cathepsin L to cleave AMC, a fluorophore, from Z-FR-AMC, a nonfluorescent synthetic substrate that is extensively used in research involving cysteine cathepsins such as cathepsin L and K.

The construction of AMC curves was necessary in order to convert the relative fluorescence units into the concentration of the fluorophore. The curve allowed the validation of enzymatic activities, kinetic and inhibition assays. Also, kinetic parameters, velocities and rates could be compared to standard concentration units that can be found in the literature. Figure 24 and Figure 25 show the results of the construction of AMC curves. Results of the second linear regression give an equation that can correlate RFU and AMC concentration. The equation is $RFU = 188.6 \text{ AMC (in } \mu\text{M)} + 30.4$ and the coefficient of determination (r^2) were also calculated (0.9931).

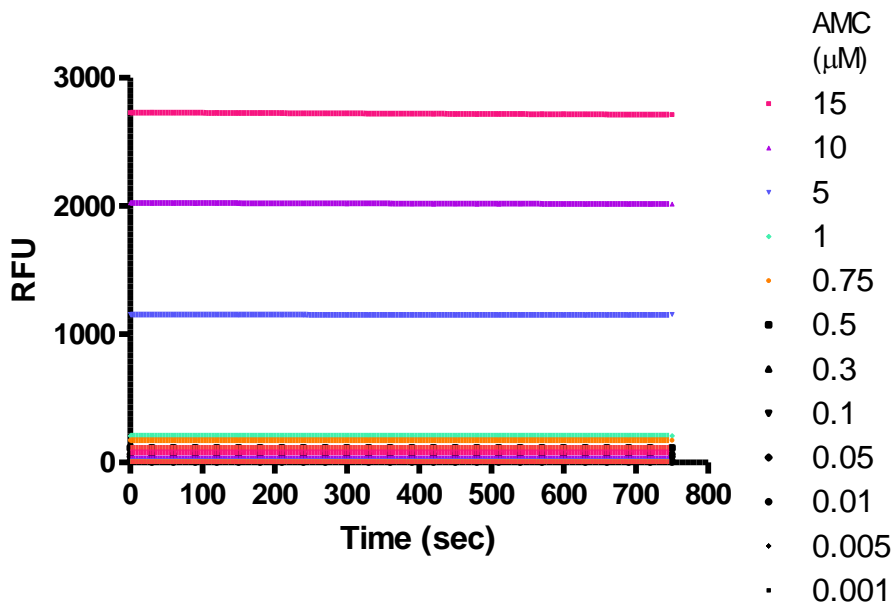


Figure 24. Fluorescence Response AMC vs. Time

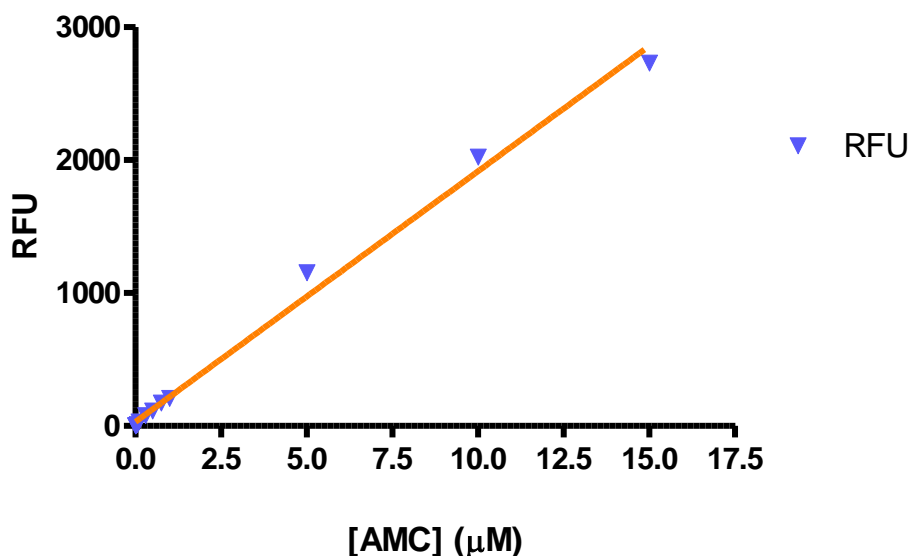


Figure 25. AMC Calibration Curve

Determination of K_M , V_{MAX} and k_{CAT}

Determination of kinetic constants was based on the steady-state analysis assumption. A large concentration of the substrate (up to 15-fold K_M value) was used for studies related to cathepsin L. Cathepsins L catalytic activity showed a linear behavior even at low substrate concentrations (0.2 μM). The cleavage of the substrate (Z-FR-AMC) was monitored for at least 5 minutes. Complete details of this experiment can be found in the Material and Methods Section of this chapter. The determination of K_M , V_{MAX} and k_{CAT} was possible with experiments that observe the catalytic activity of a fixed concentration of cathepsin L (1 nM) but vary the concentration of the substrate Z-FR-AMC (0.2 -14 μM). Experiments were carried out in triplicate. Catalytic rates were calculated by applying linear regression of the data. ([AMC] is the dependent variable and time (seconds) is the independent variable). Then, a nonlinear regression analysis of the Michaelis-Menten equation (Eq. 1.1) was performed to calculate K_M and V_{MAX} values with the aid of commercially available software (GraphPad 5.0). The k_{CAT} constant value

was determined using equation 2.1. The velocity v_o is the initial rate velocity at a specific substrate concentration. The V_{MAX} is the maximum velocity, K_M is the Michaelis-Menten constant, $[S]$ is the substrate concentration, and k_{CAT} is the catalytic rate constant of the reaction.

$$v_o = \frac{V_{MAX}[S]}{K_M + [S]} \quad (1.1) \quad k_{CAT} = \frac{V_{MAX}}{[cathepsin L]} \quad (2.1)$$

K_M , V_{MAX} and k_{CAT} values were found to be $1.1 \pm 0.17 \mu\text{M}$, $1.56 \pm 0.04 \text{ AMC nM/s}$ and 1.56 s^{-1} , respectively.³²⁶

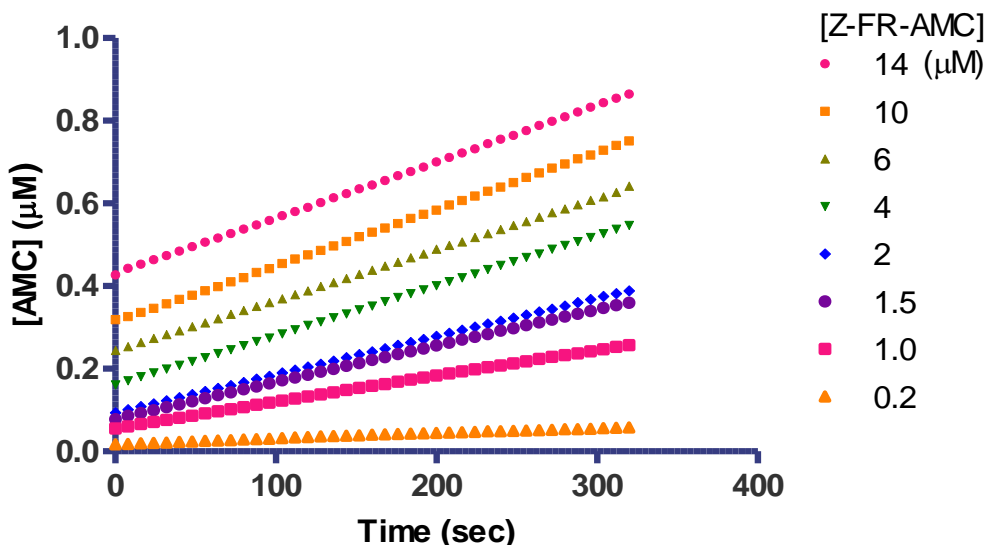


Figure 26. Catalytic Activity of Cathepsin L Using Z-FR-AMC as the Fluorogenic Substrate

Determination of Inhibitory Efficacy of Thiosemicarbazone Analogs at 10 μM

The first set of experiments was performed at a set concentration (10 micromolar) of individual compounds to verify if they were cathepsin L inhibitors. If the catalytic activity of the cysteine protease was inhibited by 50% or more by a fixed concentration (final concentration: 10 μM) of the potential inhibitor. Three independent sets of

experiments of untreated ([I]: 0 μM) and treated samples ([I]:10 μM) were preincubated with 1 nM cathepsin L for 5 minutes at 25 $^{\circ}\text{C}$.

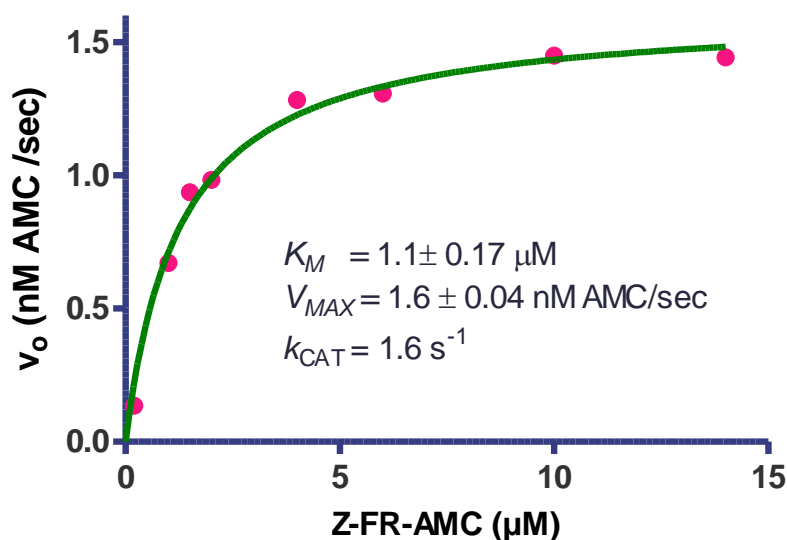


Figure 27. Determination of K_M and V_{MAX} for Human Cathepsin L Using Z-FR-AMC as a Fluorogenic Substrate

Inhibitory activities were monitored when reactions were started by adding 50 μM Z-FR-AMC as a fluorogenic substrate. Reactions demonstrated a linear behavior for the first five minutes. Catalytic rates of uninhibited and inhibited samples were calculated by linear regression of the data ([AMC]: dependent variable and time (seconds): independent variable). Compounds that exhibited inhibition of more than 50% cathepsin L activity at 10 μM in solution were further analyzed to determine their exact IC_{50} values. If the ratio $v_i/v_o \leq 0.5$, then the compounds were not considered potential inhibitors and an approximated IC_{50} value ≤ 10000 nM was assigned to them.

Determination of IC_{50} Values

A library of more than one hundred fifty thiosemicarbazone analogs was analyzed to verify their efficacy to inhibit the catalytic activity of cathepsin L. The synthetic

compounds were synthesized under the guidance of Dr. Kevin G. Pinney by several members of his research group. The synthesis of these compounds has been previously published.^{3,6,7,10-12} A 96-well microplate fluorometric based assay was utilized to determine the inhibitory activity of each of these inhibitors. Uninhibited cathepsin L catalytic activity showed linear behavior when 50 μM Z-FR-AMC was used for reaction times that were 5 minutes long. The determination of the IC_{50} values was carried out with experiments that observed the inhibitory capacity of the synthetic compounds when a fixed concentration of cathepsin L (1 nM) was preincubated for 5 minutes at 25 $^{\circ}\text{C}$. The final concentration of each compound varied between 10 pM and 10 μM . Experiments were carried out in triplicate. Catalytic rates of uninhibited and inhibited samples were calculated by linear regression of the data. ([AMC]: dependent variable and time (seconds): the independent variable). Data followed a typical sigmoidal dose response, and therefore, a nonlinear regression of the equation 1.2 was performed to calculate IC_{50} values with the aid of commercially available software (GraphPad 5.0). The value Y represents the inhibited activity (normalized relative to control), X is $\log([\text{inhibitor}])$ in M. The velocities v_{min} and v_{max} represent cathepsin L preincubation with the highest and lowest inhibitor concentrations respectively (10 μM and 0 pM or control). The Hill slope value is the slope of the sigmoidal curve. $\text{IC}_{50} \pm \text{S.E.}$ values represent the average and standard errors of at least three experiments. Structure-activity relationship (SAR) is shown in Tables 28 to 45 that group these inhibitors by functional groups or families.

$$Y = \frac{v_{\text{MIN}} + (v_{\text{MAX}} - v_{\text{MIN}})}{1 + 10^{(\log(\text{IC}_{50}-X) * \text{Hillslope})}} \quad (1.2)$$

Table 28. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12}

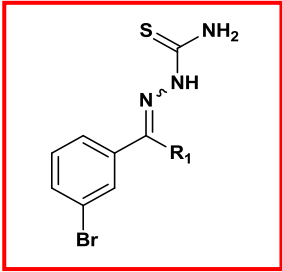
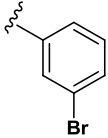
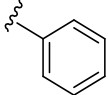

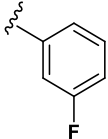
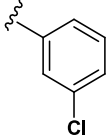
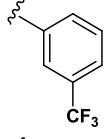
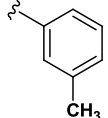
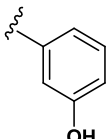
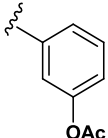
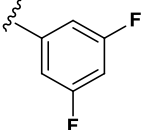
Compound	Structure	R ₁	IC ₅₀ ± S.E. (nM)
1			16.7 ± 1.0
2			≥ 10000
3 ⁸			832.9 ± 16.9
4			250.3 ± 18.9
5			131.4 ± 4.0
6			46.5 ± 2.8
7			224.4 ± 2.8
8			188.7
9			150.8 ± 16.0
10			59.4 ± 0.7

Table 28. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
11			520.9 ± 24.6
12			415.2 ± 35.6
13			96.0 ± 4.0
14			23.83 ± 0.9
15			1610.0 ± 150.0
16			1200.0 ± 130.0
17			79.6 ± 2.7
18			327.1 ± 13.8
19			2156.0 ± 70
20			≥ 10000

Table 28. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S., (nM)
21			≥ 10000
22			118.0 ± 4.0
23			63.2 ± 4.6
24			2600.0 ± 546
25			609.7 ± 48
26			113.6 ± 10.8
27			83.3 ± 3.3
28			233.3 ± 29.0
29			1000 ± 90
30			>10000

Table 28. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
31			126.1 ± 0.6
32			≥ 10000
33			≥ 10000
34			232.4 ± 19.3
35			814.5 ± 78.9
36			931.3 ± 124.5
37			369.7 ± 7.8
38			≥ 10000
39			≥ 10000
40			9650 ± 1.65

Table 28. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

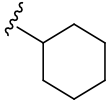
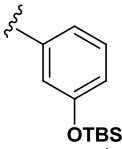
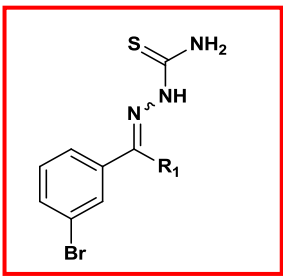
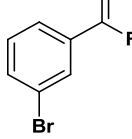
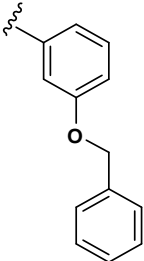
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
41			≥ 10000
42			≥ 10000
43			≥ 10000
44			≥ 10000

Table 29. Inhibition of Human Cathepsin L by *meta*-Substituted Propanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

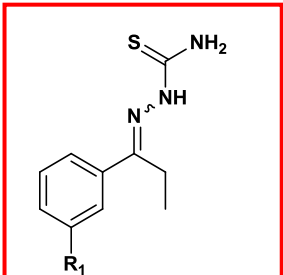
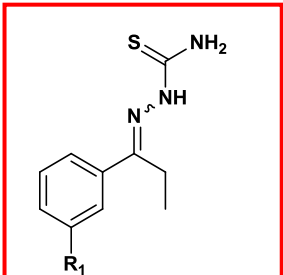
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
45		NO ₂	1905 ± 206
46		NH ₂	≥ 10000

Table 30. Inhibition of Human Cathepsin L by *para*-Bromo Functionalized Benzophenone Thiosemicarbazones. For Synthesis: ^{3,6,7,10-12}

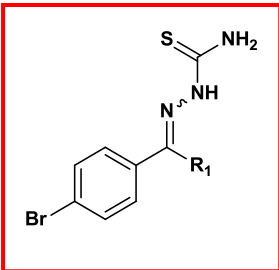
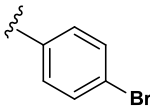
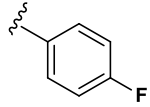
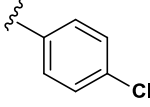
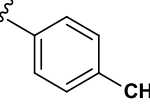
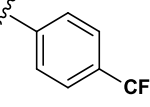
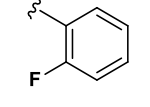
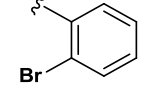
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
47			≥ 10000
48			3320 ± 260
49			≥ 10000
50			2828 ± 92
51			≥ 10000
52			2423 ± 576
53			≥ 10000

Table 31. Inhibition of Human Cathepsin L by Dihalogen-substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

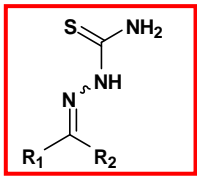
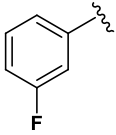
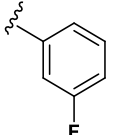
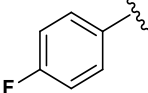
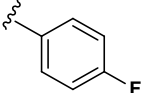
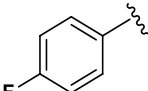
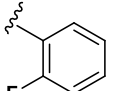
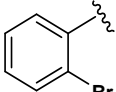
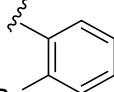
Compound	R_1	R_2	$IC_{50} \pm S.E.,(nM)$
			
54			4871 ± 89
55			≥ 10000
56			2458
57			ND

Table 32. Inhibition of Human Cathepsin L by 3,3'-Dibromo-*N*-Substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

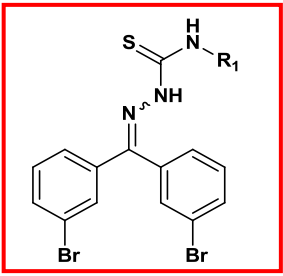
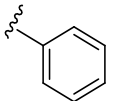
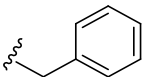

Compound	Structure	R_1	$IC_{50} \pm S.E.,(nM)$
58			≥ 10000
59			≥ 10000
60			≥ 10000

Table 33. Inhibition of Human Cathepsin L by Thiosemicarbazones Containing a Phenyl Group. For Synthesis of Compounds: ^{3,6,7,10-12}

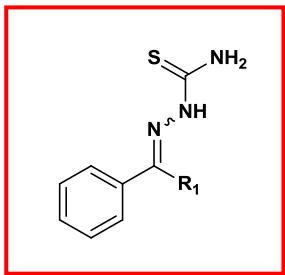
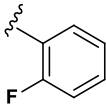
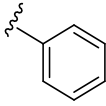
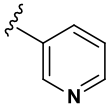
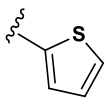
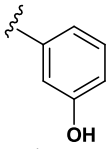
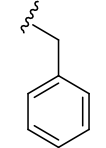
<i>Compound</i>	<i>Structure</i>	<i>R₁</i>	<i>IC₅₀ ± S.E., (nM)</i>
61			≥ 10000
62			≥ 10000
63			≥ 10000
64			≥ 10000
65			≥ 10000
66			5410 ±

Table 34. Inhibition of Human Cathepsin L by Di-halogenated or Monohalogenated Ketones. For Synthesis of Compounds: ^{3,6,7,10-12}

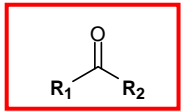
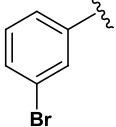
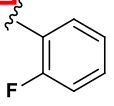
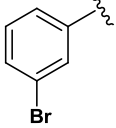
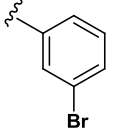
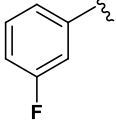
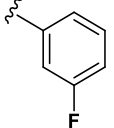
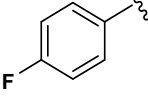
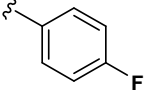
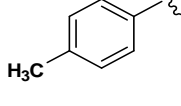
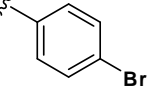
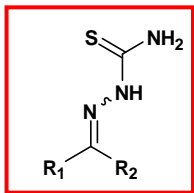
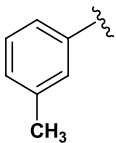
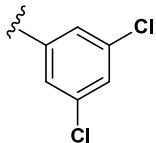
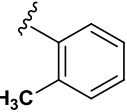
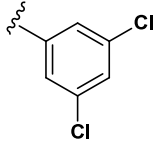
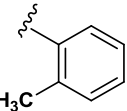
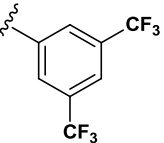
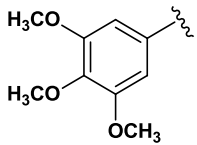
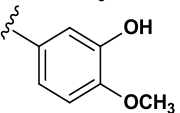
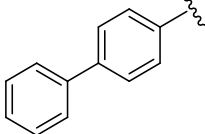
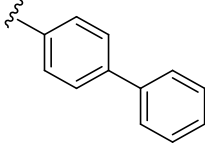
Compound	X ₁	X ₂	IC ₅₀ ± S.E., (nM)
			
67			≥ 10000
68			≥ 10000
69			≥ 10000
70			≥ 10000
71			≥ 10000

Table 35. Inhibition of Human Cathepsin L by Substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

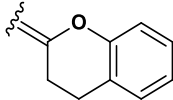
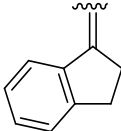
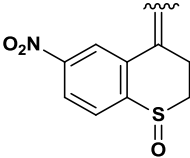
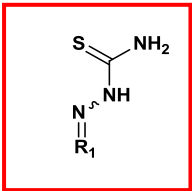
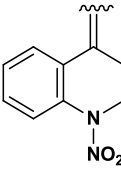
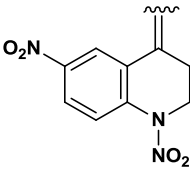
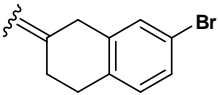
Compound	R_1	R_2	$IC_{50} \pm S.E., (nM)$
			
72			≥ 10000
73			86.4 ± 6.0
74 ³²⁷			101.4 ± 4.8
75			3600 ± 250
76			ND (Insoluble)

*ND: Not Determined

Table 36. Inhibition of Human Cathepsin L by Functionalized Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
77			$\geq 10000^*$
78			$\geq 10000^*$
79			$\geq 10000^*$
80			≥ 10000
81			≥ 10000
82			≥ 10000
83			≥ 10000
84			$\geq 10000^*$
85			≥ 10000
86			≥ 10000

Table 36. Inhibition of Human Cathepsin L by Functionalized Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}(Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
87			≥ 10000*
88			≥ 10000*
89			≥ 1740 ± 50
90			≥ 10000
91			ND
92			228.4 ± 7.2
93			≥ 10000*

* [Z-FR-AMC]: 5 μM; ND: not determined

Table 37. Inhibition of Human Cathepsin L by Functionalized Annulone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

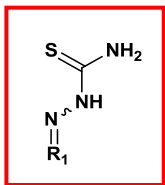
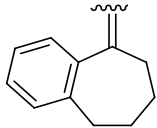
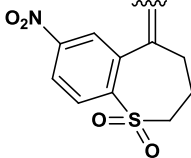
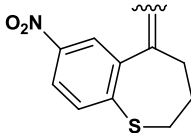
Compound	Structure	R_1	$IC_{50} \pm S.E.,(nM)$
94			≥ 10000
95			≥ 10000
96			658.2 ± 114.0

Table 38. Inhibition of Human Cathepsin L by Substituted Quinolone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

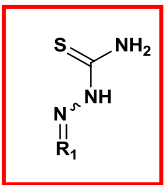
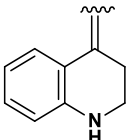
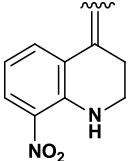
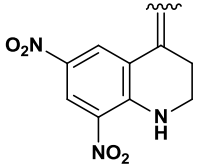
Compound	Structure	R_1	$IC_{50} \pm S.E.,(nM)$
97			≥ 10000
98			≥ 10000
99			337.9 ± 9.6

Table 38. Inhibition of Human Cathepsin L by Substituted Quinolone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)


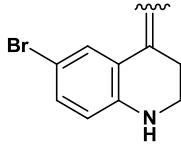
<i>Compound</i>	<i>Structure</i>	<i>R₁</i>	<i>IC₅₀ ± S.E.,(nM)</i>
100			163.8 ± 16.4

Table 39. Inhibition of Human Cathepsin L by Non-Thiosemicarbazone Based Analogs. For Synthesis of Compounds: ^{3,6,7,10-12}

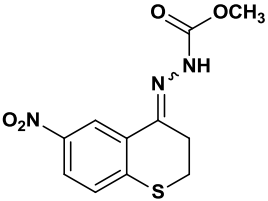
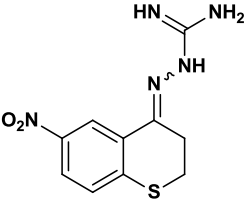
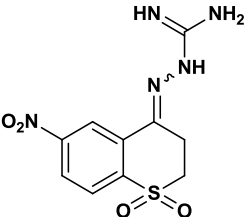
<i>Compound</i>	<i>Structure</i>	<i>IC₅₀ ± S.E.,(nM)</i>
101		≥ 10000
102		≥ 10000
103		≥ 10000

Table 39. Inhibition of Human Cathepsin L by Non-Thiosemicarbazone Based Analogs.
For Synthesis of Compounds: ^{3,6,7,10-12}(Continued)

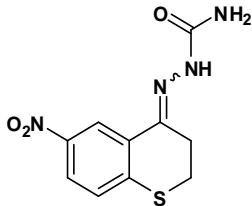
Compound	Structure	IC ₅₀ ± S.E.,(nM)
104		≥ 10000

Table 40. Inhibition of Human Cathepsin L by Substituted Tetralone
Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

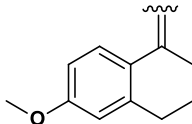
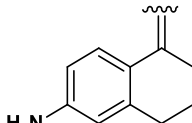
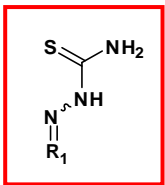
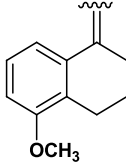
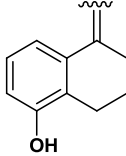
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
105			≥ 10000
106			≥ 10000
107			≥ 10000
108			≥ 10000
109			≥ 10000

Table 41. Inhibition of Human Cathepsin L by Functionalized Chromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

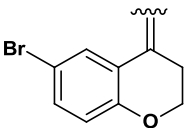
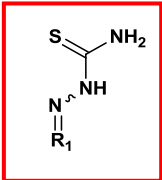
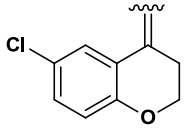
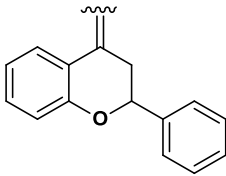
<i>Compound</i>	<i>Structure</i>	<i>R₁</i>	<i>IC₅₀ ± S.E., (nM)</i>
110			303.0 ± 24.7
111			369.4 ± 4.5
112			≥ 10000

Table 42. Inhibition of Human Cathepsin L by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

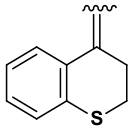
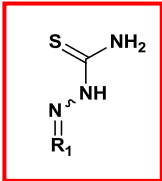
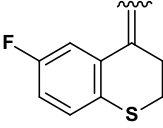
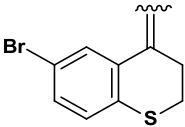
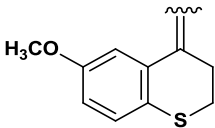
<i>Compound</i>	<i>Structure</i>	<i>R₁</i>	<i>IC₅₀ ± S.E., (nM)</i>
113			≥ 10000*
114			741.8 ± 23.3
115			152.3 ± 2.7
116			≥ 10000

Table 42. Inhibition of Human Cathepsin L by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

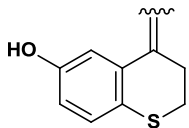
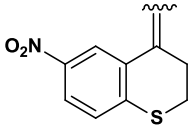
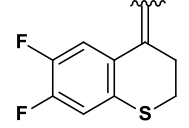
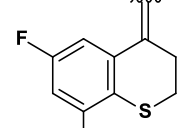

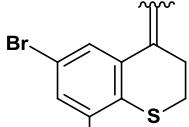
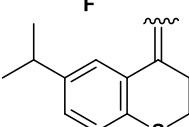
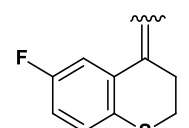
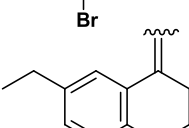
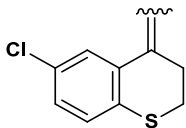
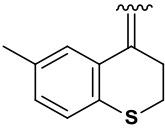
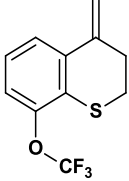
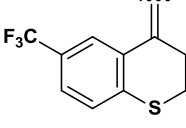
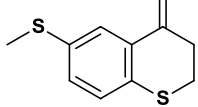
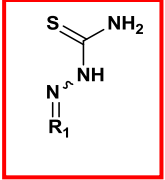
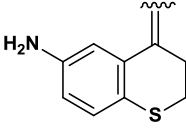
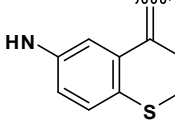
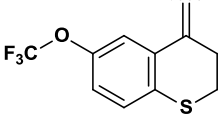
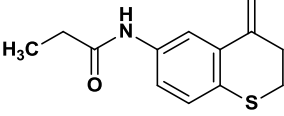
Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
117			≥ 10000
118			67.85 ± 4.5
119			54.0 ± 7.6
120			1500 ± 100.1
121			≥ 10000
122			434.2 ± 14.8
123			≥ 10000
124			≥ 10000
125			2720 ± 240

Table 42. Inhibition of Human Cathepsin L by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
126			228.4 ± 11.6
127			213.8 ± 16.4
128			≥ 10000
129			284.1 ± 10.7
130	 		≥ 10000
131			≥ 10000
132			≥ 10000
133			255.5 ± 26.2
134			≥ 10000

*[Z-FR-AMC]: 5 μM

Table 43. Inhibition of Human Cathepsin L by Substituted Sulfone Thiosemicarbazones.
For Synthesis of Compounds: ^{3,6,7,10-12}

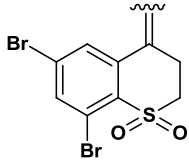
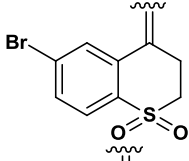
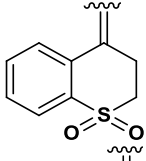
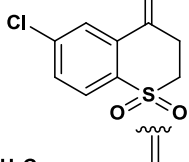
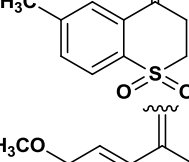
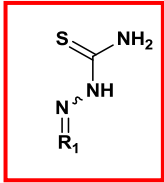
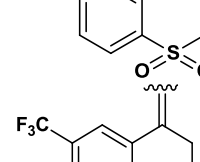
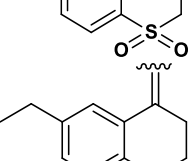
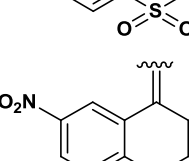
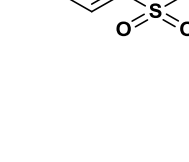
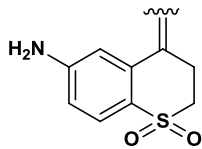
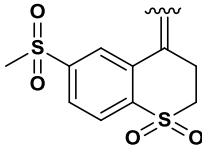
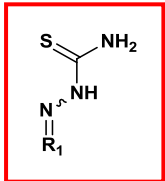
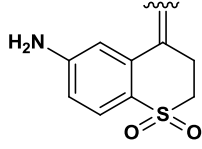
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
135			355.9 ± 11.8
136			573.9 ± 37.4
137			≥ 10000*
138			ND
139			≥ 10000
140			≥ 10000
141			259.8 ± 34.4
142			6521 ± 490
143			112

Table 43. Inhibition of Human Cathepsin L by Substituted Sulfone Thiosemicarbazones.
 For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
144			3650 ± 378
145			≥ 10000
146			≥ 10000
147			≥ 10000
148			≥ 10000
149			1117 ± 102.7
150			3960 ± 40
151			≥ 10000
152			≥ 10000

Table 43. Inhibition of Human Cathepsin L by Substituted Sulfone Thiosemicarbazones.
For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
153			≥ 10000
154			≥ 10000
155			≥ 10000

*ND: 5 μM

Table 44. Inhibition of Human Cathepsin L by Functionalized Benzoyl-Benzophenone Thiosemicarbazones*

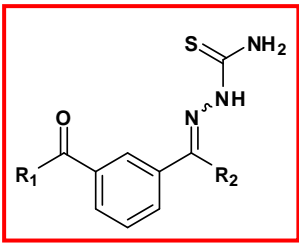
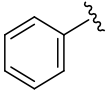
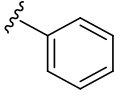
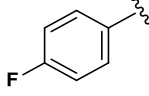
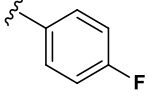
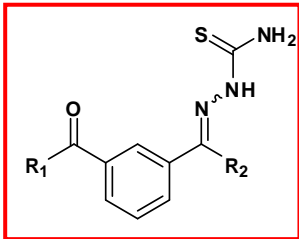
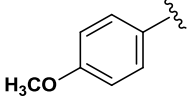
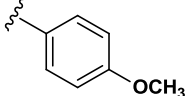
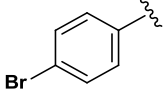
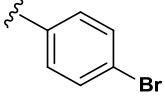
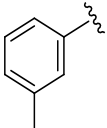
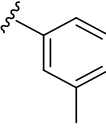
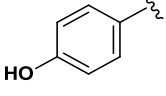
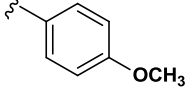
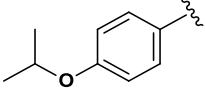
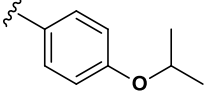
Compound	R ₁	R ₂	IC ₅₀ ± S.E., (nM)
			
156			9.85 ± 0.60
157			24.3 ± 1.2

Table 44. Inhibition of Human Cathepsin L by Functionalized Benzoyl-Benzophenone Thiosemicarbazones* (Continued)

Compound	R_1	R_2	$IC_{50} \pm S.E., (nM)$
			
158			586.9 ± 46.3
159			≥ 10000
160			7823 ± 736
161			≥ 10000
162			≥ 10000

*Compounds **156-162** have been synthesized by members of the Kevin G. Pinney laboratory (Baylor University)

Table 45. Inhibition of Human Cathepsin L by Substituted-Benzoyl-Benzophenone Thiosemicarbazones*

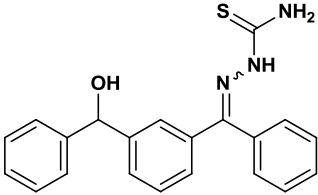
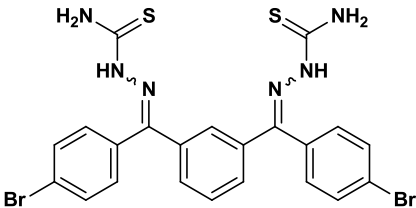
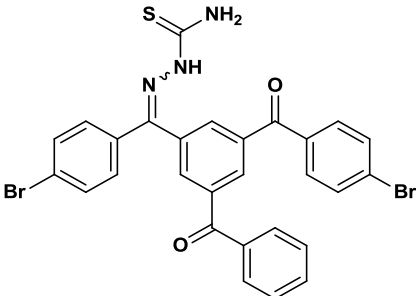
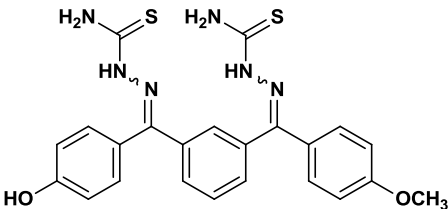
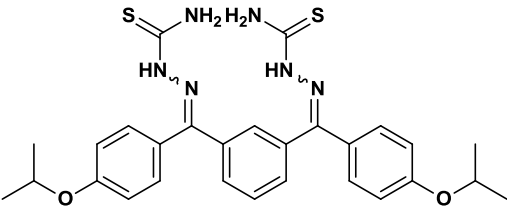
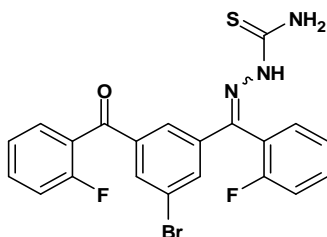
Compound	Structure	$IC_{50} \pm S.E.,(nM)$
163		23.8 ± 0.8
164		≥ 10000
165		25.0 ± 0.23
166		≥ 10000
167		≥ 10000

Table 45. Inhibition of Human Cathepsin L by Substituted-Benzoyl-Benzophenone Thiosemicarbazones* (Continued)

Compound	IC_{50} (\pm S.E.), nM
168	8.12 \pm 0.5



*Compounds **163-168** have been synthesized by members of the Kevin G. Pinney laboratory (Baylor University)

Structure-Activity Relationship (SAR) of Thiosemicarbazones as Cathepsin L Inhibitors

More than 150 thiosemicarbazone (TSC) and ketone-derived synthetic compounds were analyzed during the preliminary screening in order to determinate the potency of these compounds. Approximately half of the samples showed a significant inhibitory potency, and IC_{50} values were calculated. A selected group of eight different compounds were excellent cathepsin L inhibitors with IC_{50} values between 8 and 55 nM. Eleven compounds demonstrated to be good inhibitors ($50 \leq IC_{50} \leq 100$ nM). One fifth of the library (36 analogs) had a moderate activity. IC_{50} values of these synthetic compounds were ranged between 100 and 1000 nM.

Potent cathepsin L inhibitors. Figure 28 summarizes the structures of the compounds with IC_{50} values less than 55 nM. Three analogs belong to the subfamily of *meta*-brominated benzophenone TSC analogs. Fluorine, bromine, and trifluoromethyl, are all halogenated moieties and the substituents in this subfamily. The second subgroup consists of four unsubstituted and substituted benzoyl benzophenone thiosemicarbazones.

Interestingly, the best two compounds found in the family of TSC are benzoyl-benzophenone TSC with IC_{50} values less than 10 nM. The list of this series is completed with a 6,7-difluoro thiochromanone and a benzoyl benzhydrol thiosemicarbazone.

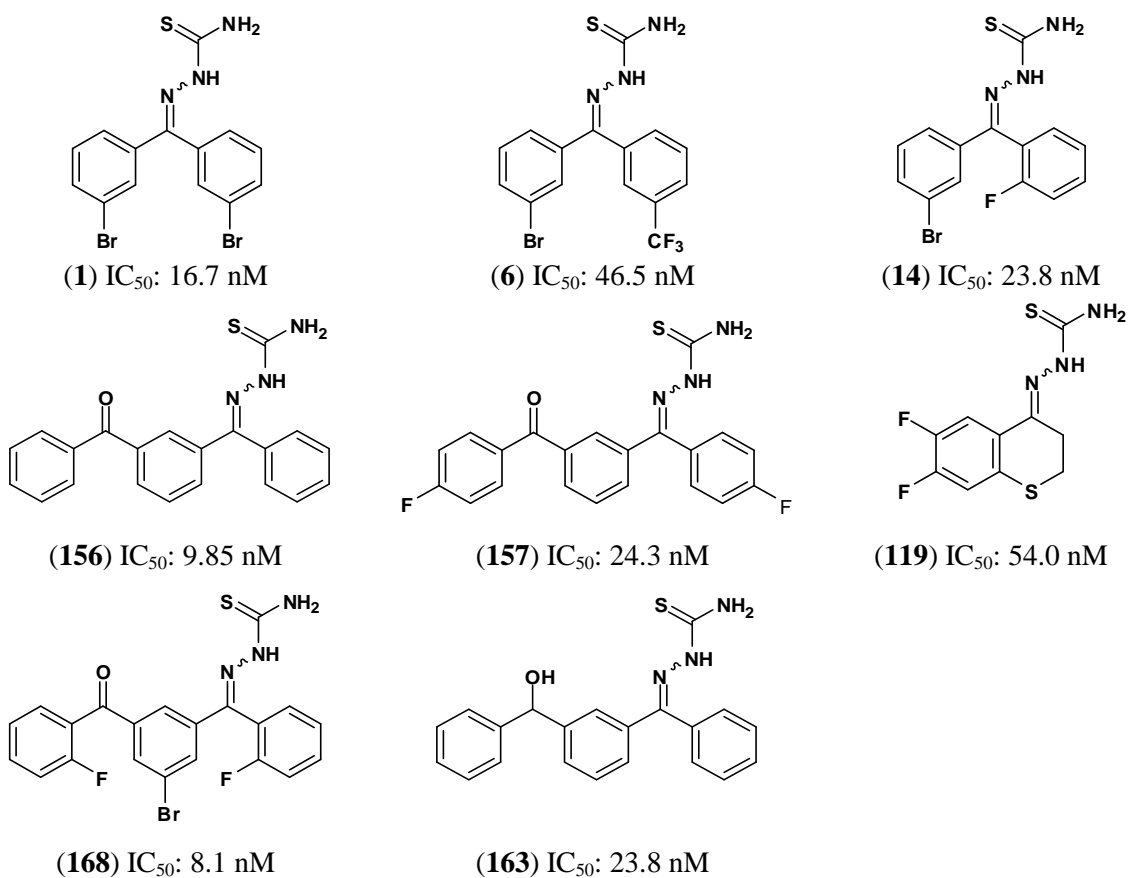


Figure 28. Thiosemicarbazone analogs with Potent Inhibitory Activity against Human Cathepsin L⁹⁻¹²

Compounds with IC_{50} less than 100 nM. Eleven compounds proved to have good inhibitory activity (less than 100 nM). In addition to the eight compounds in Figure 28, these compounds included: 3-bromo-3',5'-difluorobenzophenone TSC, 3-bromo-3',5'-ditrifluoromethylbenzophenone TSC, 3-bromo-3'-fluorobenzophenone TSC, 3-bromo-

3',4',5',6'-fluorobenzophenone TSC, 2'-methyl-3',5'-difluoro benzophenone TSC, 6'-nitrothiochromanone TSC, 6'-nitrosulfone TSC, and four others.

Compounds with moderate activity. Thirty six TSC analogs showed IC₅₀ values between 100 and 1000 nM. The list included, but was not limited to: *meta*-brominated benzophenone TSCs, substituted tetrahydroquinoline TSCs halogenated chromanone TSCs, some naphthalene, sulfides, and sulfone TSCs and some substituted benzoyl benzophenone TSCs.

General Remarks of the Structure-Activity Relationship

Further analysis of the structure-activity relationship revealed that thiosemicarbazone inhibitory activity can be enhanced or reduced dramatically by moiety substitution. The series of substituted *meta*-brominated benzophenone thiosemicarbazones is the largest subgroup among the compounds that were tested. It consists of forty-four different analogs with inhibitory activities that varied between 16.7 and ≥ 10000 nM. Halogenated substituents greatly enhanced the activity of the thiosemicarbazones. Compounds **1**, **2**, **4**, **5** (Table 46) are thiosemicarbazones sharing the main moiety (monobromobenzophenone thiosemicarbazone) but varying the *meta*-halogenated substituent. Compound **2** is used as a comparison to complete the series. Analog **2** does not possess any inhibitory activity towards cathepsin L. The activity is inversely proportional to the atomic size of the substituent. The brominated analog is 15-fold more active than its fluorinated analog.

Table 46. Inhibition of Human Cathepsin L by 3-Bromo-3'-Halogen Benzophenone Thiosemicarbazones

R_1							
		<		<		<	
1	5	4	2				
IC_{50} (nM)	16.7	131.4	250.3	≥ 10000			

Five fluorinated-substituted benzophenone TSC analogs (**4**, **10**, **22**, **23**, and **25**; Table 47) were also compared. Polyfluorination of the benzyl rings indicates that cathepsin L-inhibitory activity of the compounds can be increased with the addition of the electronegative fluoro-substituents. However, a deeper comparison between the two difluoro benzophenone TSCs also reveals that position of the substitutions plays a key role in the activity of these analogs. The *ortho*-difluoro analog is 10 times less active than the *dimeta* counterpart.

The effect of heteroatomic substituents was also evaluated (Table 48). Various groups were substituted at the *meta*- position of one of the phenyl rings. The group includes trifluoromethyl, methyl, hydroxyl, and acetate moieties. Three compounds showed modest activities compared to the trifluoromethyl analog, which is almost five times more potent than the methylated analog. Notably, compound **8** has improved solubility characteristics and was further investigated.

Table 47. Inhibition of Human Cathepsin L by 3-Bromo-Fluorinated-benzophenone Thiosemicarbazones

R_1		>		>	
	10		23		22
				>	
					4
				>	
					25
IC_{50} (nM)	59.4		63.2		118.1
					250.3
					609.7

Table 48. Inhibition of Human Cathepsin L by 3-Bromo-3'-Heteroatomic groups Benzophenone Thiosemicarbazones

R_1				
		<		<
	6		8	
				<
			9	
				<
			7	
IC_{50} (nM)	46.5		131.4	
			150.8	
			224.4	

A small trifluoromethyl series of three compounds (**6**, **11**, and **13**; Table 49) also revealed that substituted compounds often show better activity against cathepsin L when compared to monosubstituted. *Meta*-compounds are also better cathepsin L inhibitors than *para*-compounds. Additionally, the position of the substituent enhances or decreases the activity of these synthetic compounds.

Table 49. Inhibition of Human Cathepsin L by 3-Bromo-Trifluoromethyl Benzophenone Thiosemicarbazones

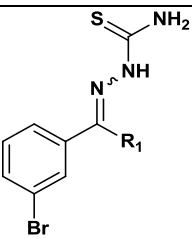
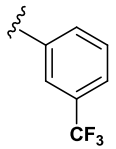
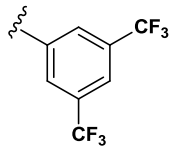
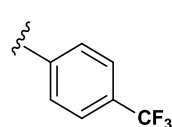
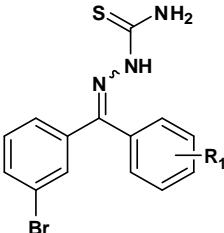
R_1			
			
	6	13	11
IC ₅₀ (nM)	46.5	96.0	520.9

Table 50. Effect of the Position of Substituents in the Inhibitory Activity of 3-Bromo-Benzophenone Thiosemicarbazones

	R_1	<i>ortho</i>	<i>meta</i>	<i>para</i>
	-F	(14) 23.9	(4) 250.3	(17) 79.6
	-Cl	(15) 1610.0	(5) 131.4	(18) 327.1
	-Br	(24) 2600	(1) 16.7	(21) ≥ 10000
	CH ₃	(20) ≥ 10000	(7) 224.4	(19) 2156.0

However, the effect of position of the substituent can be analyzed in depth with the series of twelve compounds that are monosubstituted with halogen and aliphatic groups (F, Cl, Br, and CH₃). Table 50 compares IC₅₀ for these compounds. The inhibitory activity of the compounds increases with more electronegative substituents if they are positioned in the *ortho* position. On the other hand, the trend is the opposite for halogen based the substituents are found in the *meta* positions. Furthermore, the *meta*-brominated analog is the third most potent inhibitor found in this library. In general, *meta*-substituted TSCs analogs are better inhibitors than *ortho*- and *para*-substituted TSCs with the exception of fluoro-substituted benzophenone thiosemicarbazones.

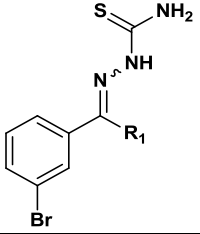
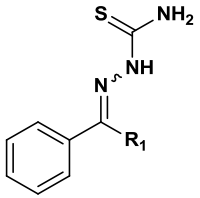
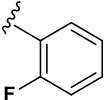
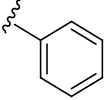
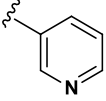
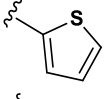
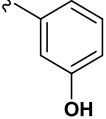
Compounds **61-65** also demonstrated the importance of the presence of bromine in one of the phenyl rings in order to enhance the potency of thiosemicarbazones. A comparison of unsubstituted and bromo-substituted analogs unsubstituted and halogenated benzophenones, phenols, pyridines, and thiophenes; revealed that none of the unsubstituted compounds showed significant inhibitory activity. Similar trends were observed with substituted hydroquinolines, and naphthalenes (Table 51).

The activity of TSCs was greatly enhanced by the bromination of the benzophenone in the *meta* position (up to 41 times more potent in the case of the fluorinated analogs) with the exception of the chlorinated derivative which showed no activity in either case (Table 52).

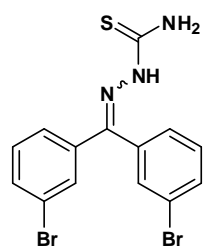
Bromination of the benzophenone group is also critical for the activity of TSC as potential cathepsin L inhibitors. A closer examination of halogenated compounds with similar chemical structures reveals that a different halogen substituent, such as fluorine, greatly reduces the activity of the compounds. The 3,3'-difluorobenzophenone TSC (**54**) is almost 300 times less active than 3,3'-dibromobenzophenone TSC (**1**). Furthermore, both 4,4'-difluorobenzophenone TSC (**54**) and 4,4'-dibromobenzophenone TSC (**47**) showed no activity against cathepsin L, showing that *para*-disubstitution with halogen substituents cannot be used as a possible route for the design of similar compounds.

Compounds **58, 59, and 60** (Table 32) demonstrated that modification in the thiosemicarbazone moiety was detrimental in the activity of the synthetic compounds. None of them showed significant activity towards cathepsin L.

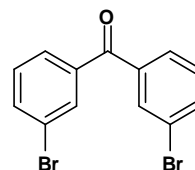
Table 51. Comparison between Brominated and Unbrominated Benzophenone, Thiophene and Pyridine Thiosemicarbazones for Cathepsin L

R_1		
	(14) 23.8	(61) ≥ 10000
	(2) ≥ 10000	(62) ≥ 10000
	(29) 1000	(63) ≥ 10000
	ND	(64) ≥ 10000
	(8) 188.7	(65) ≥ 10000

Screening of building blocks also demonstrated that TSC is one of the key features for a successful inhibition of cathepsin L. Compound **1**, the reference compound, is a potent inhibitor with an IC_{50} value of 16.7 nM. On the other hand, the dibromosubstituted ketone did not show any activity under the same conditions (Figure 29). A similar situation was observed with 3-bromo-2'-fluoro TSC and its ketone counterpart..



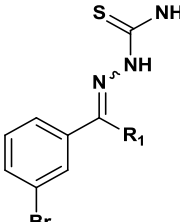
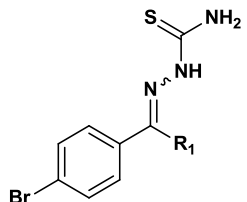
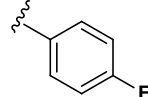
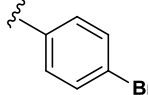
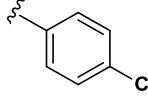
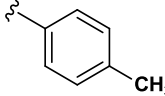
(1) IC₅₀: 16.7 nM



(68) ≥ 10000

Figure 29. Comparison between Inhibitory activities of **1** and **68**

Table 52. Comparison of the Inhibitory Activity between *meta*- and *para*-Bromine-substituted Benzophenone Thiosemicarbazones

<i>R</i> ₁		
	(17) 79.6	(48) 3320
	(21) 327.1	(47) ≥ 10000
	(18) ≥ 10000	(49) ≥ 10000
	(19) 2156.0	(50) 4570

A series of non benzophenone thiosemicarbazone analogs was explored (Table 36). The series includes fluorenes, naphthalenes, indenes, annulenes, chromanones, aminoquinolines, and others. The majority of these derivatives were inactive compounds

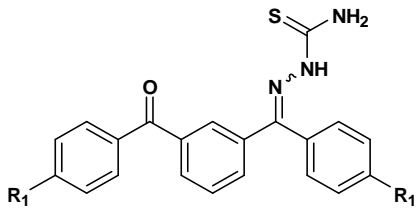
and the IC₅₀ values were not be determined. Only **88** and **92** were capable to inhibit cathepsin L.

Thiochromanones (Table 42) and sulfones (Table 43) are important subclasses of thiosemicarbazones that were screened in this library. Their IC₅₀ values ranged between 46.5 and 10000 nM. In general, sulfides showed better inhibitory activities when compared with sulfones. Small size substitution in the thiochromanone moiety in various positions series reveals that thiochromanone TSCs become better inhibitors with nitro-aromatic or *poly*-fluorinated groups. *Mono*-halogenation, on the other hand, does not improve their activity yielding modest activities. Once again, the position of the substituent plays a critical role in the activity of thiosemicarbazones. A comparison of two sets of halogenated sulfones and thiochromanones were compared. First, 6,7-difluorothiochromanone TSC (**119**, IC₅₀: 54 nM) is 200 times more efficient than its analog, a 6,8-*difluorothiochromanone* TSC (**120** IC₅₀: 1500 nM). Additionally, 6,7-*difluorosulfone* TSC TSC (**144**, IC₅₀: 3650 nM) is 3 times more potent than 6,8-*difluorosulfone* TSC (**146**, IC₅₀ ≥ 10000 nM). These observations also supported the effect of the substituent's position. Similarly, 6-fluoro-8-bromothiochromanone TSC (**124**) and 6-bromo-8-fluorothiochromanone TSC (**122**), two isomers, showed inhibitory activities that were ≥ 10000 and 434.2 nM, respectively.

Lastly, the recent discovery of a new potent thiosemicarbazone with an IC₅₀ value of 9.85 nM (compound **156**), led to the synthesis of a new generation of thiosemicarbazones (unpublished results). Chemically, compound **156** is a benzoyl benzophenone thiosemicarbazone. The small subclass of TSCs that was screened is composed of thirteen analogs with similar structures. Unsubstituted and *para*-substituted

analogs were also examined (Table 53). The unsubstituted analog showed an extraordinary potency with an IC_{50} less than 10 nM, followed by a difluorobenzoyl benzophenone TSC and diacetate analog (24.3 and 587 nM). Interestingly, the dibromobenzoyl benzophenone analog did not prove to be a good inhibitor for cathepsin L. The presence of two thiosemicarbazones moieties did not prove to be beneficial for the activity of these synthetic compounds.

Table 53. Inhibition of Human Cathepsin L by Difunctionalized Benzoyl Benzophenone Thiosemicarbazones

	R_1				
	H (156)	F (157)	OCH ₃ (158)	OCH(CH ₃) ₂ (162)	Br (159)
IC_{50} (nM)	9.9	24.3	587	≥ 10000	

Finally, a polyhalogenated benzoyl benzophenone thiosemicarbazone was one of the last compounds to be screened. Compound **168** showed an extraordinary potency of 8.1 nM, becoming the most potent cathepsin L inhibitor found in this library of thiosemicarbazone analogs.

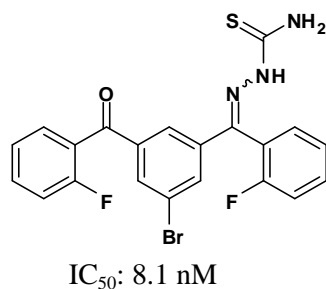


Figure 30. Chemical Structure of **168**

Advanced Kinetic Studies

Two compounds, **1** and **8** proved to be potent and good cathepsin L inhibitors (IC_{50} values are 16.7 and 188.7 nM respectively) and were further characterized in order to understand their mechanism of action. Additionally, **8** offers the possibility of a derivatization that could increase the solubility of the thiosemicarbazone. Both compounds are structurally related. Their scaffold consists of a *para* brominated benzophenone thiosemicarbazone. However, the major difference between **1** and **8** is the presence of different *para*substituents (**1**: -Br and **8**: -OH) in the second (B) phenyl ring. A more detailed analysis was carried out using the phenolic analog (**8**).



Figure 31. Chemical Structures of **1** and **8**

The characterization of compounds that could be used as potential therapeutic agents involved numerous evaluations. Understanding the kinetic effect of these inhibitors helped to identify their mechanism of actions. Fluorometric based assays were utilized to study various assay parameters inhibition activities, determination of K_I , and reversibility of thiosemicarbazone inhibitors, mechanism of inhibition and mode of inhibition.

*Kinetic Analysis of 3-Bromo-3'-Hydroxybenzophenone Thiosemicarbazone (8)
as a Cathepsin L Inhibitor*

Effect of Inhibitor Concentration on Cathepsin L Progress Curves

Cathepsin L (1 nM) was added to six different concentration curves ranging from 0 to 20 μM of compound **8**. Reactions were initiated by the rapid addition of Z-FR-AMC (Final concentrations between 1 and 50 μM). The release of AMC from the nonfluorescent substrate was monitored every 3 seconds for fifty minutes. Figure 32- Figure 36 show the effect of inhibitor concentration on cathepsin L progress curves at different substrate concentrations. Uninhibited and inhibited reactions were monitored for 3000 seconds. The first conclusion, based on visual observations, is the clear inhibition dependence with time. Then, data were fitted to equation 1.7, by nonlinear regression analysis using GraphPad 5.0. P is the concentration of product (μM), v_o and v_s are the initial and steady-state velocities ($\mu\text{M/s}$), t is the time in seconds and k_{obs} the rate constant for conversion of the initial velocity v_o to the steady state velocity v_s . The rate constant (k_{obs}) units are given in s^{-1} . The equation was entered into the computational software, knowing that P and t are the dependent and independent variables, while keeping the velocities and the rate constant as unknowns. For each case, the constraints for their calculation were to give positive values (i.e. $k_{\text{obs}} \geq 0$). It is also worth noting that equation 1.7 is only valid when substrate depletion is insignificant.¹⁸⁸ Therefore, some points were excluded in every case for data fitting. Velocities, rates, r^2 and points analyzed for each substrate concentration are shown in Tables 54-58.

$$P = v_s t \frac{(v_o - v_s)}{k_{\text{obs}}} (1 - e^{-k_{\text{obs}} t}) \quad (1.7)$$

Table 54. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin L Progress Curves with 50 μM Z-FR-AMC

[I] (μM)	20	10	5	1	0.5
v_s ($\mu\text{M/s}$)	0.0007067	0.001048	0.001150	0.001340	0.001376
v_o ($\mu\text{M/s}$)	0.4482	~ 12.38	~ 6.261	~ 5.213	~ 5.182
k_{obs} (s^{-1})	0.5569	~ 17.64	~ 9.394	~ 8.232	~ 8.597
r^2	0.9904	0.9988	0.9990	0.9988	0.9987
Points analyzed	1000	1000	1000	1000	1000

Table 55. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin L Progress Curves with 25 μM Z-FR-AMC

[I] (μM)	20	10	5	1	0.5
v_s ($\mu\text{M/s}$)	0.0003545	0.0008308	0.0009205	0.001256	0.001283
v_o ($\mu\text{M/s}$)	0.002994	0.2028	~ 0.2786	~ 2.064	~ 2.577
k_{obs} (s^{-1})	0.003522	0.4363	~ 0.6231	~ 6.293	~ 7.384
r^2	0.9734	0.9926	0.9953	0.9994	0.9993
Points analyzed	1000	1000	1000	1000	1000

Table 56. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin L Progress Curves with 10 μM Z-FR-AMC

[I] (μM)	20	10	5	1	0.5
v_s ($\mu\text{M/s}$)	8.407e-005	0.0001831	0.0002328	2.674e-014	0.001182
v_o ($\mu\text{M/s}$)	0.001703	0.001567	0.001577	0.001591	~ 0.3222
k_{obs} (s^{-1})	0.002457	0.001048	0.0008123	0.0002494	~ 1.800
r^2	0.9793	0.9956	0.9967	0.9982	0.9972
Points analyzed	1000	1000	1000	1000	1000

Table 57. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin L Progress Curves with 5 μM Z-FR-AMC

[I] (μM)	20	10	5	1	0.5
v_s ($\mu\text{M/s}$)	2.732e-005	8.012e-006	4.627e-005	$\sim 1.888\text{e-}16$	2.040e-15
v_o ($\mu\text{M/s}$)	0.001111	0.001048	0.0009878	0.001053	0.001114
k_{obs} (s^{-1})	0.003824	0.001566	0.0009347	0.0004653	0.0003643
r^2	0.9617	0.9898	0.9953	0.9976	0.9989
Points analyzed	667	667	667	667	1000

Table 58. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin L Progress Curves with 1 μM Z-FR-AMC

[I] (μM)	20	10	5	1	0.5
v_s ($\mu\text{M/s}$)	1.327e-005	2.622e-005	4.346e-005	8.961e-5	3.944e-5
v_o ($\mu\text{M/s}$)	0.0006334	0.0004182	0.0004304	0.0004607	0.0003939
k_{obs} (s^{-1})	0.01499	0.005331	0.003741	0.002667	0.001218
r^2	0.9273	0.9816	0.9907	0.9965	0.9974
Points analyzed	402	402	402	402	402

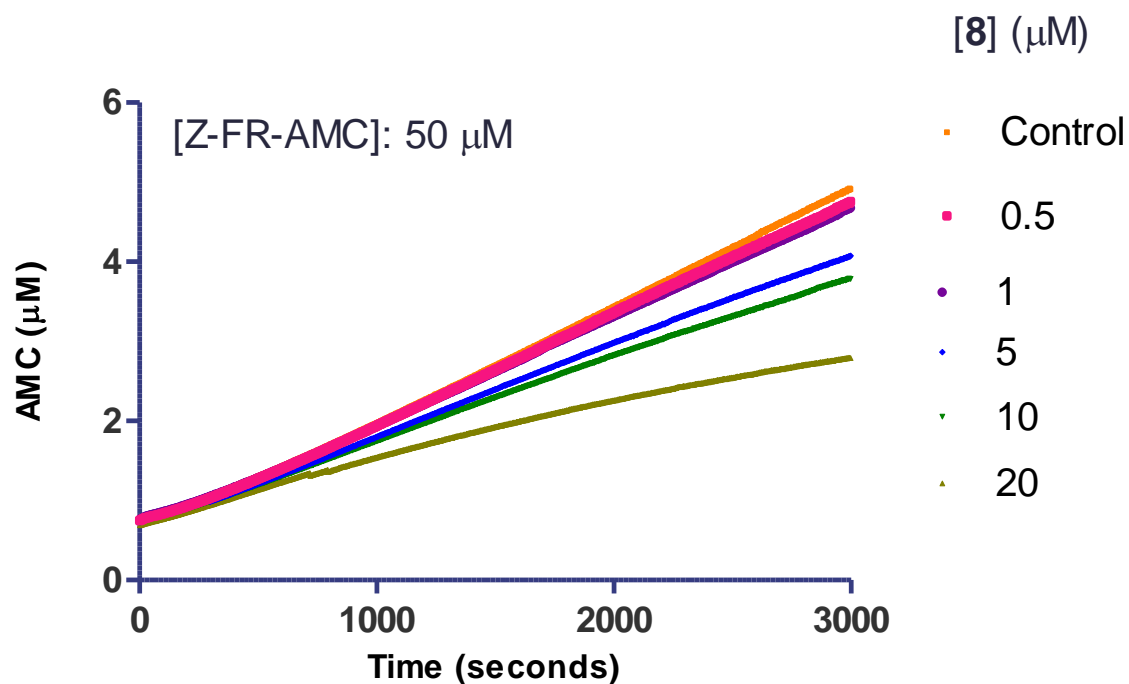


Figure 32. Cathepsin L Progress Curves with 8 Using 50 μM Z-FR-AMC

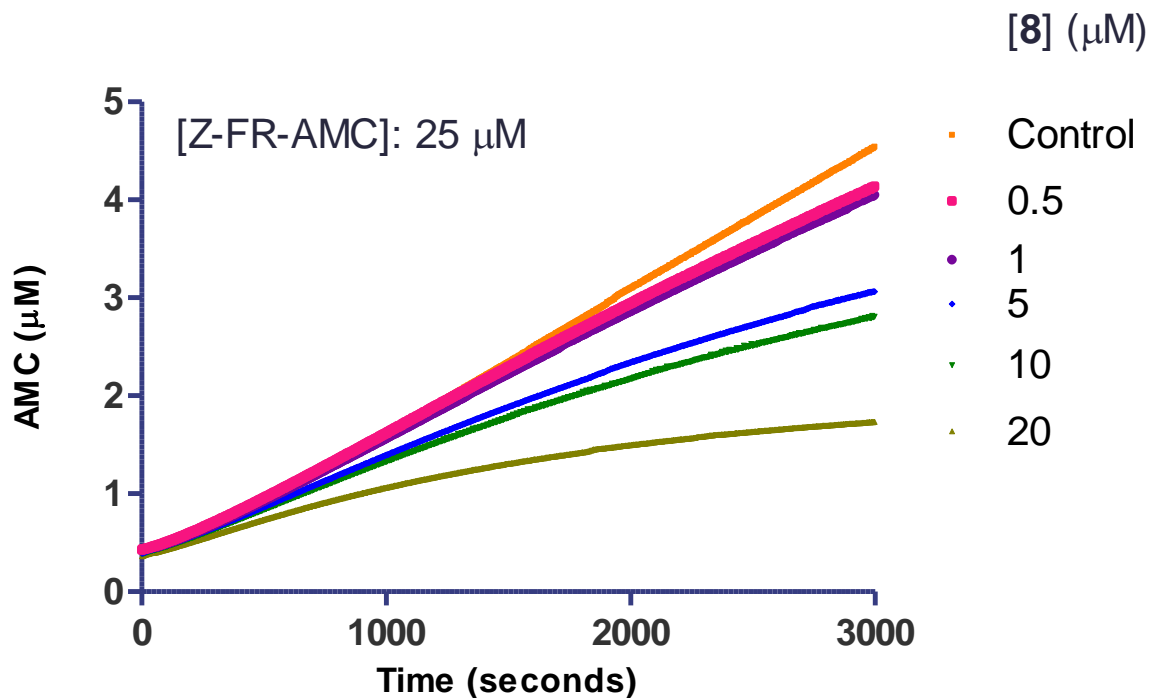


Figure 33. Cathepsin L Progress Curves with **8** Using 25 μM Z-FR-AMC

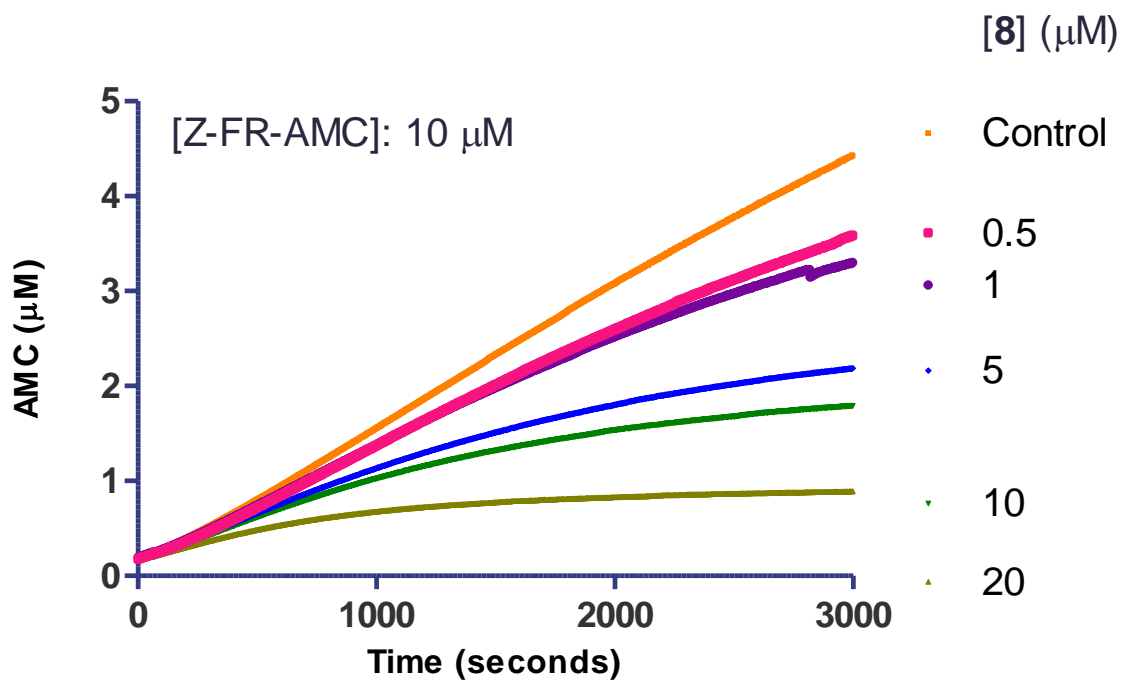


Figure 34. Cathepsin L Progress Curves with **8** Using 10 μM Z-FR-AMC

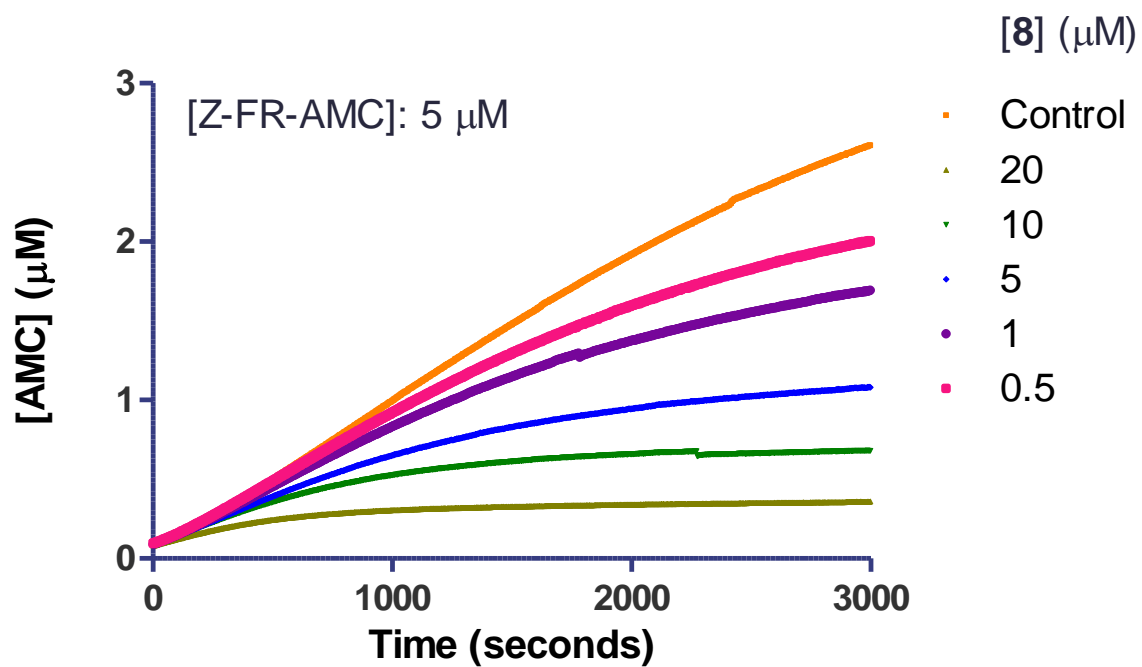


Figure 35. Cathepsin L Progress Curves with **8** Using 5 μM Z-FR-AMC

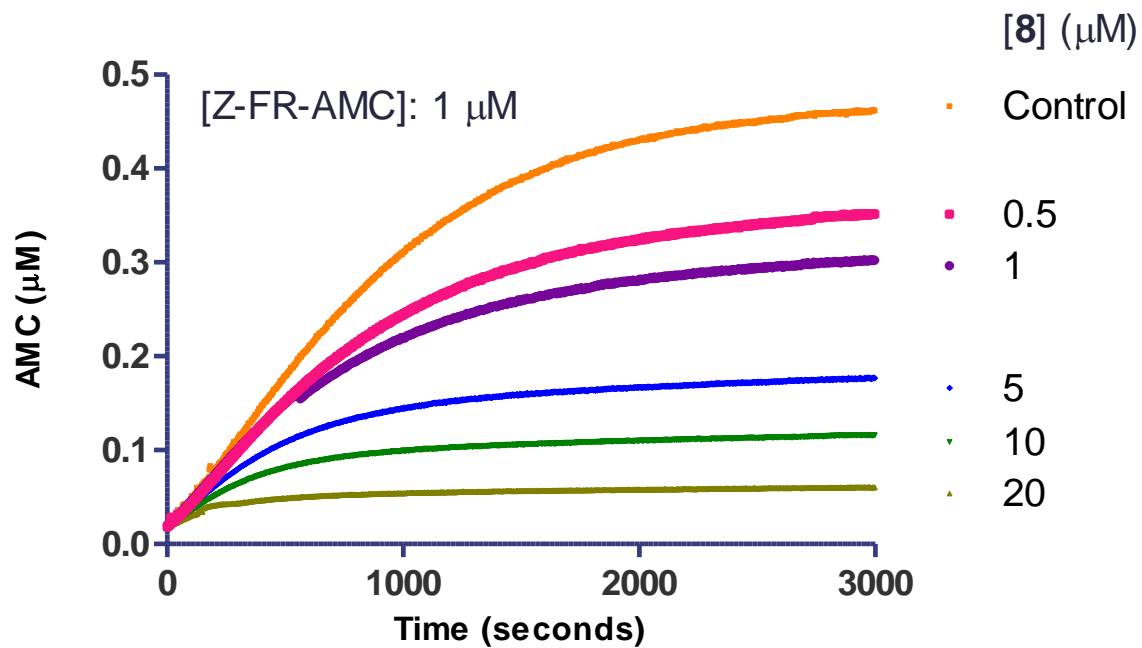


Figure 36. Cathepsin L Progress Curves with **8** Using 1 μM Z-FR-AMC

Approximate values for the rate constants, and velocities were obtained for every substrate concentration (25 and 50 μM). Data were fitted into Equation 1.7, but due to the elevated amount of substrate present in solution, compared to the concentration of the phenolic analog, the results were not reasonable. A better fit was acquired with lower concentrations of Z-FR-AMC (1-10 μM). We decided to investigate the mechanism of inhibition of **8** as a slow binding inhibitor of cathepsin L, due to the clear time-dependence of the progress curves. There are two possible mechanisms that are known for slow binding inhibitors. Figure 37 shows the two proposed mechanisms that could be observed using the phenolic bromobenzophenone thiosemicarbazone (3,3'-Br-PhOH-TSC) as a slow-binding cathepsin L inhibitor. Mechanism A summarizes a simple reversible inhibition with k_{on} and k_{off} values relatively small. The parameters k_{on} and k_{off} are the rate constants for the formation and dissociation of the cathepsin L-3,3'-Br-PhOH-TSC complex. Mechanism B offers an extra step which is more complicated to monitor or verify. This is a more general approach where it is assumed the cathepsin L -3,3'-Br-PhOH-TSC complex undergoes an auto-isomerization, or a possible covalent modification of the enzyme due to the presence of the inhibitor.

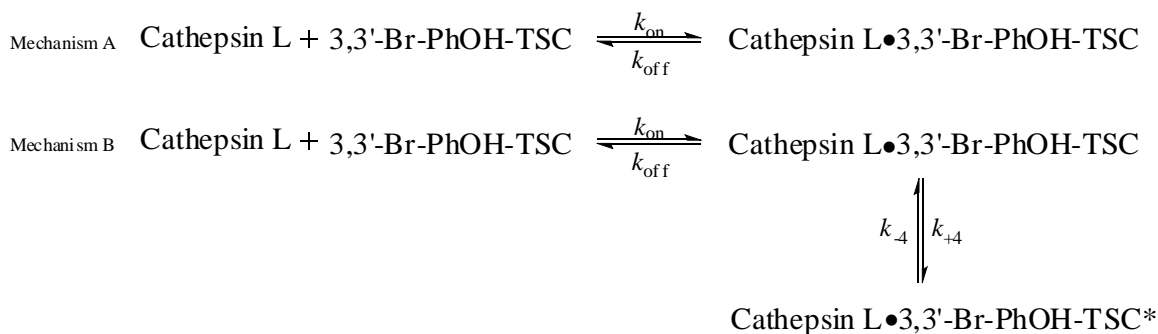


Figure 37. Possible Mechanisms of Inhibition of Human Cathepsin L by **8**

Initial velocities and k_{obs} values were plotted versus inhibitor concentration. The first plot (v_o vs [I]) is utilized to determine the mechanism of inhibition that compound **8** follows under *in vitro* conditions. Figure 38 and Figure 39 show the v_o vs [I] and k_{obs} vs [I] plots. Initial velocities of the reactions with **8** at different substrate concentrations (1-10 μM) are the same as the reactions without **8**. The independence of v_o with respect to the inhibitor concentration is a clear indication that a simple reversible inhibition (Mechanism A, Figure 37) is carried out for slow-binding inhibitors.

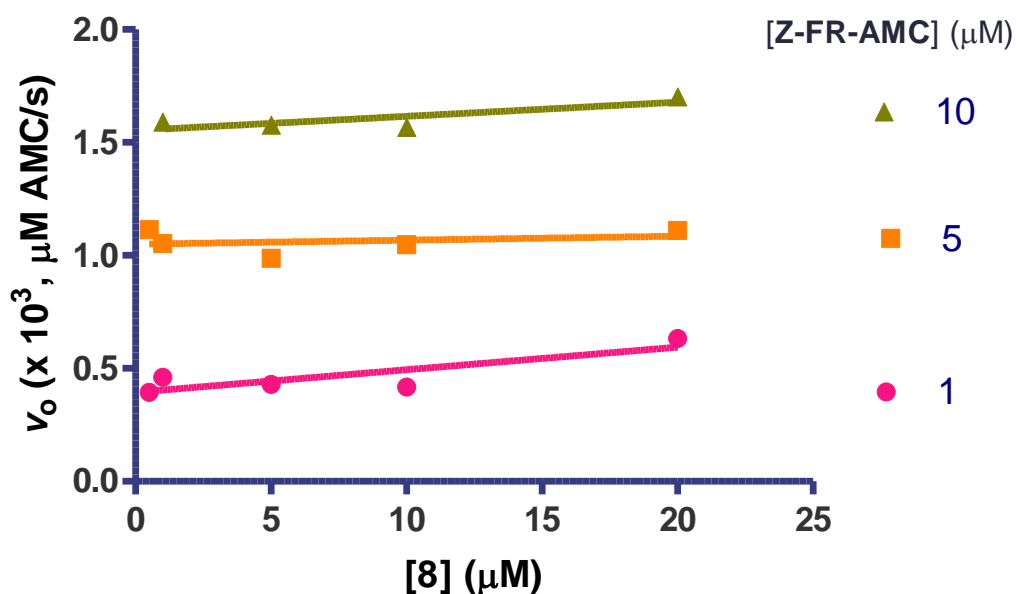


Figure 38. Calculated v_o from Eq. 1.7 Cathepsin L Progress Curves with **8**

Figure 39 shows the linear trend between k_{obs} and [I]. This figure would confirm that a simple reversible mechanism can be observed between cathepsin L and **8**. Interestingly, many groups have reported similar plots, but their final remarks suggest a covalent modification.^{328,329} For example, Hang investigated the behavior of non-nucleoside inhibitors of hepatitis C virus (HCV) polymerase. Their results also confirmed the behavior of these compounds as slow binding inhibitors.

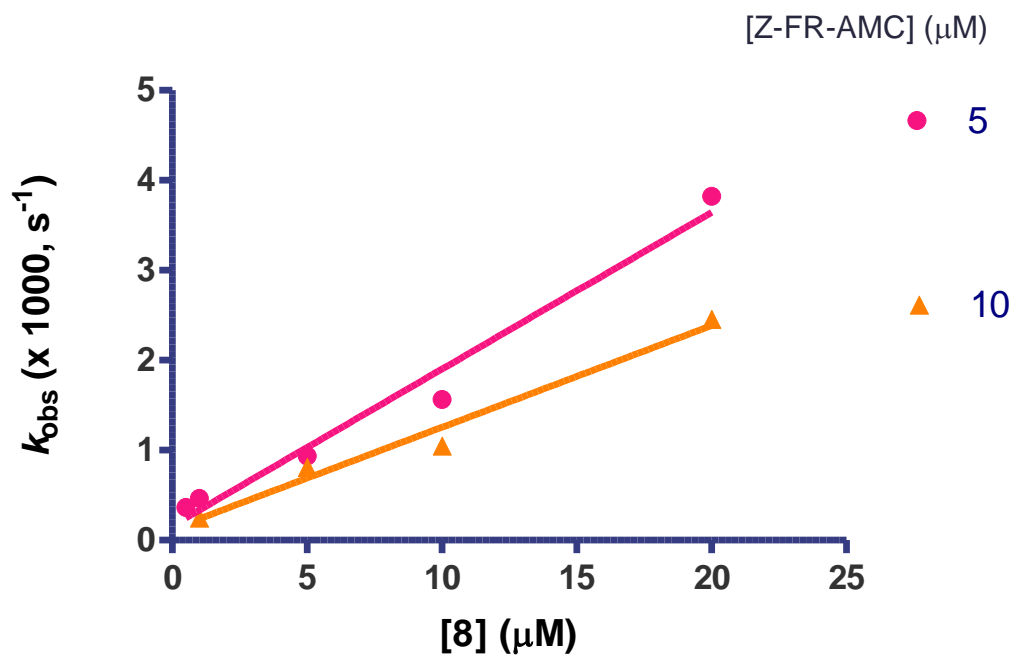


Figure 39. Calculated k_{obs} from Eq. 1.7 Cathepsin L Progress Curves with **8**

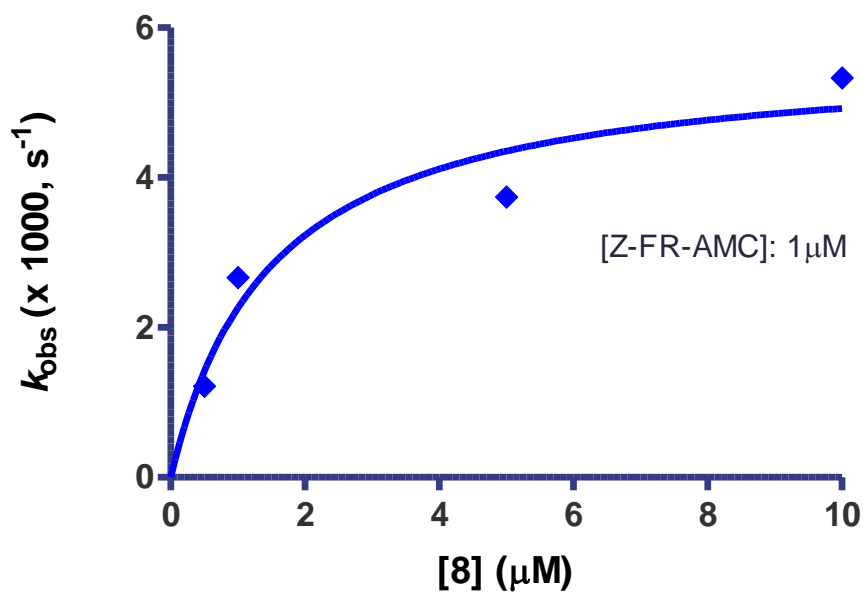


Figure 40. Calculated k_{obs} from Eq. 1.7 Cathepsin L Progress Curves with **8** using 1 μM Z-FR-AMC

Linear regression analysis was performed using GraphPad 5.0 in order to analyze the data (5 and 10 μM Z-FR-AMC). Results can be seen in Table 55. The k_{on} and k_{off}

values can be then calculated using equations 2.1 and 2.2.³²⁸ However, a closer examination of the data showed that lower concentrations ([Z-FR-AMC]: 1 μM) there was a hyperbolic relation between k_{obs} and [I] (Figure 40). These results indicated that, instead, a possible covalent modification might occur, but further experiments will need to be performed to validate this statement. Non-linear regression analysis was performed using GraphPad 5.0 in order to analyze the data (1 μM Z-FR-AMC). These results indicate that the mechanism of inhibition for slow binding inhibitors (i.e. simple reversibility, isomerization, or covalent modification) is substrate dependent. Higher concentrations of substrate force the inhibition to a simple mechanism. In contrast, the inhibitor might form a complex species with the enzyme when lower concentrations of substrate are present.

$$k_{\text{obs}} = k_{\text{on}}^{\text{app}} [I] + k_{\text{off}} \quad (2.1)$$

$$k_{\text{obs}} = \frac{k_{\text{on}} [I]}{\left(1 + \frac{[S]}{K_M}\right)} + k_{\text{off}} \quad (2.2)$$

Table 59. Calculated k_{on} and k_{off} for Selected Substrate Concentrations

[Z-FR-AMC] (μM)	10	5
$k_{\text{on}}^{\text{app}} (\mu\text{M}\cdot\text{s})^{-1}$	0.17	0.11
$k_{\text{on}} (\mu\text{M}\cdot\text{s})^{-1}$	0.96	1.14
$k_{\text{off}} (\text{s}^{-1})$	0.16	0.12
r^2	0.9770	0.9761

The results that are summarized in table 55 also confirm the slow release of the inhibitor from the cathepsin L:3,3'-Br-PhOH-TSC complex. k_{on} rates are 6 fold higher

than k_{off} kinetic constants. The k_{on} and k_{off} values indicate the dissociation of the enzyme-inhibitor complex happens at a much slower rate than the formation of the species.

Effect of Preincubation Studies on Cathepsin L Inhibition Assays Using 8

Progress curves provided strong evidence that **8** is a slow-binding inhibitor. Therefore, we explored the influence of preincubation time on the inhibitory potency of thiosemicarbazones and their IC_{50} values. It is often found that reported inhibitory assays differ in this specific parameter in the literature literature. Therefore, it is quite difficult to compare IC_{50} values because it is an assay dependent result. Compound **8** IC_{50} values were determined at seven different preincubation times ranging between 0 and 240 minutes. Inhibitor final concentration varied between 0 and 10 μ M. Figure 41 shows the results of the studies.

Table 60. Effect of Preincubation Time on IC_{50} Values Using **8**

<i>Pre-incubation times</i> (minutes)	<i>$IC_{50} \pm$ Standard Error (nM)</i>
240	46.97 ± 1.99
120	79.72 ± 2.03
60	96.84 ± 4.17
30	131.2 ± 2.05
5	188.7 ± 18.7
1	6081 ± 334.3
0	$\geq 10,000$

The effect of preincubation on the potency of **8** as seen in Table 60 is quite evident. The phenolic bromobenzophenone thiosemicarbazone (**8**) showed no inhibitory activity when cathepsin L was not in preincubated with the compound. Compound **8** has poor activity with one minute preincubation time. However, the trend changed dramatically when the preincubation time was increased to 5 minutes. The inhibitory

activity of **8** increased 32-fold with a remarkable IC_{50} less than 200 nM (188.7 nM). The potency of **8** modestly increased with longer preincubation times. Finally, the best potency activity was found when **8** was preincubated for four hours with cathepsin L, the activity of the phenolic analog increased 4-fold to give a value of 47 nM. These results confirmed the strong dependence of IC_{50} value determination with respect to the preincubation time parameter. Controls (i.e. uninhibited cathepsin L) were monitored at every preincubation time. There was no significant loss of catalytic activity at longer preincubation times.

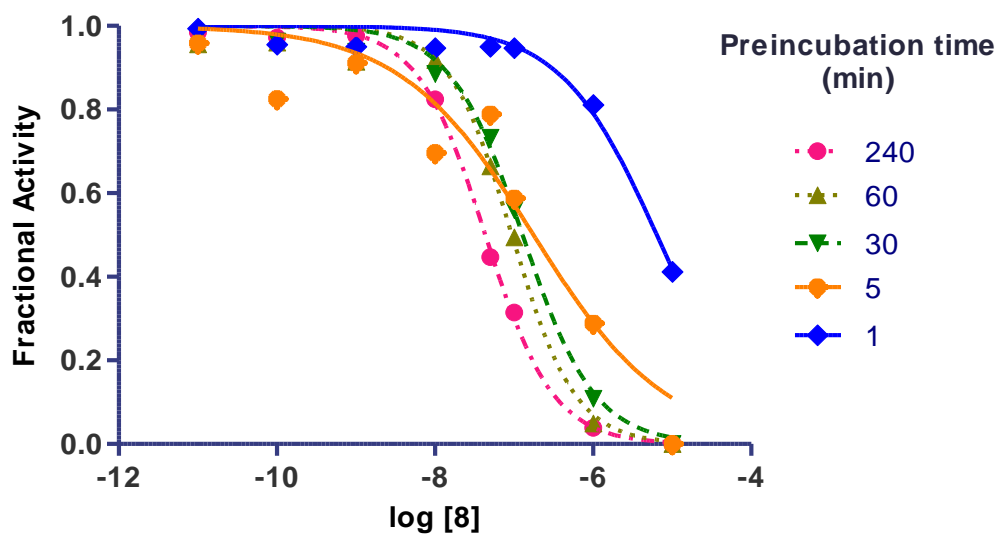


Figure 41. Effect of Preincubation Time on IC_{50} Values of **8** against Cathepsin L

Determination of K_i^{app} Using Morrison's Quadratic Equation. Effect of Preincubation Time Using 8.

The data obtained from the preincubation was further analyzed. The possibility that **8** was a tight-binding inhibitor was analyzed with Morrison's quadratic equation. (See equation 1.11).

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

The rates v_i and v_o are the inhibited and uninhibited cathepsin L velocities (RFU/s); $[E]_T$ (nM) is, the total concentration of enzyme found in solution (free enzyme and inhibitor-enzyme complex); $[I]_T$ (nM) is the total concentration of inhibitor present in solution (free inhibitor and inhibitor-enzyme complex); and K_I (nM) is the inhibition constant, often referred as the dissociation constant. The equation may be solved to give two possible answers. However, the equation is written so that there is only one possible answer that fits physiological conditions (i.e. $K_I^{app} > 0$). GraphPad 5.0 was used to fit the data after data manipulation for every preincubation time point. Inhibitor, substrate, and enzyme concentrations, as well as K_M , K_I^{app} , were all in micromolar units (μ M). The residual activity (or v_i/v_o) was normalized to 1 (i.e. v_o : 1 and $0 \leq v_i \leq 1$). Normalized residual activity and $[I]$ were defined as the dependent and independent variables respectively. Nonlinear regression was applied using the following conditions: $[S]$: 50 μ M, K_M : 1.1 μ M, $[E]_T$: 0.001 μ M and v_o : 1. Normalization and nonlinear conditions were used to three or four independent experiments per preincubation time. Average and standard errors can be seen in Table 57 and Figure 57.

Table 61. Effect of Preincubation Time in K_I^{app} Values of **8** against Cathepsin L

<i>Pre-incubation times</i> (minutes)	$K_I^{\text{app}} \pm \text{Standard Error (nM)}$
1	121.5 \pm 8.1
5	3.7 \pm 0.3
30	2.7 \pm 0.2
60	2.0 \pm 0.1
120	1.6 \pm 0.1
240	1.1 \pm 0.07

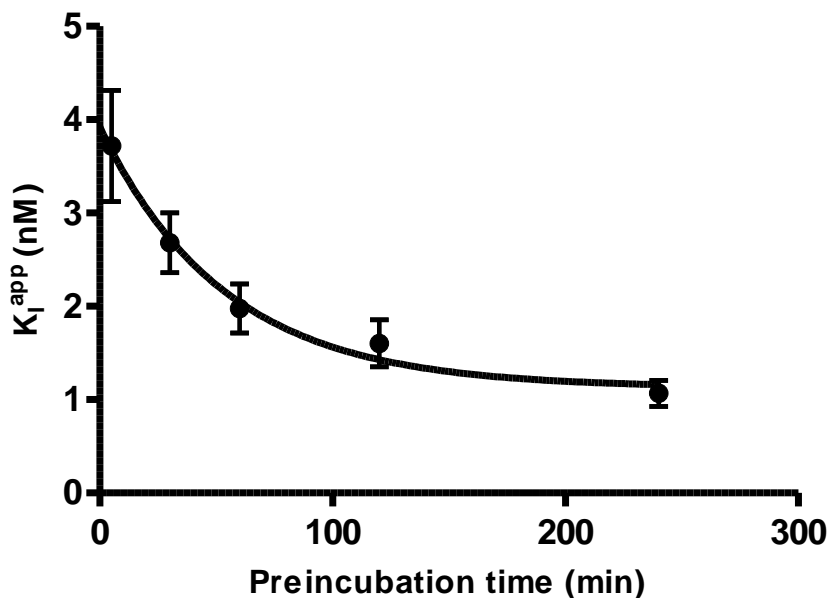


Figure 42. Effect of Preincubation Time in K_I^{app} Values of **8** against Cathepsin L

The K_I^{app} values are shown in Table 61. It is apparent that the inhibition constants are also time-dependent. One minute of preincubation shows a good inhibition constant with a value of 121.5 nM. Furthermore, the potency is extremely enhanced (more than 30-fold) for the calculated value at standard conditions (preincubation time: 5 minutes). The best inhibition constant value was obtained at the longest preincubation time (4 hours). The apparent K_I was 120-fold better when compared with the corresponding value at one minute preincubation time ($K_I^{\text{app}} = 1.1$ nM)

Cathepsin L Reversibility Studies

Compound **8** was found to be a time-dependent inhibitor in the preincubation studies. Therefore, we decided to explore if this specific compound was a reversible inhibitor of cathepsin L. A mixture containing 100 X cathepsin L and 10 X $IC_{50}(\text{preincubation time: 4 hours})$, which are 100 and 470 nM respectively, were incubated at 25 °C for four hours. The inhibition of cathepsin L by **8** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. Experiment was set up in order to monitor these reactions almost immediately after adding cathepsin L to the assay. Final conditions are 1 nM cathepsin L, 4.7 nM of the phenolic bromobenzophenone thiosemicarbazone (**8**), and 50 μM . Additionally, a control experiment (cathepsin L with DMSO as control vehicle) was also carried out. Figure 38 shows the release of AMC for the first 5000 seconds. However, uninhibited and inhibited reactions were followed for a total time of four hours. Cathepsin L was able to recover its catalytic activity after the rapid dilution with assay buffer containing Z-FR-AMC. Apparent substrate depletion can be observed in the uninhibited reaction after 5000 seconds. Thus, linear regression was applied to the first 5000 seconds of the reactions to determine cathepsin L activity. Cathepsin L activity for both reactions were found to be 1.62 and 0.44 nM AMC/s for control and inhibited reactions. The results showed cathepsin L was recovered up to 30% after 5000 seconds of reaction. Furthermore, a closer inspection of the early points showed that cathepsin L recovered its activity within the first 150 seconds of the reaction up to a 17%. Cathepsin L activity for both reactions were 1.76 and 0.31 nM AMC/s. (See Figure 43 and Figure 44).

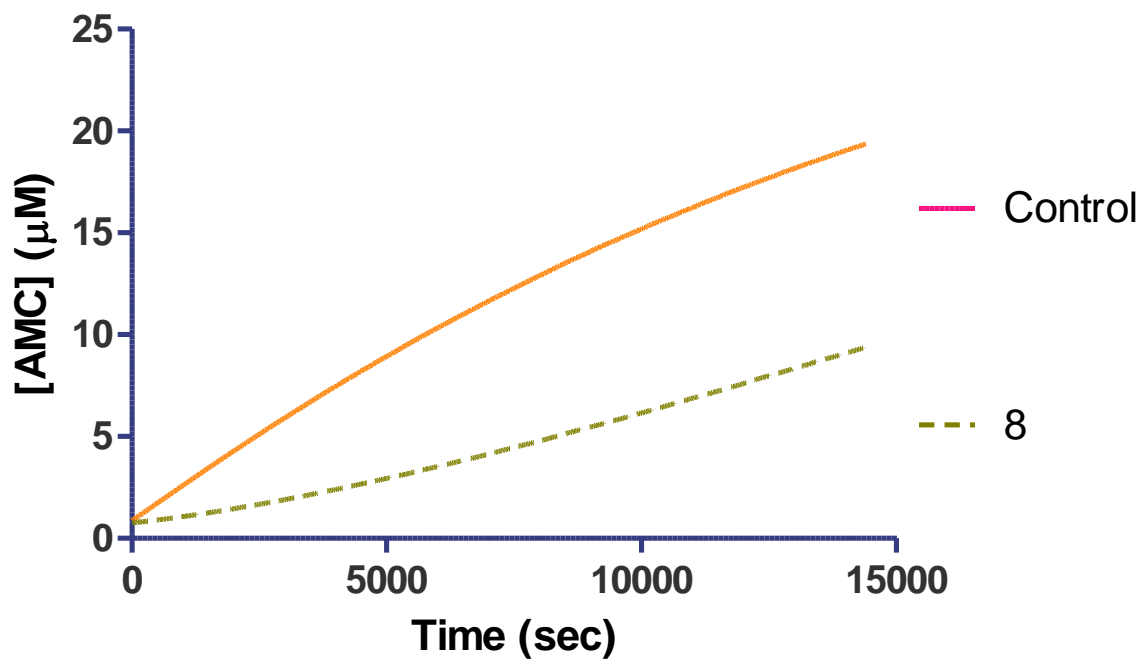


Figure 43. Cathepsin L Reversibility Studies with **8** Using 50 μM Z-FR-AMC

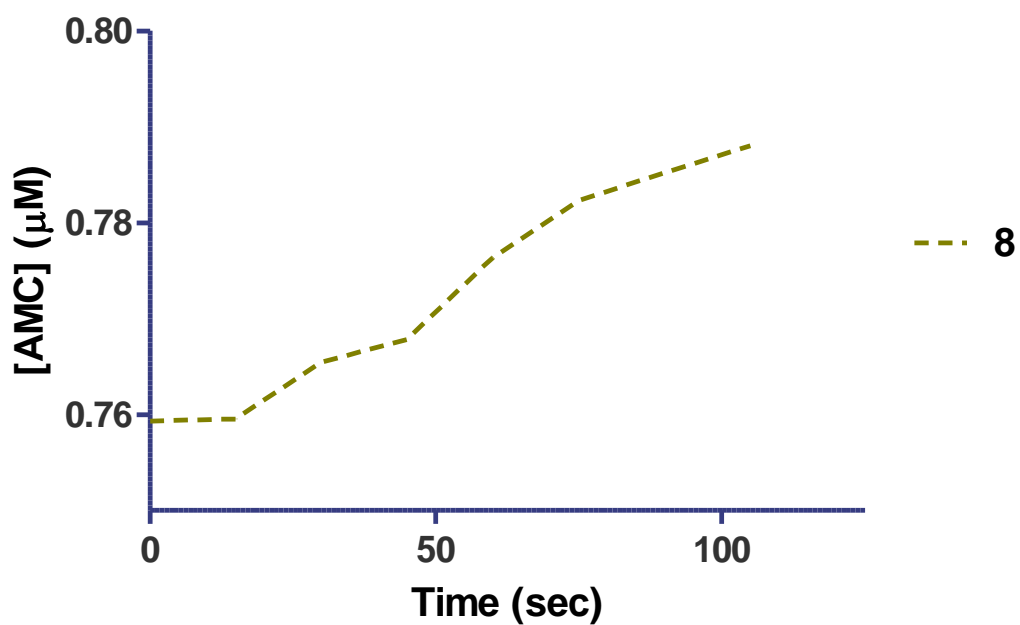


Figure 44. Cathepsin L Reversibility Studies with **8** Using 50 μM Z-FR-AMC

Effect of Substrate Concentration (Z-FR-AMC) on IC₅₀ Values

We finally tried to determine the mode of inhibition of thiosemicarbazones as slow-binding inhibitors of cathepsin L. Therefore, we investigated the effect of substrate concentration on IC₅₀ values.¹⁷⁷ Classical methods for the determination of mode of inhibition (steady-state kinetics) fail when studying a slow, tight-binding inhibitor. In these cases, Copeland offers a suitable alternative for the investigation. IC₅₀ values decrease hyperbolically if the compound is an uncompetitive inhibitor. In the case of competitive inhibitors, IC₅₀ values increase in a linear trend with higher substrate concentrations. Three substrate concentrations (Final concentrations: 50, 10, and 5 μM) were used to determine IC₅₀ values for compound **8**. Cathepsin L and the series of inhibitors were incubated for four hours. Results can be found in table 58 and figure 40.

Table 62. Effect of Substrate Concentration in IC₅₀ Values of **8** against Cathepsin L

<i>[Z-FR-AMC] (μM)</i>	<i>IC₅₀ ± Standard Error (nM)</i>
50	46.97 ± 1.99
10	31.0 ± 2.1
5	28.4 ± 1.1

The effect of substrate concentration was investigated using **8** as a lead compound of the series of thiosemicarbazones in order to investigate its mode of inhibition. A linear behavior can be observed with the values. Lowering the amount of substrate in solution gives a subtle, yet evident increase in the potency of the phenolic analog. According to Copeland, a positive linear behavior (i.e. IC₅₀ ∝ [S]) is an indication the compound acts as a competitive inhibitor; that is, both substrate and the compound compete for the cathepsin L active site. However, the results cannot be taken as conclusive and more experiments were necessary to confirm these findings.

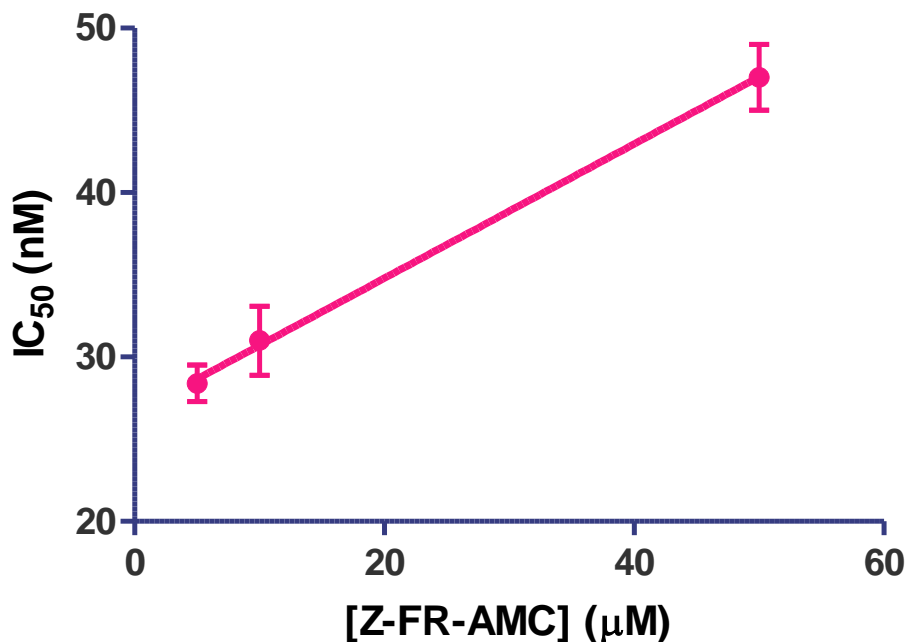


Figure 45. Effect of Substrate Concentration in IC_{50} Values of **8** against Cathepsin L

Effect of Substrate Concentration (Z-FR-AMC) on Cathepsin L Progress Curves

The effect of substrate concentration in IC_{50} values suggested the phenolic bromobenzophenone thiosemicarbazone (**8**) was a competitive inhibitor of cathepsin L. Thus, we decided to explore other alternatives to validate previous experiments. A fixed concentration of **8** was used to monitor the effect of this compound on cathepsin L progress curves when using different substrate concentrations. Then, by using non-linear regression analysis (equation 1.7), the parameters k_{obs} , v_o , and v_s were determined. Compound **8** (0.5 μM) and seven Z-FR-AMC concentrations (ranging from 0.5 and 20 μM) were set up in order to verify the effect of substrate on inhibited cathepsin L progress curves. Reactions were started by the addition of cathepsin L (final concentration: 1 nM). The release of AMC was monitored every five seconds for sixty minutes. Figure 46 shows a typical result of this experiment. The results once again

showed the strong time dependence of **8** when inhibiting cathepsin L. Substrate depletion is negligible for every substrate concentration as seen in Figure 46. Data were fitted into equation 1.7, (see Effect of inhibitor concentration in cathepsin L progress curves). P is the concentration of product (μM), v_i and v_s are the initial and steady-state velocities ($\mu\text{M/s}$), t is the time in seconds, and k_{obs} is the rate constant for conversion of the initial velocity v_o to the steady state velocity v_s . The k_{obs} units are given in s^{-1} . Similar constraints were also given (i.e. $k_{\text{obs}} \geq 0$). Velocities, rates, r^2 and points analyzed for each substrate concentration are shown in tables Table 63, Table 64 and Table 65.

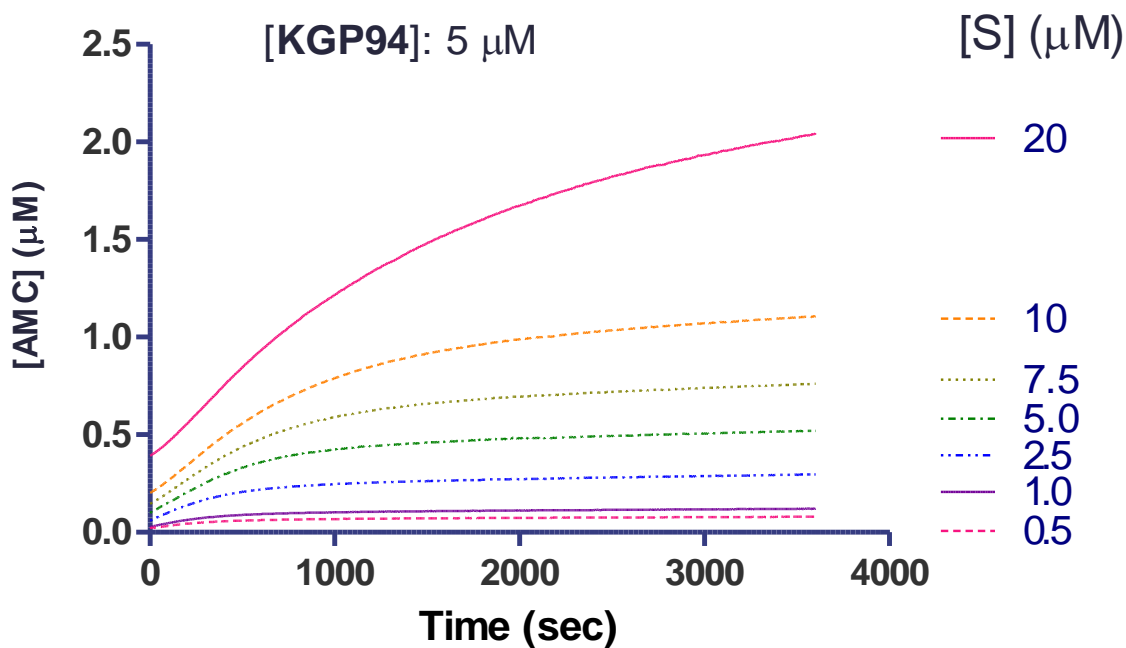


Figure 46. Cathepsin L Progress Curves with $5 \mu\text{M}$ **8** Using Z-FR-AMC

Table 63 shows the average and standard error of calculated k_{obs} as a function of $[S]$. Data were graphed as a function of the unitless substrate/Michaelis-Menten constant $[S]/K_M$, where K_M : $1.1 \mu\text{M}$ as previously determined. The rate constant k_{obs} follows an

inversely hyperbolic trend. The rate constant values increased as the concentration of Z-FR-AMC is decreased in solution. This is a clear indication that **8** is a cathepsin L competitive inhibitor with respect to the fluorogenic substrate. A positive hyperbolic trend can be seen for the initial velocities with higher substrate concentrations. (See Table 63 and Figure 47).

Table 63. Effect of [Z-FR-AMC] in k_{obs} Values When Using **8** against Cathepsin L

[Z-FR-AMC] (μM)	$k_{\text{obs}} \pm \text{Standard Error} (x 10^3, \text{s}^{-1})$
20	2.58 ± 0.31
10	2.15 ± 0.04
7.5	2.50 ± 0.05
5	3.03 ± 0.01
2.5	4.22 ± 0.09
1	5.72 ± 0.18
0.5	6.63 ± 0.37

Table 64. Effect of [Z-FR-AMC] in v_o Values When Using **8** against Cathepsin L

[Z-FR-AMC] (μM)	$v_o \pm \text{Standard Error} (x 10^3, \mu\text{M/s})$
20	2.81 ± 0.13
10	1.96 ± 0.08
7.5	1.77 ± 0.12
5	1.44 ± 0.07
2.5	1.04 ± 0.02
1	0.65 ± 0.09
0.5	0.46 ± 0.07

Table 65. Effect of [Z-FR-AMC] in v_s Values When Using **8** against Cathepsin L

[Z-FR-AMC] (μM)	$v_s \pm \text{Standard Error} (x 10^4, \mu\text{M/s})$
20	2.91 ± 0.11
10	0.90 ± 0.04
7.5	0.60 ± 0.09
5	0.42 ± 0.09
2.5	0.20 ± 0.002
1	0.10 ± 0.01
0.5	0.06 ± 0.008

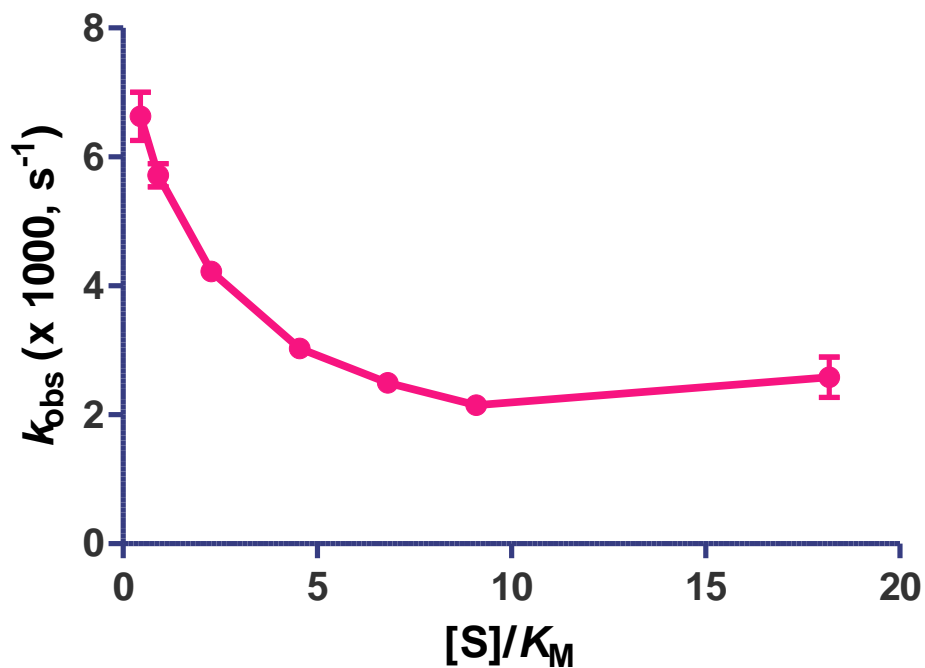


Figure 47. Effect of [Z-FR-AMC] in k_{obs} Values When Using **8** against Cathepsin L

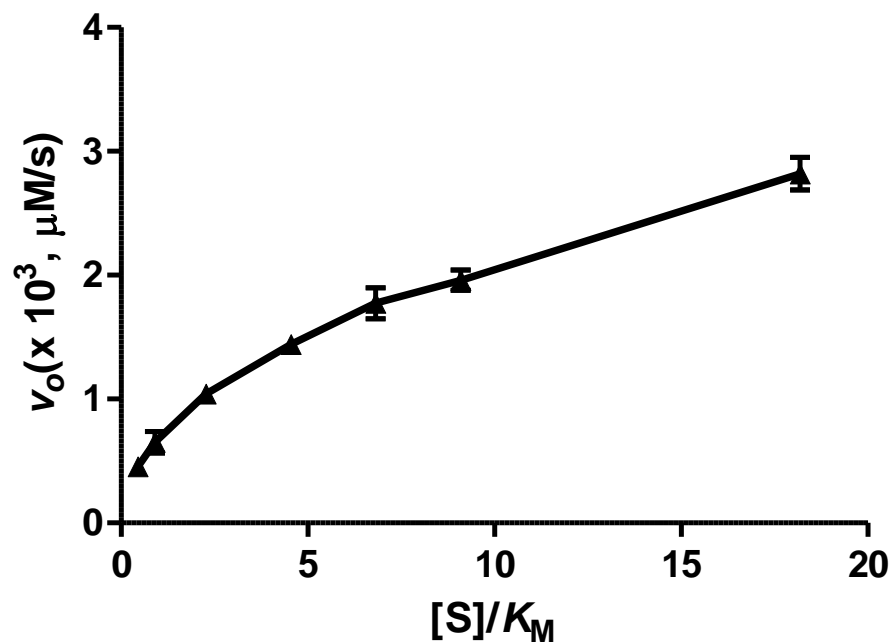


Figure 48. Effect of [Z-FR-AMC] on v_o Values When Using **8** against Cathepsin L

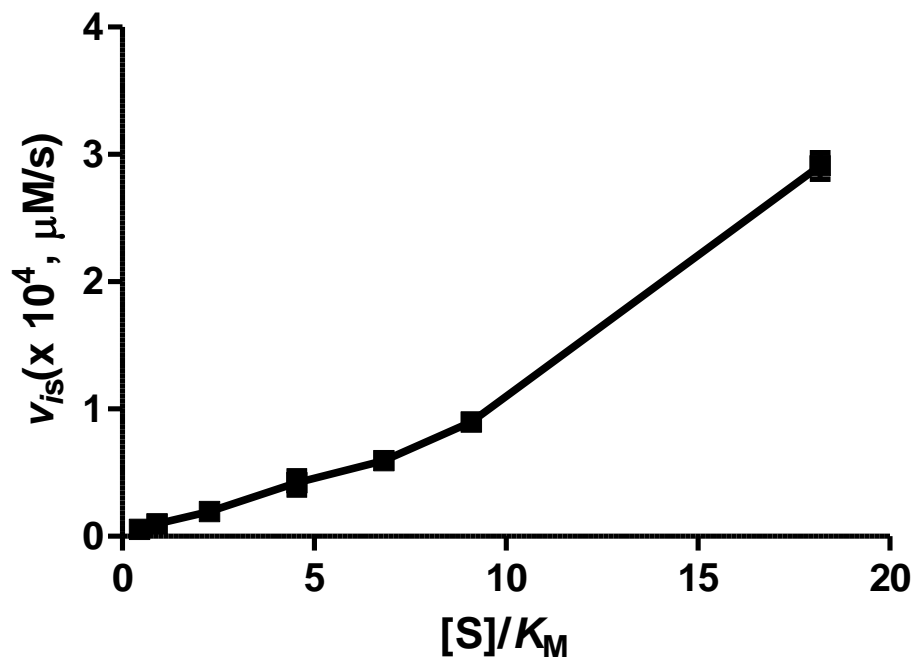


Figure 49. Effect of [Z-FR-AMC] on v_s Values When Using **8** against Cathepsin L

Inhibition of Cathepsin L Collagenase Activity by Thiosemicarbazone Derivatives

Cathepsin L is an extremely powerful hydrolase. It is considered one of the most powerful cathepsin among the cathepsin family which also includes, cathepsin K, B, S, and V. It has also been shown that cathepsin L is also capable of degrading high molecular weight proteins. Cathepsin L plays a vital role in the degradation of the extracellular matrix under normal and pathological conditions. Collagen is a major component of the ECM and a substrate for cathepsin L. Thus, we explored the catalytic activity of cathepsin L by using a natural substrate, type I collage from human skin. We also tested the ability of one of the lead compounds (**8**, IC₅₀: 188.7 nM), to inhibit the collagenase catalytic activity of cathepsin L. Untreated and treated samples were incubated at 37 °C under acidic conditions (pH 5.5). Four sets of samples were prepared for this experiment. Two sets were not treated with **8** (DMSO was used as control vehicle). The third and fourth sets were treated with 50 μM of compound **8** in 2%

DMSO. The effect of preincubation activity on the catalytic activity of cathepsin L and the inhibitory potency of compound **8** was also tested. Thus, two fixed preincubation times (0 and 120 minutes) were chosen to carry out these experiments.

*Inhibition of Cathepsin L Collagenase Activity by Thiosemicarbazone Derivative **8**. (Preincubation Time: 2 Hours).*

A sample containing three microliters of 1.2 μ M cathepsin L in CLI and 2.5 μ l of 260 μ M compound **8** were preincubated as previously described for two hours.

Individual samples were prepared to complete a series of six samples. Then, 7.5 μ l of 0.4 mg/ml type I collagen in acetic acid were added to the cathepsin L-inhibitor mixture. Every sample was carefully mixed and placed in a 37°C water bath. Reactions were monitored between 0 and 20 hours, stopped with 2 μ l LDS sample buffer and heated at 90 °C for every time point. Inactivated samples were immediately stored at -80 °C to preserve them.

Untreated and treated samples were loaded onto 4-12% Bis-Tris SDS gel. A sample of molecular weight standards, one sample with type I collagen only, and one sample of cathepsin L only were also loaded as reference controls. These last two samples were inactivated after two hours of preincubation time. Electrophoresis was performed at 200 V for 60 minutes. The gel was rinsed off with water and placed into a 1X SYPRO® solution, a fluorescent protein staining dye. The gel was stained for a minimum of one hour at room temperature. The destaining process was made by washing the gel once with water and twice with 7.5% acetic acid. Finally, a digital image of the gel was obtained by using a fluorescence scanner imager.

*Inhibition of Cathepsin L Collagenase Activity by Thiosemicarbazone Derivative 8.
(Preincubation Time: 0 hours).*

A sample containing three microliters of 1.2 μ M cathepsin L in CLI , 2.5 μ l of 260 μ M compound **8** and 7.5 μ l of 0.4 mg/ml type I collagen were added to start the reaction. Figure 50 shows a picture of the collagenase inhibition of cathepsin L by **8** when they were preincubated for two hours. Similarly, Figure 51 shows the conditions for a similar experiment when there was no preincubation time.

Figure 50 shows the results of degradation of type I collagen from human skin using human cathepsin L soluble in acidic solutions.³³⁰ The activity of the protease was stopped at different time points (0-20 hours). The figure shows the progress of the reaction for the first six hours after collagen I was added to the cathepsin L-inhibitor mixture that was previously incubated for two hours. The first lane shows the molecular marker (MM), which shows ten different bands ranging from 12 to 225 kDa. The second lane is type I collagen that was not treated with cathepsin L after two hours of preincubation time in the vehicle control (2% DMSO). Reported literature establishes that collagen is large protein that is divided into three defined chains: α , β , and γ heavy chains and is crosslinked and polymerizes into fibrils. The molecular weights of the heavy chains are approximately $80 \leq MW_{\alpha} \leq 125$ kDa for α chain ; $160 \leq MW_{\beta} \leq 250$ kDa for β chain; and $240 \leq MW_{\gamma} \leq 375$ kDa for γ chain.^{331,332} Purity and integrity of the sample are some of the biggest obstacles prior the development of the experiment. Collagen I extractions usually require the use of potent proteolytic enzymes, such as pepsin under acidic conditions, in order to increase the solubility of the macromolecule.³³³ Due to its size (MW: 300 kDa), type I collagen solubility is very poor in polar solvents, such as water.³³⁴ However, solubility of the protein is enhanced if

acidic solvents are used such as acetic acid. Results show the presence of three major bands with high molecular weights that are consistent with the reported values for α , β , and γ heavy chains in the literature. Approximate molecular weights are: $\alpha \approx 102$ kDa, $\beta \approx 150$ kDa and $\gamma \approx 200$ kDa (lane 2). The third and fourth bands of Figure 50 represent untreated and treated samples that were stopped after 20 minutes of the natural substrate. Treated samples contained type I collagen, and cathepsin L in DMSO, while treated samples consisted of type I collagen, cathepsin L, and compound **8** in DMSO. The untreated sample showed a slight degradation process even as little as 20 minutes. The α , β , and γ bands look more defined than their respective control. The degradation is more evident in the case of the α bands (lane 3). The degradation process is evident after 90 and 180 minutes. All of the bands α , β , and γ look well defined and there is evidence that heavy bands compared to untreated collagen at time 0. (lane 2). A comparison between treated (lanes 5 and 7) and untreated (lanes 6 and 8) samples reveals that compound **8** was able to inhibit the collagenase activity of cathepsin L. Longer preincubation times ($t \geq 6$ hours) indicate that uninhibited cathepsin is able to completely degrade 3 μg of type I collagen. However, that is not the case for the treated samples with 50 μM of compound **8**. These results indicate that the phenolic brominated benzophenone thiosemicarbazone is capable to inhibit cathepsin L collagenase activity up to a fifty percent.

Additionally, Figure 51 shows the results of samples containing cathepsin L with **8** (50 μM) and type I collagen that was immediately added without preincubation time. Results are consistent with previous observations where the protease was preincubated with **8** for two hours. Cathepsin L was able to degrade the fibrillar protein in a lesser extent. However, untreated samples were not able to completely degrade three

micrograms at longer preincubation times. A small sample of α and β chains can be seen after 20 hours of reactions, but γ chains were completely degraded. Inhibition can be observed because the protease was not capable to degrade the γ chain. Finally, cathepsin L samples (275 nM) can be seen at the bottom of both gels. The bands showed an apparent molecular weight between 24 and 31 kDa, which is consistent with reported values (MW: 29 kDa).

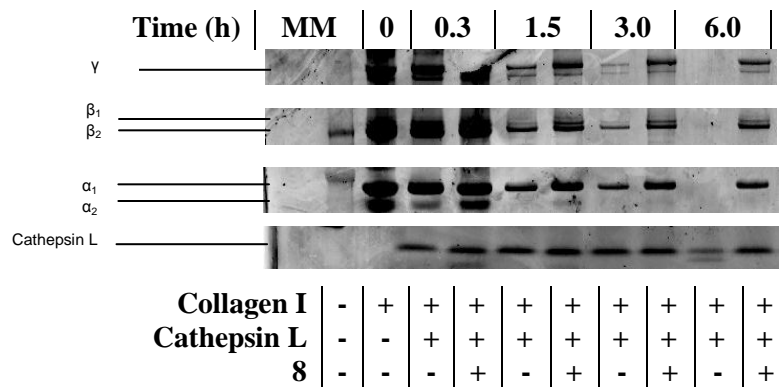


Figure 50. Inhibition of Collagenase Activity of Cathepsin L by **8**, Preincubation Time: 2 Hours.

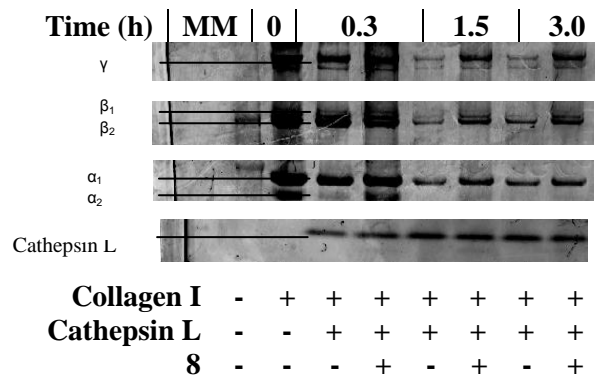


Figure 51. Inhibition of Collagenase Activity of Cathepsin L by **8**, No Preincubation time

Molecular Docking of 8 with Cathepsin L

In a collaborative work with Dr. Shen-En Chen, compound **8** was docked in in cathepsin L using computational software (Insight II, 2005). The crystal structure was originally reported by Guncar and coworkers²²⁸ and the crystal structure was deposited in the protein data bank (PDB ID1ICF).³³⁵ Briefly, the macromolecule was minimized by deleting the original ligand (thyroglobulin type-1). Also, the hydrogen atoms were added to the structure and were set to pH 5.5. The Cys25-His163 thiolate-imidazolium ion pair was generated by changing the setting the hybridization state of the species to sp^2 . Finally, water molecules were added in order to calculate electronic potentials. Compound **8** was also prepared by changing the hybridization states of its nitrogen atoms (sp^2); solvated, and minimized.⁴

Overall, two important hydrogen bonds between the inhibitor and cathepsin L were found. The first hydrogen bond was between the NH terminal of **8** and the carboxamide chain residue of Gln19. The second hydrogen bond was between the side chain of Asp162 and the phenolic moiety of **8**. Gln19 is part of the oxyanion hole, while Asp162 forms a wall on the right side of the S_2 pocket (See Crystal Structure of Cathepsin L, page 48). These two amino acid residues help placing **8** in close proximity to the active site of cathepsin L. Figure 52 shows the orientation of cathepsin L S_1 and S_1' pockets and the relative position of **8** with respect of cathepsin L. Specifically, the S_1' (facing down) was found to interact with the bromophenyl group of **8**. The molecular docking suggested the formation of a transient tetrahedral intermediate (i.e. covalent modification).

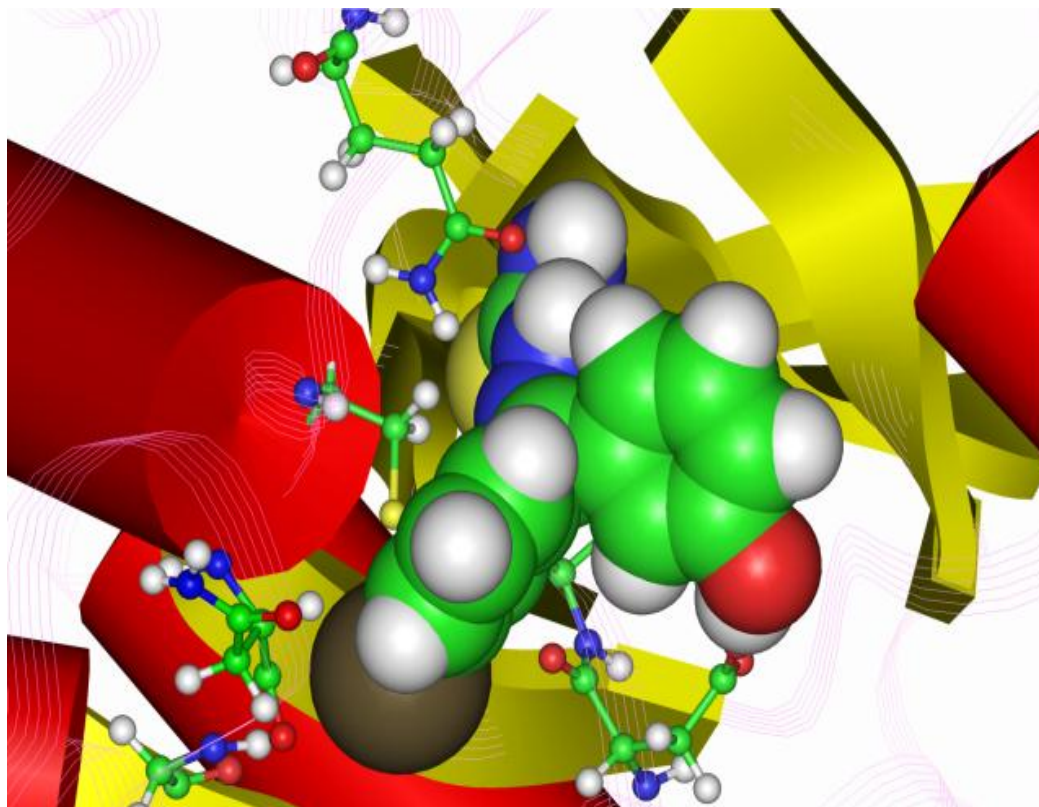


Figure 52. Molecular Modeling of Cathepsin L with **8** (Image courtesy of Dr. Shen-En Chen, Baylor University)

Proposed Mechanism of Inhibition of Cathepsin L by 8

Advanced kinetic analysis and molecular docking of **8** as good cathepsin L inhibitors were used to propose a mechanism of inhibition of the protease by this thiosemicarbazone analog. Figure 53 shows a two dimensional representation of the most important hydrogen bonds between the inhibitor and two residues of the protease (Gln19 and Asp162) and a covalent bond between Cys25 and thiocarbonyl carbon of **8**. The possibility of formation of a thiocarbamoylated enzyme cannot be ruled out.

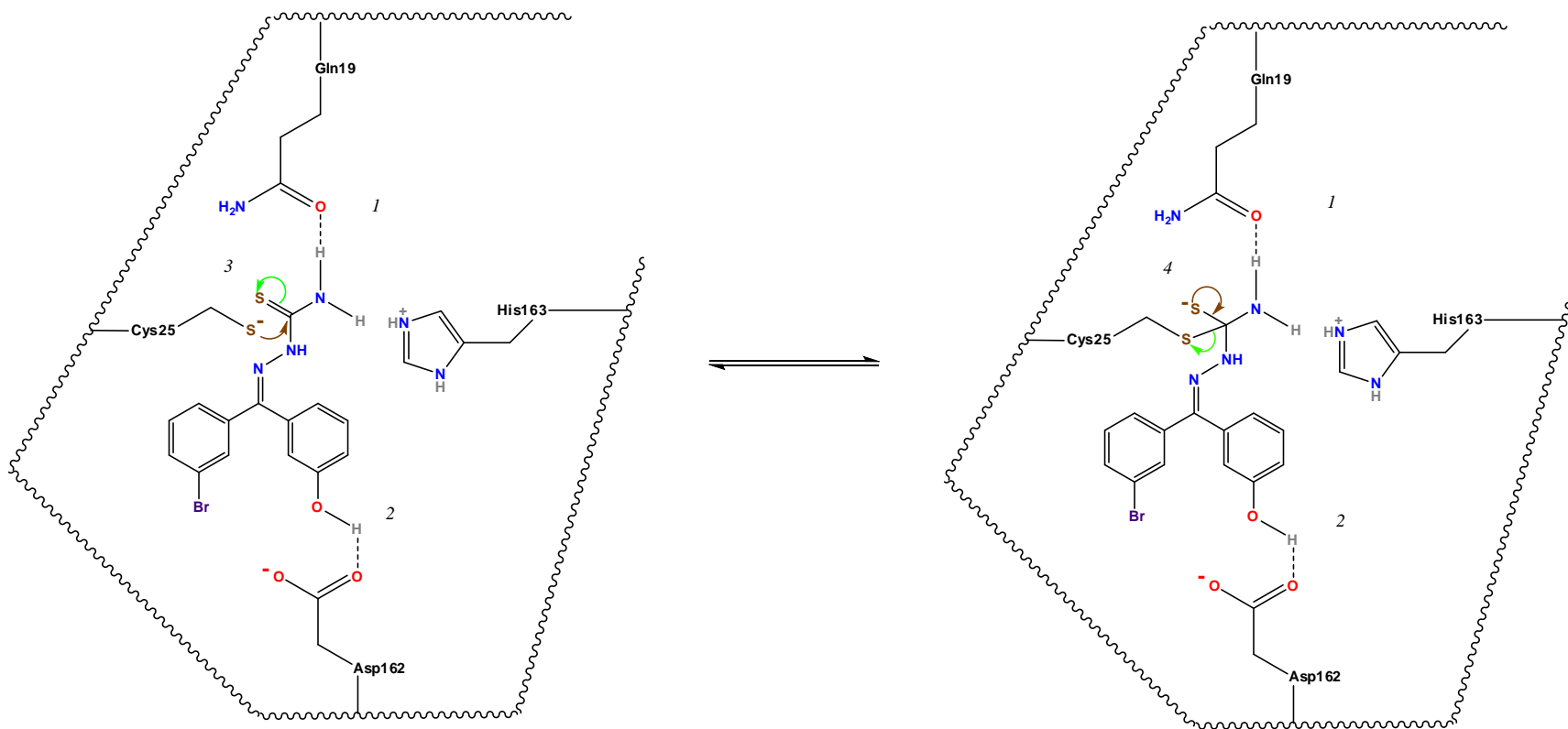


Figure 53. Two-Dimensional Representation of the Proposed Mechanism of Inhibition of Cathepsin L by **8**. 1. Hydrogen Bond between Carbonyl Oxygen of Gln19 Side Chain and KPG94 NH Moiety. 2. Hydrogen Bond between **8** Phenolic Group and Asp162 Carboxylic Acid Side-Chain. 3. The Cys25 Thiolate is in Position to Attack the **8** Thiocarbonyl Carbon Forming a Reversible Transient Covalent Bond (4) and Orients **8** at the Active Site.

*Kinetic Analysis of 3,3'-Dibromobenzophenone Thiosemicarbazone (1)
as a Cathepsin L Inhibitor*

Effect of Inhibitor Concentration on Cathepsin L Progress Curves

Cathepsin L (1 nM) was added to six different concentration curves ranging from 0 to 10 μM of **1**. Reactions were initiated by the rapid addition of Z-FR-AMC (final concentration: 5 μM). The release of AMC from the nonfluorescent substrate was monitored every 3 seconds for fifty minutes. Figure 54 shows the effect of inhibitor concentration on cathepsin L progress curves. Based on visual observations, the inhibition of cathepsin L by **1** is time-dependent. Data were fitted to equation 1.7, by nonlinear regression analysis using GraphPad 5.0. P is the concentration of product (μM), v_i and v_s are the initial and steady-state velocities ($\mu\text{M/s}$), t is the time in seconds and k_{obs} the apparent first-order rate constant for the steady state that is formed between all the species present in the reaction (enzyme, inhibitor, enzyme-inhibitor complex, and substrate). The k_{obs} units are given in s^{-1} . The equation was entered into the computational software, knowing that P and t are the dependent and independent variables, while keeping the velocities and the rate constant as unknown. For each case, the constraints for their calculation were set to give a positive value (i.e. $k_{\text{obs}} \geq 0$). It is also worth noting that equation 1.7 is only valid when substrate depletion is insignificant. Therefore, some points were excluded in every case for data fitting. Velocities, rates, r^2 , and points analyzed for each substrate concentration are shown in Table 66 and Figure 54.

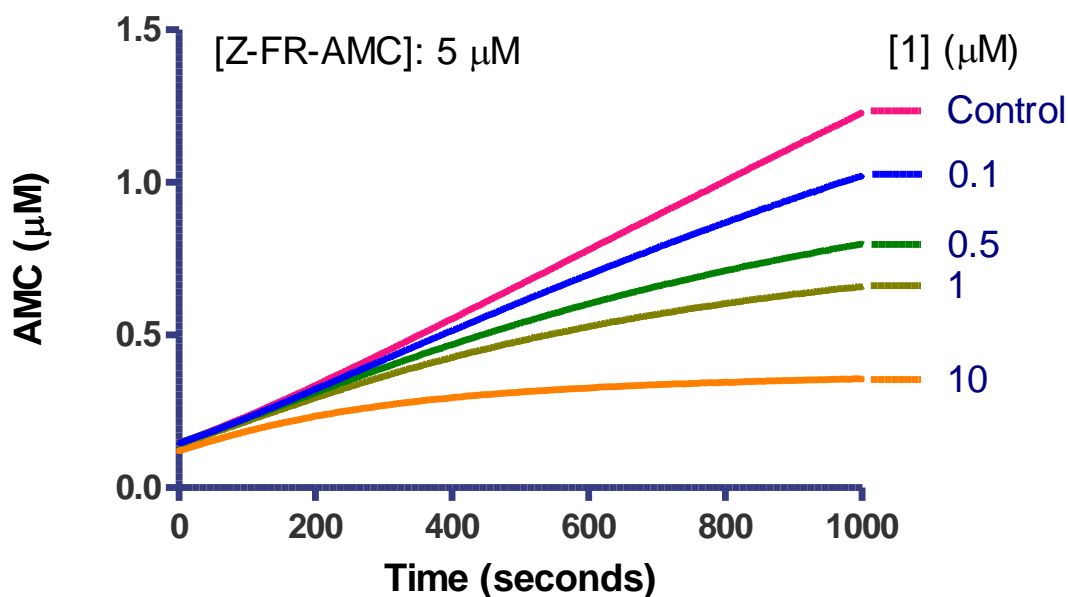


Figure 54. Cathepsin L Progress Curves with **1** using 5 μM Z-FR-AMC

Data sets were fitted into equation 1.7. Ambiguous results were obtained for reactions low inhibitor concentration or uninhibited sets ($[I] \leq 0.1 \mu\text{M}$). Only three concentrations (10, 1, and 0.5 μM) were able to be resolved. Further investigation needs to be done to make stronger conclusions about the mechanism of inhibition of analog **1**.

Effect of Preincubation Studies on Cathepsin L Inhibition Assay Using 1

Progress curves provided strong evidence that **1** is a slow-binding inhibitor. Therefore, we explored the influence of preincubation time on the inhibitory potency of thiosemicarbazones and their IC_{50} values. Compound **1** IC_{50} values were determined at 7 different preincubation times ranging between 0 and 240 minutes. Substrate concentration was 5 μM . Inhibitor final concentration varied between 0 and 10 μM . Table 66 and Figure 55 show the results of the studies.

Table 66. Calculated Kinetic Parameters from Eq.1.7 for Inhibited Cathepsin L Progress Curves with **1** Using 5 μM Z-FR-AMC

[I] (μM)	10	5	0.5
v_s ($\mu\text{M/s}$)	0.0001502	0.0004666	0.0006583
v_i ($\mu\text{M/s}$)	0.004320	0.005051	0.06016
k_{obs} (s^{-1})	0.01850	0.02006	0.3207
r^2	0.9117	0.9736	0.9867
Points analyzed	335	335	335

Table 67. Effect of Preincubation Times on IC_{50} Values of **1** against Cathepsin L

Pre-incubation times (minutes)	$\text{IC}_{50} \pm \text{Standard Error (nM)}$
0	752 ± 17.77
1	155.9 ± 8.44
5	8.26 ± 0.8
15	7.504 ± 0.8
30	6.98 ± 0.7
60	3.08 ± 0.8
120	2.75 ± 0.3
240	2.48 ± 0.1

The effect of preincubation in the potency of **1** as seen in Table 67 is quite evident. The dibromobenzophenone thiosemicarbazone showed a potent activity when cathepsin L was not preincubation with the compound (752 nM). Compound **1** improved its activity five times with only one minute preincubation time. The trend changed dramatically when the preincubation time was increased to 5 minutes. The inhibitory activity of **1** increased 19-fold with a remarkable IC_{50} less than 10 nM (8.3 nM). The potency of **1** was modestly increasing with longer preincubation times. Finally, the best potency was found when **1** was preincubated for four hours with cathepsin L. The activity of the brominated analog increased 4-fold to give a value of 2.5 nM. These results confirmed the strong dependence of IC_{50} value determination with respect to the preincubation time parameter for **1**. Controls (i.e. uninhibited cathepsin L) were

monitored at every preincubation time. There was no significant loss of catalytic activity at longer preincubation times.

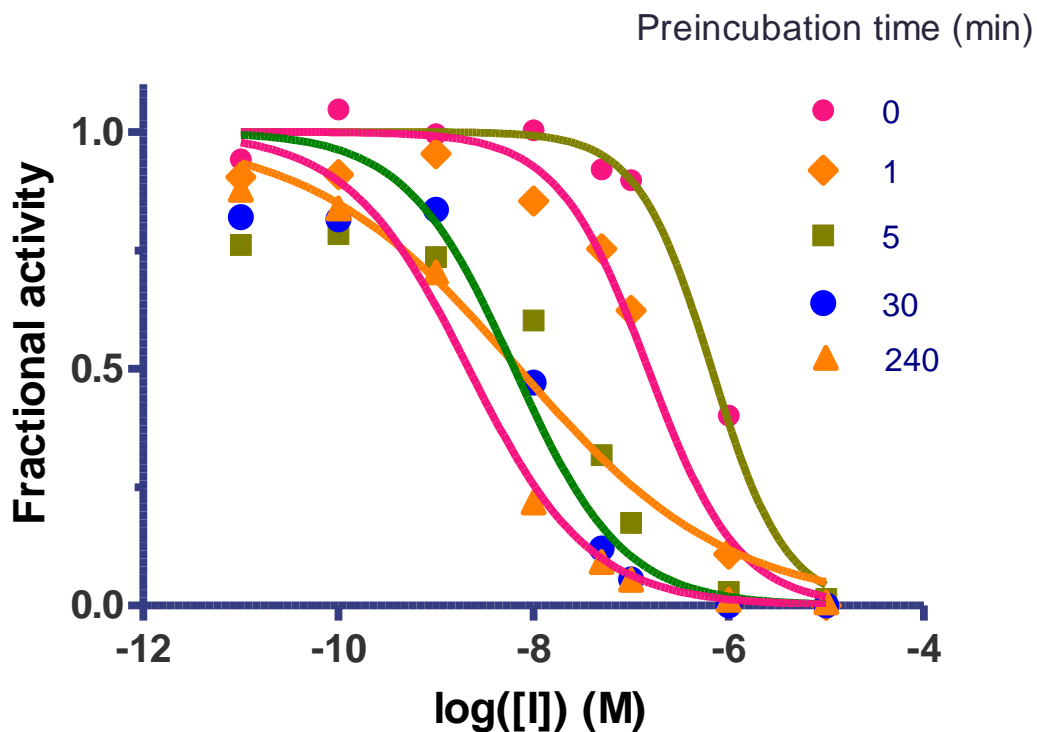


Figure 55. Effect of Preincubation Time on IC₅₀ Values of **1** against Cathepsin L

*Determination of K_i^{app} using Morrison's Quadratic Equation. Effect of Preincubation Time Using **1**.*

The data obtained from the preincubation time was further analyzed. The possibility that **1** was a tight-binding inhibitor was analyzed with the Morrison's quadratic equation. (See equation 2.3). The velocities v_i and v_o are the inhibited and uninhibited cathepsin L velocities (RFU/s), $[E]_T$ (nM) is the total concentration of enzyme found in solution (free enzyme and inhibitor-enzyme complex), $[I]_T$ (nM) is the total concentration of inhibitor present in solution (free inhibitor and inhibitor-enzyme complex), and K_I (nM) is the inhibition constant, often referred as the dissociation

constant. The equation may be solved to give two possible answers. However, the equation is written so that only there is only one possible answer that fits the physiological conditions (i.e. $K_I^{\text{app}} > 0$). GraphPad 5.0 was used to fit the data for every preincubation time point. Inhibitor, substrate, and enzyme concentrations, as well as K_M , K_I^{app} , were all in micromolar units (μM). The residual activity (or v_i/v_o) was normalized to 1 (i.e. $v_o: 1$ and $0 \leq v_i \leq 1$). Normalized residual activity and $[I]$ were defined as the dependent and independent variables, respectively. Nonlinear regression was applied using the following conditions: $[S]: 50 \mu\text{M}$, $K_M: 1.1 \mu\text{M}$, $[E]_T: 0.001 \mu\text{M}$ and $v_o: 1$. Normalization and nonlinear conditions were used to three or four independent experiments per preincubation time. Average and standard errors can be seen in Table 64 and Figure 49. The K_I^{app} values are shown in Table 68 and Figure 56. It can also be seen that the apparent inhibition constants are also time-dependent. No preincubation results in a good inhibition constant with a value of 138 nM. Furthermore, the potency is extremely enhanced (more than 38-fold) for the calculated value at a long preincubation time (1 hour). The apparent K_I was 125-fold better when compared with the corresponding value at no preincubation time ($K_I^{\text{app}}: 1.1 \text{ nM}$).

Table 68. Effect of Preincubation Time in K_I^{app} values of **1** against Cathepsin L

<i>Pre-incubation times</i> (minutes)	$K_I^{\text{app}} \pm \text{Standard Error (nM)}$
0	138 ± 13.1
1	24.7 ± 1.2
5	3.6 ± 0.6
15	2.1 ± 0.2
30	1.5 ± 0.1
60	1.1 ± 0.1
120	ND
240	ND

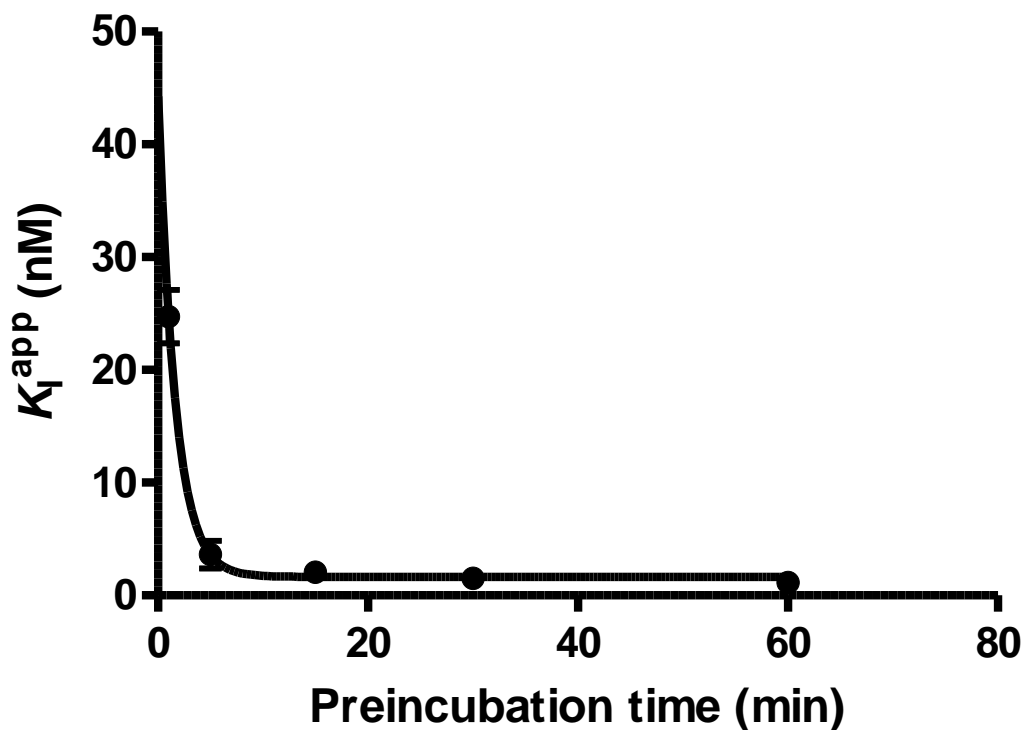


Figure 56. Effect of Preincubation Time in K_I^{app} Values of **1** against Cathepsin L

Cathepsin L Reversibility Studies

Compound **1** was found to be a time-dependent inhibitor in the preincubation studies. However, we decided to explore if this specific compound was a reversible inhibitor of cathepsin L. A mixture containing 100 X cathepsin L and 10 X $\text{IC}_{50(\text{preincubation time: 5 minutes})}$, which are 100 and 170 nM, were incubated at 25 °C between 0 and 60 minutes. The inhibition of cathepsin L by **1** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. Final conditions are 1 nM cathepsin L, 1.7 nM of dibromobenzophenone thiosemicarbazone (**1**), and 50 μM . Additionally, a control experiment (cathepsin L with DMSO as control vehicle) was also set up. Complete experimental details can be found in the material and methods sections. Figure 57 shows the release of AMC for the first 3000 seconds. Uninhibited and inhibited reactions were followed for a total time of four hours.

Cathepsin L was able to recover its catalytic activity after the rapid dilution with assay buffer containing Z-FR-AMC. Apparent substrate depletion can be observed in the uninhibited reaction after 5000 seconds. A linear regression was applied to the first 3000 seconds of the reactions to determine cathepsin L activity. Cathepsin L activity for the untreated reaction was found to be 0.71 nM AMC/s. Similarly, the inhibited reaction when the system was preincubated for one hour was 0.20 nM AMC/s. The results showed cathepsin L was inhibited more than 80% with **1**. Enzymatic activity can be seen within the first 300 seconds (Figure 51).

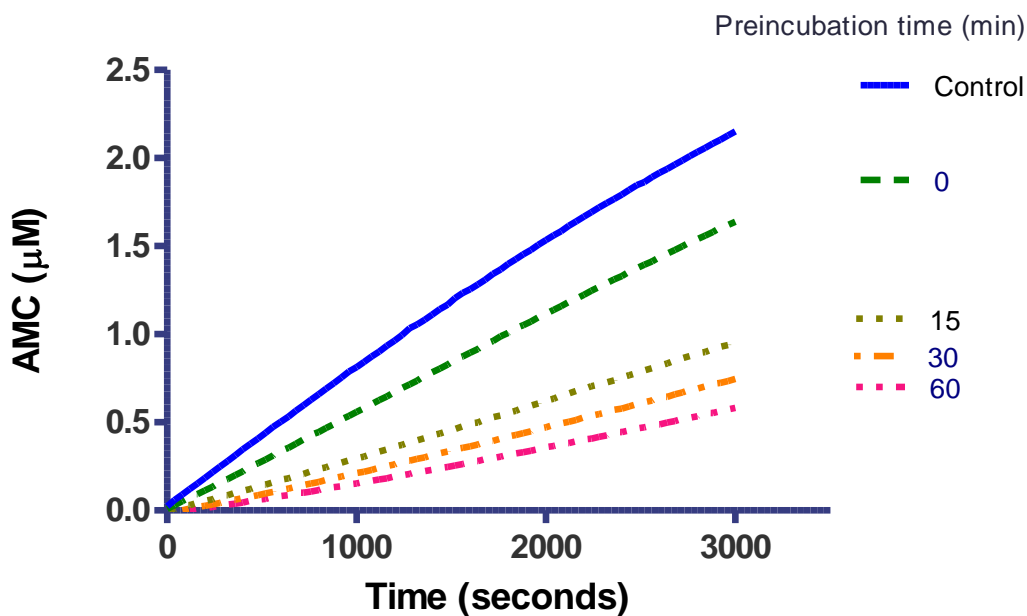


Figure 57. Cathepsin L Reversibility Studies with **1** Using 50 μM Z-FR-AMC

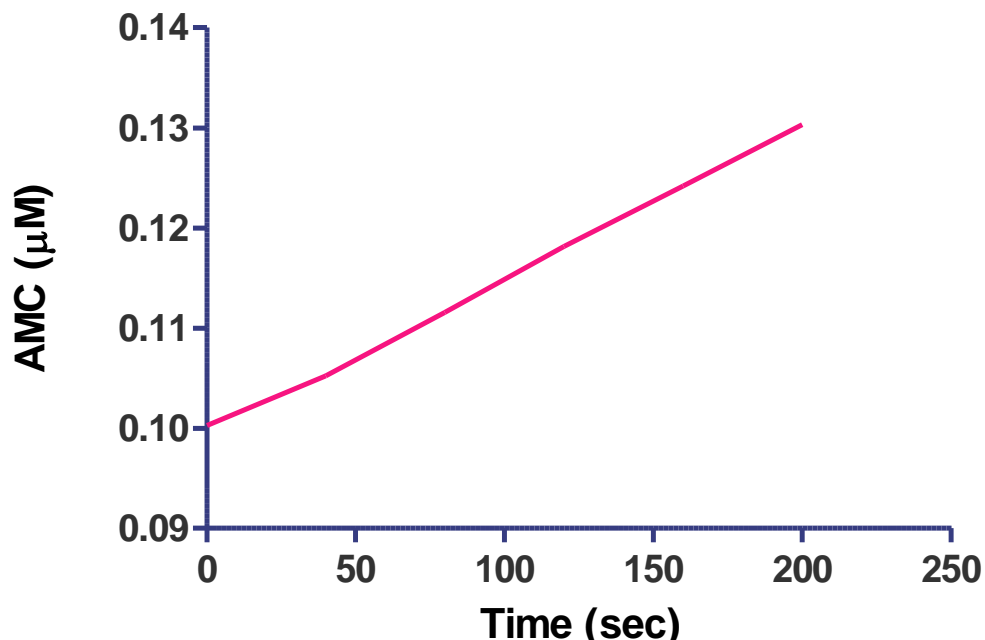


Figure 58. Cathepsin L Reversibility Studies with **1** Using 50 μM Z-FR-AMC. Preincubation time: 1 hour.

*Effect of Substrate Concentration (Z-FR-AMC) in IC_{50} Values Using **1***

We finally tried to determine the mode of inhibition of the one of the most potent of the thiosemicarbazone inhibitors of cathepsin L. Therefore, we investigated the effect of substrate concentration to verify if these synthetic compounds are competitive, noncompetitive, or uncompetitive inhibitors. Three substrate concentrations (Final concentrations: 50, 20, and 5 μM) were used to determine IC_{50} values for compound **1**. Cathepsin L and the series of inhibitors were incubated for five minutes. Results can be found in Table 69 and Figure 59.

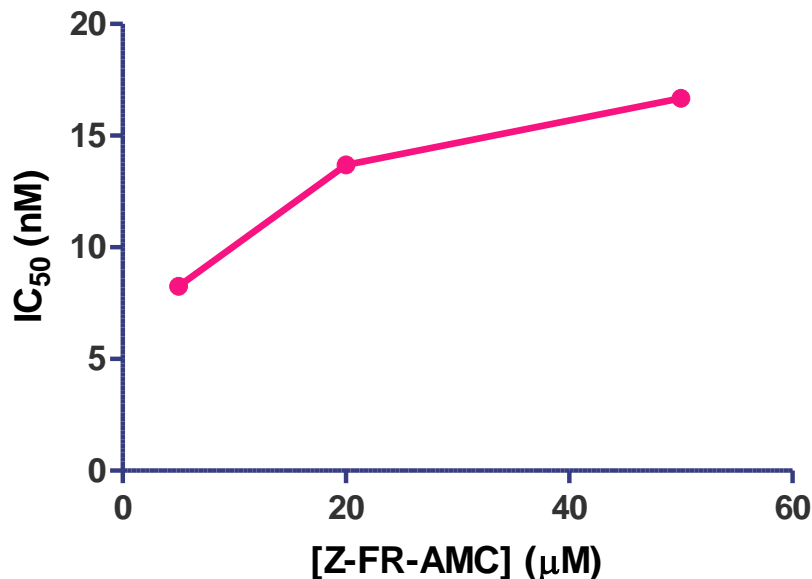


Figure 59. . Effect of Substrate Concentration in IC₅₀ values of **1** against cathepsin L

Table 69. Effect of Substrate Concentration on IC₅₀ Values of **1** against Cathepsin L

<i>[Z-FR-AMC] (μM)</i>	<i>IC₅₀ ± Standard Error (nM)</i>
50	16.7 ± 1.0
20	13.7 ± 1.7
5	8.3 ± 0.8

The effect of substrate concentration was investigated using **1** as the lead compound of the series of thiosemicarbazones. A pseudo linear behavior can be observed with the values. Lowering the amount of substrate in solution gives a subtle, yet evident increase in the potency of the dibrominated analog. According to literature, a positive linear behavior (i.e. $IC_{50} \propto [S]$) is an indication of the compound acts as competitive inhibitor; that is, both substrate and the compound compete for cathepsin L active site. However, the results cannot be taken as conclusive and more experiments were necessary to carry out to confirm these findings.

Cell Culture Experiments

Preliminary screening of a library of more than 150 thiosemicarbazones provided interesting and promising results. More than 6 compounds were found to potent cathepsin L inhibitors, (See Table 28-Table 45). It was also shown that one of the compounds, the phenolic brominated benzophenone thiosemicarbazone was capable to inhibit the collagenase activity of cathepsin L *in vitro*. Thus, we decided to explore the potency of some of the most promising compounds to arrest the invasion and motility of MDA-MB-231 cells, a human breast cancer cell line that has been previously investigated and showed high levels of cathepsin L, invasiveness and motility properties. Figure 60 shows five of the most promising compounds that were used to perform cell culture experiments. Two analogs belong to the *parabrominated* benzophenone thiosemicarbazone series (Compounds **1** and **8**), and the remaining three compounds are part of the benzoylbenzophenone subfamily. E-64 (Table 8, page 31) , a nonspecific cysteine protease irreversible inhibitor was used as a positive control to verify the activity of the compounds. E-64 was selected due to being highly selective towards cysteine proteases and not reacting with other enzymes. E-64 has also been used in cell studies due its permeability characteristics.³³⁶

Invasion and migration assays were carried out using MDA-MB-231 cells that have been harvested with DMEM media supplemented with 10% FBS. Maintenance and subculture procedures have been described in the material and procedure secondstion of this chapter. Three separate T75 flasks with MDA-MB-231 cells (passages 13, 13, and 13) were used to carry out invasion and migration experiments. Briefly, stock solutions

of the compounds and the positive control were made according to Table 25 and Table 26. Stock solution concentrations for each compound were 50 and 20 μM in 4% DMSO.

MDA-MB-231 cells were trypsinized when they were 90% confluent. DMEM media with low FBS was used to neutralize the enzymatic activity of the solution. Cell counting was used to calculate the cell density of each stock solution (cells/ml). Then, DMEM without FBS was added to each stock solution to make a dilution with a concentration of 200,000 cells/ml. Invasion inserts (inserts with a Matrigel® layer) and migration inserts, (inserts without a Matrigel® layer with an 8 μm pore) were rehydrated with no-FBS DMEM media at 37 °C in 5% CO_2 /air environment. Media was removed and discarded. Immediately, inserts were placed into 24-well companion plates that contained DMEM media supplemented with 10% FBS. Every insert (for migration and invasion assays) contained a mixture with cells and two concentrations for every compound. Final concentrations were 50,000 cells, 2% DMSO, and 0, 10 (or 25) μM of compounds **1**, **8**, **156**, **157**, **168** and E-64. Samples were incubated for 24 hours at 37 °C in 5% CO_2 /air environment. After incubation time, media was removed, and the inner sides of the membranes were cleaned to remove cells that did not invade or migrate. Then, membranes were stained with DiffQuik staining kit with azure and xanthenes dyes. Membranes were air-dried and placed on slides. Manual cell counting was performed using a Zeiss inverted microscope. Slides were placed onto the instrument, and were observed under a 40X objective. Cell that migrated the 8 μm membrane or cells that invaded the Matrigel® layer have a dark purple shade with a round shape. The purple coloration is due that cytoplasmic matter and nuclei are sensitive to both dyes. The dark color is also easier to follow and distinguish from other particles.

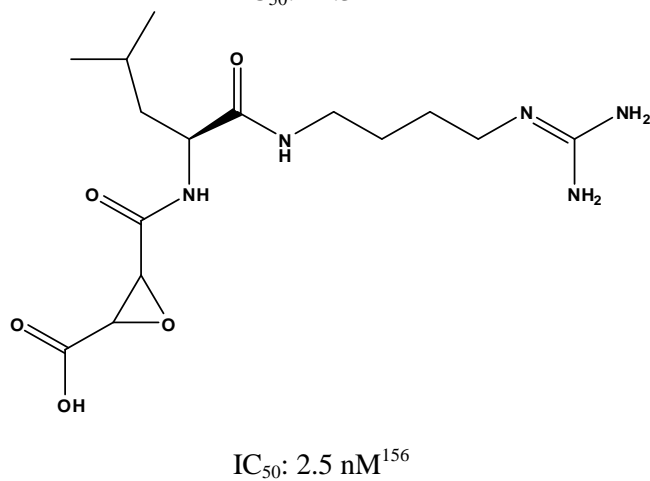
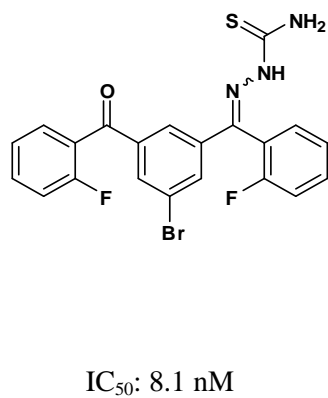
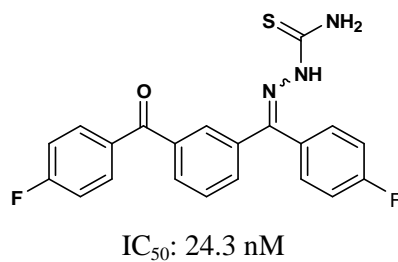
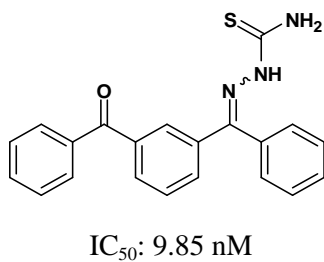
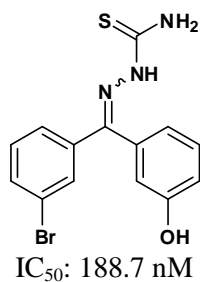
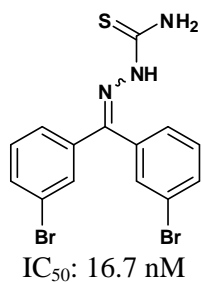


Figure 60. Chemical Structures of **1**, **8**, **156**, **157**, **158** and E-64.



Figure 61. Invasion and Migration Assay Samples using MDA-D231 cells treated with **1, 8, 156, 157, 168** and E-64

Figure 61 shows the samples that were preserved in glass slides for further investigation. Manual cell counting was performed using a Zeiss inverted microscope. Samples were placed onto the instrument, and were observed under a 40X objective. Cells that migrated the 8 μm membrane or cell that invaded the Matrigel® layer have a dark purple shade with a round shape. The purple coloration is due that cytoplasmic matter and nuclei that are sensitive to both dyes. The dark color is also easier to follow and distinguish from other particles. On the other hand, pores look clear with a round shape, similar to a ‘bubble trapped in the membrane’. Only purple dots were manually counted, but there were some cells that did not migrate/invade entirely and have a dark

shadow semi-round shape. These cells were also taken into account for the manual cell counting process.

Ten representative areas were taken for every sample. Eight areas located in the outer periphery of the sample were counted. Also, two more areas located at the center of the membrane were counted. Figure 18 shows a scheme that describes approximate positions for every counted area. Eighty-four samples were observed and analyzed. The list includes triplicates of controls with 2% DMSO (control vehicle), controls with no DMSO, and 10 or 25 μ M of compounds **1**, **8**, **156**, **157**, **168** and E-64 (positive control). Figure 62, Figure 63, Figure 64, Figure 65, Figure 66, Figure 67 show panels of representative experiments of some of the samples that were analyzed.

Results were tabulated and total numbers of samples were calculated. Table 70, Table 71, Table 72, Table 73 show the final results of manual cell counting. Furthermore, the percentages of cell invasion and migration that were inhibited by the several compounds are defined in Equations 2.4, 2.5, 2.6 and 2.7.

$$\% \text{ Invasion} = \frac{\text{Cells that invaded the Matrigel(R) in the presence of compound}}{\text{Cells that invaded the Matrigel (R) with no compound}} \quad (2.4)$$

$$\% \text{ Migration} = \frac{\text{Cells that migrated the Matrigel(R) in the presence of compound}}{\text{Cells that migrated the Matrigel (R) with no compound}} \quad (2.5)$$

$$\% \text{ Inhibition (invasion)}: 100 - \% \text{ Invasion} \quad (2.6)$$

$$\% \text{ Inhibition (migration)}: 100 - \% \text{ Migration} \quad (2.7)$$

Table 70. Cell Migration (25 μ M) Results

25 μ M	Cell Count							
	Vehicle (2% DMSO)	No vehicle	# 1	#8	#156	#157	#168	E-64
E1	1002	1004	29	119	136	328	336	603
E2	807	727	76	168	339	367	532	541
E3	1101	766	67	137	121	635	332	657
Average	970	832.3	57.3	141.3	198.7	443.7	400.0	600.3
S.E.	86.4	86.6	14.4	14.3	70.3	96.3	66.0	33.5
% \pm S.E.	100 ± 8.9	85.8 ± 8.9	5.9 ± 1.5	14.6 ± 1.5	20.4 ± 7.2	45.74 ± 9.9	41.2 ± 6.8	61.9 ± 3.5

Table 71. Cell Migration (10 μ M) Results

10 μ M	Cell Count							
	Vehicle (2% DMSO)	No vehicle	# 1	#8	#156	#157	#168	E-64
E1	1002	1004	96	222	576	360	750	904
E2	807	727	286	350	578	668	727	871
E3	1101	766	87	252	492	640	653	754
Average	970	832.3	156.3	274.7	548.7	547.7	710	843
S.E.	86.4	86.6	64.9	38.7	28.3	10.3	29.3	45.5
% \pm S.E.	100 ± 8.9	85.8 ± 8.9	16.1 ± 6.7	28.3 ± 4.0	56.6 ± 2.9	66.8 ± 1.1	73.2 ± 3.0	86.9 ± 4.7

Table 72. Cell Invasion (25 μ M) Results

25 μ M	Cell Count							
	Vehicle (2% DMSO)	No vehicle	# 1	#8	#156	#157	#168	E-64
E1	714	728	70	86	583	361	402	209
E2	889	819	102	69	535	270	316	167
E3	699	632	89	114	476	237	90	160
Average	767.3	726.3	87.0	89.7	531.3	289.3	269.3	178.7
S.E.	61.0	54.0	9.3	13.1	30.9	37.1	93.0	15.3
% \pm S.E.	100 ± 7.9	94.7 ± 7.0	11.3 ± 1.2	11.7 ± 1.7	69.3 ± 4.0	37.7 ± 4.8	35.1 ± 12.1	23.3 ± 2.0

Table 73. Cell Invasion (10 μ M) Results

10 μ M	Vehicle (2% DMSO)	No vehicle	# 1	#8	#207	#244	E#312	E-64
E1	714	728	128	279	586	735	219	347
E2	889	819	155	139	644	690	231	213
E3	699	632	222	248	636	540	249	367
Average	767.3	726.3	168.3	222.0	622.0	655.0	233.0	309.0
S.E.	61.0	54.0	27.9	42.5	18.2	59.0	8.7	48.4
% \pm S.E.	100 ± 7.9	94.7 ± 7.0	21.9 ± 3.6	28.9 ± 5.5	81.1 ± 2.4	85.4 ± 7.7	30.4 ± 1.1	40.3 ± 6.3

Six different compounds were tested to see their ability to inhibit cell invasion and migration using a well established colorimetric method to quantify these two characteristics. Five of these compounds are thiosemicarbazones analogs, and E-64 is a non-specific irreversible cysteine inhibitor that is used as a positive control. MDA-MB-231 cells are known to show high levels of invasiveness and motility.³³⁷

Cell migration was achieved with the set of six compounds that were tested. In general, substituted *para*-brominated benzophenone thiosemicarbazones were the best compounds to delay cell migration. Compounds **1** and **8** showed an extraordinary capacity to delay cell migration. More than 85% of the cells (compared to control) did not migrate the 8 μ M membranes. Furthermore, the benzoylbenzophenone subfamily was also able to significantly arrest cell migration. The unsubstituted analog (compound **156**) inhibited almost 80% of cell migration. Overall, this group of selected thiosemicarbazones demonstrated to have a better inhibitory activity compared to the positive control.

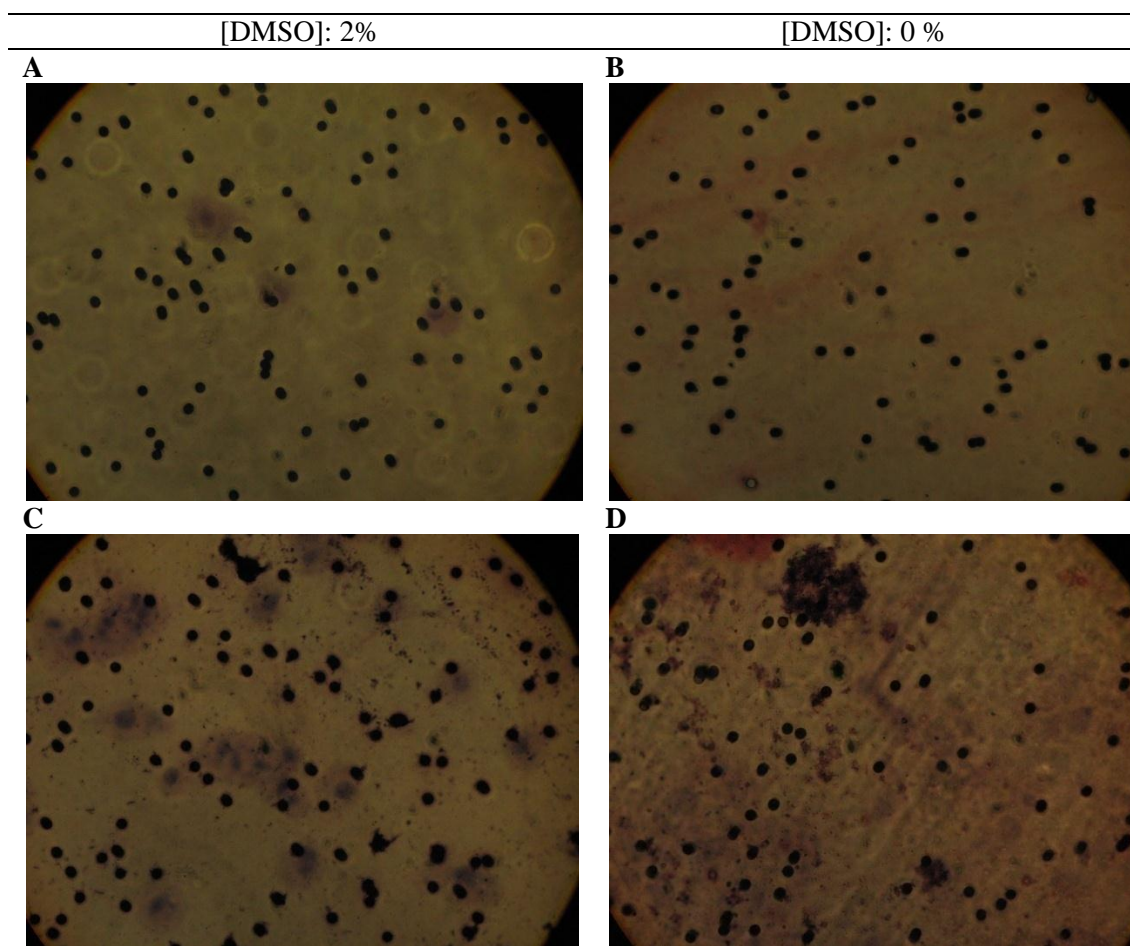


Figure 62. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays Using Non-Treated MDA-MB-231 Breast Cancer Cells. **A.** Invasion Assay, Control 2% DMSO. **B.** Invasion Assay, Control 0% DMSO. **C.** Migration Assay, Control 2% DMSO. **D.** Migration Assay, Control 0% DMSO.

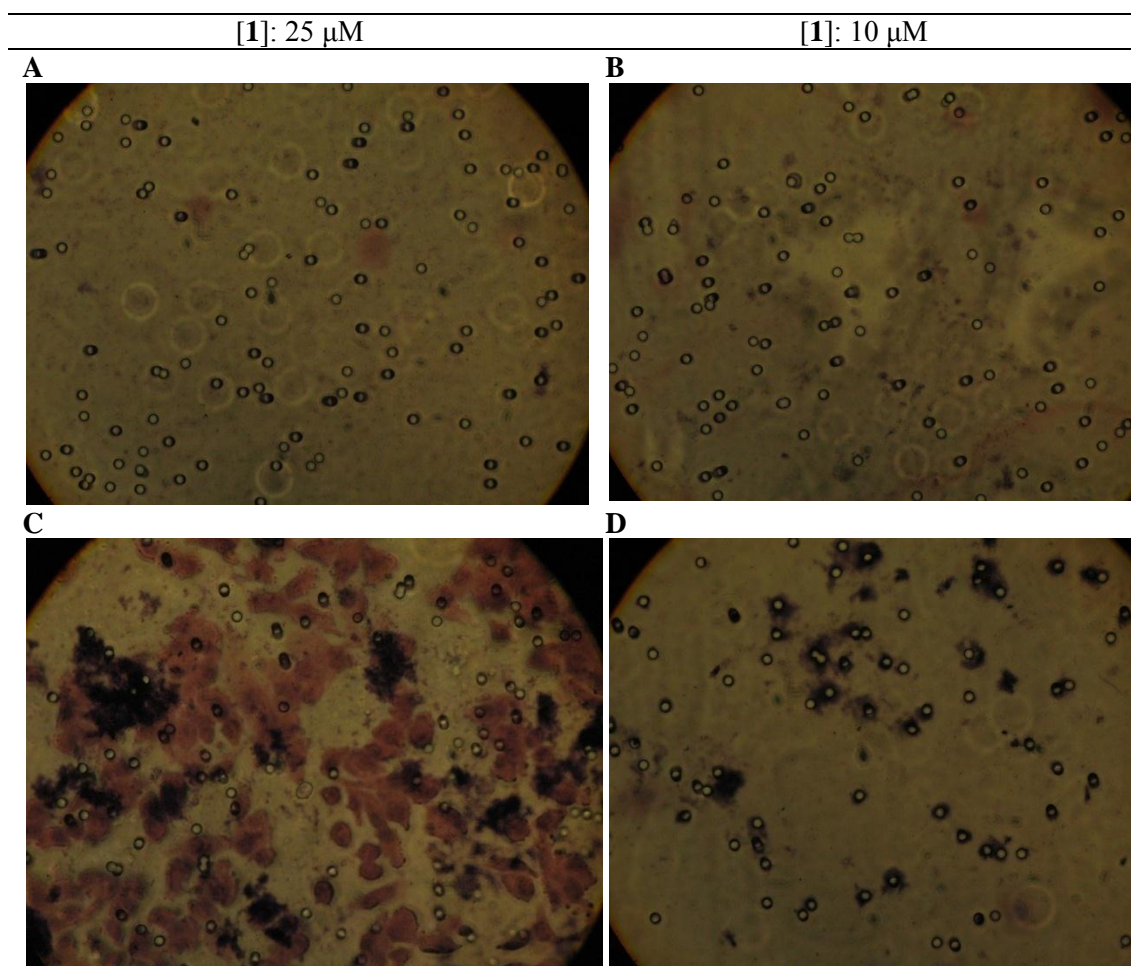
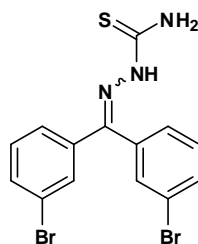


Figure 63. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays using MDA-MB-231 Breast Cancer Cells treated with **1**. A. Invasion Assay, 25 μ M of **1**. B. Invasion Assay, 10 μ M of **1**. C. Migration Assay, 25 μ M of **1**. D. Migration Assay, 10 μ M of **1**

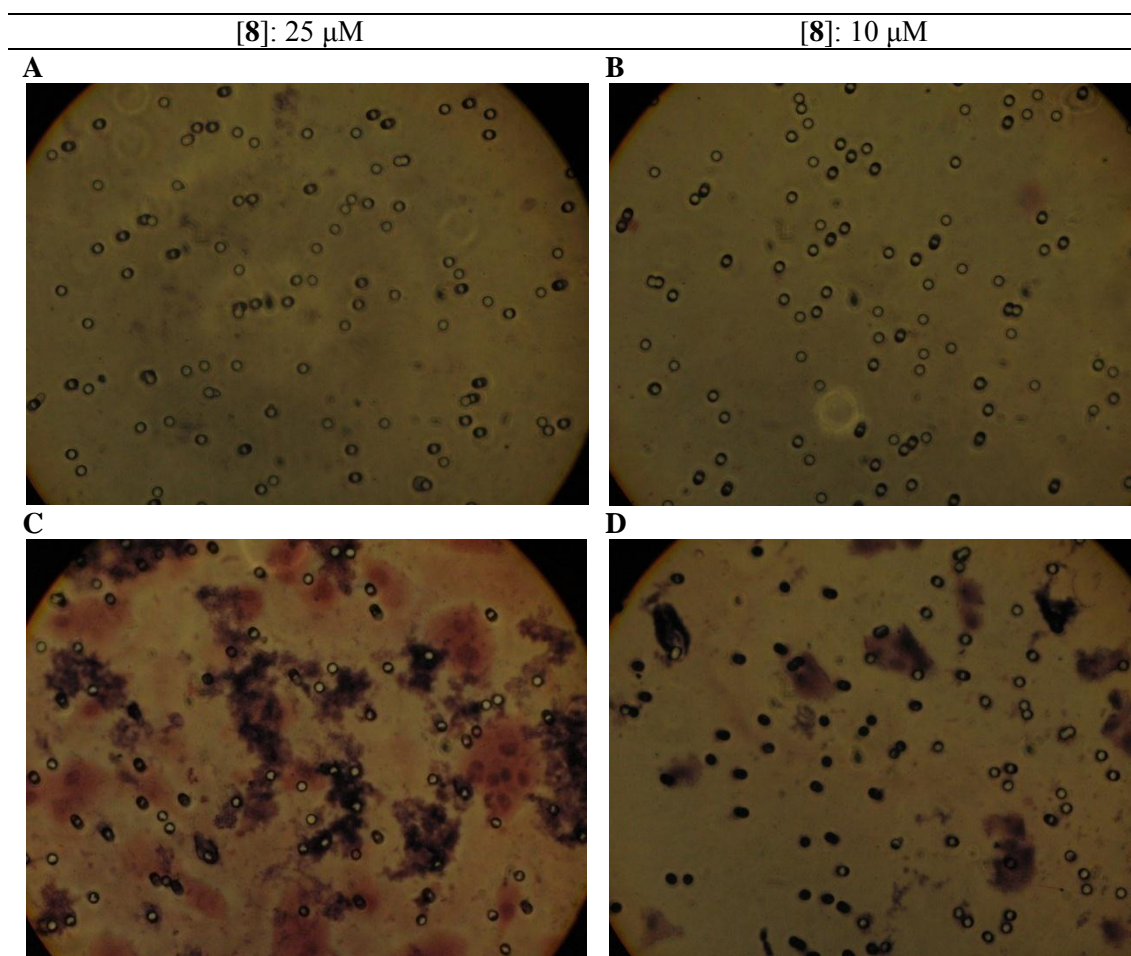
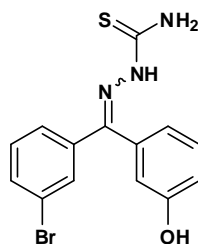


Figure 64. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays using MDA-MB-231 Breast Cancer Cells treated with **8**. A. Invasion Assay, 25 μ M of **8**. B. Invasion Assay, 10 μ M of **8**. C. Migration Assay, 25 μ M of **8**. D. Migration, 10 μ M of **8**

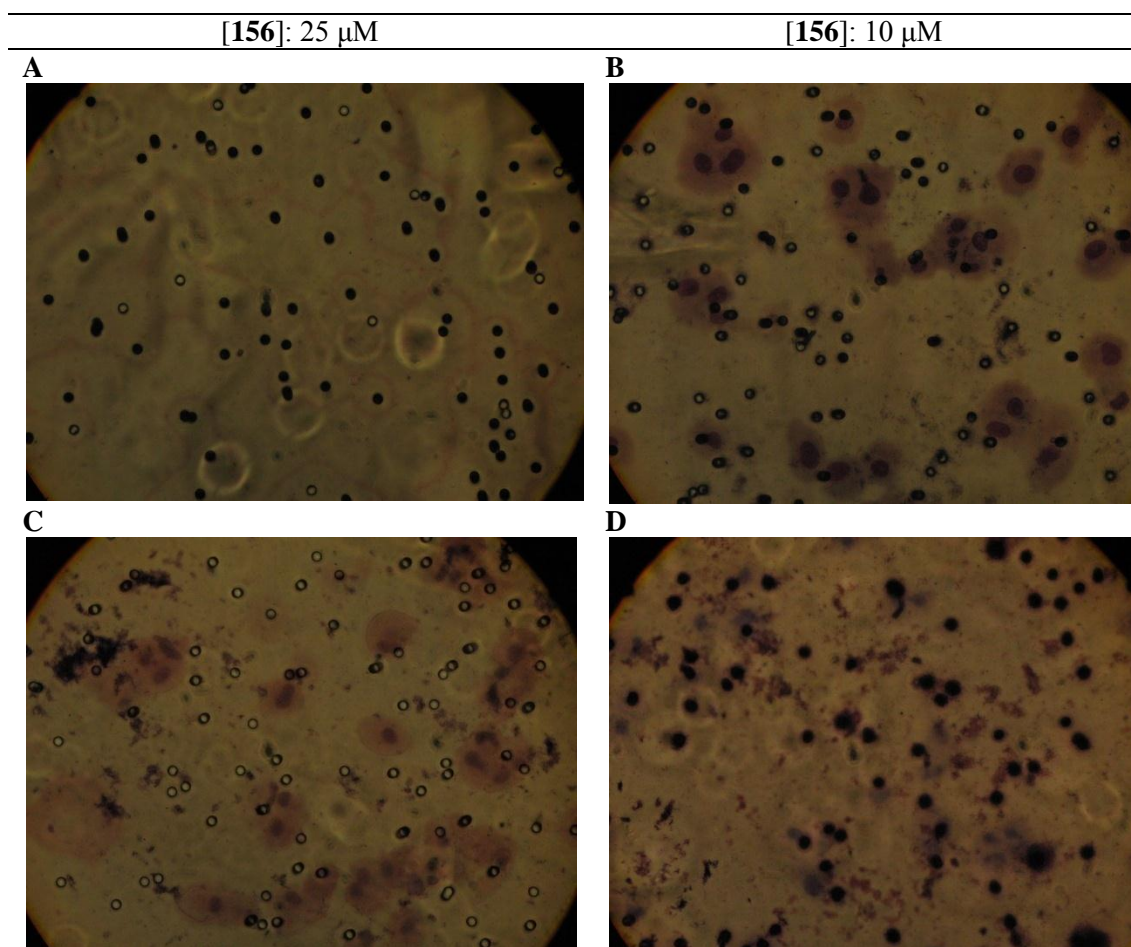
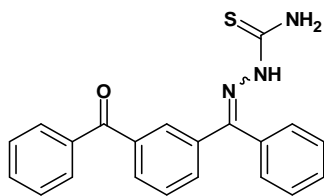


Figure 65. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays using MDA-MB-231 Breast Cancer Cells treated with **156**. A. Invasion Assay, 25 μ M of **156**. B. Invasion Assay, 10 μ M of **156**. C. Migration Assay, 25 μ M of **156**. D. Migration Assay, 10 μ M of **156**

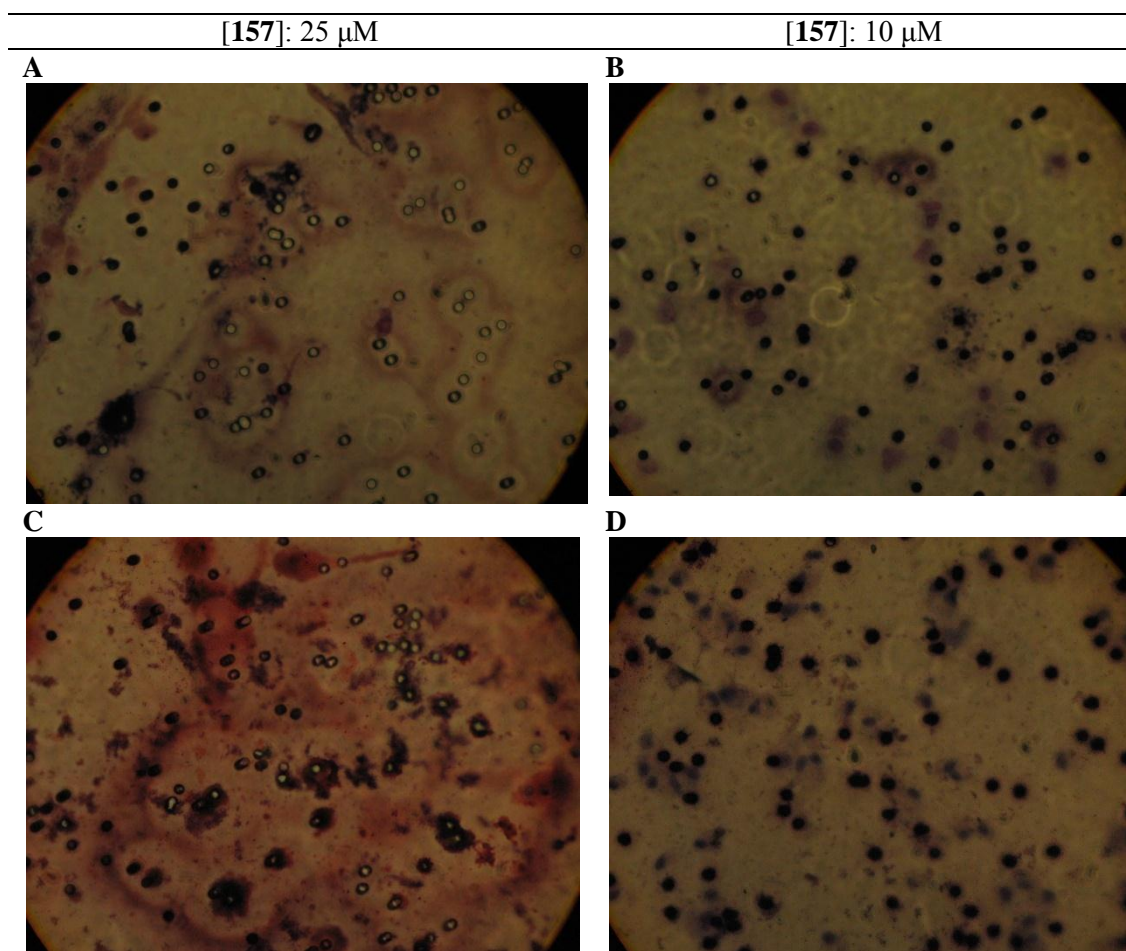
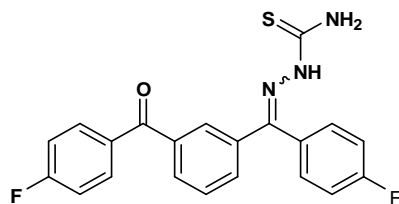


Figure 66. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays using MDA-MB-231 Breast Cancer Cells treated with **157**. A. Invasion Assay, 25 μ M of **157**. B. Invasion Assay, 10 μ M of **157**. C. Migration Assay, 25 μ M of **157**. D. Migration Assay, 10 μ M of **157**

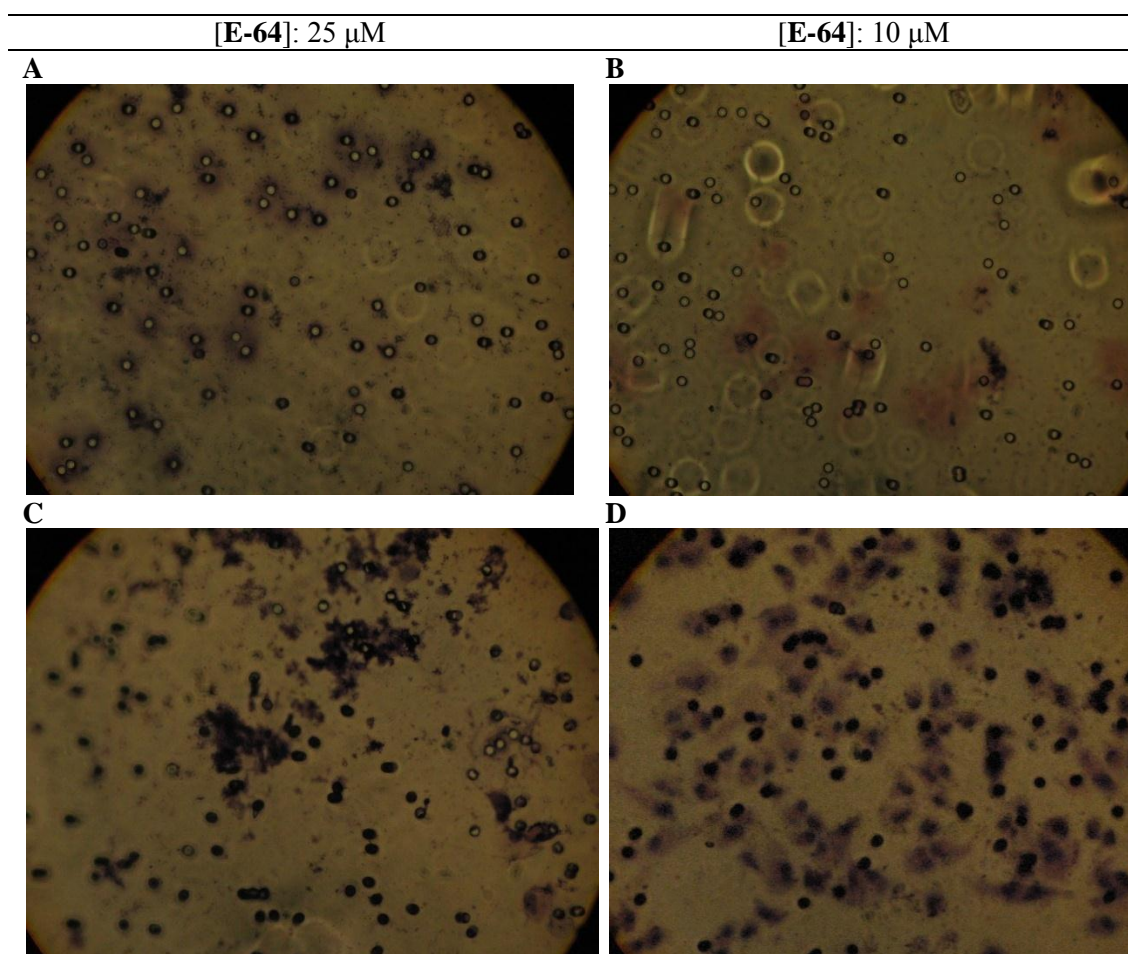
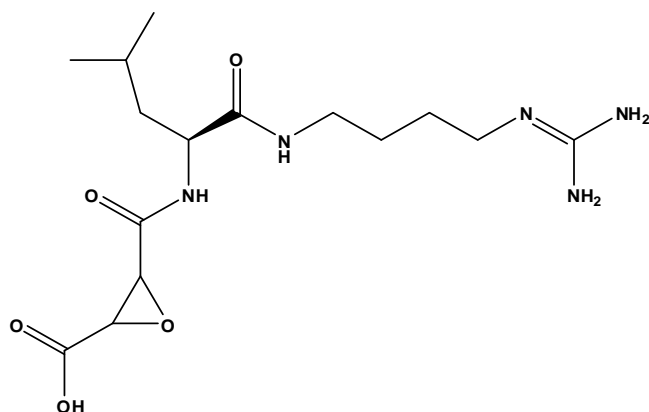


Figure 67. Representative Samples of Cell Invasion (A and B) and Cell Migration (C and D) Assays using MDA-MB-231 Breast Cancer Cells treated with E-64. A. Invasion Assay, 25 μ M of E-64. B. Invasion Assay, 10 μ M of E-64. C. Migration Assay, 25 μ M of E-64. D. Migration Assay, 10 μ M of E-64

Compound **157**, the difluorobenzoyl benzophenone thiosemicarbazone is almost 1.5 times more potent than E-64. Results may also indicate that thiosemicarbazones have a good permeability when a proper solvent vehicle concentration is used (2% DMSO).

Compounds **1** and **8** also showed to be the best compounds of this series of analogs (more than 88% inhibition). Furthermore, the hetero halogenated analog of the benzoylbenzophenone series was able to significantly inhibit cell invasion. However, the concentration effect does not seem to enhance the potency of this compound. It was the only one of the subfamily that proved to be a better cell invasion agent when compared to E-64 results. Finally, the unsubstituted benzoyl benzophenone did not show a significant activity when inhibiting MDA-MB-231 cell invasion.

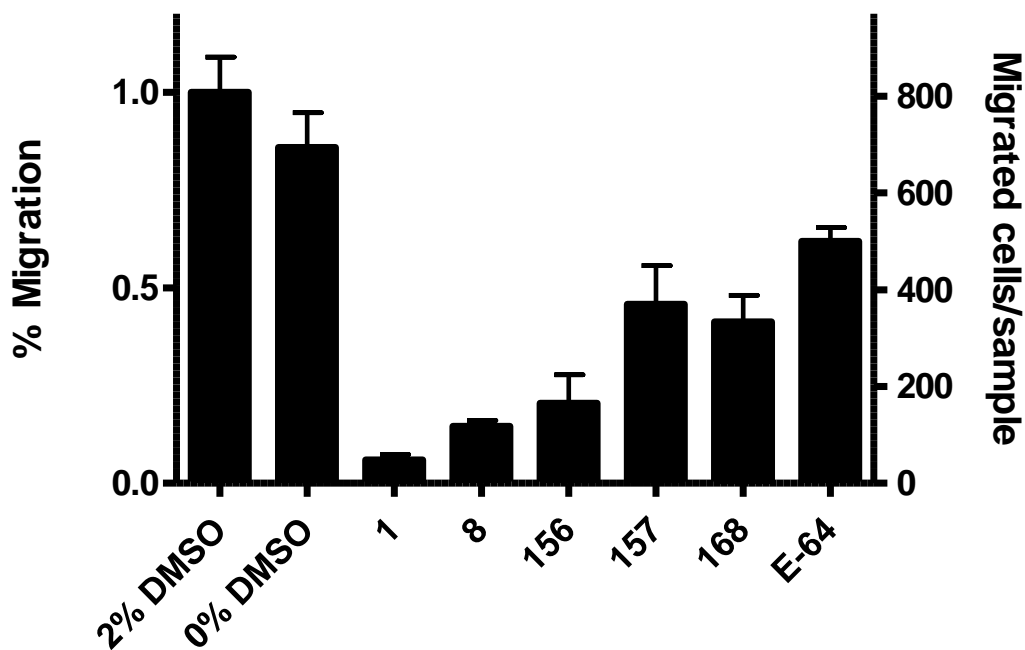


Figure 68. Cell Migration (25 μ M) Results

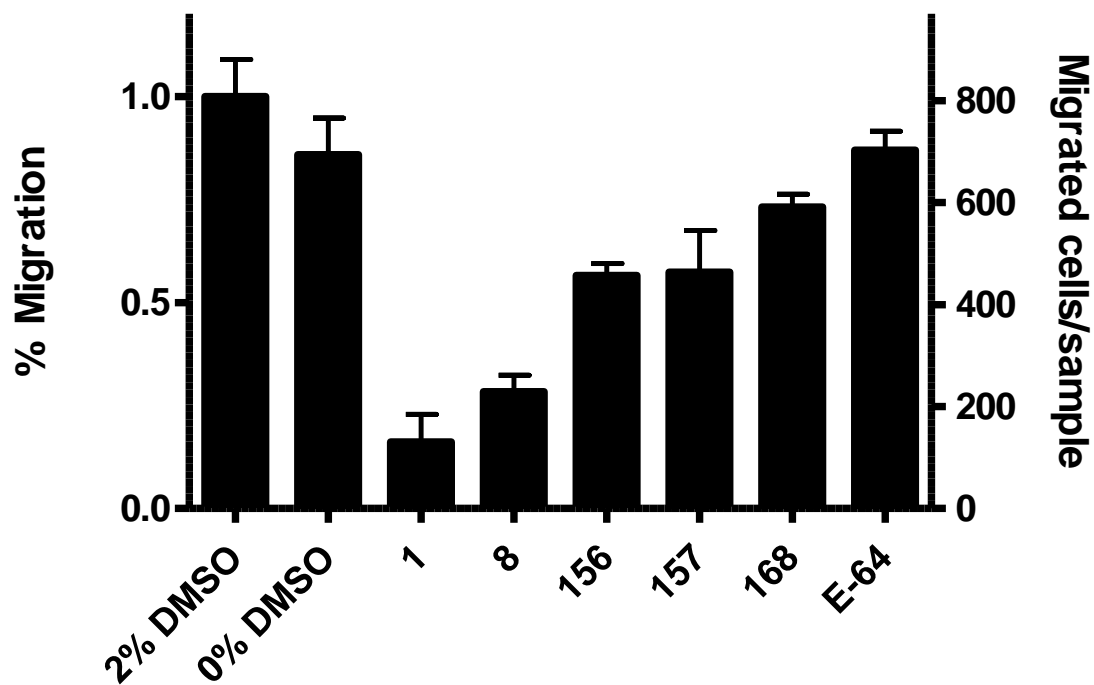


Figure 69. Cell Migration (10 μ M) Results

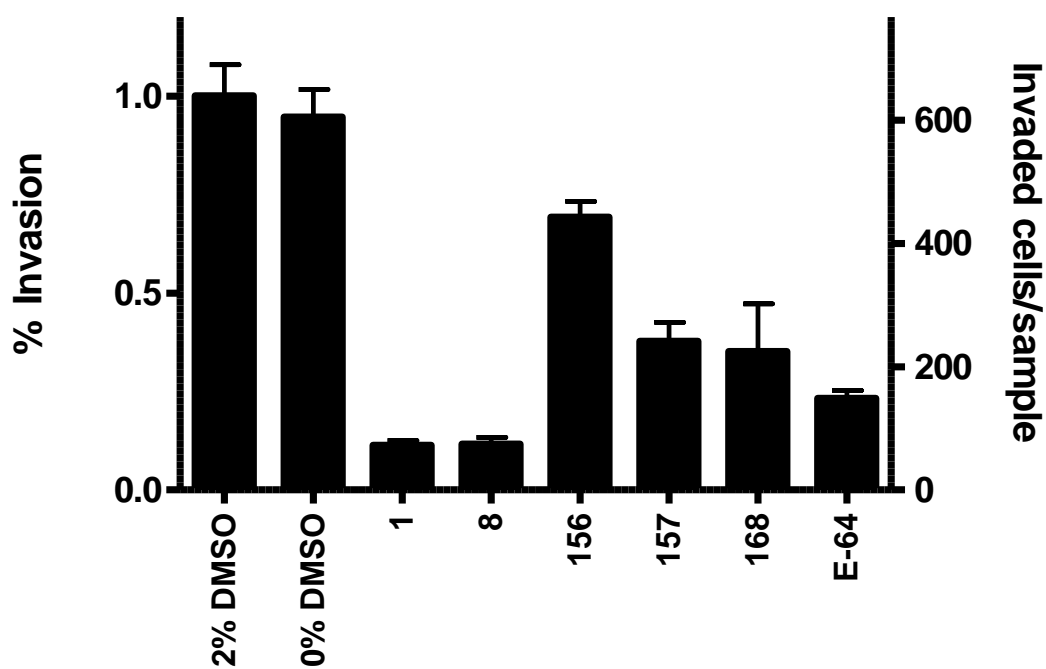


Figure 70. Cell Invasion (25 μ M) Results

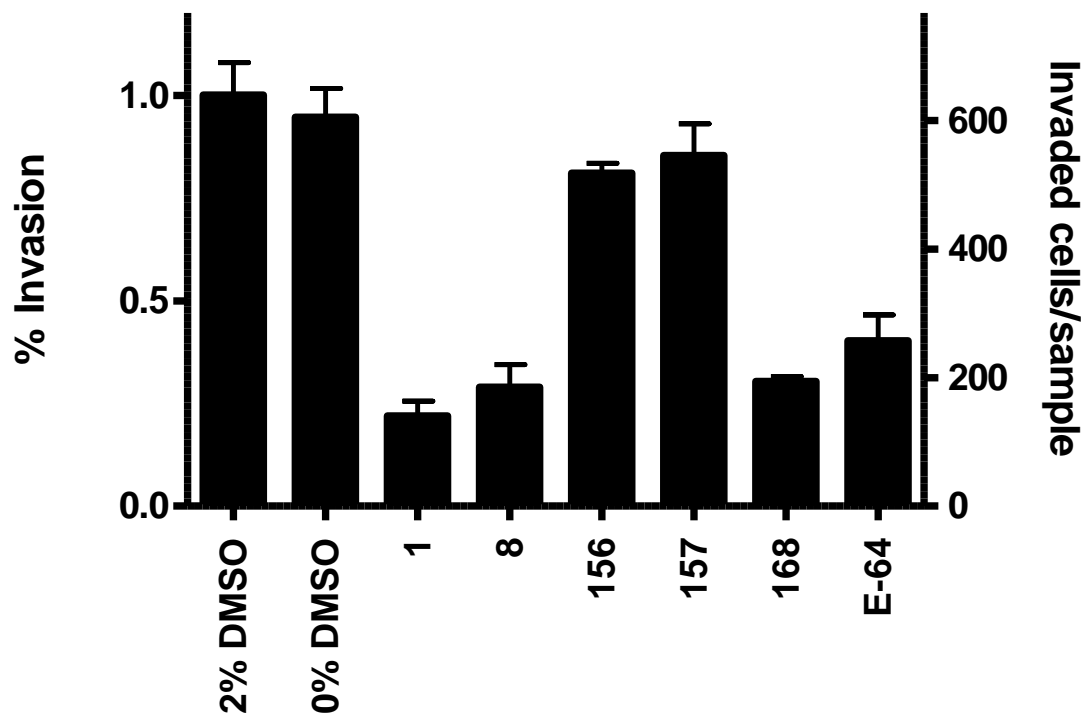
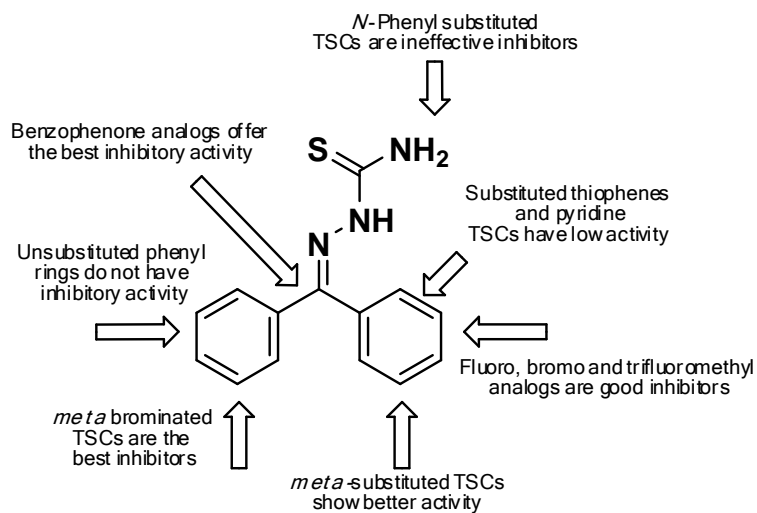


Figure 71. Cell Invasion (10 μ M) Results

Conclusions

A medium size library of 168 thiosemicarbazones analogs were evaluated as inhibitors of recombinant, cathepsin L from human liver. Eight analogs arose as excellent cathepsin L inhibitors ($IC_{50} \leq 50$ nM). The examined library included functionalized benzophenone, thiochromanone, sulfones, and benzoyl benzophenone thiosemicarbazones. Figure 72 shows the Structure-Activity Relationship of thiosemicarbazones based on their chemical structures.

A.



C.

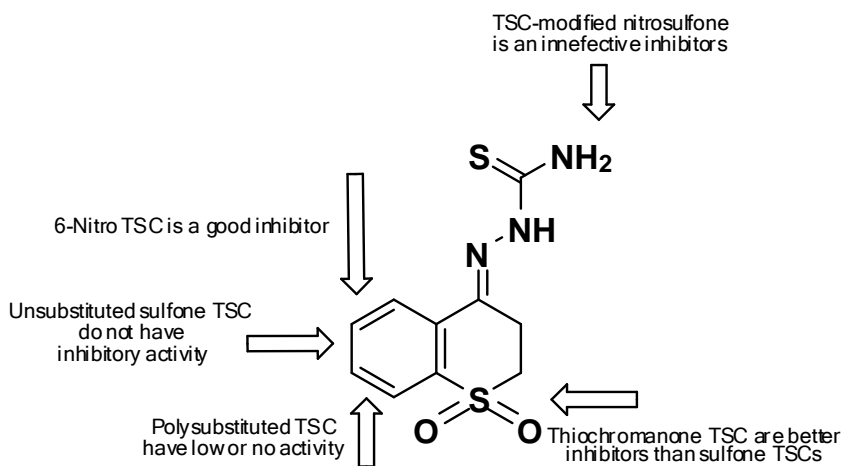
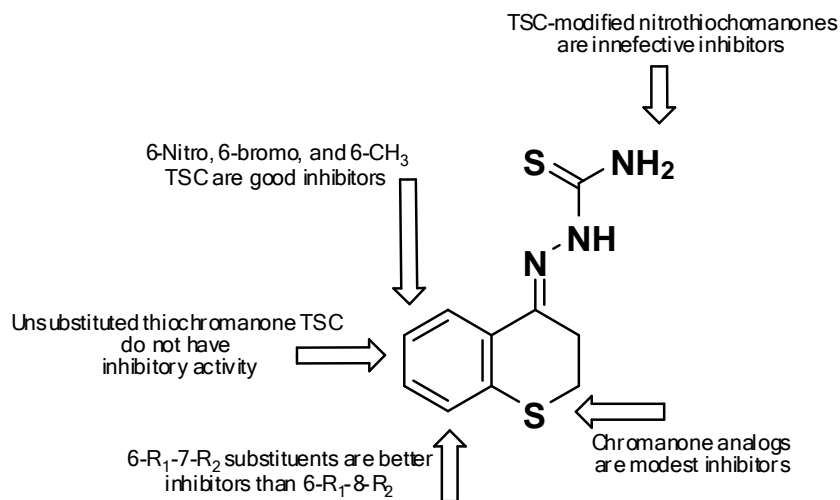


Figure 72. Structure-Activity Relationship for TSCs as Cathepsin L Inhibitors. **A.** Benzophenone TSCs. **B.** SulfoneTSCs

A.



B.

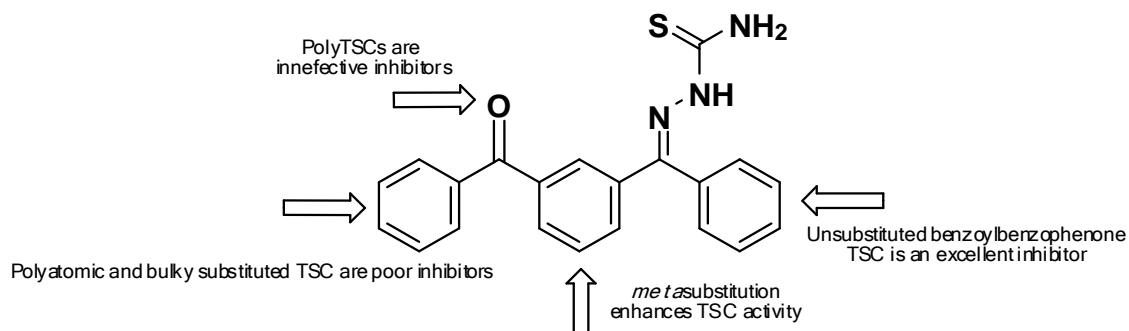


Figure 73. Structure-Activity Relationship for TSCs as Cathepsin L Inhibitors. **A.** Thiochromanone TSCs. **B.** Benzoylbenzophenone TSCs

Two lead compounds, **1** and **8** were characterized by using advanced kinetic studies to determine kinetic parameters, mechanisms and mode inhibition of these non-peptidic inhibitors. Both compounds were determined to be slow, time-dependent, reversible, competitive inhibitors of the fluorogenic substrate Z-FR-AMC. Kinetic studies with **8**, are consistent the phenolic brominated benzophenone thiosemicarbazone inhibits cathepsin L by simple reversible mechanism. However, a two-step mechanism is suggested at lower substrate concentrations. The inhibition of cathepsin L collagenase activity was also studied and inhibited when **8** was used in *in vitro* studies using type I

collagen from human skin as a natural substrate. Compound **8** was able to inhibit the catalytic activity of cathepsin L more than 50% after 6 hours of reaction.

Finally, five compounds (**1**, **8**, **156**, **157**, **168**) were used with MDA-MB-231 cells, (adenocarcinoma breast cancer), to test their ability to inhibit cell invasion and cell migration of this cell line. Functionalized benzophenone thiosemicarbazones showed the most potent activities among the selected group by inhibiting more than 85% of cells invading Matrigel®, a mixture containing collagen, fibronectin and laminin, which resembles the complex extracellular matrix found in human tissues.

CHAPTER THREE

Evaluation of Thiosemicarbazones as Cathepsin K Inhibitors

Nomenclature, Classification and Historical Background

Human cathepsin K (EC 3.4.22.38) is a hydrolase and belongs to the subclass of peptidases, enzymes that act on peptidic bonds.³³⁸ Similarly to cathepsin L, cathepsin K is also classified as an endopeptidase, that is, a hydrolase that only acts on peptidic bonds located within amino acid sequences.³³⁹ Basically, cathepsin K is a cathepsin L-like enzyme as well.³⁴⁰

The official name of the protein is cathepsin K (CK). CK is also known as cathepsin O or cathepsin O2.¹⁹³ The discovery of cathepsin K is more recent than its cathepsin L analogue. Reports of the presence of cathepsin K were reported by Tezuka and coworkers in 1994 when they cloned this protease from rabbit osteoclasts.³⁴¹ Cathepsin K is so far, one of the most studied cysteine proteases in mammalian organisms due to its extraordinary versatility and hydrolytic capacities. These aspects will be discussed in this chapter.

Localization of cathepsin K

Interestingly, cathepsin K is a lysosomal cathepsin L-like protease and its distribution in human tissues is more limited compared to cathepsin L. This might be due to its highly capacity to degrade macromolecules among tissues and organs. Specifically, expression of cathepsin K has been found in osteoclasts (these cells are found in bone matrix and are responsible of the degradation of the proteins found in bone tissue),

macrophages, epithelial cells of several systems (respiratory, and gastrointestinal), urinary tract of human fetus, skin, ear, lung, ovary, testis, and colon.^{105,342–347}

Natural Substrates of Cathepsin K

Cathepsin K (CK) is one of the most potent of the cysteine cathepsins. It plays a critical role in bone resorption as it will be discussed later in this chapter. The activity of this cysteine protease has been well investigated in numerous *in vitro* studies.

Research has proved that extracellular matrix components are the preferred substrates for cathepsin K.³⁴⁸ Cathepsin K is capable of degrading aggrecan, elastin, and osteonectin among other proteins.^{342,349,350}

Perhaps collagen is cathepsin K's special substrate. Cathepsin K collagenase activity has been verified by using multiple collagen substrates including types I, II, III, and IV collagen.^{351–354} Types I and II collagen are very resistant to degradation.³⁵⁵ Few enzymes are able to utilize them as substrates. The selected group includes MM1, MMP8, MMP13, MMP14, and cathepsin K.³⁵⁶ The uniqueness of cathepsin K among the cathepsins resides in its extraordinary capacity for degrading both the telopeptides-globular structures located at the ends of native molecules of collagen- and the triple helix chains.^{222,351,357,358} Other cysteine cathepsins fail in this task. Cysteine cathepsins have telopeptidase activity, while MMPs are only able to degrade the triple helices of collagens. The difference between CK and the rest of the papain family is found in its catalytic active site. Cathepsins L, K, V, and S belong to the cathepsin L-like subfamily. They all share similar structural characteristics, but there are differences in their substrate specificity. Cathepsin L tends to favor aromatic hydrophobic residues for peptidic cleavage at the P2 position (i.e. N-Phe-X-C); on the other hand, cathepsins S and K

might cleave any aliphatic residues (i.e. N-Leu-X-C). The main difference between cathepsin L and K is their specificity at the P2 position. Cathepsin L can accept both aliphatic and aromatic residues, while cathepsin K's position only favors aliphatic amino acids residues in the P2 position (i.e. N-Leu-X-C or N-Ile-X-C). Furthermore, cathepsin K is able to cleave peptides with proline and glycine at the P2 and P3 positions, which are unusual residues for substrates of cysteine cathepsins.^{225,359} This explains why proline-rich macromolecules, such as type I and II collagens, are excellent cathepsin K substrates.³⁶⁰

Biological Roles of Cathepsin K

Cathepsin K distribution is selective in human tissues. Therefore, its physiological function is limited to a specific number of processes.

The most important, and perhaps the best known, process is the participation of cathepsin K in bone resorption. Type I collagen is one of the most abundant fibrillar proteins in the bone matrix. This fibrillar protein has a molecular size of almost 300 kDa. Type I collagen (TIC) is composed of three helical chains, and telopeptides, the end of these chains.³⁶¹ MMPs with collagenase activity lack telopeptidase activity. This is not the case for cathepsin K, which has both activities combined (collagenase, telopeptidase).^{351,362,363}

The Collagen Family: Type I and Type IV Collagens

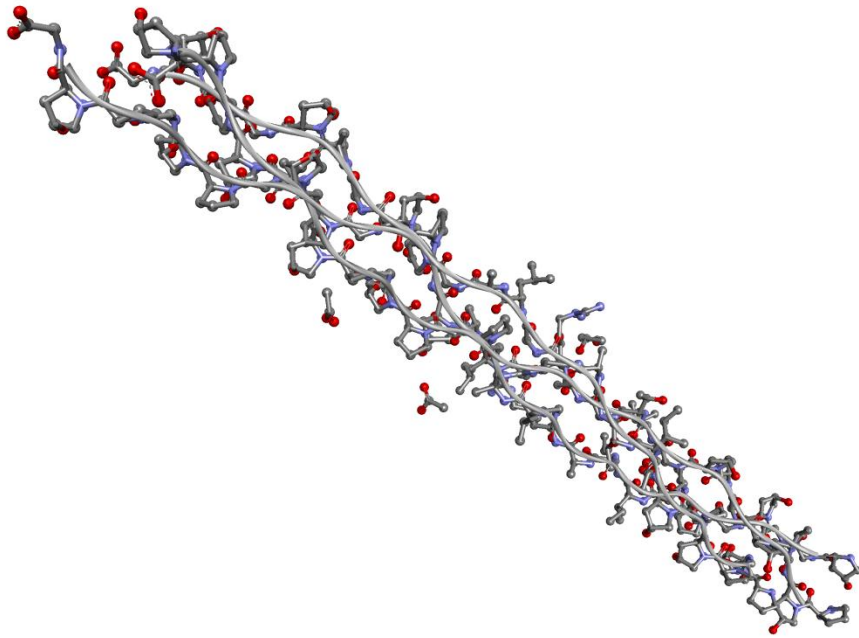
The collagen family comprises, to date, twenty-eight types of this fibrillar protein.³³³ However, the collagens share twenty-five α -chains. There are several characteristics that make collagens one of the most studied proteins. Even though the collagen are practically insoluble, their versatility is very well known and are used in

cosmetics, biomaterials, tumorigenic studies, drug delivery, ulcer treatment among other applications.^{364,365} Collagens' amino acid sequence is quite simple, yet efficient: Gly-X-Y. X and Y are usually proline and hydroxyproline (Hyp) residues. The roles of each amino acid, glycine, proline and hydroxyproline are well known. Glycine side chain, a hydrogen atom gives the characteristic compactness to the molecule. The side chain is able to fit into the tight conformation formed by three helical chains. Also, hydrogen bonds between NH and C=O groups of two adjacent helical chains backbone are formed. Both, proline and hydroxyproline provide the characteristic kinks found in the molecule, due to the fixed angle formed in the amino acid structure (C-N).²²⁴

Tropocollagen, or simply collagen is made of three left handed α -chains that form a right-handed macromolecule. The main function of the collagens is to provide support and anchor surrounding cells within the body, such as connective tissues, bones, and extracellular matrix. It was believed that fibroblasts, connective tissues cells, were the only ones responsible for the expression of collagen. Nevertheless, additional research has demonstrated that other epithelial cells can also express several types of collagens. Figure 74 shows the crystal structure of type I collagen (PDB ID: 1CAG)³⁶⁶ and the non collagen domain of type IV collagen (PDB ID: 1T61)³⁶⁷

Collagens can be classified by their α -chain composition (homotrimeric or heterotrimeric), and quaternary structures (or form of polymerization): fibrillar, fibril associated with interrupted triple-helices (FACIT), short chain, basement membrane, multiplexins, membrane associated with interrupted triple-helices (MACIT).^{333,368}

A.



B.

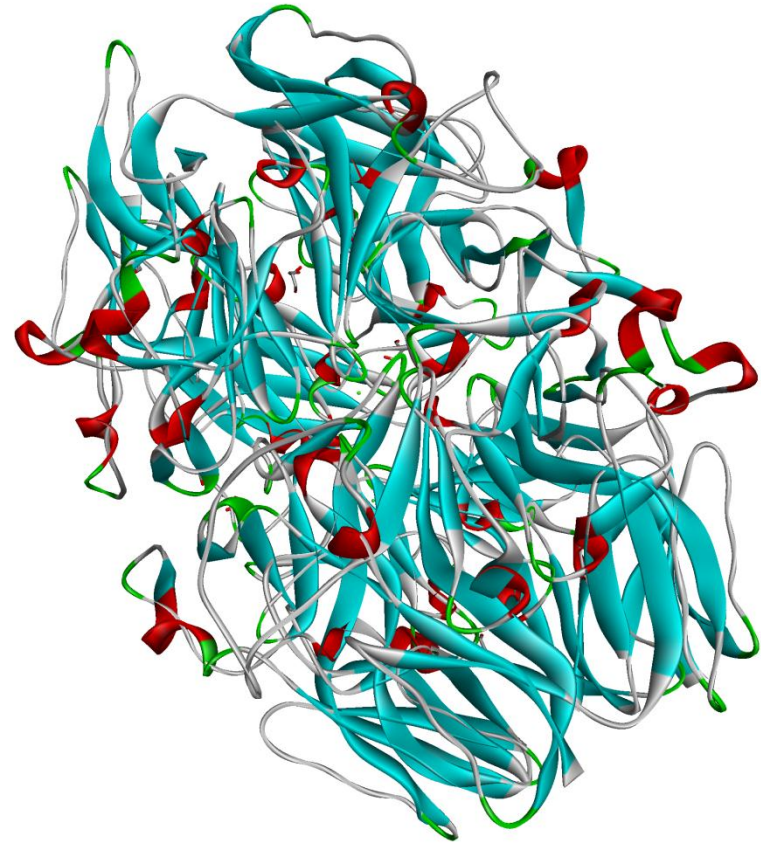


Figure 74. Crystal Structure of Collagen. **A** Type I Collagen Triple Helix Structure (PDB ID: 1CAG). **B**. Type IV Collagen NC1 Domain from Human Placenta (PDB ID: 1T61).

Type I collagen (TIC) is considered a heterotrimeric, fibril-forming macromolecule with a molecular formula of $[\alpha 1(I)]_2 \alpha 2(I)$. It is the most abundant member of this family (up to 90%) and is widely found in extracellular matrix, bone, skin, tendons, ligaments, cornea, etc. The average length of each α -chain is approximately 1400 residues (approximately 300 nm, and 0.5- 3 μ m diameter in their aggregated form).³⁶⁸

Type IV collagen (TIVC) is also a heterotrimeric, network forming macromolecule with a molecular formula of $[\alpha 1(IV)]_2 \alpha 2(IV)$. TIVC is only found in the basal lamina, which is a layer of the extracellular matrix.

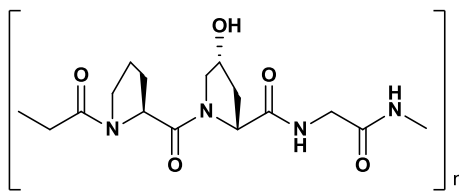


Figure 75. Representative Amino Sequence of the Collagens³⁶⁹. The sequence represents Pro-Hyp-Gly

Similar to type I collagen, TIVC's three α -chains contain 1400 residue per chain. Each one of them is composed of three different domains: an amino-terminal, the Gly-Pro-Hyp main chain and the C-terminal which contains approximately 230 residues that have non-collagen characteristics. The N-terminal domain is rich in cysteine and lysine residues. Both amino acids are responsible for the interchain crosslinking between four triple helical molecules through disulfide bonds and lysine-hydroxylysine bonds. The strong interactions between the triple helices provide the macromolecule high stability and low susceptibility to proteolytic degradation. Perhaps the unique feature of TIVC is the presence of a pattern of interruptions that occur in the main chain (~ 21-26). The

interruptions provide the macromolecule the ability to form two-dimensional networks rather than usual 3-D structures that are more common to find within other members of the collagen family, such as TIC. Khoshnoodi offered an extensive review about mammalian collagen IV.³⁷⁰

The Effect of Chondroitin 4-Sulfate on Cathepsin K Collagenase Activity

The collagenase activity of cathepsin K is greatly enhanced by the presence of chondroitin 4-sulfate (C4-S), a glycosaminoglycan.³⁷¹ In contrast, the macromolecule acts as an inhibitor of other member of the cathepsin family, like cathepsin L and S, which share structural similarities with cathepsin K.³⁷² The crystal structure of cathepsin K complexed with E-64 and C4-S was solved by Li and coworkers (Figure 76).³⁷³ Their findings can be summarized as follows:

1. E-64 binds to Cys25 via covalent modification. (Cys25 thiol group – E64 epoxide ring. E-64 occupies cathepsin K's S2 and S3 pockets)
2. C4-S binds to the R-domain of cathepsin K which is located on the opposite side of the protease's active site.
3. Hydrogen bonds and ion-pairs are the responsible of stabilizing the complex formed by cathepsin K and C4-S.
4. The backbones and/or side chains of D6, R8, K9, K10, I171, E172 and N190 interact with C4-S moieties.
5. The minimal ratio between C4-S and cathepsin K was calculated to be 1:3.

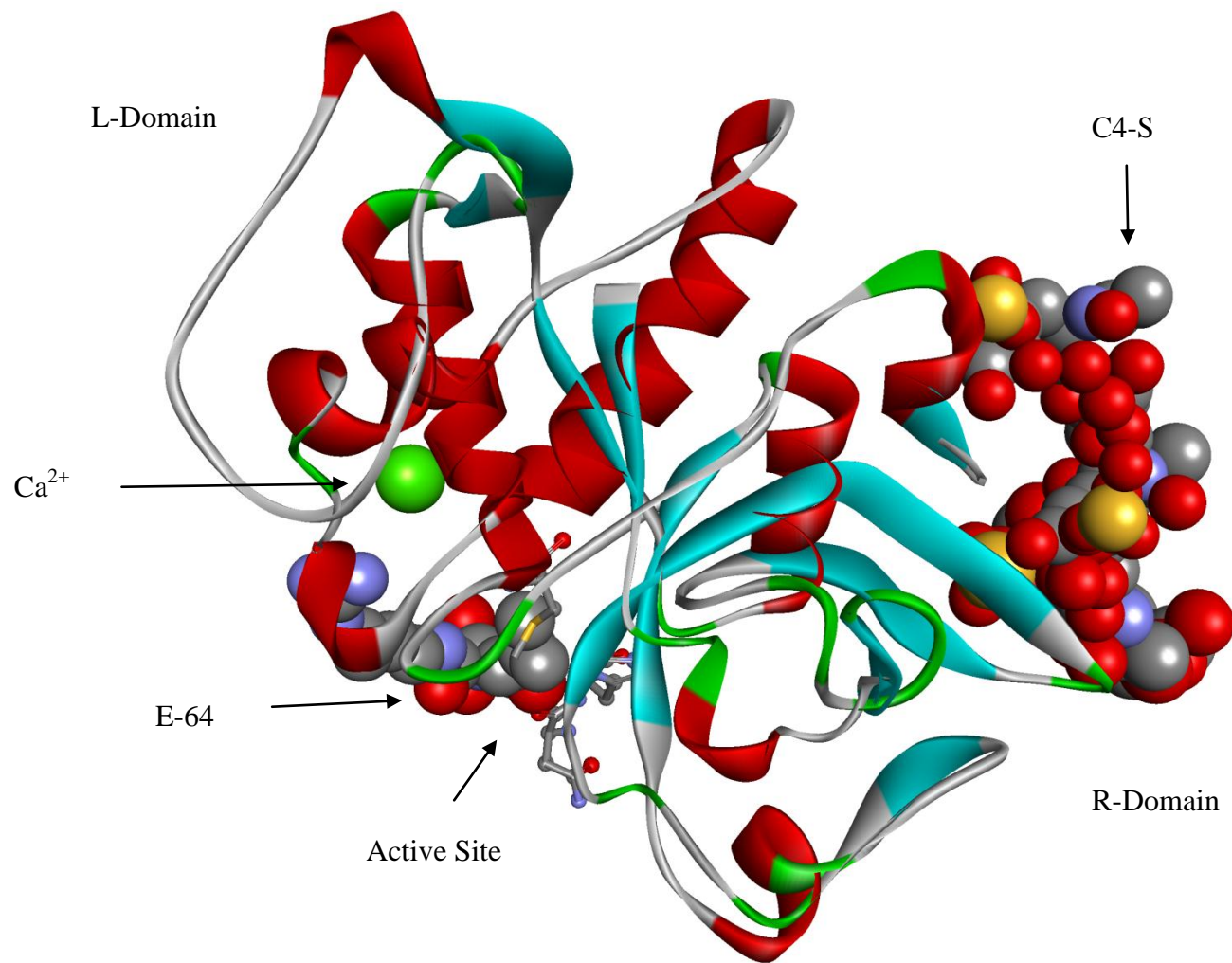


Figure 76. Crystal Structure of Cathepsin K Complexed with E-64 and Chondroitin 4-Sulfate (PDB ID: 3C9E)

Catalytic Activity of Cathepsin K and Substrate Specificity

Cathepsin K possesses similar characteristics to cathepsin L. Their amino acid sequence is 51% identical.¹³⁵ Thus, it is not surprising that share similar substrate specificities. However, there are subtle differences as pointed earlier in this chapter. Cathepsin K prefers substrates with amino acid residues containing aliphatic and hydrophobic (i.e. Ala, Ile, Leu) residues at the second position prior to the peptidic cleavage. Furthermore, proline is a very unusual residue for peptidic cleavage at the P2 position. Few enzymes are able to cleave these residues.³⁷⁴ Cathepsin K is the only cysteine cathepsin that has a preference for proline at the aforementioned position (X-Pro---X).^{359,375}

Structure of Cathepsin K

Cathepsin K is a homodimer containing 215 residues per monomer with a molecular weight between 25 and 29 kDa.¹⁹³ Amino acid sequences of both cysteine cathepsins (L and K) have been identified and confirmed. Literature reports identities between 50 and 60 percent.¹³⁵ Figure 77 shows the amino acid sequence of mature cathepsin K containing 215 residues.³⁷⁶ The three most abundant residues are glycine, asparagine and arginine (24, 19, and 19 residues respectively). On the other hand, the least abundant residues are histidine (2 residues), tryptophan and methionine (four residues each). Similar to cathepsin L, hydrophobic residues account for almost one third of the total composition of cathepsin K. Also, 44 percent of the composition is entirely made of polar, uncharged residues. The secondary structure also reveals that eight alpha helices accounting for 27% of the structure. However, 23% of the sequence forms beta sheets distributed in seventeen strands.³⁷⁷

1 APDSVDYRKKGYVTPVKNQGQCGSCWAFSSVGALEGQLKKKTGKLLNLSP
 51 QNLVDCVSENDGCGGGYMTNAFQYVQKNRGIDSEAYPYVVGQEESCMYNP
 101 TGKAAKCRGYREIPEGNEKALKRAVARVGPVSVDAIDSLTSFQFYSKGVY
 151 YDESCNSDNLNHAVLAVGYGIQKGNKHWIKNWSWGENWGNKGYILMARNK
 201 NNACGIANLASFPKM

Figure 77. Amino Acid Sequence of Cathepsin K (PDB ID: 3KX1).³⁷⁶

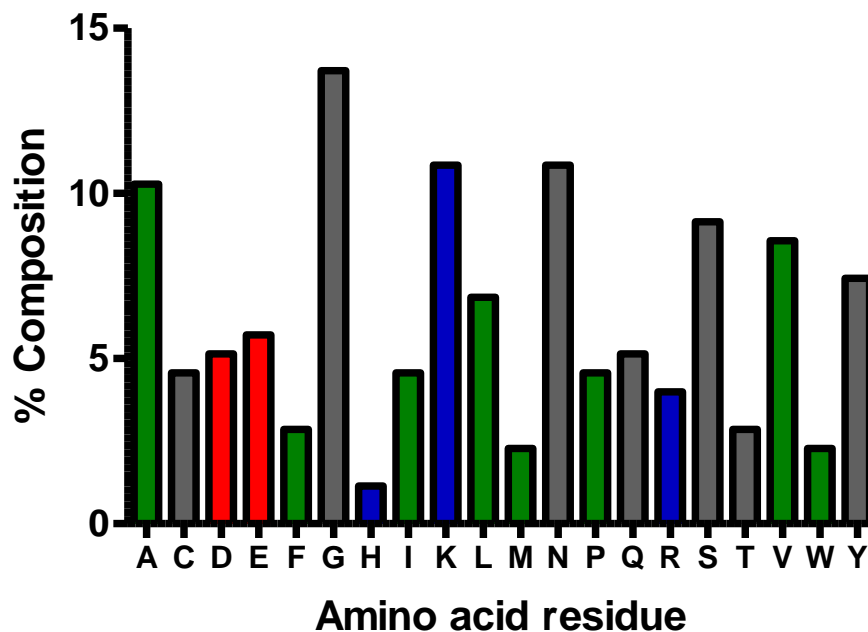


Figure 78. Composition of Human Cathepsin K. Legend= Blue: Basic Residues; Red: Acidic Residues; Green: Hydrophobic (Non Polar) Residues; Gray: Neutral, Polar Residues

Table 70. Amino acid Composition of Cathepsin K.

AA	#	%	AA	#	%	AA	#	%
Ala	18	8.37	Ile	8	3.72	Arg	7	3.26
Cys	8	3.72	Lys	19	8.84	Ser	16	7.44
Asp	9	4.19	Leu	12	5.58	Thr	5	2.33
Glu	10	4.65	Met	4	1.86	Val	15	6.98
Phe	5	2.33	Asn	19	8.84	Trp	4	1.86
Gly	24	11.16	Pro	8	3.72	Tyr	13	6.05
His	2	0.93	Gln	9	4.19	Total	215	100

Crystal Structure of Cathepsin K

One of the first crystal structures of human cathepsin K was resolved by McGrath and coworkers in 1996, two years after the protease was discovered in rabbits.¹²⁴ The macromolecule was crystallized with an inhibitor, and important interactions were determined between cathepsin K and the inhibitor. Mature cathepsin K is a macromolecule with 215 amino acids. The approximate size of the crystal was determined as 55 X 35 X 37 Å. In the structure, Cys25 and His159, two of the members of the catalytic triad appeared in the form of an ion pair (thiolate-imidazolium pair). The third residue, Asn175 was also found but protected by Trp77 (papain numbering). The overall structure contains several pockets; some of them are well defined while others were poorly formed. Seven of these pockets interact with potential cysteine protease inhibitors.

Eight amino acid residues located on the surface of the molecule could not be fully characterized. These residues are Ser4, Lys41, Lys44, Glu59, Lys77, Lys119, Arg127, and Lys147. However, comparisons with similar cysteine cathepsin structures revealed little deviation between them. Important interactions were found at the active site of cathepsin K. For example, one hydrogen bond is formed between the carboxyl oxygen of Asn175 and one of the nitrogens of the imidazole ring of His159. The crystal structure also reveals a close proximity between one of the His159 nitrogens and the thiol group of Cys25 (*N-S*: 3.65 Å). The importance of this inhibitor-cathepsin K complex resides in the interaction of the inhibitor (vinyl sulfone) that binds to the macromolecule as a substrate. The Cys25 thiol moiety attacks a vinyl carbon of the carbon that mimics the substrate carbonyl carbon in a possible substrate. The crystal structure of cathepsin K

inhibited by the vinyl sulfone is shown in Figure 79. Also, Figure 80 shows the crystal structure of uninhibited cathepsin K.^{376,377} Three hydrogen bonds are formed between inhibitor and cathepsin K Gly65, Gly66, and Asn158 residues. Additionally, van der Waal contacts were detected with Gly64, Gly65, Cys63, and Gly23. The almost-hydrophobic S2 subsite is formed by six residues, (Tyr67, Met, 68, Ala133, Leu157, Ala160, and Leu205). The presence of Leu205 makes the S2 pocket be extremely shallow for bulky residues, such as phenylalanine. On the other hand, the S3 pocket is made only with two residues: Tyr67 and Asp61. The P1' pocket is made with side chains of Asn158, Trp177, and Ala136. Trp177, the protective group of Asn175, appears to be in close proximity to the substrate as well. Finally, the key signature of cysteine proteases, a disulfide bond between two cysteine residues was found in cathepsin K (Cys153 and Cys200). Figure 80 shows the crystal structure of human cathepsin K. Three active residues (Cys25, His159, and Asn175) are shown in ball and stick form. The structure is colored according to the following: α -helices are red, β -sheets are cyan, turns are green, and coils are white PDB: 3KX1.³⁷⁶

General Considerations of Procathepsin K

Lalonde and coworkers elucidated the crystal structure of human procathepsin K.¹³⁸ Again, the macromolecule is formed by a 99-residue (the propeptide region) and the mature form of 215 residues). The mature form was found to have similar characteristics without major conformational changes when the propeptide is present. The complete amino acid sequence is presented in Figure 82. The proregion has a globular-shaped form and is made with four secondary structures: 3 α -helices and 1 β -sheet. The proregion is located above the β -sheet domain of the mature cathepsin K.

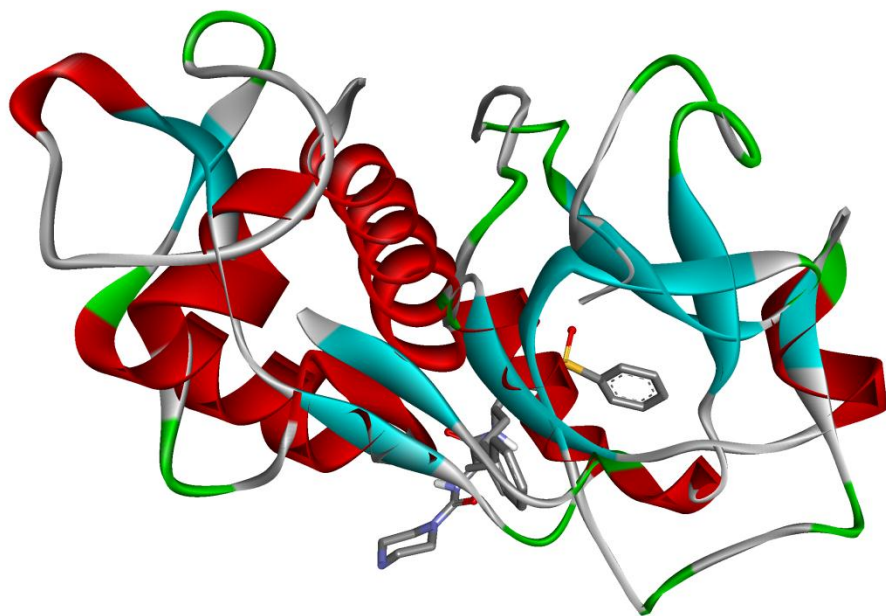
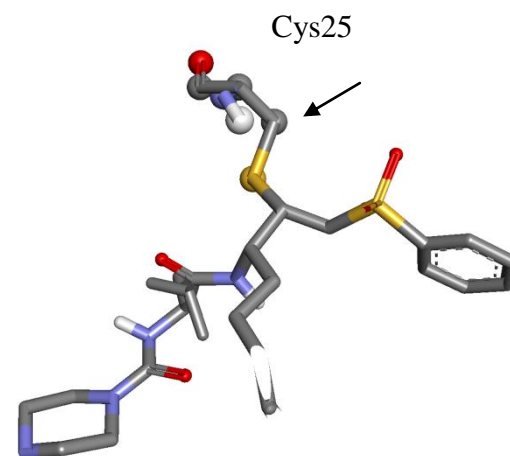
A**B**

Figure 79. Crystal Structure of Human Cathepsin K with Z-Phe-Tyr(OBut)-COCHO, a Reversible, Slow, Tight Binding Inhibitor. **A.** Overall Structure. **B.** Covalent Modification of Cys25 by the Inhibitor (PDB ID: 1MEM)

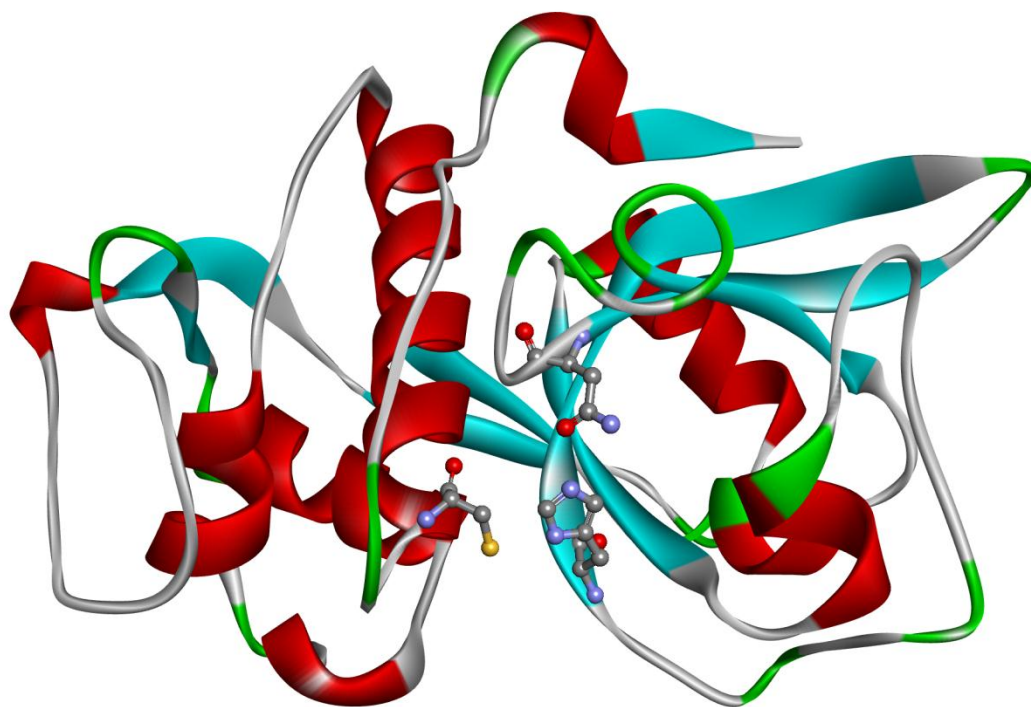


Figure 80. Crystal Structure of Human Cathepsin K (PDB: 3KX1)

The first helix of the propeptide region is an 11-residue almost found to be orthogonal compared to the second helix (27-residues). This change in direction forces the tripeptidic β -sheet to be in close to cathepsin K's β -sheet domain forming a 2-stranded β -sheet. Finally, the fourth helix (9 residues) is found right next to the active site forcing the cleft to appear stretched. The investigation also found that the propeptide can be divided into three major groups: the globular domain (69 residues), active site-binding segment (8 residues) and the C-terminal segment with 18 segments.

The globular domain of the propeptide shows hydrophobic and hydrogen bond interactions with mature cathepsin K. For example, the hydrophobic sides of helices 2 and 3 make contact with the nonpolar cathepsin K surface (residues 236-251 in the mature form). The responsible of the active site stretching in the mature form is called

the propeptide binding loops and is a 6-residue α -helix. They also found six hydrogen bonds between main chain and side-chains. Phe243 from the mature form also shows van der Waals interactions between Asn38, Ile42, Leu58, Asn61, Leu63 and Gly64 (all of them form part of the propeptide). Conversely, the propeptidic Tyr56, a hydrophobic residue, interacts with 6 residues found in the mature form (Tyr244, Gly247, Val248, Tyr249, Tyr250 and Asp251). Figure 82 shows the crystal structure of human procathepsin K. Three active residues (Cys25, His159, and Asn175) are shown in ball and stick form. The structure is colored according to the following: α -Helices are red, β -sheets are cyan, turns are green, and coils are white (PDB ID: 1BY8).¹³⁸

Eight residues of the propeptide (p74-p81, p indicates is part of the propeptide) are located close to the active site. This is the region where several electrostatic and hydrogen bonds are found. pThr76 and pMet75 occupy S' subsites, above the catalytic Cys25. On the other hand, pGly77, Leu78 and Lys79 occupy the S subsites. pLys74 also forms hydrogen bonds with Cys20 and Glu19 (mature form numbering). pMet75 interacts with two residues of the mature form: Gln18 and Trp182. Lastly, another hydrogen bond is made between Thr76 and Gln17 of the mature form. A full description of the structure can be found in their presented work.

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1  LYPEEILDTHWELWKKTHRKQYNNKVDEISRRLIWEKNLKYISIHNLEAS
51 LGVHTYELAMNHLGDMTSEEVVQKMTGLKVPLSHSRSNDTYIPEWEGRA
101 PDSVDYRKKGYVTPVKNQGCGSCWAFSSVGALEGQLKKKTGKLLNLSQ
151 NLVDCVSENDGCGGGYMTNAFYVQKNRGIDSEDAYPYVGQEESCMYNPT
201 GKAAKCRGYREIPEGNEKALKRAVARVGPVSVDAIDASLTSFQFYSGVYY
251 DESCNSDNLNHAFLAVGYGIQKGNKHWIKNWGENWGNKGYILMARNKN
301 NACGIANLASFPKM

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Figure 81. Amino Acid Sequence of Procathepsin K (PDB ID: 3KX1).¹³⁸ The Propeptide, a 99-Residue Polypeptide, is Shown in Red.



Figure 82. Crystal Structure of Human Procathepsin K (PDB ID: 1BY8)

Inhibitors of Cathepsin K

Research related to cathepsin K has found numerous inhibitors, natural and synthetic. To date, there are two compounds that have been successfully tried in early clinical phases, and they still are under investigation.

Natural Inhibitors of Cathepsin K

Cathepsin K propeptide is perhaps one of the best inhibitors of this enzyme showing a K_I value of 4 nM. However, it does not show selectivity towards this protease because it also inhibits cathepsins L and S.¹³⁵ Mammalian organisms express glycosaminoglycans in bone tissues. These macromolecules dramatically increase the

collagenase activity of cathepsin K by forming complexes with the protease.

Interestingly, these molecules do not have any effect on the collagenase activity of cathepsins L and S.³⁵⁷ Cystatins, stefins, kininogens, thyropins, and serpins are powerful cathepsin K inhibitors. *In vitro* experiments were carried out demonstrating the inhibitory efficacy of these molecules. For example, the K_I values for Stefin B, Cystain C, L-kininogen and H-kininogen varied between 1 and 15 pM. Finally, α -macroglobulin, a well known inhibitor of other cysteine proteases (cathepsins B, H and L) has not been tested to verify its inhibitory activity towards cathepsin K.

Synthetic Inhibitors of Cathepsin K in Preclinical Trials

The importance of cathepsin in bone resorption has propelled the search for synthetic inhibitors of this cysteine protease. A selected group of these compounds have reached clinical phases in the treatment of osteoporosis. *In vitro* studies show derivatives of cyclohexanones, dihydrofuranones, ketoamides, sulfonamidoketone, ketooxadiazole, and peptidic nitriles are excellent cathepsin K inhibitors that exhibit showing relative selectivity towards the protease.³⁸⁶ Figure 83 presents selected compounds that showed inhibitory activity towards this protease.

Synthetic Inhibitors of Cathepsin K in Clinical Trials

The number of synthetic inhibitors in clinical trials is outstanding. Four compounds have reached clinical trials and some of them have shown promising results.^{158,387} Balicatib is a nitrile-derived compound developed by Novartis. *In vitro* studies showed this compound is a selective inhibitor of cathepsin K.

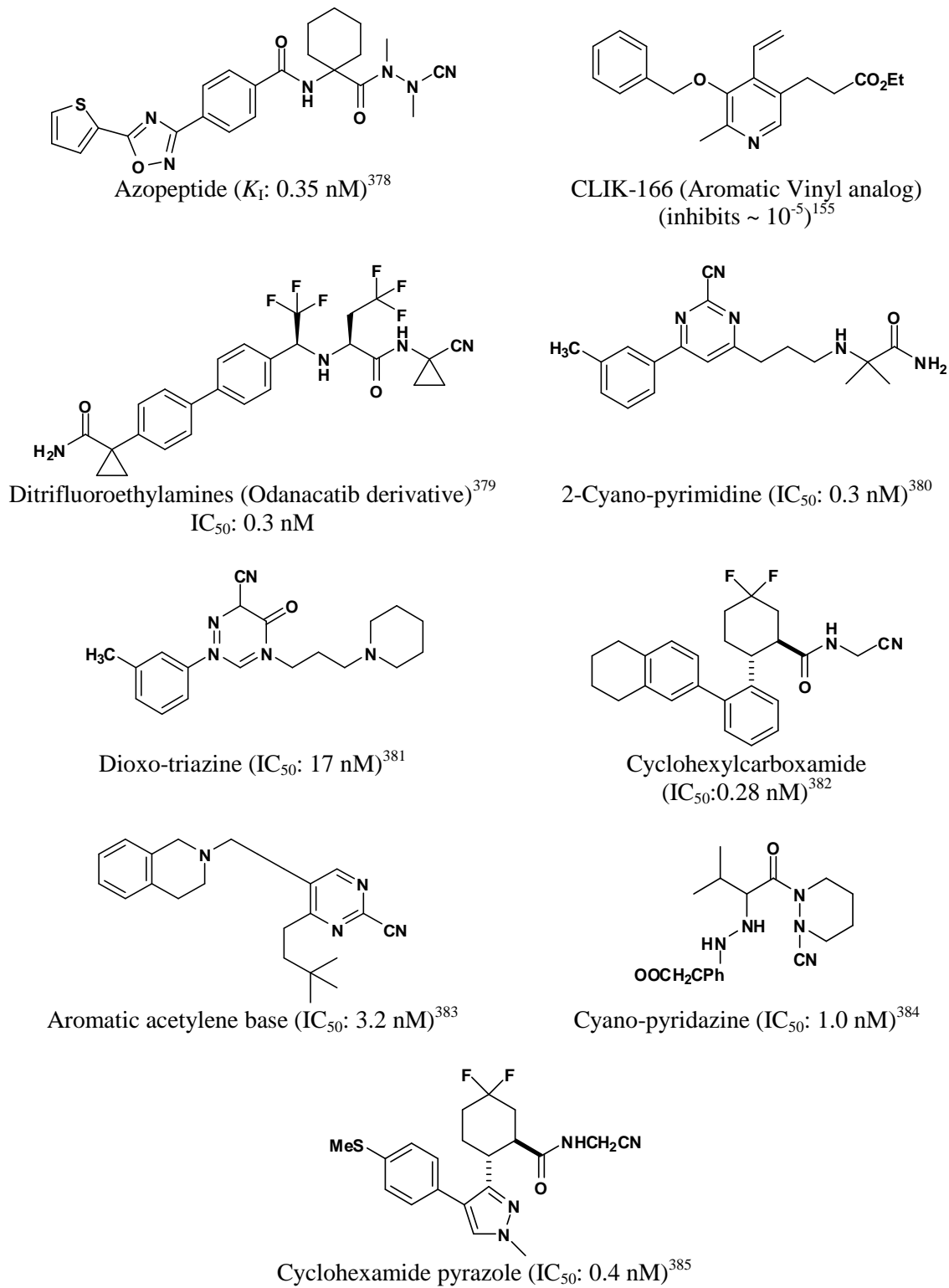


Figure 83. Selected Cathepsin K Inhibitors Reported between 2007 and 2012

Interestingly, the compound loses its selectivity in cell culture studies by inhibiting other cysteine cathepsins. Phase II studies are completed with female patients with osteoarthritis showing promising results in a one year study where bone resorption marker levels were decreased. Clinical trials were discontinued due to adverse side effects seen in patients' skin.³⁸⁸

Glaxo Smith Kline had a cathepsin K inhibitor in its clinical trials (Relacatib, an α -heteroatom cyclic ketone). This inhibitor shows no selectivity against cathepsin L-like proteases (cathepsins L, K, V). Studies were stopped during phase I trials due to possible interactions with acetaminophen, ibuprofen, and atorvastatin. These compounds are usually prescribed to people with osteoporosis and osteoarthritis).³⁸⁹

Ono Pharmaceutical has ONO-5334, a compound with modest results in phase II trials. Their results showed a dramatic decrease in bone resorption markers only in patients that were treated with high doses of the compound. However, reports of side effects were less severe in treated patients.³⁹⁰

The most promising inhibitor of cathepsin K is Odanacatib. The nitrile-based analog is one of Merck's most studied compounds. The compound shows a high selectivity against cathepsin K. The efficacy of this compound is due to the reversible thioimide formed between odanacatib's nitrile moiety and cathepsin K Cys25 thiol group.^{391,392} The advantages of this compound are that patients tend to tolerate Odanacatib quite well; with minor side effects, and have a long half life, between 3-4 days. Daily doses (2.5 mg/patient/day) showed a significant decrease in bone resorption markers. Odanacatib is considered an excellent candidate for osteoarthritis patients because it does not interfere with the bone formation processes, a situation that has been

found with other candidates such as alendronate and denosumab. Currently, Odanacatib is under investigation in Phase III clinical trials.^{393–396}

Importance of Cathepsin K in Medicine

Cathepsin K is one of the most studied cysteine proteases along with Cathepsin L and B. Its powerful hydrolytic activity that degrades components of the ECM is unique and fascinating. Cathepsin K is mainly involved in bone remodeling and resorption. However, it is well known that overexpression of this protease in specific tissues and processes promote the development of serious pathological diseases (See Table 9). In summary, Cathepsin K is involved in thyroidal and bone-related conditions. Bone related diseases have major implications in health issues across the country. Prostate, breast, and skin cancer tend to metastasize to the bone and patient survival probabilities tend to decrease because current treatments fail to delay or arrest bone metastasis. Additionally, overexpression of proteases is an indication of excessive osteoclast activity. This proteolytic activity led to pathological conditions such as osteoporosis and osteoarthritis. Osteoporosis is characterized for excessive fragility in bone tissue and it is observed more often in female patients that are 55 years old. Overall, an alarming number of 10 million people older than 50 years will develop osteoporosis. Indirect and direct costs of people with osteoporosis represent more than 19 billion of dollars.

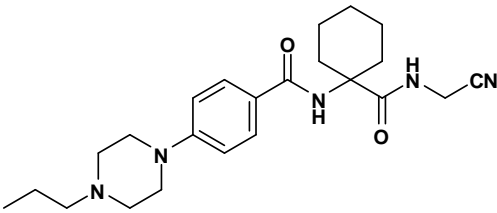
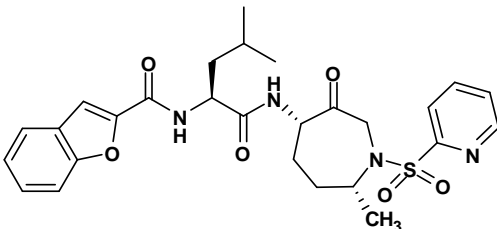
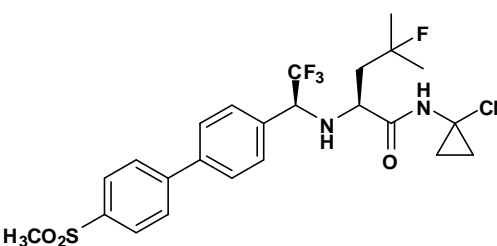
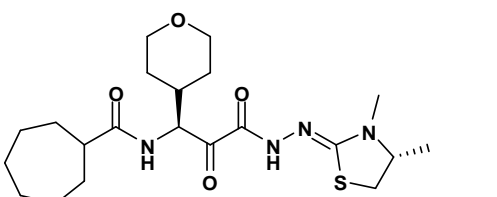
Compound	Name	Phase	CathK	CathB	CathL	CathS	CathV	CathF
	Balicatib	II	IC ₅₀ : 1.4	IC ₅₀ : 4800	IC ₅₀ : 503	IC ₅₀ : 65000	Not determined	Not determined
	Relacatib	I	K _I : 0.041	K _I : 13	K _I : 0.068	K _I : 1.6	K _I : 0.063	Not determined
	Odanacatib	III	IC ₅₀ : 0.2	IC ₅₀ : 1034	IC ₅₀ : 2295	IC ₅₀ : 60	IC ₅₀ : 762	IC ₅₀ : 795
	ONO-5334	I	K _I : 0.1	K _I : 32	K _I : 1.7	Not determined	Not determined	Not determined

Table 74. IC₅₀ and K_I Values of Cathepsin K Inhibitors in Clinical Trials.^{397,398} Values are in nM

Regulation of Cathepsin K in Osteoclasts

Little is known about cathepsin K and its implication in signal transduction and its participation in cellular processes. Most of the work done related to the protease involves its proteolytic activity in osteoclasts by degrading numerous fibrillar proteins that form part of the extracellular matrix. However, the expression of cathepsin K gene in osteoclasts is tightly regulated by numerous activators and inhibitors. Troen offers a comprehensive explanation of the cell signaling involving activators or inhibitors of the expression of the enzyme.³⁹⁹

RANKL is the main regulator of cathepsin K due to its participation in the differentiation and activity of the osteoblasts. Other stimulators include vitamin D, glucocorticoids, interleukins, prostaglandin, histamine, etc. Therefore, it is not surprising that RANKL inhibitors also suppress the activity and expression of cathepsin K.⁴⁰⁰

The receptor activator of NF- κ B ligand stimulates cathepsin K mRNA expression in human osteoclasts. Additionally, studies have demonstrated the direct inhibition of the expression of cathepsin K. Calcitonin, IFN- γ , estradiol, and calcium are some examples that have been studied.

Cathepsin K and Thyroid-Related Diseases

The thyroid is one of the most important organs regulating hormonal processes. It requires specialized proteins for hormone processing. Cathepsin K is one of the members of a selected group with thyroglobulinase activity. High concentrations of cathepsin K have been found in patients with thyroid cancer when compared to a sample of healthy

people. Animal studies also revealed that Cathepsin K deficiency is translated into longer small intestines.^{111,401}

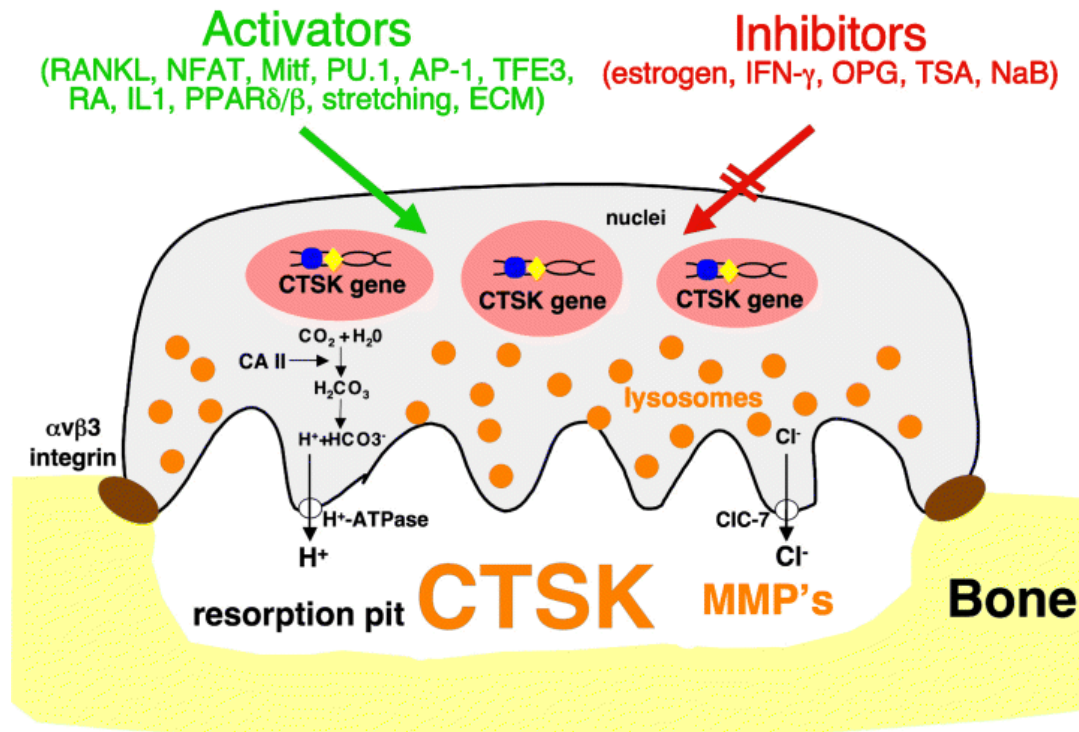


Figure 84. Regulation of Cathepsin K (CTSK) Gene Expression in Osteoclasts (Reproduced from Troen, page 169)

Cathepsin K and Bone Resorption

Bone remodeling is intrinsically related to bone resorption, a function that is tightly regulated by osteoclasts. It is also proven that abnormal osteoclast activity is responsible for pathological conditions such as osteoporosis, osteoarthritis, and bone metastasis. In summary, the role of these specialized cells is the degradation of organic bone material and minerals found in bone tissues.

The Resorption Cycle

The osteoclastic activity can be summarized in what is called the resorption cycle. Kalervo and coworkers explained in recent reviews the steps of this cycle.⁴⁰² Full reviews can also be found elsewhere.⁴⁰³

1. Osteoclasts migrate to the 'resorption cycle' and attach to the bone tissue.
2. Osteoclasts undergo a self-polarization and form new membrane domains
3. Osteoclasts dissolve hydroxyapatite, the mineral component of bones
4. Osteoclasts degrade the organic bone material
5. Cells perform a housekeeping process by removing the degraded material
6. Cells undergo apoptosis or migrate to a non-resorbing stage

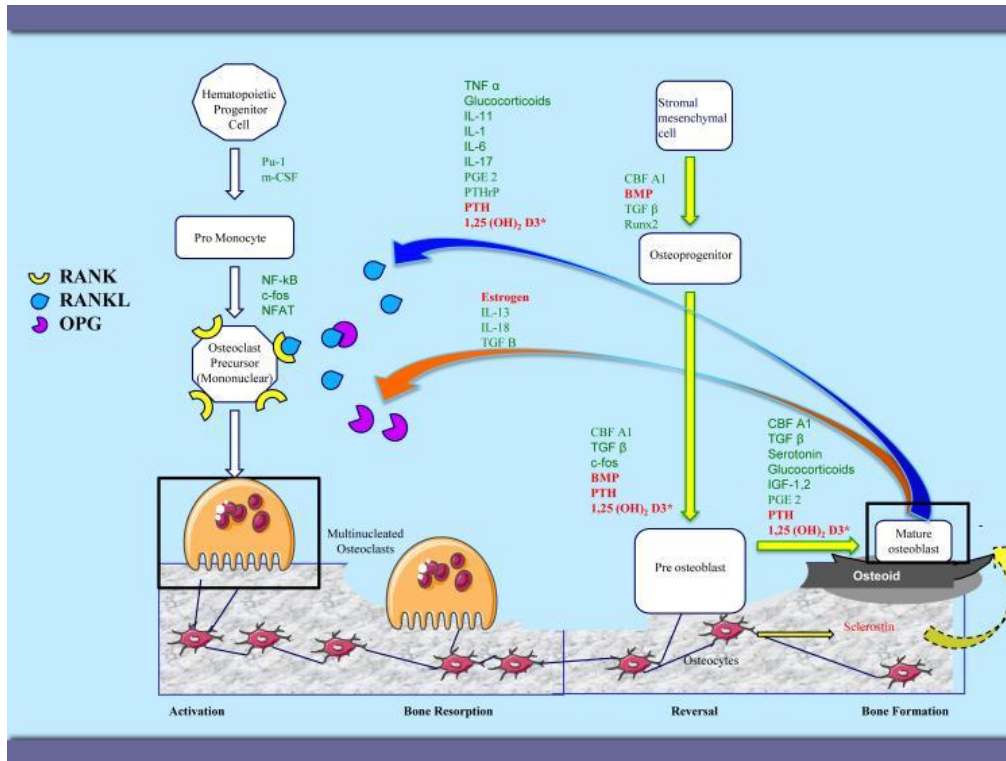


Figure 85. Bone Remodeling Showing the Various Stages and the Factors Involved. Also Shown is the Development of Osteoblasts and Osteoclasts from Precursors. (reproduced from Gallager, page 303⁴⁰⁴)

It has been showed that osteoclast expresses significant levels of cathepsin K in normal conditions. The activity of cathepsin K and other important metalloproteases is required in the fourth step. These enzymes are responsible for the proteolytic degradation of the organic bone material. Specifically, MMP-9 and cathepsin K are the main players in the resorption cycle. Furthermore, the expression of cathepsin K accounts for 98% of the cysteine proteases found in osteoclasts. Its activity has been found in the lysosomes, ruffled border and resorption lacuna (See The resorption Cycle, page 210).⁴⁰⁵

Cathepsin K and Bone Remodeling

Bone remodeling comprises a complex cycle where bone tissues are ‘formed’ and ‘degraded’. Both processes are necessary in healthy bone tissues and help in bone growth, bone fracture, etc. This is a lengthy process taking approximately six months to be completed. Proteases, such a metalloproteases and cysteine proteases are important players in the normal development of bone processes. Overexpression of cathepsin K in animal studies revealed an accelerated degradation of the organic bone material, a clear indication of osteoporosis.⁴⁰⁶ On the other hand, pycnodysosis is a pathological disease characterized by short stature and osteopretosis. Pycnodysosis is the result of a mutation in the genetic code that produces sub-expression of Cathepsin K.⁴⁰⁷

Role of Cathepsin K in Osteoporosis

Extensive research has been carried out in the search for selective and efficient cathepsin K inhibitors. Examples of these compounds in clinical trials have been discussed early in this chapter. All of these compounds were developed as potential therapeutic agents against osteoporosis. Osteoclasts are the main cells responsible for the development of osteoporosis in patients suffering this condition. Currently, the reatment

of osteoporosis is limited to antiresorptive and anabolic agents. The list of antiresorptive agents (agents that inhibit the formation of new bone tissue) includes estrogen-based therapy, selective estrogen receptor modulators, calcitonin treatment, and bisphosphonates. In addition, anabolic agents include parathyroid hormone and strontium ranelate, a treatment where the agent is given to patients with a positive outcome in phase III clinical trials.⁴⁰⁴

Role of Cathepsin K in Arthritis

It is not surprising that cathepsin K is also implicated in the development of rheumatoid arthritis in human and animal studies. Two types of stromal cells, fibroblast and chondrocytes, are cells that are strongly linked to this disease. Table 12 shows the expression of cysteine proteases in some of these stromal cells. Furthermore, the expression of cathepsin K has also been found in fibroblasts and chondrocytes.^{160,398,408}

Cathepsin K and Atherosclerosis

One of the least investigated diseases where the cathepsin K role is crucial is atherosclerosis. The role of cysteine cathepsins in cardiac research has been recently reviewed due to their importance in cardio-vascular conditions.⁴⁰⁹ Investigators have demonstrated that cathepsin K is involved in the deterioration of blood vessels walls and the destabilization of plaques.^{410,411}

Bone Metastasis

Bones are one of the most common sites affected by metastasis. Thus, research has also been devoted to study the biology of bone metastasis and its implication in cancer development.^{300,412} Breast and prostate cancer, the leading types of cancer in

female and male patients, usually metastasize to bone as their primary metastatic site. Once again, osteoclasts are mainly involved in the metastatic colonization of the bone. However, metastasis in female and male patients with breast or prostate cancer is different. For example, bone metastasis in female cancer is predominantly osteolytic, while osteogenic lesions are found in male patients. Cysteine cathepsins and metalloproteases are also responsible for the facilitation of bone metastasis. These two types of proteases greatly influence and modify the tumor microenvironment promoting the dispersion and generation of new metastatic lesions. Specifically, overexpression of cathepsin K in osteoclast cells in breast cancer samples facilitates cell migration, cell invasion, adhesion to a vessel, and extravasation to the final destination (metastatic organ). The search for excellent antimetastatic agents is not exclusively against cathepsin K. In depth investigations are also being carried out for agents inhibiting the activity of Src, TFG β , MMPs, RANKL ET-1, uPA, DKK-1, WNT, etc. Figure 86 shows some of the most important targets in bone metastasis research that are currently in extensive studies and their most important inhibitors.

Cathepsin K and Skin Cancer

The evidence of cathepsin K expression in normal skin tissue has led to the evaluation of cathepsin K in melanoma and other skin cancer diseases. This form of cancer is extremely aggressive and also metastasizes to bone as well. Recent studies showed higher levels of cathepsin K in squamous cell carcinoma in more than 100 samples of benign and malignant skin cancer tumors. The invasiveness of this type of cancer was also investigated and concluded that the high aggressiveness that

characterizes this type of tumor is partially due to the overexpression of cathepsin K in fibroblast cells.^{350,354,413}

Implications of Cathepsin K in Prostate Cancer

Cathepsin K activity was detected by several techniques, including reverse-transcription –polymerase chain reaction, western blotting, and immunochemistry in prostate cancer cell lines and tumors from patients. However, there are no recent studies linking the proteolytic activity of cathepsin K to prostate cancers.⁴¹⁴ Treatments with denosumab, a RANKL inhibitor, and bisphosphonates have shown modest results in patients with metastasized prostate cancer.⁴¹⁵ In addition, SRC, cABL kinases, and cathepsin K inhibitors have been under investigation to verify their potency against prostate cancer.

Cathepsin K and Breast Cancer

The importance of cathepsin K in breast cancer progression is more evident. There is strong evidence that numerous breast carcinomas samples express cathepsin K. Inhibition studies with animals showed that cathepsin K played an important role in bone metastasis. BT474 cells (invasive ductal breast carcinoma) were implanted in mice. A cathepsin K inhibitor significantly reduced osteolytic lesions when a cathepsin K inhibitor was administered in clinical doses.⁴¹⁶ Additional investigation links the activity of cathepsin K in angiogenesis and expression of pro-inflammatory cytokines as well.⁴¹⁷

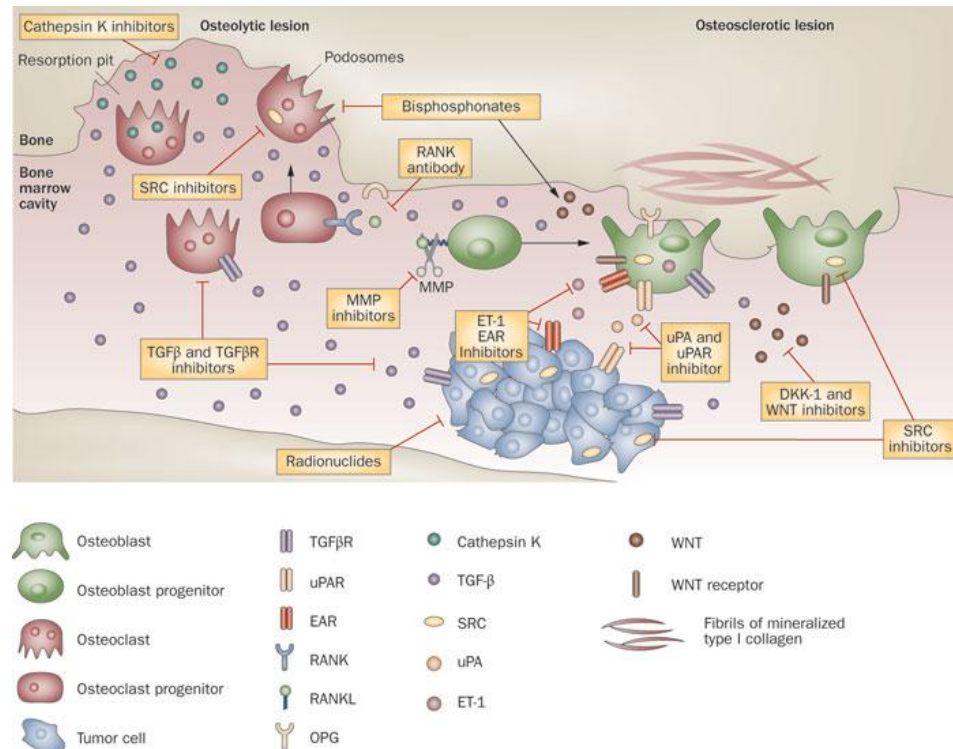


Figure 86. Bone-targeted therapy in Metastatic Lesions. (reproduced from Sturge, page 359)

Role of Collagens in Cell Signaling and Cancer

It has been demonstrated that both type I and type IV collagens participate in numerous cell-related processes including cell adhesion, migration, survival proliferation and differentiation.³⁷⁰ Type IV is one of the main substrates of numerous types of cells (i.e. hepatocytes, keratinocytes, and tumor cells such as breast and prostate carcinoma, melanoma, fibrosarcoma and gliomas).

Cell attachment to the collagens is via integrins, which act as mediators between cells and their surrounding microenvironment. In general, integrins can bind to specific peptide sequences found in ECM proteins. The most important integrin-binding sites found within the collagens are located in the main triple-helical chains and the non-collagen domains (NC1D).

The two most important integrin receptors of type I and type IV collagen triple helical chains are $\alpha_1\beta_1$ and $\alpha_2\beta_1$. One of the most important integrins binding recognition sites is the Arg-Gly-Asp peptide and is also found in collagens. However, cell adhesion in type IV is independent because integrins cannot reach that site found deep in the triple helical chain of the macromolecule. Both integrins belong to the β_1 subfamily. The first integrin, $\alpha_1\beta_1$ shows more affinity toward type IV collagen, while the latter one prefers type I collagen. Research showed that deletion of the binding group reduced fibroblastic cell adhesion using type IV collagen as a basement. Other integrins that interact with the main chain are $\alpha_3\beta_1$, $\alpha_{11}\beta_1$, and $\alpha_{10}\beta_1$.

The non collagen domain (NC1D) also increases cell adhesion in collagen. It might also inhibit angiogenesis.

Additionally, collagen degradation is crucial in the diagnosis of diseases. The presence of bone turnover markers in serum and urine has lead to promising investigation and early treatment of diseases including osteoporosis, bone turnover, bone loss, renal failure and Paget's disease.⁴¹⁸

Degradation of type I collagen, the major component of the extracellular matrix produces more than ten biomarkers. Three of them are the aminoterminal (NTX) and the carboxyterminal (CTX-I, and ICTP) cross linked telopeptides of type I collagen. Specific bone resorption markers are especially investigated due to their presence in samples of patients with bone metastasis.⁴¹⁹ Patients with breast or prostate cancer have showed elevated amounts of CTX-I or NTX-I in their samples.^{420,421}

Material and Methods for the Biological Evaluation of Thiosemicarbazones Derivatives as Inhibitors of Cathepsin K

Materials

The comprehensive list of materials and equipment is listed in chapter two under the Materials and Equipment section. Recombinant, human procathepsin K (0.9 µg/µl) was purchased from Enzo. The proenzyme was expressed in insect cells according to the manufacturer. Type IV collagen (0.327 mg/ml) from human placenta was a product from BD Biosciences. Type IV collagen was extracted and purified using proteolytic enzymes. Chondroitin Sulfate A (≥ 60 % pure) sodium salt from bovine trachea is a product from Sigma. Tubulin from calf brain (1.25 mg/ml) was purified by members of the Trawick Group under the guidance and supervision of Dr. Mary-Lynn Trawick.

NuPAGE® MES SDS Running Buffer (20X) and 4-12% Bis-Tris SDS-PAGE are products from Invitrogen. Western blotting detection kit was purchased from Thermo Scientific. The detection kits includes: wash buffer (30X), blocking buffer (1X), PVDF membranes, molecular marker and two secondary antibodies (goat anti-mouse Cy-3 labeled and goat anti-rabbit Cy-5 labeled). Mouse anti-human cathepsin K that can recognize both procathepsin and active cathepsin K was obtained from Sigma. Trans-Blot® SD Semi-Dry Electrophoretic Transfer Cell was from BioRad.

Preparation of Buffers and Stock Solutions

Stock Solutions. Protocol details about 1 M sodium acetate buffer, pH 5.5, 40 mM EDTA, 80 mM DTT, 4 mM AMC, 29.94 mM Z-FR-AMC, thiosemicarbazone derivatives stock solutions, preparation of inhibitor dilutions, and TSC derivatives stock

solutions can be found in chapter two in the preparation of buffers and stock solutions.

Synthesis of thiosemicarbazones have been previously published.^{3,6,7,10-12}

Preparation of 180 mM sodium acetate buffer, pH 5.5. One liter of this solution was prepared by diluting 180 ml of 1 M buffer stock solution with 820 ml of water. Acetic acid glacial and/or sodium hydroxide 5 M (MW: 40 g/mol) was used to adjust the pH of the buffer. This solution was stored at 4 °C to increase its shelf life.

Preparation of 32.5 mM sodium acetate buffer, pH 3.5. A fresh solution of this buffer was prepared by diluting 3.5 mg of solid sodium acetate anhydrous and 43.9 µl of acetic acid glacial in a 25 ml volumetric flask. Acetic acid glacial or sodium hydroxide 5 M (MW: 40 g/mol) were used to adjust the pH of the buffer. This buffer is also referred as solution B.

Procathepsin K. Ten micrograms of human procathepsin K (0.9 µg/µl) were kept at -80 °C to preserve the integrity of the protein.

Preparation of inhibitors dilutions. Each inhibitor was serially diluted in pure DMSO to provide seven different concentrations (named A-G) that were ten-fold diluted. The concentrations of these dilutions varied between 2 mM (solution A) and 2 nM (solution G). Each dilution was diluted ten-fold in a solution that contained 50% DMSO in water (named 1-8). Concentrations of the aqueous solutions varied between 200 µl and 200 pM. An eighth solution was made by using solution C (20 µM) to prepare an intermediate solution (1 µM in 50% DMSO). Solutions made in pure DMSO were stored

for two weeks at (-80 °C). Aqueous solutions were made prior to the experiment and kept at (4 °C).

Preparation of 1 M NaCl. A sample of 584.4 mg of sodium chloride was dissolved in 10 ml of water. This solution was stored at 4 °C to increase its shelf life.

Preparation of 10 mM EDTA. A sample of 37.2 mg of EDTA was dissolved in ten milliliters of water. Alternatively, 2.5 ml of 40 mM EDTA were diluted with 7.5 ml of water. The dilution was stored at 4 °C to increase its shelf life.

Preparation of 60 mM EDTA. A stock concentration of this solution was prepared by dissolving 22.34 mg (60 µmol) of EDTA (MW: 372.29 g/mol) in one milliliter of water. Fifty milliliters of this solution (744.5 mg; 2 mmol) was prepared for multiple experiments. The solution of 40 mM EDTA was stored at 4 °C.

Preparation of 1.9 % chondroitin sulfate A. The solution was prepared by dissolving 0.19 grams of Chondroitin Sulfate A (C4-S) with 10 milliliters of 2 M sodium acetate buffer. The solution was vigorously mixed at room temperature to dissolve the contents. The solution was made prior the experiments.

Preparation of 3% chondroitin 4-sulfate in 50% DMSO. Five milliliters of this solution were prepared by dissolving 0.15 g of C4-S with 2.5 ml of pure DMSO and 2.5 ml of water. The solution was prepared prior to the experiment. Final conditions were 3% C4-S and 50% DMSO.

Preparation of inhibitor dilutions with C4-S. Each inhibitor was serially diluted in pure DMSO to provide seven different concentrations (named A-G) that were ten-fold diluted. The concentrations of these dilutions varied between 2 mM (solution A) and 2 nM (solution G). Each dilution was diluted ten-fold in a solution that contained 50% DMSO in water with C4-S (named 1-8), by adding 90 μ l of 3% Chondroitin 4-Sulfate in 50% DMSO and 10 μ l of stock solution (A-G). Concentrations of the aqueous solutions varied between 200 μ l and 200 pM. An eighth solution was made by using solution C (20 μ M) to prepare an intermediate solution (1 μ M in 50% DMSO). Solutions made in pure DMSO were stored for two weeks at (-80 °C). Aqueous solutions were made prior to the experiment and kept at (4 °C). Final concentrations are 55% DMSO and 2.7% C4-S.

Solution A. Five milliliters of this solution were made by mixing 2.5 ml of 1 M NaCl, 0.5 ml 10 mM EDTA and 2 ml of water. Conditions of this solution are: 0.5 M NaCl and 1 mM EDTA. The dilution was stored at 4 °C to increase its shelf life.

Activation of procathepsin K under acidic conditions at room temperature. The proenzyme was activated by mixing 1 μ l of 0.9 μ g/ μ l (33.3 μ M) procathepsin K, 1 μ l 32.5 mM NaOAc pH 3.5 (solution B) and 4 μ l of solution A. Total volume is 6 μ l. Conditions for activation of the enzyme are: 5.6 μ M of mature cathepsin K, 5.4 mM NaOAc pH 3.5, 0.7 mM EDTA and 333.3 mM NaCl. The mixture was incubated in a siliconized tube at room temperature (24 °C) between 30 and 180 minutes. Activation time varied depending of the proenzyme batch. Activation of the proenzyme was done prior to any experimentation with the mature enzyme.

Preparation of cathepsin K assay buffer. Cathepsin K assay buffer (will be referred as assay buffer for future references in this chapter) contained 3 mM EDTA, 3 mM DTT, 0.01% Brij 35 and 180 mM NaOAc pH 5.5. Ten milliliters of assay buffer were made by adding 750 μ l of 40 mM EDTA, 375 μ l of 80 mM DTT, 4.33 μ l of Brij 35 and 8.9 ml of 180 mM NaOAc pH 5.5 in a plastic 15 ml conical tube. Table 75 summarizes a typical calculation sheet to make different volumes of the assay buffer. The solution was stable and used for a maximum of 24 h.

Table 75. Preparation Table for Cathepsin K Assay Buffer

Assay Buffer (ml)	40 mM EDTA (μ l)	80 mM DTT (μ l)	Brij 35 (μ l)	180 mM NaOAc (μ l)
5	375	187.5	2.2	4436
10	750	375	4.3	8871
20	1500	750	8.7	17743
40	3000	1500	17.3	35482

Preparation of cathepsin K stock solution. A stock solution of cathepsin K (CK) contained 2.5 mM EDTA, 2.5 mM DTT, 1.5 nM cathepsin K and 150 mM NaOAc pH 5.5. One milliliter of cathepsin K stock solution was made by adding 62.5 μ l of 40 mM EDTA, 31.2 μ l of 80 mM DTT, 2.7 μ l of 5.6 μ M activated cathepsin K, 833 μ l of 180 mM NaOAc pH 5.5 and 70.2 μ l of water in disposable glass test tubes. Table 76 summarizes a typical calculation sheet to make different volumes of cathepsin K stock solution. The solution was made prior to every kinetic or inhibition experiment.

Table 76. Preparation Table for Cathepsin K Stock Solution

CK (ml)	40 mM EDTA (μ l)	80 mM DTT (μ l)	Brij 35 (μ l)	180 mM NaOAc (μ l)	Cathepsin K (μ l)	Water (μ l)
1	62.5	31.2	0	833	2.7	70.2
2	125	62.5	0	1666	5.4	140.4
4	250	125	0	3332	10.8	280.8

Preparation of Z-FR-AMC stock solution. A stock solution of Z-FR-AMC (500 μM) in 15 % DMSO was prepared by adding 20 μl of 29.94 mM Z-FR-AMC, 160 μl of DMSO, and 1020 μl of water in order to have a final volume of 1200 μl . The solution was made in a plastic dark 15 ml conical tube. Table 77 shows representative calculations to prepare various amounts of the substrate solution. The stock solution was stable and used for a maximum of 24 hours.

Table 77. Preparation Table for 500 μM Z-FR-AMC for Cathepsin K Assays

500 μM Z-FR-AMC (ml)	29.94 mM Z-FR-AMC (μl)	DMSO (μl)	Water (μ)
1	16.7	133.3	850.1
2.5	41.8	333.3	2125.2
5	83.5	666.6	4250.4
10	167.1	1333.1	8500.7

Assay buffer for reversibility studies (AKR). Three milliliters of AKR were prepared by mixing 900 μl of 500 mM NaOAc pH 5.5, 1.0 μl of Brij 30, 93.8 μl of 80 mM DTT, 187.5 μl of 40 mM EDTA, 115 μl of DMSO, 5.1 μl of 29.94 mM Z-FR-AMC and 1698 μl of water. AKR was prepared prior to the experiment in a 15 ml plastic conical tube kept in the dark. Final conditions of AKR are: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM EDTA, 0.01% Brij 30, 50.5 μM Z-FR-AMC, and 4% DMSO. Table 78 summarizes these conditions.

Table 78. Preparation Table for Cathepsin K Assay Buffer for Reversibility Studies

Volume (ml)	40 mM EDTA (μl)	80 mM DTT (μl)	Brij 35 (μl)	500 mM NaOAc (μl)	29.94 mM Z-FR- AMC (μl)	DMSO (μl)	Water (μl)
3	187.5	93.8	1	900	5.1	115	1698

Preparation of cathepsin K stock solution for reversibility studies (CKR). One hundred and forty-four microliters of CKR were prepared by mixing 86.4 μ l of 500 mM NaOAc pH 5.5, 9.0 μ l of 80 mM DTT, 18 μ l of 40 mM EDTA, 7.8 μ l of 5.6 μ M activated cathepsin K, and 22.8 μ l of water in a glass test tube. Final conditions of CSR are: 300 mM NaOAc pH 5.5, 5 mM EDTA, 5 mM EDTA, and 300 nM cathepsin K. Table 79 summarizes preparation tables for CKR.

Table 79. Preparation Table for Cathepsin K Stock Solution for Reversibility Studies

	Volume (ml)	40 mM EDTA (μ l)	80 mM DTT (μ l)	Brij 35 (μ l)	500 mM NaOAc (μ l)	Cathepsin K (μ l)	Water (μ l)
CKR	0.144	18	9	0	86.4	7.8	22.8

Cathode buffer. One liter of cathode buffer (25 mM Tris base; 40 mM glycine; 0.1% SDS) was prepared by dissolving 3.03 grams of Tris base, 3 grams of glycine and 1 gram of SDS with water. The solution was then filtered using a 0.2 μ m filter and stored at 4 °C. The recorded pH of the solution was 8.8.

Anode I buffer. Half liter (500 ml) of anode buffer (300 mM Tris base; 15 % methanol) was prepared by dissolving 18.2 grams of Tris base, and 75 ml of HPLC grade methanol with water. The solution was filtered with a 0.2 μ m filter and stored at 4 °C. The recorded pH of the solution was 10.4.

Anode II buffer. Half liter (500 ml) of anode buffer (25 mM Tris base; 15 % methanol) was prepared by dissolving 1.5 grams of Tris base, and 75 ml of HPLC grade methanol with water. The solution was filtered with a 0.2 μ m filter and stored at 4 °C.

The recorded pH of the solution was 9.7. Table 80 and Table 81 summarize conditions and procedures for cathode, anode I and anode II buffers.

Table 80. Concentrations of Cathode, Anode I and Anode II Buffers

Buffer	Tris Base (mM)	Gly (mM)	Methanol (%)	SDS (%)	Total (ml)
Cathode	25	40	0	0.1	1000
Anode I	300	0	15	0	500
Anode II	25	0	15	0	500

Table 81. Preparation Table for Cathode, Anode I and Anode II Buffers

Buffer	Tris Base (g)	Gly (g)	Methanol (ml)	SDS (g)	Total (ml)
Cathode	3.03	3.00	0	1	1000
Anode I	18.2	0	75	0	500
Anode II	1.51	0	75	0	500

Preparation of NuPAGE® MES SDS running buffer (1X). Fifty milliliters of the NuPAGE® MES SDS Running Buffer (20X) were diluted with 950 milliliters of ultrapure water in a volumetric flask. Final volume of the solution was one liter. Final conditions of the buffer (1X) were: 50 mM MES, 50 mM Tris Base, 0.1% SDS, and 1 mM EDTA. The pH of the buffer was 7.3 and was stable for up to 6 months when stored at 4 °C.

Preparation of cathepsin K calibration curves for detection limits by using western blotting. Four stock dilutions of inactive cathepsin K (pCKSS) ranging between 10 and 150 ng/μl were prepared by diluting procathepsin K stock solution (0.9 μg/μl ≡ 900 ng/μl) with water as described in Table 82. Then, fourteen samples ranging between 5 and 200 ng were set up by using the former stock solutions. Samples were brought up

to a final volume of 10 μl by adding 2.5 μl of loading buffer 4X Dilution volumes, and masses are shown in Table 83.

Table 82. Preparation Table of Procathepsin K Stock Solutions Detection Limit by Western Blotting Studies

pCKSS	Procathepsin K (ng/ μl)	Dilution Factor	From S (μl)	Water (μl)
S	900	(Stock)	1	0
S1	150	1:6	1	5
S2	50	1:18	1	17
S3	25	1:36	1	35
S4	10	1:90	1	89

Table 83. Preparation Table of Procathepsin K Samples Ranging Between 5 and 200 ng

Sample	Procathepsin K (ng)	From pCKSS	From CKSS (μl)	Loading Buffer (μl)	Water (μl)
1	5	S4	0.5	2.5	7
2	10	S4	1	2.5	6.5
3	15	S4	1.5	2.5	6
4	20	S4	2	2.5	5.5
5	25	S3	1	2.5	6.5
6	30	S3	1.2	2.5	6.3
7	35	S3	1.4	2.5	6.1
8	40	S3	1.6	2.5	5.9
9	45	S2	1.8	2.5	5.7
10	55	S2	1.1	2.5	6.4
11	65	S2	1.3	2.5	6.2
12	100	S1	2	2.5	5.5
13	150	S1	1	2.5	6.5
14	200	S1	1.33	2.5	7

Preparation of 12 mM Sodium Acetate Buffer, pH 3.5. A fresh solution of this buffer was prepared by diluting 369 μl of 32.5 mM sodium acetate buffer, pH 3.5 to a final volume of 1000 μl by adding 631 μl of water. Glacial Acetic acid or sodium hydroxide 5 M (MW: 40 g/mol) were used to adjust the pH of the buffer. This buffer is also referred as modified solution B.

Solution A with DMSO (SA-C). Five milliliters of this solution were made by mixing 3.21 ml of 1 M NaCl, 0.67 ml 10 mM EDTA, 385 μ l DMSO and 738 μ l of water. Conditions of this solution are: 0.64 M NaCl, 1.34 mM EDTA, and 7.68 % DMSO.

Solution A with 1 in DMSO (SA-I). Five milliliters of this solution were made by mixing 3.21 ml of 1 M NaCl, 0.67 ml 10 mM EDTA, 375 μ l DMSO, 9.61 μ l of 20 mM **1** diluted in DMSO and 738 μ l of water. Conditions of this solution are: 0.64 M NaCl, 1.34 mM EDTA, 7.68 % DMSO and 38.5 μ M **1**.

Activation of procathepsin K with DMSO as vehicle control. The proenzyme was activated by mixing 3 μ l of 0.9 μ g/ μ l (33.3 μ M) procathepsin K, 45 μ l 12 mM NaOAc pH 3.5 (modified solution B) and 52 μ l of solution SA-C. Total volume is 100 μ l. Conditions for activation of the enzyme are: 1 μ M of mature cathepsin K, 5.4 mM NaOAc pH 3.5, 0.7 mM EDTA, 333.3 mM NaCl and 4% DMSO. The mixture was incubated in a siliconized tube at 3 °C between zero and seven hours.

Activation of procathepsin K with 1 and DMSO as vehicle control. The proenzyme was activated by mixing 3 μ l of 0.9 μ g/ μ l (33.3 μ M) procathepsin K, 45 μ l 12 mM NaOAc pH 3.5 (modified solution B) and 52 μ l of solution SA-I). Reaction volume is 48 μ l. Conditions for activation of the enzyme are: 5.6 μ M of mature cathepsin K, 5.4 mM NaOAc pH 3.5, 0.7 mM EDTA, 333.3 mM NaCl, 20 μ M **1**, and 4% DMSO. The mixture was incubated in a siliconized tube at 3 °C between zero and seven hours.

Inhibition of cathepsin K proteolytic activity by thiosemicarbazone derivatives. Three stock solutions of human type I collagen ([ACI]: 0.4 mg/ml), human type IV collagen([ACIV]: 0.327 mg/ml) and Tubulin ([TUB]: 1.25 mg/ml) were obtained or

prepared as recently described in the materials and methods of chapters two and three. The three macromolecules were used as cathepsin K substrates. Two inhibitor stock solutions with final concentrations of 20 μM were prepared by mixing 6.25 μl 20 mM stock solution of the inhibitor in DMSO, 243.75 μl of DMSO and 250 μl of water. The experiment was performed by the addition of mature cathepsin K (activated under standard procedures) to a solution containing sodium acetate buffer pH 5.5, EDTA, DTT, C4-S, inhibitor, DMSO and substrate. Final volumes were 12.5 microliters in every case. Similarly, another group of control samples (no inhibitor) were also set up. The degradation of type I collagen, type IV collagen and tubulin were performed at 37 $^{\circ}\text{C}$ and monitored between 0 and 6 hours. Volumes, concentrations of stock solutions and final conditions of the reactions are summarized in Tables 83, 84 and 85. Inactivation of untreated and treated samples, electrophoresis and staining procedures using SYPRO[®] were described in chapter two in the section material and methods.

Table 84. Assay Conditions for the Degradation of Type IV Collagen by Cathepsin K

Reagent	Stock Solution	Volume (μl)	Final Conditions
EDTA	60 mM	0.54	2.6 mM
DTT	80 mM	0.41	2.6 mM
NaOAc	1 M	1.00	80 mM
C4-S	1.9%	1.00	0.15%
Inhibitor (in DMSO)	250 μM	1.00	20 μM
DMSO	50 %		4 %
Type IV Collagen	0.327 $\mu\text{g}/\mu\text{l}$	8.00	2.6 μg (mass)
Cat K	1728 nM	0.50	70 nM
	Total	12.5	

Table 85. Assay Conditions for the Degradation of Tubulin by Cathepsin K

Reagent	Stock Solution	Volume (μ l)	Final Conditions
EDTA	60 mM	0.54	2.6 mM
DTT	80 mM	0.41	2.6 mM
NaOAc	1 M	1.00	80 mM
C4-S	1.9%	1.00	0.15%
Inhibitor (in DMSO)	250 μ M	1.00	20 μ M
DMSO	50 %		4 %
Tubulin	1.25 μ g/ μ l	2.4	3 μ g (mass)
Cat K	1728 nM	0.50	70 nM
Water		5.60	-
	Total	12.5	

Table 86. Assay Conditions for the Degradation of Type I Collagen by Cathepsin K

Reagent	Stock Solution	Volume (μ l)	Final Conditions
EDTA	60 mM	0.54	2.6 mM
DTT	80 mM	0.41	2.6 mM
NaOAc	1 M	1.00	80 mM
C4-S	1.9%	1.00	0.15%
Inhibitor (in DMSO)	250 μ M	1.00	20 μ M
DMSO	50 %		4 %
Type I Collagen	0.4 μ g/ μ l	7.5	3 μ g (mass)
Cat K	1728 nM	0.50	70 nM
Water		5.60	-
	Total	12.5	

Experimental Section

Activation of Procathepsin K

The protocol for the activation of procathepsin K was slightly modified as suggested from the manufacturer. Activation of the proenzyme needed to be carried out under acidic conditions. Several parameters were optimized prior to the beginning of the screening of thiosemicarbazones as potential cathepsin K inhibitors. The parameters that

were optimized were: incubation time, enzyme concentration, substrate effect, stability of the mature enzyme after activation, and assay condition effects.

Assay Optimization

Five different cathepsin K assays were compared in order to determine the optimal condition to obtain the highest cathepsin K catalytic activity. Table 87 summarizes final conditions that were studied and compared. Assays A and B compare the effect of DTT, a reducing agent, in cathepsin K activity. Assays A and C compare the effect of Brij 30, a detergent, in cathepsin K activity. Finally, assays A and D compare the effect of DMSO, an aprotic solvent, in cathepsin K activity, without the presence of Brij 30. Four proenzyme activation times were arbitrarily set between 50 and 200 minutes for these studies.

Table 87. Cathepsin K Assay Buffer Conditions

Assay	EDTA mM	DTT mM	BRIJ %	NaOAc mM	DMSO %	Cathepsin K nM	Z-FR- AMC μM
A	2.5	2.5	0	150	2	3.6	50
B	2.5	20	0	150	2	3.6	50
C	2.5	2.5	0.01	150	2	3.6	50
D	2.5	2.5	0	150	4	3.6	50
E	2.5	2.5	0.01	150	4	3.6	50

Stability of Activated Cathepsin K

Literature also reports activated cathepsin K is extremely unstable and loses its enzymatic activity fairly quickly. Therefore, we tested the stability of cathepsin K after the protease was fully activated. Stability studies were done for twelve time periods between 0 and 55 minutes after activation time.

Enzyme Concentration Effect

Four different cathepsin K concentrations were also evaluated to determine an optimal catalytic activity. Activated cathepsin K final concentrations were 1, 1.5, 2 and 2.5 nM using the conditions that were previously described. Additionally, every cathepsin K concentration was tested at seven different activation times (between 1 and 4 hours).

Kinetic Cathepsin K Assay

Kinetic analysis of cathepsin K was carried out by using a Thermo Fluoroskan microplate reader, 3691 96-well black microplates. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 10 μ l of 50% DMSO solution, 20 μ l cathepsin K stock solutions, and 20 μ l of Z-FR-AMC stock solution. A mixture containing assay buffer, and 50% DMSO, was preincubated at 25 °C during 5 minutes using 96-black microplates. Triplicate sets of the catalytic activity of cathepsin K were monitored by adding a concentration of Z-FR-AMC to every reaction. The production of AMC was monitored for 5 minutes at 25 °C using excitation and emission references of 355 and 460 nm respectively. Readings were taken every 8 seconds for five minutes. The final concentrations of the kinetic assay are: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, and 1.5 nM activated cathepsin K. Final concentrations of Z-FR-AMC varied between 1 and 75 μ M.

Preliminary Inhibition Studies

Thiosemicarbazone analogs (provided by Dr. Kevin G. Pinney's laboratory)^{3,6,7,10-12} were prescreened to determine if they have inhibitory activity against cathepsin K. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay

buffer, 50 μ l of water, 10 μ l of 50% DMSO or 10 μ l of dilution “1” (final concentration: 10 μ M), 20 μ l cathepsin K stock solutions, and 20 μ l of Z-FR-AMC stock solution. The enzyme-inhibitor mixtures (180 μ) assay buffer, 50% DMSO or inhibitor, and cathepsin K was preincubated at 25 °C during 5 minutes using 96-black microplates. Reactions were started by adding 20 μ l of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five minutes. Final reactions were started by adding 20 μ l of Z-FR-AMC. The final concentrations of the preliminary inhibitory studies are: 150 mM NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 1.5 nM cathepsin K, 10 μ M of the screened inhibitor and 50 μ M of Z-FR-AMC. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Compounds that did not have cathepsin K inhibitory activity more than 50% (i.e. $v_i/v_o \leq 0.5$) were considered as ‘inactive’ compounds and a general IC₅₀ value greater than 10000 nM was assumed. Compounds that inhibited cathepsin K inhibitory activity more than 50% were further considered for cathepsin K inhibitory studies and an exact IC₅₀ was determined.

Cathepsin K Inhibition Assay

Analysis of cathepsin K and its inhibitors was carried out by using a modified protocol of the kinetic cathepsin K assay. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 50 μ l of water, 10 μ l of 35% DMSO or 10 μ l of inhibitor dilutions, 20 μ l cathepsin K stock solutions, and 20 μ l of Z-FR-AMC stock solution. An 180 μ l mixture containing assay buffer, 50% DMSO or inhibitor, and cathepsin K was preincubated at 25 °C during 5 minutes using 96-black microplates. Reactions were started by adding 20 μ l of Z-FR-AMC. The release of AMC by inhibited

and uninhibited samples were monitored for five min. The final concentrations of the inhibitory cathepsin K assay are: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 1.5 nM cathepsin K and 50 μ M of Z-FR-AMC. Final concentrations of the inhibitors varied between 10 μ M and 10 pM. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Table 89 summarizes final volume and final conditions for the kinetic and inhibitory cathepsin K assays.

Construction of AMC Calibration Curve

The calibration curve obtained for cathepsin L experiments (Chapter 2) was used for similar cathepsin K experiments. Complete details and results of these experiments can be found in chapter 2 under the subsection “Construction of AMC calibration curve”.

Table 88. Final Conditions for Cathepsin K Kinetic and Inhibitory Assays

NaOAc (mM)	EDTA (mM)	DTT (mM)	Brij 35 (%)	DMSO (%)	Cathepsin K (nM)	Z-FR-AMC (μ M)
150	2.5	2.5	0.01	4	1.5	1 - 75

Table 89. Preparation Table for Kinetic, Inhibitory Cathepsin K Assays and Construction of AMC Calibration Curves

Reagent (μ l)	Kinetic assay	Inhibitory assay	AMC Curve
Assay Buffer	150	150	150
Control	10	0	10
Inhibitor	0	10	0
CK	20	20	20
Z-FR-AMC/AMC	20	20	20
Total	200	200	200

Effect of Inhibitor Concentration on Cathepsin K Progress Curves

Final concentrations, conditions, and volumes are similar as previously described in the cathepsin K inhibition assay. Assay buffer, inhibitors (Final concentrations varied

between 100 nM and 10 μ M of **1**) and Z-FR-AMC (final concentration: 50 μ M) were added to the 96-well black plates (volume of the substrate-inhibitor mixture: 180 μ l). Then, 20 μ l of cathepsin K stock solution were added immediately without preincubation time. Readings were taken every 3 seconds for fifty minutes.

Effect of Inhibitory Activity of Cathepsin K Inhibitors on Preincubation Studies

Preincubation studies with compound **1** were carried out by a slight modification of the cathepsin K inhibition assay. Different sets of mixtures containing assay buffer, inhibitor and cathepsin K were preincubated at various periods of times between 0 and 180 minutes. Reactions were taken every 25 seconds for five minutes and carried out in triplicate.

Determination of K_i^{app} using Morrison's Quadratic Equation

Data that were obtained in the effect of preincubation studies was further analyzed. The possibility of **1** to be a tight binding inhibitor was examined by fitting the data by a nonlinear regression using Morrison's quadratic equation.

Cathepsin K Reversibility Studies

Three milliliters of assay buffer for reversibility studies (AKR) were prepared (see section Materials and methods). Twenty five microliters of cathepsin K (100X: 150 nM) assay buffer for reversibility studies (ASL) were pre-incubated with an equal amount of a concentrated solution of the inhibitor (100X: 0.1 IC₅₀) at 25 °C for sixty minutes. Then, two microliters of the enzyme-inhibitor mixture were rapidly mixed with 198 μ l of AKR in order to start the reaction. Total reaction volume was 200 μ l. Readings were

taken every twenty five seconds for four hours. Final concentrations are similar as described previously. Table 90 describes required volumes for this experiment.

Table 90. Preparation Table for Cathepsin K Reversibility Assay

Reagent (μ l)	Reversibility studies
Assay Buffer	198
Inhibitor/CK	2
Total	200

Effect of Substrate Concentration (Z-FR-AMC) on IC_{50} Values

The effect of [Z-FR-AMC] was studied with compound **1**. Minor modifications of the cathepsin K inhibition assay were carried out. Different sets of mixtures containing assay, inhibitor and cathepsin K were preincubated at a specific preincubation time (5 minutes). Reactions were initiated with the addition of different concentrations of Z-FR-AMC. Readings were taken every 25 sec for five minutes and carried out in triplicate.

Detection Limits of Procathepsin K by Fluorescent Western Blotting

Fourteen samples of human cathepsin K ranging between 5 and 200 ng were prepared as previously described. Samples also contained sample buffer (1X) and water for a final volume of 10 μ l each.

Selection of the Antibody. The selection of the antibody was extremely important for the immunodetection of procathepsin K and mature cathepsin K. In summary, the ideal antibody could be able to detect both special simultaneously. Monoclonal human anti-cathepsin K (host: mouse) was chosen due to its ability to recognize specifically mature cathepsin K. The epitope of the antibody recognizes the 14-residue peptide with

the sequence: RGYREIPEGNEKAL (residues 222-235 in mature cathepsin K). This antibody can also recognize procathepsin K since both species share that specific peptide in their amino acid sequences. Figure 87 shows the polypeptide sequence that can be recognized by the antibody.

```

1  LYPEEILDTHWELWKKTHRKQYNNKVDEISRRLIWEKNLKYISIHNLEAS
51 LGVHTYELAMNHLGDMTSEEVVQKMTGLKVPLSHSRSDTLYPEWEGRA
101 PDSVDYRKKGYVTPVKNQGCSCWAFSSVGALEGQLKKKTGKLLNLSPO
151 NLVDCVSENDGCGGGYMTNAFQYVQKNRGIDSEDAYPYVQEEESCMYNPT
201 GKAAKCRGYREIPEGNEKALKRAVARVGPVSVDAIDASLTSFQFYSGVYY
251 DESCNSDNLNHAFLAVGYGIQKGNKHWIKNWGENWGNKGYILMARNKN
301 NACGIANLASFPKM

```

Figure 87. Amino Acid Sequence of Procathepsin K (PDB ID: 3KX1).¹³⁸ Legend: Red: Propeptide; Green: Mature Cathepsin K; Purple: Antibody Recognition Site

Electrophoresis. The set of samples were heated at 90 °C for ten minutes and loaded into 4-12% Bis-Tris SDS-PAGE gels. Electrophoresis of the gel containing was performed using MES running buffer (1X) containing SDS at 200 V for 35 minutes. A fluorescent marker (2.0 µl) was also loaded as a reference.

Equilibration of the Blotting Sandwich. Six pieces of blotting paper, low fluorescent PVDF membrane and gel were equilibrated in cathode, anode I and anode II buffers. In summary, the SDS-PAGE gel was equilibrated in cathode buffer for 30 minutes; three pieces of blotting paper in cathode buffer for 30 minutes; two pieces of blotting paper in anode I buffer for 30 minutes; one piece of blotting paper in anode II buffer for 30 minutes. Finally, the membrane was equilibrated in pure methanol for 2 minutes and then equilibrated in anode II buffer for 40 minutes.

Table 91 summarizes conditions for equilibration of the blotting sandwich components. The usage of methanol and SDS has been explained by the manufacturer

(Biorad). Discontinuous systems, that is, the use of three buffers in semi-dry protein transfers systems require the optimization and empirical determination of optimal conditions when transferring proteins from gels to membranes.

Methanol is the preferred solvent in protein transfer for numerous reasons. First, methanol can have the ability to remove SDS excess that is bound to the target protein. This removal promotes a higher binding affinity between the macromolecule and the membrane. One the major disadvantages of methanol is the reduction of pore size in membrane causing poor transfer efficiencies if the concentration of methanol is relative high. It is usually recommended to keep higher concentration of methanol if the target protein to transfer is relative small of medium in size. Lower concentrations of methanol may improve transfer efficiencies when transferring larger proteins (i.e. MW \geq 100 kDa).

On the other hand, lower amounts of SDS may improve the solubility of the proteins, thus facilitating the transfer of the former ones from the gel to the membrane. The most important disadvantage of SDS in discontinuous systems is the poor efficiency of protein binding to the membrane.

Table 91. Equilibration Times for the Components of the Blotting Sandwich

Buffer	Blotting paper	Gel	PVDV membrane
Methanol			2 min
Cathode	3 pc (30 min)	30 min	
Anode I	2 pc (30 min)		
Anode II	1 pc (30 min)		40 min

Assembly of the transfer stack and protein transfer. The blotting sandwich was prepared as described. All of the components were placed in the center of the anode plate as described: two pieces equilibrated in anode buffer I; one piece equilibrated in anode buffer I; PVDF membrane equilibrated with methanol and anode II buffer; SDS- PAGE

gel equilibrated in cathode buffer; and three pieces of blotting paper equilibrated in cathode buffer. Air trapped between each layer was carefully removed. Finally, the cathode electrode plate was firmly placed on top of the stack and secured. Figure 74 shows a schematic representation of the transfer stack in a semi-dry protein transfer system. Protein transfer was run at 20 V for 30 minutes.

Protein blocking and blotting. The membrane was blocked with a blocker solution containing 1% BSA at room temperature for one hour. Then, the blocked membrane was blotted in a 50 ml solution containing 1.0 µg/ml of mouse anti-human cathepsin K ([Stock solution]: 2 mg/ml; dilution factor: 1:2000) in blocking solution (3% BSA). The blotting was incubated at 4 °C for 12 hours.

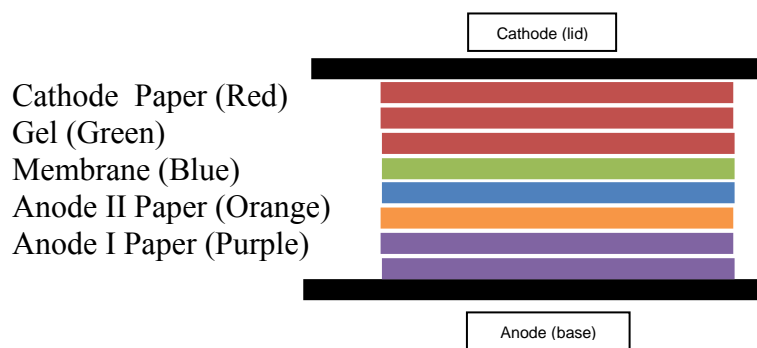


Figure 88. Schematic Representation of the Blotting Sandwich using a Semi-Dry Transfer Device

Protein detection. Unbound antibody was washed with wash buffer (PBS-Tween) three times, 10 minutes each. Then, the washed membrane was incubated using wash buffer with DyLight 649 goat anti-mouse IgG (H + L) secondary antibody (1: 2500) for one hour at room temperature. Finally, the excess of antibody was removed by washing the membrane with wash buffer six times, 5 minutes each. Fluorescent detection was

performed by using a GE typhoon FLA 9000 using an excitation wavelength of 635 nm. Conditions of the scanning were: PMT: 100 V, method: Cy5, excitation wavelength: 635 nm, pixel size 100 μm .

Inhibition of the Activation of Human Procathepsin K

Activation of human procathepsin K was slightly modified in order to extend activation time of procathepsin K. Two sets of samples (untreated and treated with compound **1**) were prepared as previously described. Three micrograms of procathepsin K (3 μl ; stock solution: 0.9 $\mu\text{g}/\mu\text{l}$) were incubated at 3 degrees activation buffer with DMSO (control). Similarly, three micrograms of procathepsin K (3 μl ; stock solution: 0.9 $\mu\text{g}/\mu\text{l}$) were incubated at 3 degrees activation buffer with **1** and DMSO (treated samples). Final conditions were: 1 μM (27 $\text{ng}/\mu\text{l}$) active cathepsin K, 0.7 mM EDTA, 333.3 mM NaCl, 5.4 mM NaOAc pH 3.5, 4% DMSO, and 20 μM of **1** (for the treated samples). Then, samples consisting of 200 ng (7.5 μl of activated cathepsin K) of protein were taken every 45 minutes and inactivated using 1% SDS (1.5 μl) and heated at high temperatures (100 $^{\circ}\text{C}$) in order to stop the activation process. Final volumes of the samples for western blotting were 9.5 μl . Immediately, heated samples were stored at -80 $^{\circ}\text{C}$. Also, aliquots (2.5 μl) were also taken to measure cathepsin K catalytic activity using fluorometric techniques. Final concentrations in fluorometric assays were: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 DTT, 0.1 % Brij, 50 μM Z-FR-AMC, 4% DMSO, 1.5 nM cathepsin K and 30 nM **1** (for treated ones). Activation of procathepsin K was followed for a total time of 6.75 hours.

Electrophoresis and Western Blotting. Electrophoresis was carried out as previously described in this chapter under the section “Detection limits of procathepsin K by Fluorescent Western Blotting” with minor modifications. Briefly, 9.5 μ l of inactivated cathepsin K (200 ng) were mixed with 2.5 μ l of loading sample buffer (4X) containing LDS and 1 μ l of reducing agent. The set of mixtures (treated and untreated) were heated at 90 °C for ten minutes and loaded into 4-12% Bis-Tris SDS-PAGE gels. Electrophoresis of two gels containing treated and untreated samples separately was performed using MES running buffer containing SDS at 200 V for 35 minutes. A fluorescent marker (2.0 μ l) was also loaded as a control. Equilibration of the blotting sandwich, Assembly of the transfer stack and protein transfer, Protein blocking and blotting and Protein detection were carried out as previously described in this chapter.

Effect of Chondroitin 4-Sulfate on Cathepsin K Inhibition Assay

Analysis of cathepsin K and **1** was carried out by using a modified protocol of the Cathepsin K Inhibition Assay. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 50 μ l of water, 10 μ l of 35% DMSO or 10 μ l of inhibitor dilutions, containing 2.7 % C4-S, 20 μ l cathepsin K stock solutions, and 20 μ l of Z-FR-AMC stock solution. An 180 μ l mixture containing assay buffer, 50% DMSO or inhibitor, and cathepsin K was preincubated at 25 °C during 5 minutes using 96-black microplates. Reactions were started by adding 20 μ l of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five min. The final concentrations of the inhibitory cathepsin K assay are: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 1.5 nM cathepsin K, 0.14 % C4-S and 50 μ M of Z-FR-AMC. Final concentrations of the inhibitors varied between 10 μ M and 10 pM.

Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate.

Results and Discussion

Previous results obtained in collaboration results between Trawick and Pinney groups indicated that thiosemicarbazones are lead compounds as cathepsin L inhibitors¹⁰⁻¹². Therefore, the library of compounds required a complete screening to verify the potency of novel analogs against human cathepsin K and potential lead compounds that could show selectivity when compared to cathepsin L's case (Chapter Two). The Structure-Relationship Activity (SAR) was characterized after completing *in vitro* testing against cathepsin K (IC₅₀ values).

Fluorometric based assays were utilized to study various assay parameters in inhibition activities, determination of K_i , and reversibility of thiosemicarbazone inhibitors. Finally, the inhibition of the activation of procathepsin K under acidic conditions by using a thiosemicarbazone as a lead inhibitor was followed by using western blotting and fluorometric techniques.

Assay Optimization

To optimize the assay a variety of reaction conditions and activation times were explored. The catalytic activity of activated cathepsin K was measured by using Z-FR-AMC, a fluorogenic substrate at specific conditions and activation times. Triplicates or more for each condition were carried out. Averages and standard errors are presented in Table 92 and Figure 89.

Five different enzymatic conditions were tested to verify the activity of cathepsin K over time. The lowest activity was obtained when final conditions contained relatively lower concentrations of the solvent vehicle (DMSO, 2%). Another factor that enhanced the activity was the presence of detergent (Brij 30). Cathepsin K activity at longer activation times (220 min) was 1.3 fold greater when a non-ionic detergent was present. A similar trend could be observed when DMSO was 4% (Conditions D and E, Table 87).

Table 87. Cathepsin K Assay Buffer Conditions

Assay	EDTA mM	DTT mM	BRIJ %	NaOAc mM	DMSO %	Cathepsin K nM	Z-FR- AMC μM
A	2.5	2.5	0	150	2	3.6	50
B	2.5	20	0	150	2	3.6	50
C	2.5	2.5	0.01	150	2	3.6	50
D	2.5	2.5	0	150	4	3.6	50
E	2.5	2.5	0.01	150	4	3.6	50

Table 92. Comparison of Cathepsin K Activity by Varying Assay Conditions

Activation time (min)	Condition (see Table 87, page 237, for complete details)				
	A	B	C	D	E
	(Cathepsin K Activity, nM AMC/s)				
50	0.17	0.23	0.42	0.51	0.60
90	0.55	0.63	1.01	1.38	1.51
150	1.33	1.65	2.62	3.39	4.13
200	1.36	2.11	2.75	3.68	4.60

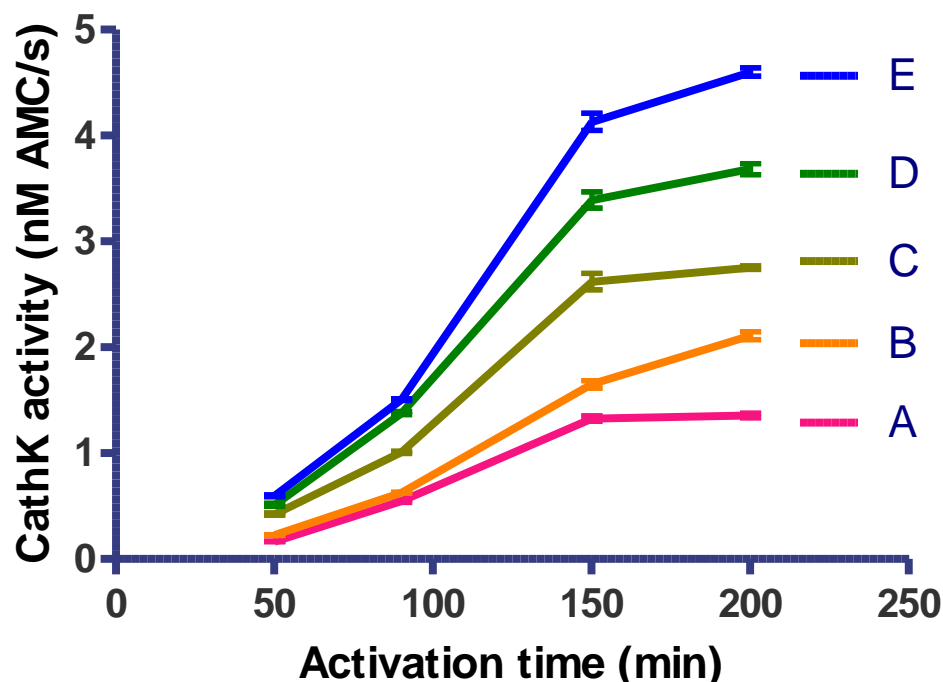


Figure 89. Comparison of Cathepsin K Activity by Varying Assay Conditions. (See Table 87 and Table 92)

Based on experimental results, condition E (150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM DTT, 0.01% Brij, 4% DMSO, 50 μ M Z-FR-AMC) was chosen as the preferred condition for future cathepsin K experiments. It was also established that activation times can vary and they can be quite lengthy. However, activation time was determined empirically for every batch of procathepsin K that was obtained. Maximum activity was obtained with condition E when procathepsin K was activated for 220 minutes (Activity: 4.6 nM AMC/s).

Stability of Activated Cathepsin K

Stability studies were performed within the first hour after cathepsin K was fully activated. Results are the average of four independent experiments per stability time and can be seen in Figure 90.

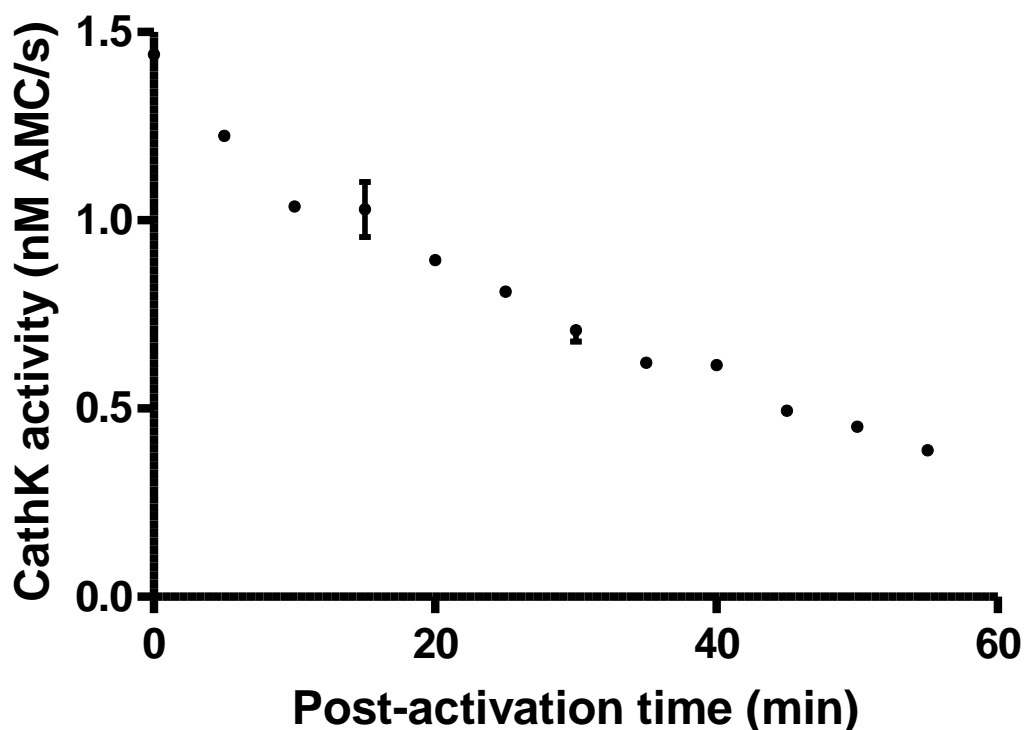


Figure 90. Stability of Activated Cathepsin K Curve

As seen in Figure 90, cathepsin K is extremely unstable under this specific activation conditions (pH: 3.5) and loses its activity at a very fast pace. Nevertheless, linear regression of this data (not shown, r^2 : 0.9612) reveals cathepsin K loses 1.3% its catalytic activity per minute. For example, cathepsin K lost 30% of its activity after 10 minutes it was activated. Therefore, it was determined that cathepsin K experiments should be carried out within the first ten minutes after activation time was reached.

Enzyme Concentration Effect

Finally, the minimum enzyme concentration was also determined to verify if a proper detection limit was obtained. Results are the average of three independent results and can be seen in Table 93 and Figure 91.

Table 93. Cathepsin K Activity vs Enzyme Concentration

Activation time (h)	Cathepsin K (nM)			
	1	1.5	2	2.5
	(nM AMC/s)			
1.00	0.07	0.25	0.60	0.75
1.50	0.24	0.48	0.72	1.42
2.00	0.32	0.67	0.97	1.35
2.50	0.39	0.79	1.33	1.65
3.00	0.46	0.49	0.71	1.20
3.50	0.14	0.46	0.66	1.15
4.00	0.19	0.32	0.85	0.96

Four cathepsin K concentrations were evaluated to optimize the activation time. Seven different activation times were tested for each concentration. Maximum activity was reached when the enzyme was activated under acidic conditions between 2.5 and 3 hours. Furthermore, the fragility of the enzyme was quite evident at longer incubation times (activation times: 3.5 and 4 hours). Activation time was set up to be between 2 and 2.5 hours. Nevertheless, activation time was determined empirically for each procathepsin K.

Cathepsin K Kinetic and Inhibition Studies

Cathepsin K studies were carried out using a 96-well microplate fluorometric based assay. Conditions, reagents, and experiments can be seen in chapter three under the subsection “Cathepsin K kinetic and inhibition studies”.

Determination of K_M , V_{MAX} and k_{CAT}

Kinetic parameters for cathepsin K were determined using GraphPad 5.0 (see Chapter Two, for a complete explanation). Z-FR-AMC, a broad specificity substrate for cysteine proteases was used as a fluorogenic substrate for the recombinant protein (Figure

23, equation 1.1, page 35). The release of AMC observed a linear trend when 0.5 μM Z-FR-AMC was used during these experiments. The determination of K_M , V_{MAX} and k_{CAT} was possible by using substrate concentrations that varied between 1 and 75 μM . Substrate inhibition was also observed when $[\text{Z-FR-AMC}] \geq 150 \mu\text{M}$. To date, this is the first report of the effect of substrate inhibition on the activity of cathepsin K.

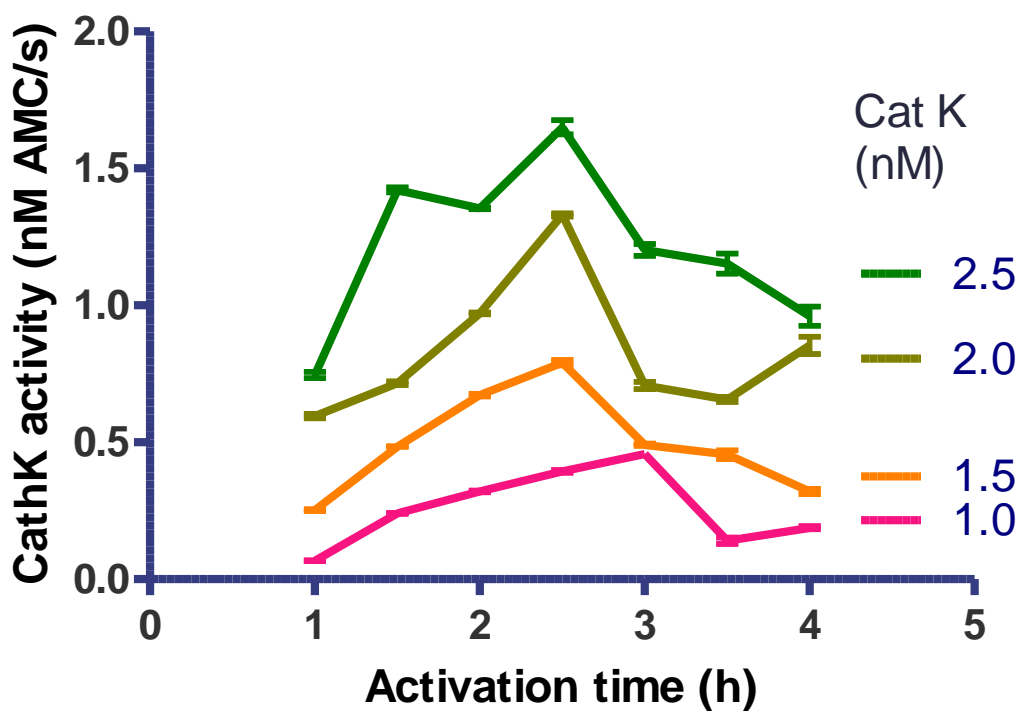


Figure 91. Cathepsin K Activity vs Enzyme Concentration

Experiments were carried out in triplicate. The parameters K_M , V_{MAX} and k_{CAT} were determined by using equations 1.1 and 3.1. Their values were found to be $13.5 \pm 2.1 \mu\text{M}$, $0.52 \pm 0.05 \text{ nM AMC/sec}$ and 0.52 s^{-1} respectively³⁶². A comparison between cathepsins K and L kinetics values reveals a considerable difference between affinities. Z-FR-AMC represents an excellent cathepsin L substrate (K_M : 1 μM); however, that is not the case for cathepsin K.

Its K_M is 10 times higher than cathepsin L and this discrepancy might be explained with substrate affinity. Cathepsin K has a specific preference for proline, and other substrates (Z-GPR-AMC) have been developed to test cathepsin K activity in other studies.³⁵⁹

$$v_o = \frac{v_{MAX}[S]}{K_M + [S]} \quad (1.1) \quad k_{CAT} = \frac{v_{MAX}}{[cathepsin K]} \quad (3.1)$$

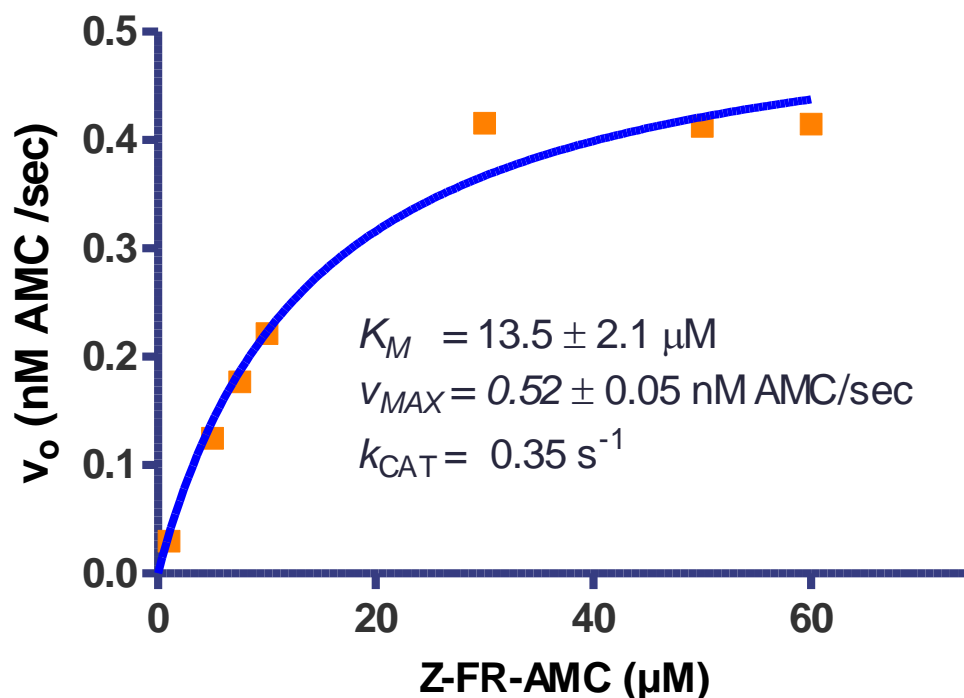


Figure 92. Determination of K_M and V_{MAX} for Human Cathepsin K Using Z-FR-AMC as a Fluorogenic Substrate

Substrate Inhibition

As previously described, substrate inhibition was observed at higher concentrations ($[Z-FR-AMC] \geq 75 \mu M$). Figure 79 shows cathepsin K activity versus substrate concentration. Inhibition at $75 \mu M$ is slightly evident, and a dramatic inhibition is seen when substrate concentration is doubled.

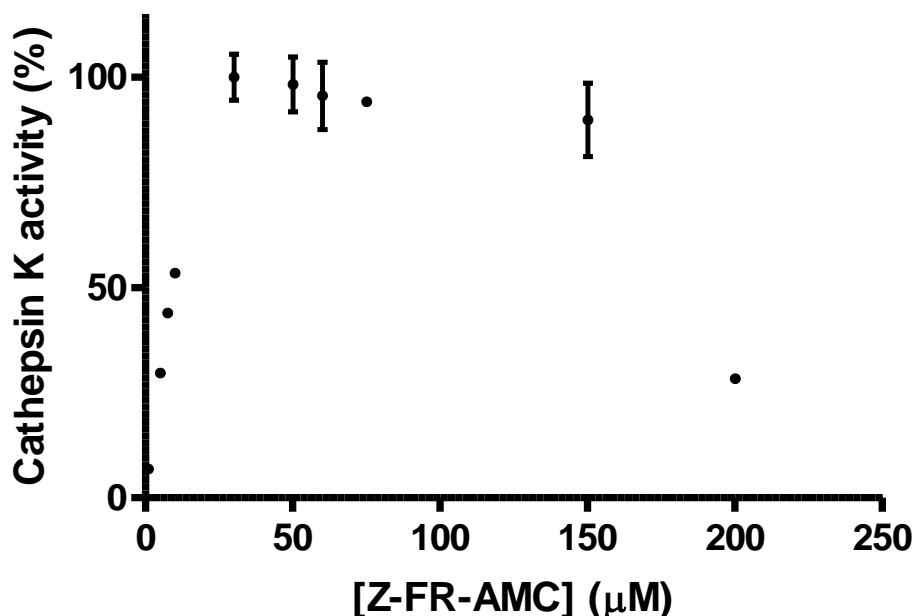


Figure 78. Effect of Substrate Inhibition on Human Cathepsin K

Determination of Inhibitory Efficacy of Thiosemicarbazone Analogs at 10 μM

The first set of experiments that were performed was to determine if individual compounds in the series of thiosemicarbazone derivatives were potential cathepsin K inhibitors, that is if the catalytic activity of the cysteine protease was inhibited by 50% or more by a fixed concentration (final concentration: 10 μM) of the potential inhibitor. Experimental details and conditions are similar to those described previously in the Material and Methods Section of Chapter Two. Three independent sets of experiments of untreated ([I]: 0 μM) and treated ([I]: 10 μM) samples were preincubated with 1.5 nM activated cathepsin K for 5 min at 25 $^{\circ}\text{C}$. Inhibitory activities were monitored when reactions were started by adding 50 μM Z-FR-AMC as a fluorogenic substrate. Reactions observed a linear behavior for the first minutes. Catalytic rates of uninhibited and inhibited samples were calculated by linear regression of the data. ([AMC]: dependent variable and time (sec): the independent variable). Compounds were

considered as potential inhibitors if more than 50% cathepsin K activity was inhibited by 10 μ M in solution. Thus, they were further analyzed to determine an exact IC_{50} value. If the ratio $v_i/v_o \leq 0.5$, then the compounds were not considered potential inhibitors and an approximated IC_{50} value of 10000 nM was assigned to them.

Determination of IC_{50} Values

Several sub-libraries of thiosemicarbazones including of *meta*-bromosubstituted benzophenone thiosemicarbazones (Compounds **1-44**) among others were analyzed to investigate their ability to inhibit cathepsin K. The synthetic compounds were synthesized under the guidance of Dr. Kevin G. Pinney by several members of his research group.^{3,6,7,10-12} A 96-well microplate fluorometric based assay was utilized to determine the inhibitory activity of each of these inhibitors. A comprehensive explanation of the experimental procedure can be found in the Material and Methods section of this chapter. Uninhibited cathepsin K catalytic activity showed a linear behavior when 50 μ M Z-FR-AMC was used for reactions times that was up to 5 minutes long. The determination of the IC_{50} is possible with experiments that observe the inhibitory capacity of the synthetic compounds when a fixed concentration of cathepsin K (1.5 nM) is preincubated for 5 minutes at 25 $^{\circ}$ C. Final concentration of each compound varied between 10 pM and 10 μ M. Experiments were carried out in triplicate. Equation 1.2 was used to determined IC_{50} values for these compounds. A complete structure-activity relationship (SAR) is shown in tables that group these inhibitors by functional groups or families (See Table 94-Table 113).

$$Y = \frac{v_{MIN} + (v_{MAX} - v_{MIN})}{1 + 10^{(\log(IC_{50}-X)*Hillslope)}} \quad (1.2)$$

Table 94. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12}

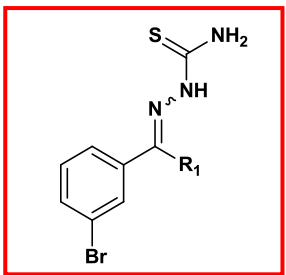
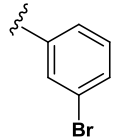
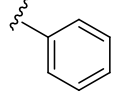
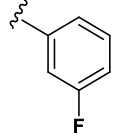
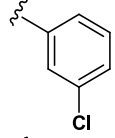
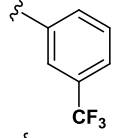
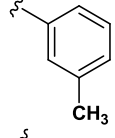
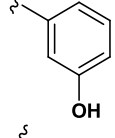
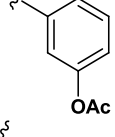
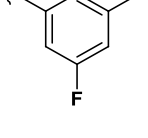
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
1			35.2 ± 2.5
2			660.4 ± 64.1
4			104.2 ± 6.2
5			259.0 ± 19.4
6			65.3 ± 5.4
7			133.3 ± 4.3
8			53.3 ± 3.4
9			98.9 ± 8.4
10			26.0 ± 1.0

Table 94. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

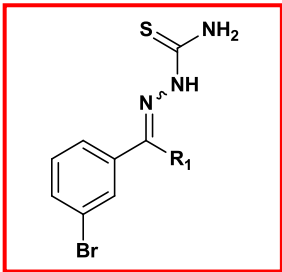
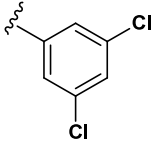
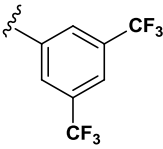
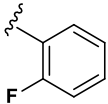
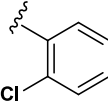
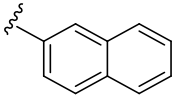
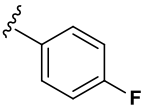
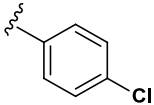
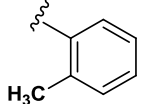
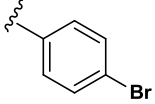
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
12			335.6 ± 13.7
13			11.3 ± 0.9
14			61.8 ± 5.5
15			312.9 ± 6.9
16			92.5 ± 2.8
17			65.6 ± 5.0
18			626.4 ± 19.3
20			838.9 ± 80.3
21		49.1 ± 2.5	

Table 94. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
22			32.6 ± 3.5
23			22.6 ± 0.7
24			1300 ± 99.6
25			107.3 ± 4.0
26			109.0 ± 5.1
27			40.3 ± 1.4
28			23.0 ± 0.4
29			66.7 ± 6.2
30			118.8 ± 8.9
31			10.2 ± 1.0

Table 94. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
32			29.0 ± 2.1
33			65.8 ± 3.7
34			15.6 ± 1.2
35			524.8 ± 31.2
36			168.5 ± 13.6
37			150.1 ± 4.3
38			≥ 10000
39			≥ 10000
40			30.3 ± 1.1
41			6154 ± 36.7

Table 94. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

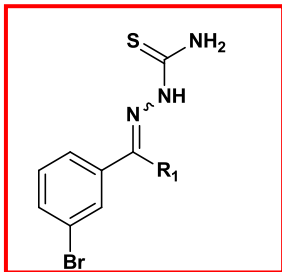
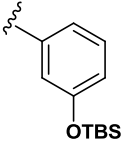
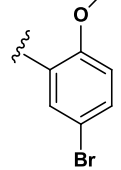
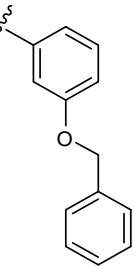
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
42			4190 ± 286.5
43			4118 ± 408.4
44			140.9 ± 3.2

Table 95. Inhibition of Human Cathepsin K by *para*-Bromo Functionalized Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

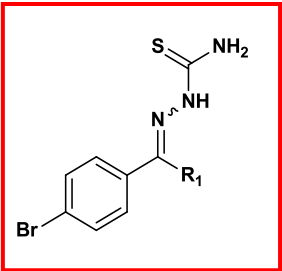
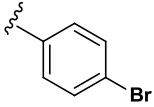
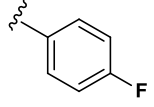
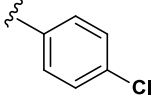
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
47			≥ 10000
48			2086 ± 178.3
49			4413. ± 333.2

Table 96. Inhibition of Human Cathepsin K by *para*-Bromo Functionalized Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

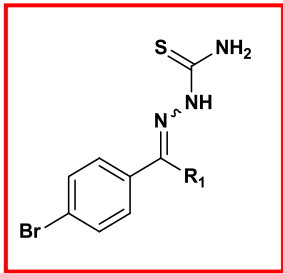
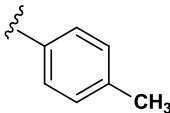
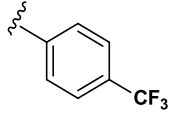
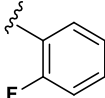
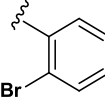
Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
50			1183 ± 81.4
51			146.5 ± 2.4
52			368.6 ± 34.6
53			≥ 10000

Table 97. Inhibition of Human Cathepsin K by Dihalogen-substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

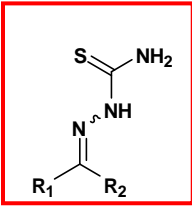
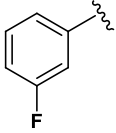
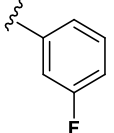
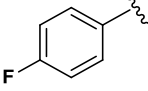
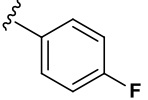
Compound	R_1	R_2	$IC_{50} \pm S.E., (nM)$
			
54			754.8 ± 22.2
55			2946 ± 261.1

Table 98. Inhibition of Human Cathepsin K by Dihalogen-substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

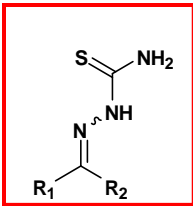
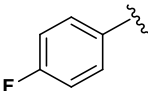
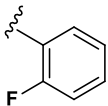
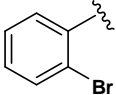
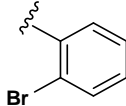
Compound	R ₁	R ₂	IC ₅₀ ± S.E., (nM)
			
56			1004 ± 71.6
57			524.8 ± 31

Table 99. Inhibition of Human Cathepsin K by 3,3'-Dibromo-*N*-Substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

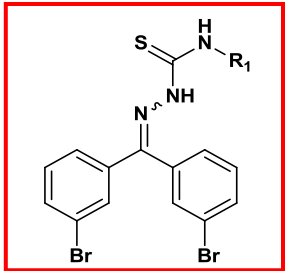
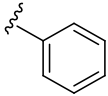
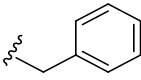

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
58			≥ 10000
59			≥ 10000
60			≥ 10000

Table 100. Inhibition of Human Cathepsin K by Thiosemicarbazones Containing a Phenyl Group. For Synthesis of Compounds: ^{3,6,7,10-12}

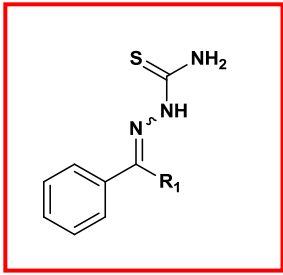
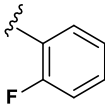
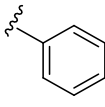
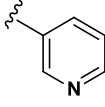
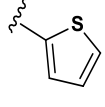
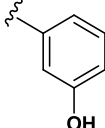
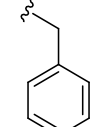
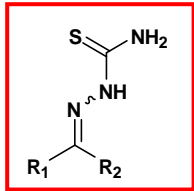
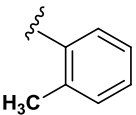
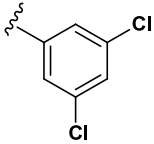
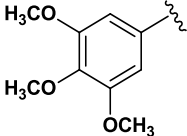
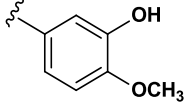
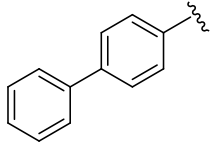
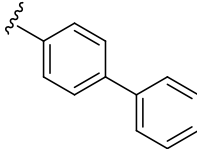
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
61			3651 ± 44.6
62			≥ 10000
63			≥ 10000
64			≥ 10000
65			≥ 10000
66			5410 ±

Table 101. Inhibition of Human Cathepsin K by Substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	R_1	R_2	$IC_{50} \pm S.E., (nM)$
			
73			57.3 ± 5.2
75			156.7 ± 14.7
76			ND (Insoluble)

*ND: Not Determined

Table 102. Inhibition of Human Cathepsin K by Functionalized Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

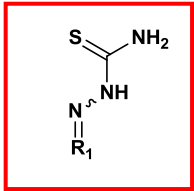
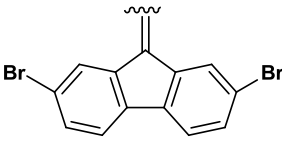
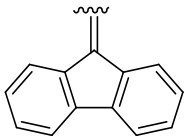
Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
80			≥ 10000
81			≥ 10000

Table 103. Inhibition of Human Cathepsin K by Functionalized Thiosemicarbazones.
For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
82			≥ 10000
85			≥ 10000
90			≥ 10000

Table 104. Inhibition of Human Cathepsin K by Substituted Quinolone
Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
98			≥ 10000

Table 105. Inhibition of Human Cathepsin K by Non-Thiosemicarbazone Based Analogs. For Synthesis of Compounds: ^{3,6,7,10-12}

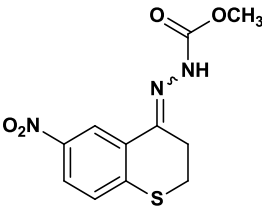
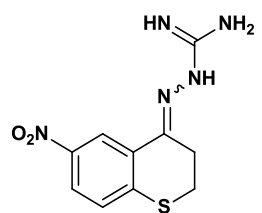
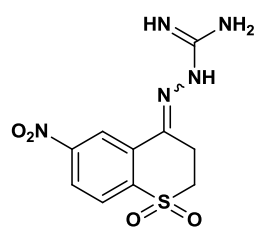
Compound	Structure	$IC_{50} \pm S.E., (nM)$
101		≥ 10000
102		≥ 10000
103		≥ 10000

Table 106. Inhibition of Human Cathepsin K by Substituted Tetralone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

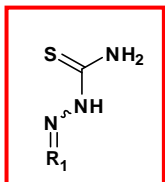
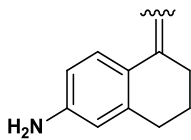
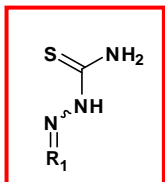
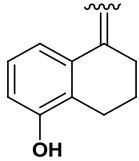
Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
106			≥ 10000
109			≥ 10000

Table 107. Inhibition of Human Cathepsin K by Functionalized Chromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
112			≥ 10000

Table 108. Inhibition of Human Cathepsin K by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
114			1069 ± 104
115			499.5 ± 34.0
117			≥ 10000
118			164.7 ± 14.9
119			176.6 ± 14.3
120			1558 ± 156

Table 108. Inhibition of Human Cathepsin K by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

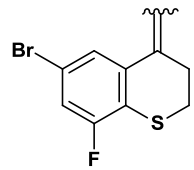
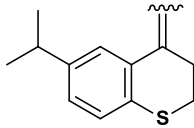
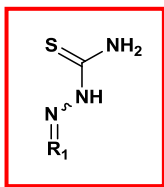
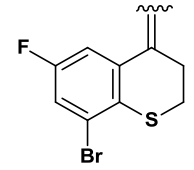
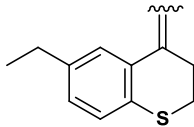
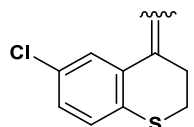
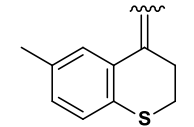
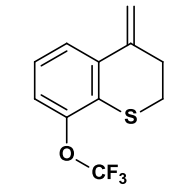
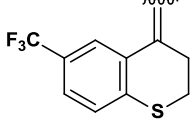
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
122			44.2 ± 2.9
123	 		21.2 ± 1.9
124			419.4 ± 21.7
125			29.0 ± 1.8
126			62.9 ± 2.3
127			80.8 ± 2.6
128			≥ 10000
129			21.5 ± 2.1

Table 108. Inhibition of Human Cathepsin K by Substituted Thiochromanone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

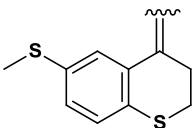
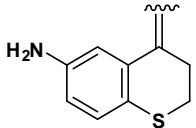
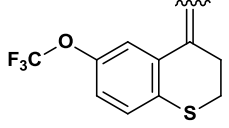
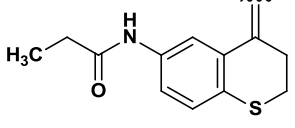
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
130			56.7 ± 4.2
132			≥ 10000
133			179.4 ± 10.7
134			≥ 10000

Table 109. Inhibition of Human Cathepsin K by Substituted Sulfone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

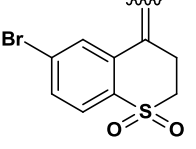
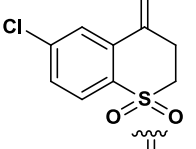
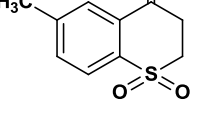
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
136			149.9 ± 7.9
138			1441 ± 48.8
139			2981 ± 41

Table 110. Inhibition of Human Cathepsin K by Substituted Sulfone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

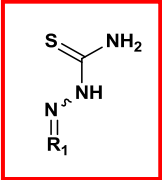
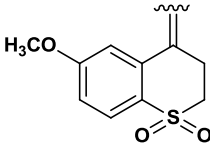
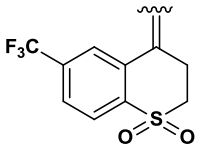
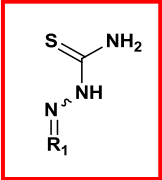
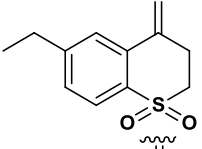
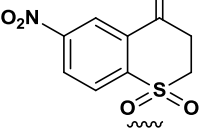
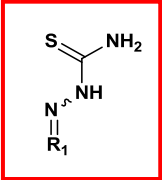
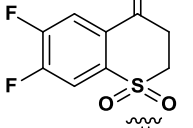
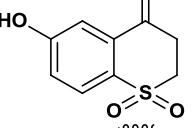
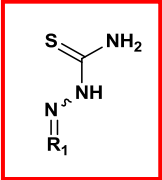
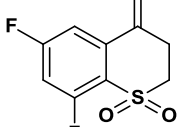
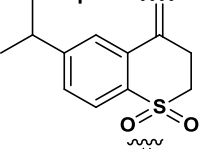
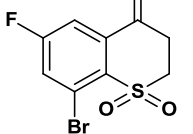
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
140			≥ 10000
141			204.2 ± 2.6
142			941.7 ± 70.0
143			≥ 10000
144			1273 ± 13.3
145			≥ 10000
146			≥ 10000
147			110.9 ± 11.6
148		≥ 10000	

Table 109. Inhibition of Human Cathepsin K by Substituted Sulfone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

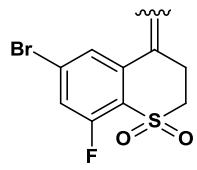
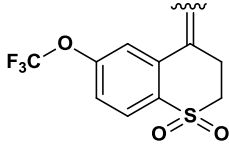
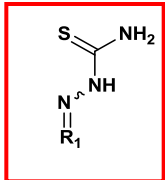
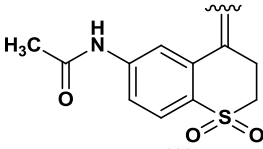
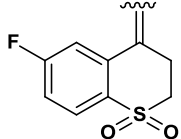
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
149			649.6 ± 37.3
150			471.4 ± 14.6
151			≥ 10000
152			5918 ± 415.7

Table 111. Inhibition of Human Cathepsin K by Functionalized Benzoyl-Benzophenone Thiosemicarbazones*

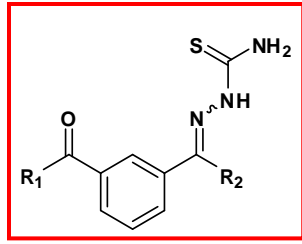
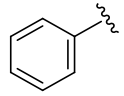
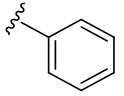
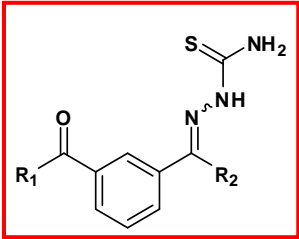
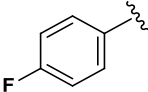
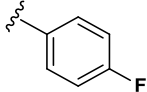
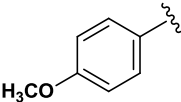
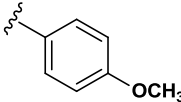
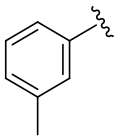
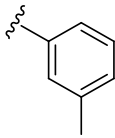
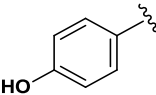
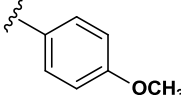
Compound	R ₁	R ₂	IC ₅₀ ± S.E.,(nM)
			
156			17.4 ± 1.4

Table 112. Inhibition of Human Cathepsin K by Functionalized Benzoyl-Benzophenone Thiosemicarbazones (Continued)

<i>Compound</i>	<i>R</i> ₁	<i>R</i> ₂	<i>IC</i> ₅₀ ± <i>S.E.</i> , (<i>nM</i>)
			
157			8500 ± 663.7
158			2092 ± 17.5
160			1034 ± 68.2
161			6162 ± 471

*Compounds **156-162** have been synthesized by members of the Kevin G. Pinney laboratory (Baylor University)

Table 113. Inhibition of Human Cathepsin K by Substituted-Benzoyl-Benzophenone Thiosemicarbazones.* For Synthesis of Compounds: ^{3,6,7,10-12}

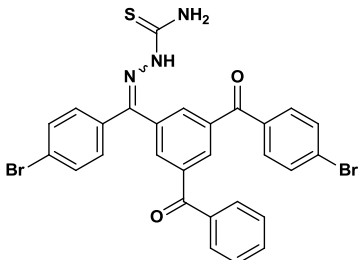
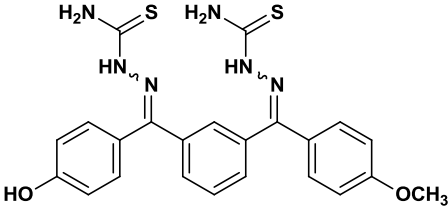
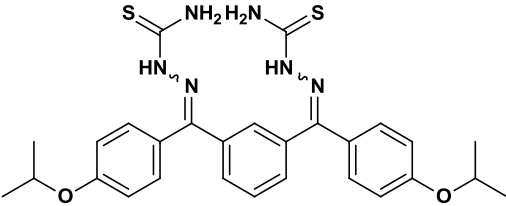
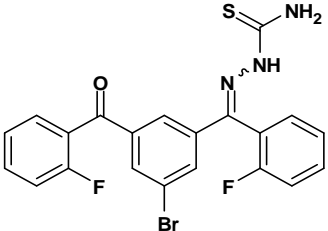
<i>Compound</i>	<i>Structure</i>	<i>IC</i> ₅₀ ± <i>S.E.</i> , (<i>nM</i>)
165		2796 ± 19.8

Table 114. Inhibition of Human Cathepsin K by Substituted-Benzoyl-Benzophenone Thiosemicarbazones.* For Synthesis of Compounds: ^{3,6,7,10-12}

<i>Compound</i>	<i>Structure</i>	<i>IC₅₀ ± S.E., (nM)</i>
166		3204 ± 356
167		≥ 10000
168		48.9 ± 3.2

*Compounds 163-168 have been synthesized by members of the Kevin G. Pinney laboratory (Baylor University)

Structure-Activity Relationship (SAR) of Thiosemicarbazones as Cathepsin K Inhibitors

Almost seventy-five percent of the library that was screened for cathepsin L was evaluated using cathepsin K as a target. A total of 120 thiosemicarbazone analogs were screened using activated cathepsin K during the preliminary screening in order to determine the potency of these compounds. A selected group of four different

compounds showed to be excellent cathepsin K inhibitors with IC_{50} between 10 and 20 nM. Eleven compounds demonstrated to be good inhibitors ($20 \leq IC_{50} \leq 100$ nM).

Thirty percent of the samples did not show a significant inhibitory potency and a general IC_{50} value greater than 10,000 nM was assigned. On the other hand, compounds that showed inhibitory activity greater than 50% were considered to calculate an IC_{50} value that was expected to be less than 10,000 nM. One fourth of the library (31 analogs) had a modest activity. IC_{50} values of these synthetic compounds were ranged between 100 and 1000 nM. Interestingly, the entire sub-library of functionalized *meta*-bromobenzaldehyde thiosemicarbazones (Table 94, page 258) showed inhibitory activity toward cathepsin K ($IC_{50} < 6.5 \mu M$).

Potent cathepsin K inhibitors. Finally, four thiosemicarbazones showed very potent inhibitory activities against this cathepsin K (CK). Figure 93 summarizes the structures of the compounds with IC_{50} values less than 20 nM. Three analogs belong to the subfamily of *metabrominated* benzophenone TSC analogs. Ditrifluoromethyl, and two brominated phenols, all halogenated moieties are the substituents in this subfamily. The fourth compound consists of the unsubstituted benzoyl benzophenone thiosemicarbazone. The IC_{50} values obtained using cathepsin L (CL) are also included in this section for comparison purposes between these two cysteine proteases. This selected group also showed good inhibitory activity towards cathepsin L. Compounds **13**, **34** and **31** were better cathepsin K inhibitors when compared to their respective cathepsin L activities (i.e. up to 15-fold more potent in the case of **34**). Interestingly, **156**, the second most potent inhibitor of cathepsin L in the library, is also one of the most potent inhibitors of cathepsin K with an IC_{50} -value less than 20 nM.

Compounds with IC₅₀ less than 100 nM. Eleven compounds proved to have good inhibitory activity (less than 100 nM). This list included: 3-3'-dibromobenzophenone TSC (**1**), 3-bromo-3'-trifluoromethylbenzophenone TSC (**6**), 3-bromo-3'-hydroxybenzophenone TSC (**8**), 3-bromo-3'-acetobenzophenone TSC (**9**), 2-methyl-3',5'-dichlorobenzophenone TSC (**73**), 6-isopropylsulfide TSC (**123**), 6-ethylsulfide TSC (**125**), 6-trifluoromethylsulfide TSC (**129**).

Compounds with modest activity. Thirty-one TSC analogs showed IC₅₀ values between 100 and 1000 nM. The list included, but not limited: *metabrominated* benzophenone (**2**, **4-5**) TSCs, sulfide (**115**, **118**, **124**), and sulfone (**149**, **150**) TSCs

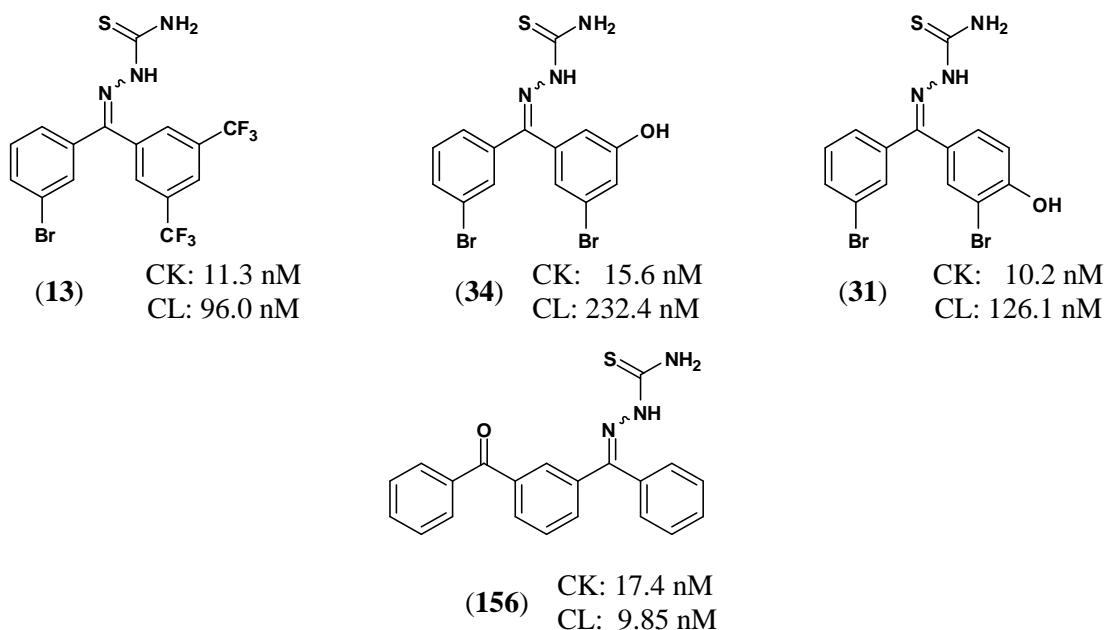


Figure 93. Thiosemicarbazone Analogs with Potent Inhibitory Activity against Human Cathepsin K⁹⁻¹²

General Remarks of the Structure-Activity Relationship

Further analysis of the structure-activity relationship revealed that thiosemicarbazone inhibitory activity can be enhanced or reduced dramatically by the substituting numerous moieties. The series of substituted *meta*-brominated benzophenone thiosemicarbazones is the largest subgroup among the compounds that were tested. Halogenated substituents greatly enhanced the activity of the thiosemicarbazones. Compounds **1**, **2**, **4**, **5** (Table 115) were compounds sharing the main moiety but varying the *meta* halogenated substituent. Compound **2** was used as a comparison to complete the series. The unsubstituted analog **2** showed modest activity compared to **1** the brominated analog with one of the best activities in the entire library. Once again, **1** was the best halogenated analog of the series with these types of substituents. The brominated analog was 3-fold more active than its fluorinated analog. The activity of these compounds between cathepsins K and L showed that **1** is the most potent inhibitor in both cases. Analog **2** was 15-fold more potent against cathepsin K when compared to cathepsin L inhibitory activity.

Five fluorosubstituted benzophenone TSC analogs (**4**, **10**, **22**, **23**, **25**, Table 116) were also compared. Fluorination of the second phenyl rings indicated that cathepsin K-inhibitory activity of the compounds can be increased with the addition of electronegative substituents, similar to the cathepsin L inhibitory activity. However, a deeper comparison between the two difluoro benzophenone TSC also revealed the position of the substitution plays a key role in the activity of these analogs. The *diortho*-fluoro analog is 4 times less active than the *dimeta*-counterpart. The activity of these inhibitors is greatly enhanced by the presence of bromine in the *meta* position of the second ring

(28). 3-bromo-2',3'-difluorobenzophenone thiosemicarbazone (28) is 2-fold more active than also 3'-difluorobenzophenone thiosemicarbazone (27).

Table 115. Inhibition of Human Cathepsin K by 3-Bromo-3'-Halogen Benzophenone Thiosemicarbazones

		R_1			
IC ₅₀ (nM)		1	4	5	2
CK	35.2	104.2	259	660.4	
CL	16.7	250.3	131.4	≥ 10000	

Table 116. Inhibition of Human Cathepsin K by 3-Bromo-Fluorinated-Benzophenone Thiosemicarbazones

		R_1				
IC ₅₀ (nM)		23	10	22	4	25
CK	22.6	26.0	32.6	104.2	107.0	
CL	63.2	59.4	118.1	250.3	609.7	

Various groups were substituted at the *meta* position of the second phenyl rings (Table 117). The groups include trifluoromethyl, methyl, hydroxyl, and acetate moieties. The best compound is the phenolic analog (8).

Table 117. Inhibition of Human Cathepsin K by 3-Bromo-3'-Heteroatomic Groups Benzophenone Thiosemicarbazones

		R_1			
IC_{50} (nM)		8	6	9	7
CK		53.3	65.3	98.9	133.3
CL		131.4	46.5	150.8	224.4

A small trifluoromethyl series of three compounds (**6**, **11**, and **13**, Table 118) also revealed that substituted compounds often demonstrate better activity against cathepsin K when compared to monosubstituted 3-bromobenzophenone thiosemicarbazones. In fact, analog **13** is the second most potent inhibitor of cathepsin K with an outstanding IC_{50} value of 11.3 nM.

In the *ortho*-substituted series, the inhibitory activity of the compounds increases with more electronegative substituents. Additionally, the exceptional case of **21**, 3-4'-dibromobenzophenone thiosemicarbazone is remarkable when comparing its activity toward cathepsins L and K. This analog shows a remarkable selectivity toward cathepsin K (Table 119).

Finally, the subfamily of substituted *meta*-brominated thiosemicarbazones was completed with unbrominated and brominated pyridines and thiophenes. Pyridine TSCs demonstrated to have good inhibitory activity while unbrominated pyridines showed no activity against cathepsin K. These small series also demonstrated that benzophenone group plays a key role in the activity of TSCs (Table 120).

Table 118. Inhibition of Human Cathepsin K by 3-Bromo-Trifluoromethyl Benzophenone Thiosemicarbazones

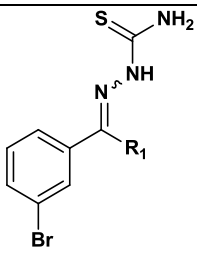
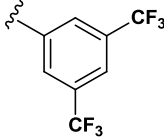
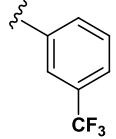
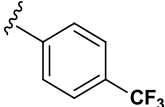
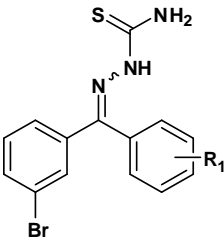
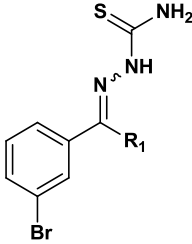
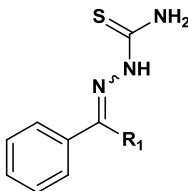
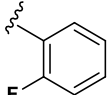
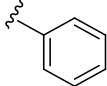
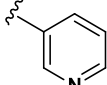
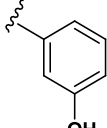
		R_1				
			<		<	
	IC ₅₀ (nM)	13		6		11
	CK	11.3		65.3		ND
	CL	96.0		46.5		520.9

Table 119. Effect of the position of substituents in the inhibitory activity of 3-Bromo-Benzophenone Thiosemicarbazones

		IC ₅₀ (nM)					
R_1		<i>ortho</i>		<i>meta</i>		<i>para</i>	
	-F	14	CK 61.8 CL 23.9	4	CK 104.2 CL 250.3	17	CK 65.6 CL 79.6
	-Cl	15	CK 312.9 CL 1610	5	CK 259.0 CL 131.4	18	CK 626.4 CL 327.1
	-Br	24	CK 1300 CL 2600	1	CK 35.2 CL 16.7	21	CK 49.1 CL ≥ 10000
	CH ₃	20	CK 838.9 CL ≥ 10000	7	CK 133.3 CL 224.4	19	CK ND CL 2156

Another interesting comparison is between *meta* and *para* brominated benzophenones (Table 121) with different halogenated and aliphatic substituents. It is evident that *meta* bromination offers the best activity and a change in this position reduces drastically the potency of any substituent. For example, 3,4'-dibromobenzophenone thiosemicarbazone is more potent than its *para* brominated counterpart, 4,4'-dibromobenzophenone that showed no activity towards cathepsin K.

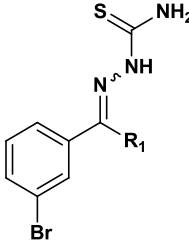
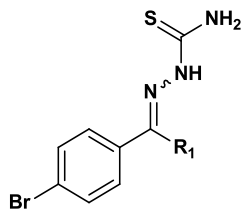
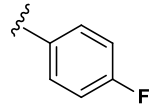
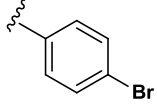
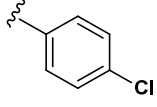
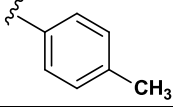
Table 120. Comparison between Brominated and Unbrominated Functionalized Benzaldehyde Thiosemicarbazones for Cathepsins K and L

R_1						
		IC ₅₀ (nM)				
	14	CK	61.8	61	CK	3651
		CL	23.8		CL	≥ 10000
	2	CK	660.4	62	CK	≥ 10000
		CL	≥ 10000		CL	≥ 10000
	29	CK	66.7	63	CK	≥ 10000
		CL	1000		CL	≥ 10000
	8	CK	53.3	65	CK	≥ 10000
		CL	188.7		CL	≥ 10000

Bromination of the benzophenone group is critical for the activity of TSC derivatives as potential cathepsin K inhibitors. An examination of halogenated compounds with similar chemical structures reveals that a different halogen substituent, such as fluorine, greatly reduces the activity of the compounds. The 3,3'-difluorobenzophenone TSC is almost 22 times less active than 3,3'-dibromobenzophenone TSC. Interestingly, 4,4'-difluorobenzophenone TSC was a better inhibitor than 4,4'-dibromobenzophenone TSC that was inactive.

Compounds **61-65** also demonstrated the importance of the presence of at least one bromine in one of the phenyl rings in order to enhance the potency of the thiosemicarbazone derivatives. In this series, only the fluorinated analog of the benzophenone thiosemicarbazone **61** showed inhibitory activity.

Table 121. Comparison of the Inhibitory Activity between *meta* and *para*Bromination of Benzophenone Thiosemicarbazones towards Cathepsins K and L

R_1						
	IC ₅₀ (nM)					
	17	CK	65.6	48	CK	2086
		CL	79.6		CL	3320
	21	CK	49.1	47	CK	≥ 10000
		CL	327.1		CL	≥ 10000
	18	CK	626.4	49	CK	4413
		CL	≥ 10000		CL	≥ 10000
	19	CK	ND	50	CK	1183
		CL	2156		CL	4570

The TSC group is a key feature for a successful inhibition of cathepsin K.

Compound **1**, the reference compound, is a potent inhibitor with an IC₅₀ value of 35.2 nM, while analogs of the thiosemicarbazone derivative were inactive (**58-60**, Table 99).

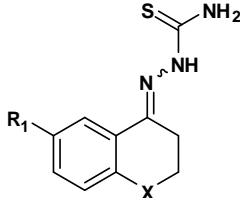


Figure 94. Comparison between Inhibitory Activities of **1** and **58**

The compounds with different scaffolds were examined (Table 102). The series included fluorenes, naphthalenes, indenes, annulenes, chromanones, aminoquinolines, among others. These compounds were inactive.

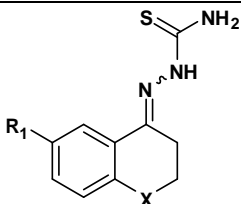
Thiochromanones and sulfones are important subclasses of thiosemicarbazones that were screened in this library. Their IC₅₀ values ranged between 21.2 and 10000 nM. In general, thiochromanones showed better inhibitory activities when compared with sulfones. The majority of thiochromanone were monosubstituted in the C6 position. The list of substituents includes halogen, hydroxyl, aliphatic, nitro, and acetate groups. Perhaps the most interesting observation is that aliphatic substituted analogs presented the best inhibitory activity (**123**, **125**, and **127**) with IC₅₀ values less than 100 nM. The potency is enhanced when the substituent is branched in both cases. However, thiochromanones are up to 37 times more potent than sulfones (Table 122). Another selective inhibitor of cathepsin K was found in this series. Compound **123** showed a good IC₅₀ values of 21 nM, while it was inactive as a cathepsin L inhibitor (See Table 122).

Table 122. Comparison of the Activity of Aliphatic Thiochromanone Versus Aliphatic Sulfone TSCs against Cathepsins K and L

IC ₅₀ (nM)	X	<i>R₁</i>								
		i-propyl		ethyl		methyl				
	S	123	CK	21.2	125	CK	29	127	CK	81
			CL	≥ 10000		CL	2720		CL	214
	SO ₂	147	CK	111	142	CK	942	139	CK	2981
			CL	≥ 10000		CL	6521		CL	≥ 10000

Halogenated compounds do not follow a defined trend (Table 123). The activity in the halogenated sulfones decreases with more electronegative substituents at the C6 position; but this does not apply to the thiochromanone analogs. The best compound of this small series is the 6-chlorothiochromanone thiosemicarbazone with an IC₅₀ value of 63 nM.

Table 123. Comparison of the Activity of Halogenated Thiochromanone vs Halogenated Sulfone TSCs against Cathepsins K and L

IC ₅₀ (nM)	X	<i>R</i> ₁								
		-F		-Cl		-Br				
	S	114	CK	1069	126	CK	62.9	115	CK	499.5
			CL	742		CL	228		CL	152
	SO ₂	152	CK	5918	138	CK	1441	136	CK	150
			CL	≥10000		CL	ND		CL	574

Substituted analogs with nitro and hydroxyl groups also were tested. Three out of four compounds showed no inhibition towards cathepsin K. The 6-nitrothiochromanone thiosemicarbazone was the only compound with an IC₅₀ value (164.7 nM, Table 124). Compound **118** also showed activity toward cathepsin L when compared to cathepsin K inhibition (2.5-fold).

Interestingly, the electrophilic trifluoromethyl thiochromanone and sulfone thiosemicarbazones (**129**, Table 125) showed remarkable activity against cathepsin K with activities as low as 21.5 nM.

Table 124. Comparison of the Activity of Nitro and Hydroxyl-Substituted Thiochromanone Versus Substituted Sulfone TSCs against Cathepsins K and L

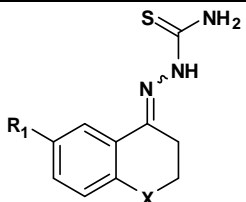
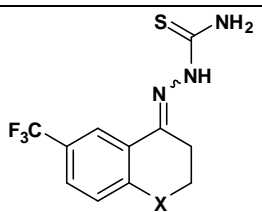
IC ₅₀ (nM)	X	R ₁					
		-NO ₂		-OH			
	S	118	CK	164.7	117	CK	≥10000
			CL	67.85		CL	≥10000
	SO ₂	143	CK	≥10000	145	CK	≥10000
			CL	112		CL	≥10000

Table 125. Comparison of the Activity of Trifluoromethyl Thiochromanone Versus Trifluoromethyl Sulfone TSCs against Cathepsins K and L

IC ₅₀ (nM)	X	
		S
	<	
	129	141
CK	21.5	204.2
CL	284.1	259.8

The position of the fluoro substituents in both thiochromanones and sulfones were also examined (Table 126). In general, dihalogenation did not improve the activity of thiochromanone. However, the 6,7-difluoro analogs (**119**, **144**) were more potent than their 6,8-difluoro analogs (**120**, **146**).

The effect of substitution using different halogen substituents was also assessed by examining halogenated isomers (Table 127). Interestingly, inhibitor activity increases when bromine is at C6, confirming what was observed with monohalogenated thiochromanones and sulfones, where brominated compounds were more active than their

fluorinated counterparts. Compound **149** is 15-fold more active than its isomer analog **148**.

Table 126. Comparison of the Activity of the Effect of the Position of Fluorosubstituents in Thiochromanone and Sulfone TSCs against cathepsins K and L

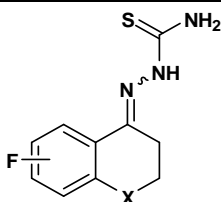
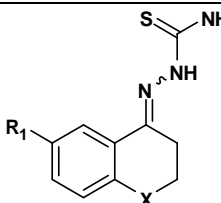
IC ₅₀ (nM)	X	<i>F</i>					
		6,7-F ₂		6,8-F ₂			
	S	119	CK	177	120	CK	1558
			CL	54		CL	1500
	SO ₂	144	CK	1273	146	CK	≥10000
			CL	3650		CL	≥10000

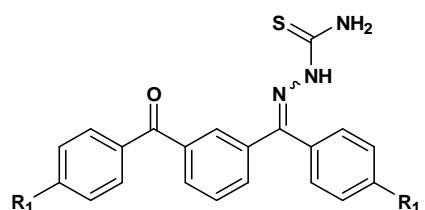
Table 127. Comparison of Halogenated Thiochromanone Isomers Versus Halogenated Sulfone Isomers against Cathepsins K and L

IC ₅₀ (nM)	X	<i>R₁</i>					
		6-fluoro-8-bromo		6-bromo-8-fluoro			
	S	124	CK	419.2	122	CK	44.2
			CL	≥ 10000		CL	434.2
	SO ₂	148	CK	≥ 10000	149	CK	649.6
			CL	≥ 10000		CL	1117

Lastly, the recent discovery of a new potent thiosemicarbazone derivative scaffold with an IC₅₀ value of 9.85 nM (Compound **156**) for cathepsin L, led to the synthesis of a new generation of thiosemicarbazones that were also tested with cathepsin K (unpublished results). Chemically, compound **156** is a benzoyl benzophenone thiosemicarbazone. The small subclass of TSC that was screened is composed of eight analogs with similar structures.

Unsubstituted and *para*-substituted analogs were also examined (Table 128). The unsubstituted analog showed an extraordinary potency with an IC₅₀ less than 20 nM (IC₅₀: 17.4 nM). Substituted analogs did not prove to have comparable activities relative to the unsubstituted benzoyl benzophenone thiosemicarbazones. IC₅₀ values varied between 1000 and 10000 nM. The most important feature of these compounds is their improved selectivity toward cathepsin L compared to their cathepsin K activity. Compound **157** is 340-fold more active toward cathepsin L. This compound could be used as a potentially good cathepsin L inhibitor with improved selectivity.

Table 128. Inhibition of Human Cathepsins K and L by Difunctionalized Benzoyl Benzophenone Thiosemicarbazones



	<i>R</i> ₁	-H (156)	-F (157)	-OCH ₃ (158)
IC ₅₀ (nM)	CK	17.4	8500	2092
	CL	9.9	24.3	587

Advanced Kinetic Studies

Previous results shown in Chapter Two indicated the 3-3'-dibromobenzophenone thiosemicarbazone (**1**) was a slow-binding, time-dependent, reversible, competitive inhibitor of cathepsin L. It also was able to arrest cell migration and cell invasion of MDA-MB-231 cells. These promising results led to the additional research using cathepsin K as a molecular target in order to characterize its mode of inhibition. Fluorometric-based assays were used to investigate and characterize compound **1**. The

possibility of the inhibition of the activation of procathepsin K was also explored by western blotting techniques.

Kinetic Analysis of 3-Bromo-3'-Bromobenzophenone Thiosemicarbazones (1) as Cathepsin K Inhibitor

Effect of Inhibitor Concentration on Cathepsin K Progress Curves

Cathepsin K (1.5 nM) was added to four reactions with different concentration of the inhibitor, ranging from 0 to 10 μM of **1**. Reactions were initiated by the rapid addition of the substrate (Z-FR-AMC, final concentration: 50 μM). The release of AMC from the nonfluorogenic substrate was monitored every 3 seconds for fifty minutes. Figure 95 shows the effect of inhibitor concentration on cathepsin K progress curve reactions. The time dependence inhibition of cathepsin K by **1** was more easily observed at lower inhibitor concentrations (100 nM), because enzymatic activity was completely inhibited at higher concentrations of **1** and the AMC release was not observed. Data sets for 100 nM were fitted to equation 1.7, by nonlinear regression analysis using GraphPad 5.0. P is the concentration of product (μM), v_i and v_s are the initial and steady-state velocities ($\mu\text{M/s}$), t is the time in seconds and k_{obs} is the rate constant for conversion of the initial velocity v_o to the steady state velocity v_s . k_{obs} units are given in s^{-1} . The equation was entered into the computational software (GraphPad 5.0), knowing that P and t are the dependent and independent variables, while keeping the velocities and the rate constant as unknown. For each case, the constraints for their calculation were to give a possible positive value (i.e. $k_{\text{obs}} \geq 0$). It is also worth noting that equation 1.7 is only valid when substrate depletion is insignificant. Therefore, some points were excluded in

every case for data fitting. Velocities, rates, r^2 , and points analyzed for 50 μM Z-FR-AMC are shown in Table 129.

$$P = v_s t \frac{(v_o - v_s)}{k_{obs}} (1 - e^{-k_{obs}t}) \quad (1.7)$$

Table 129. Calculated Kinetic Parameters from Eq.1.7 for Cathepsin K Progress Curves with 50 μM Z-FR-AMC

[I] (μM)	0.1
v_s ($\mu\text{M}/\text{s}$)	0.0005618
v_i ($\mu\text{M}/\text{s}$)	0.5440
k_{obs} (s^{-1})	0.6773
r^2	0.9102
Points analyzed	336

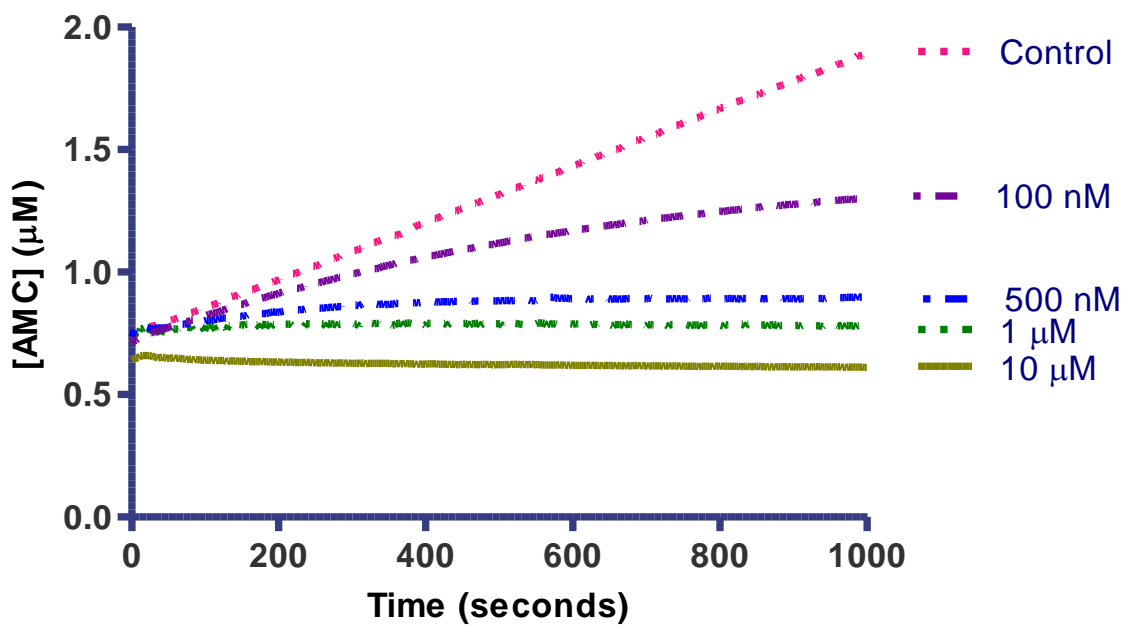


Figure 95. Cathepsin K Progress Curves with **1** Using 50 μM Z-FR-AMC

Data sets were fitted into equation 1.7. Only one concentration (0.1 μM) was able to be resolved. The results indicate **1** is a time-dependent inhibitor of cathepsin K. Additional studies will be required to confirm these results.

Effect of Preincubation Studies on Cathepsin K Inhibition Assays using 1

Experimental results using cathepsin K progress curves using **1** provided strong evidence that **1** is a time-dependent inhibitor. To confirm the time-dependence of **1**, the influence of preincubation time on the inhibitory potency of thiosemicarbazones and its IC_{50} values was studied. Compound **1** IC_{50} values were determined at 6 different preincubation times ranging between 0 and 180 minutes. Substrate concentration was 50 μM . Inhibitor final concentration varied between 0 and 10 μM . Table 130 and Figure 96 show the results of the studies.

Table 130. Effect of Pre-Incubation Times on IC_{50} Values of **1** against Cathepsin K

<i>Pre-incubation times (min)</i>	<i>$\text{IC}_{50} \pm \text{Standard Error (nM)}$</i>
0	529.3 ± 44.2
5	35.2 ± 2.4
15	$13.9 \pm .5$
45	5.1 ± 0.3
90	3.3 ± 0.5
180	1.6 ± 0.2

The effect of preincubation on the potency of **1** as seen in Table 130 is quite evident. The dibromobenzophenone thiosemicarbazone **1** had an IC_{50} of 529 nM when cathepsin K was not preincubated with the compound. Compound **1** improved its activity 15-fold with only five minutes preincubation time. The potency of **1** increased with longer preincubation times. Finally, the lowest IC_{50} value was determined when **1** was preincubated for forty-five minutes with cathepsin K. These results confirmed the strong

dependence of IC₅₀ value determination with respect to the preincubation time parameter. Controls (i.e. uninhibited cathepsin K) were monitored at every preincubation time. There was no significant loss of catalytic activity between 0 and 45 minutes. On the other hand, the activity of the enzyme decreases significantly at longer preincubation times (90 and 180 minutes), confirming the fragility of this enzyme. Maximum activity was found at 45 minutes. However, its activity was reduced by 50% and 75% for 90 and 180 minutes, respectively. Thus, the IC₅₀ values do not represent an accurate measurement of the inhibitory activity of **1**.

A general comparison between the results shown in Figure 90 and Figure 96 showed a discrepancy in results. It was previously stated activated cathepsin K is extremely unstable and lost 30% of its activity within the first ten minutes of activation. However, the pH of the activation was under acidic conditions ([NaOAc]: 5.5 mM; pH: 3.5) while preincubation studies where activated cathepsin K is kept for longer preincubation times are at pH 5.5 with inhibitor and detergent ([NaOAc]: 5.5 mM; pH: 3.5). Similar findings have also been reported. Lecaille and coworkers found that the activity of active cathepsin K is greatly enhanced under slightly acidic conditions ($5 \leq \text{pH} \leq 7$).³⁵⁹ Furthermore, the effect of DMSO in the stability of the enzyme could not be ruled out. Cathepsin K was preincubated for longer periods in the presence of the aprotic solvent ([DMSO]: 2.8%) which could have also helped in the stability of the protease.

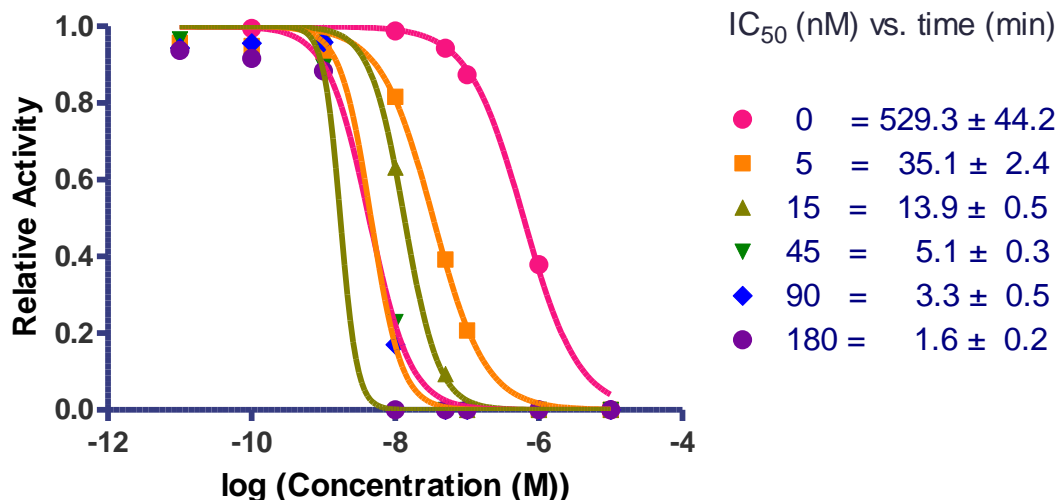


Figure 96. Effect of Preincubation Time on IC₅₀ Values of **1** against Cathepsin K

*Determination of K_i^{app} using Morrison's Quadratic Equation. Effect of Preincubation Time using **1***

The data obtained from the preincubation were further analyzed. The possibility that **1** was a tight-binding inhibitor was analyzed with the Morrison's quadratic equation which describes this behavior. (See equation 1.11).

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

The rates v_i and v_o are the inhibited and uninhibited cathepsin K velocities (RFU/s); $[E]_T$ (nM), the total concentration of enzyme found in solution (free enzyme and inhibitor-enzyme complex); $[I]_T$ (nM) is the total concentration of inhibitor present in solution (free inhibitor and inhibitor-enzyme complex) and K_I (nM) is the inhibition constant, often referred as the dissociation constant. The equation may be solved to give two possible answers. However, the equation is written so that only there is an only

possible answer that fits physiological conditions (i.e. $K_I^{\text{app}} > 0$). GraphPad 5.0 was used to fit the data by non-linear regression after data manipulation for every preincubation time point. Inhibitor, substrate, and enzyme concentrations, as well as K_M , and, K_I^{app} , were all in micromolar units (μM). The residual activity (or v_i/v_o) was normalized to 1 (i.e. v_o : 1 and $0 \leq v_i \leq 1$). Normalized residual activity and [I] were defined as the dependent and independent variables respectively. Nonlinear regression was applied using the following conditions: [S]: 50 μM , K_M : 11.8 μM , $[\text{E}]_T$: 0.0015 μM and v_o : 1. Normalization and nonlinear conditions were used for three or four independent experiments per preincubation time. Average and standard errors can be seen in Table 131 and Figure 97.

Table 131. Effect of Preincubation Time in K_I^{app} Values of **1** against Cathepsin K

<i>Pre-incubation times (min)</i>	<i>$K_I^{\text{app}} \pm \text{Standard Error (nM)}$</i>
0	103.7 ± 12.8
5	6.5 ± 0.2
15	2.2 ± 0.1
45	0.8 ± 0.02
90	0.5 ± 0.05
180	0.3 ± 0.01

The data also show the apparent inhibition dissociation constants (K_I^{app}) are also time-dependent. No preincubation allows having a good inhibition constant with a value of 104 nM. Furthermore, the potency is extremely enhanced (more than 15-fold) for the calculated value at standard preincubation time (5 minutes). The apparent K_I for maximum enzymatic activity (preincubation time: 45 minutes; K_I^{app} : 0.8 nM) was 130-fold better when compared with the corresponding value at no preincubation time.

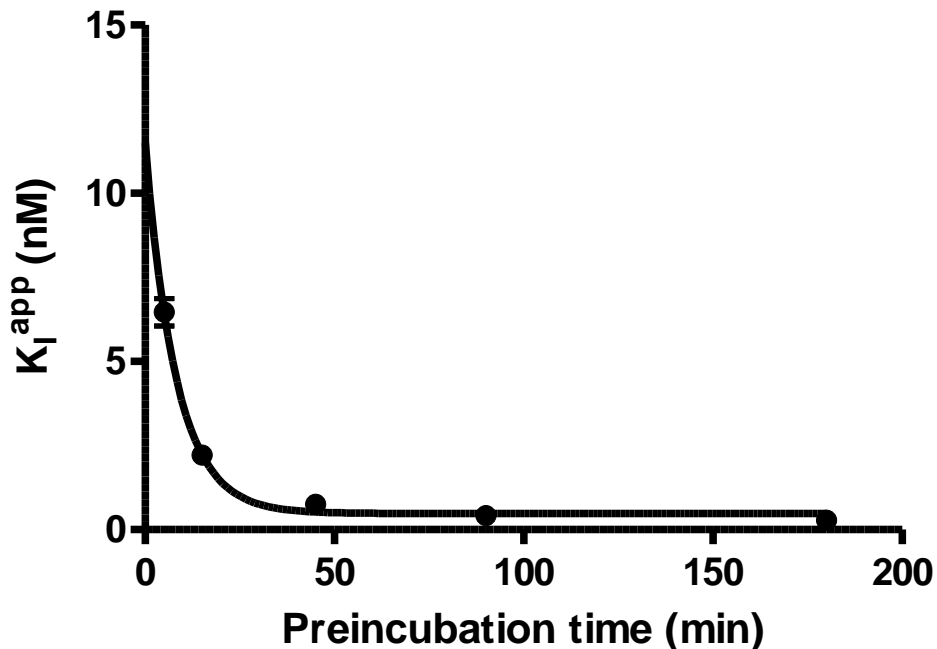


Figure 97. Effect of Preincubation Time in K_I^{app} Values of **1** against Cathepsin K

Cathepsin K Reversibility Studies

Compound **1** was found to be a time-dependent inhibitor in the preincubation studies. The next stage in the characterization of **1** as a cathepsin K inhibitor was to demonstrate if the benzophenone thiosemicarbazone inhibits cathepsin K in a reversible fashion. A mixture containing 100 X cathepsin K and 10 X IC_{50} (preincubation time: 5 min), which are 150 and 350 nM respectively, were incubated at 25 °C for 60 minutes. The inhibition of cathepsin K by **1** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. Final conditions were 1.5 nM cathepsin K, 1.7 nM of dibromobenzophenone thiosemicarbazone (**1**) and 50 μ M Z-FR-AMC. Additionally, a control experiment (cathepsin K with 4% DMSO as a control vehicle) was also set up. Complete experimental details can be found in the Material and Methods sections. Figure 98 shows the release of AMC for the first 2000 seconds. However,

uninhibited and inhibited reactions were followed for a total time of four hours. A concentrated solution of cathepsin K was able to recover its catalytic activity very slowly after the rapid dilution with assay buffer containing Z-FR-AMC. Thus, a linear regression was applied to the first 2000 seconds of the reactions to determine cathepsin K activity. Cathepsin K activity for untreated reaction was found to be 837 pM AMC/s. Similarly, the inhibited reaction when the system was preincubated for one hour was 87 nM AMC/s. The results showed cathepsin K recovered its activity less than 10% when **1** was present in solution. Enzymatic activity can be seen within the first 300 seconds.

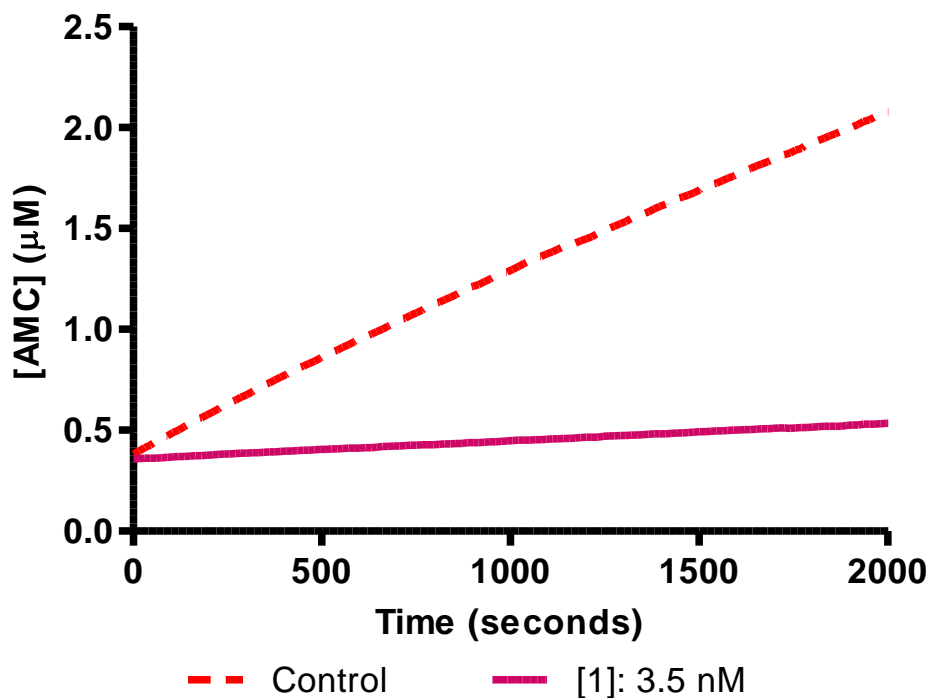


Figure 98. Cathepsin K Reversibility Studies with **1** Using 50 μM Z-FR-AMC

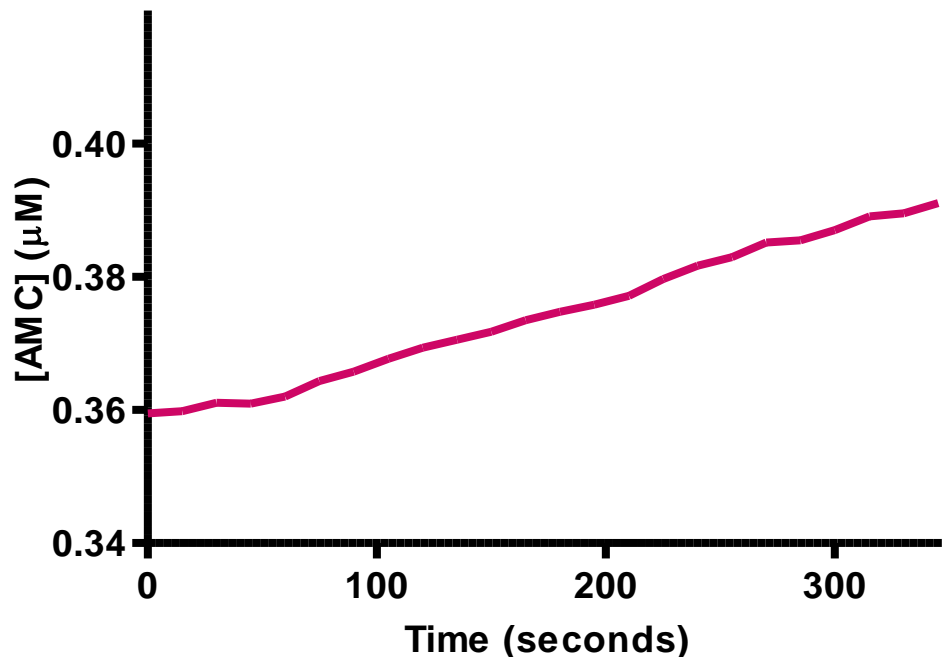


Figure 99. Cathepsin K Reversibility Studies with **1** using 50 μM Z-FR-AMC

Effect of Substrate Concentration (Z-FR-AMC) on IC_{50} Values

We finally tried to determine the mode of inhibition of **1** as a cathepsin K inhibitor. Therefore, we investigated the effect of substrate concentration to verify if these synthetic compounds were competing for the same site on the enzyme. Three substrate concentrations (final concentrations: 50, 25, and 10 μM) were used to determine IC_{50} values for compound **1**. Cathepsin K and the series of inhibitors were incubated for five minutes. Results can be found in Table 132 and Figure 100.

Table 132. Effect of Substrate Concentration on IC_{50} values of **1** against cathepsin K

<i>[Z-FR-AMC] (μM)</i>	<i>$IC_{50} \pm \text{Standard Error (nM)}$</i>
50	35.1 ± 2.5
25	25.8 ± 1.7
10	17.2 ± 1.1

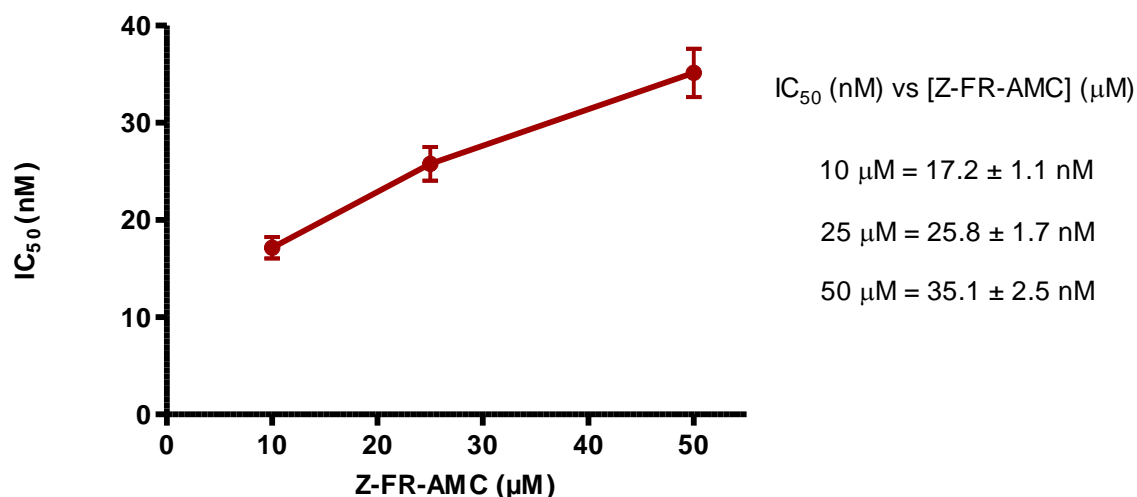


Figure 100. Effect of Substrate Concentration on IC₅₀ Values of **1** against Cathepsin K

The effect of substrate concentration was investigated using **1** as one of the lead compound of the series of thiosemicarbazones. Lowering the amount of substrate in solution gives a subtle, yet evident increase in the potency of the dibrominated analog (**1**). According to literature, a positive linear behavior (i.e. $IC_{50} \propto [S]$) is an indication the compound acts as competitive inhibitor, that is, both substrate and the compound compete for cathepsin K active site.

Detection Limits of Procathepsin K by Fluorescent Western Blotting

Finally, we decided to follow the conversion of procathepsin K to its mature form, using molecular biological techniques. However, it was first necessary to determine the detection limits of both species, procathepsin K and mature cathepsin K when using Western Blotting. The objective of this experiment was the detection limits for the activation of procathepsin K under acidic conditions at room temperature. For this experiments, Western Blotting by using fluorescent-labeled antibodies were used. Inactive human procathepsin K was detected in the nanomolar range using the protocol

previously described in this chapter. Fourteen samples ranging from 5 ng to 200 ng were loaded in a 4-12 % SDS-PAGE gel, blotted and detected. Figure 101 shows the detection limits of procathepsin K using fluorescent Western Blotting. All the samples, with the exception of 5 ng (lane 2) were detected. The intensity of the bands was increased with the protein sample concentration. The western blotting image also revealed that small amounts of procathepsin K had been cleaved and mature cathepsin K was present (lanes 9: 45 ng -14: 200 ng).

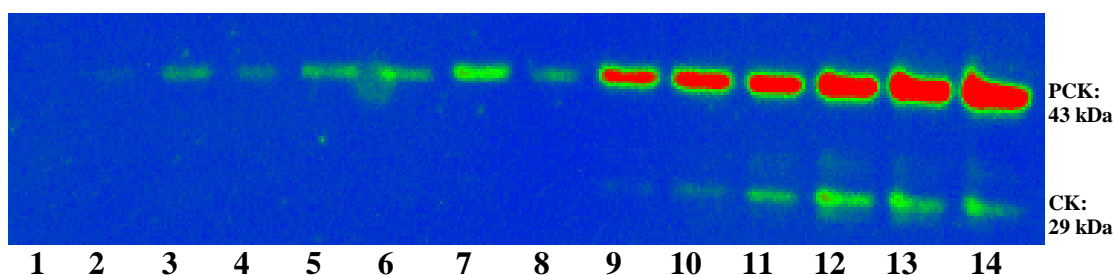


Figure 101. Western Blotting of the Detection of Procathepsin K. Legend: **1**: 5 ng (not detected); **2**: 10 ng; **3**: 15 ng; **4**: 20 ng; **5**: 25 ng; **6**: 30 ng; **7**: 35 ng; **8**: 40 ng; **9**: 45 ng; **10**: 55 ng; **11**: 65 ng; **12**: 100 ng; **13**: 150 ng; **14**: 200 ng; PCK: procathepsin K (mW~ 43 kDa); CK: active cathepsin K (MW ~29 kDa)

Inhibition of the Activation of Human Procathepsin K

Cysteine proteases are expressed as proenzymes in mammalian tissues.

Cathepsins can be activated under acidic conditions (i.e. lower pH) or by using a procathepsin as a substrate in a hydrolytic reaction (See Chapter 1). The objective of this experiment was to investigate the possibility of the inhibition of procathepsin K under acidic conditions in the presence of **1**, a slow-binding, reversible, competitive inhibitor of mature cathepsin K.

The inhibition of the activation of human procathepsin K was followed at low temperatures. Previous reports have found the activation time increases when activation under acidic conditions is performed at low temperatures. The reaction was followed by using two different techniques: western blotting and using enzyme activity assays with Z-FR-AMC. Briefly, two samples of procathepsin K were activated at 4 °C at pH 3.5. Activation was followed for 6.75 hours and two aliquots were taken every 45 minutes to be monitored using enzymatic assays and western blotting. Final conditions during the activation process are: 1 μM active cathepsin K, 0.7 mM EDTA, 333.3 mM NaCl, 5.4 mM NaOAc pH 3.5, 4% DMSO, and 20 μM of **1** (for the treated samples). Samples for western blotting were inactivated with 1.0 % SDS and heat, while samples for the fluorometric enzyme activity assay were used as a stock solution for standard kinetic assays. Final conditions for this assay are: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 DTT, 0.01 % Brij, 1.5 nM inhibited (or control) cathepsin K, 4% DMSO, and 50 μM Z-FR-AMC. Standard procedures using semi-dry protein transfer and a fluorescent western blotting kit were utilized to detect 200 ng of protein that could have been activated or inhibited. Results of western blotting experiments can be seen in Table 108.

Figure 104-A shows the first 270 minutes of activation at the specified conditions without any inhibitor. Three bands can be observed at time 0 minutes. Calculated molecular weights were 43, 35 and 28 kDa. Two of these bands correspond to molecular weights of procathepsin K (MW ~ 43 kDa) and mature cathepsin K (MW ~ 29 kDa) respectively according to suggested values found in literature. The third band (MW: 35 kDa) has also been found in other experimental procedures and suggests that cathepsin K activation is a multistep process where multiple cleavage points produce the final mature

macromolecule. The presence of a band at 28 kDa suggests that procathepsin K had already started the activation process. As previously stated, activation of cathepsin K can be a spontaneous process under acidic conditions whereby an active molecule of cathepsin K can facilitate the activation of other proenzyme molecules. Results confirmed activation of the proenzyme is a lengthy process and can take several minutes. The presence of the 43 kDa band can be observed between 2.5 and 3 hours. The activation of the enzyme produces two clear bands at 35 and 28 kDa. The formation of a transient mature cathepsin K is evident and stable for the first three hours.

The presence of this band was not observed during the rest of reaction as seen in Figure 104-A. The third band (28 kDa) was observed during the entire experiment and the intensity of the bands was increasing at longer activation time points suggesting that the proenzyme was completely activated between 5 and 6 hours.

Table 108-B shows the experimental results of the inhibition of the activation of proenzyme by analog **1**. Once again, three well defined bands with molecular bands were observed (43, 35, and 28 kDa) corresponding to proenzyme, transient mature cathepsin K and fully activated cathepsin K, respectively. However, the biggest difference between both experiments is that the presence of the proenzyme was detected for periods of times longer than 3 hours. The intensity second band (35 kDa) was not as evident as in the case of the untreated samples. Finally, the bands corresponding to the fully activated cathepsin K were barely seen at time points up to 180 minutes. Activated cathepsin K started to appear after two hours. These findings suggest that analog **1** was able to inhibit the activation of procathepsin K. It is unclear whether the synthetic compound is able to inhibit the activation of the proenzyme by direct inhibition of the macromolecule, or if

the compound inhibits only activated cathepsin K, which is partially responsible of the activation of other molecules that are in the proform.

The first set of samples was used to test the activity of mature cathepsin K under acidic conditions at low temperatures using standard fluorometric enzymatic assays.

Figure 103 depicts the untreated and treated samples respectively. Briefly, untreated and treated samples were not preincubated in order to measure the activity of activated enzyme under low temperatures. Reactions started by the addition of 500 μM Z-FR-AMC. Final conditions for cathepsin K and Z-FR-AMC are 1.5 nM and 50 μM , respectively. Measurements were taken every 6 seconds for 1200 seconds.

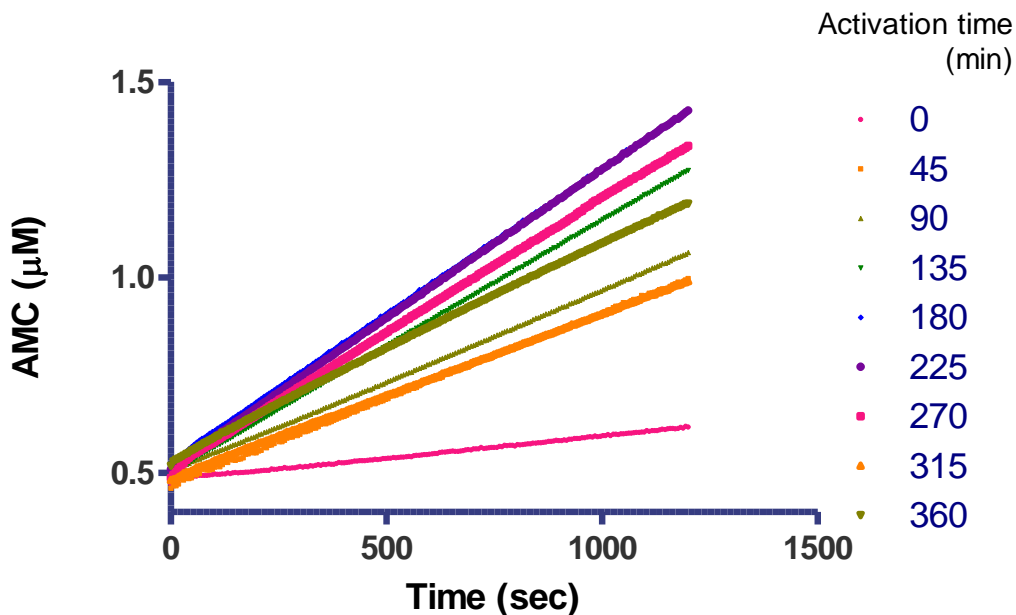


Figure 102. Activation of Procathepsin K Under Acidic Conditions at 3 °C

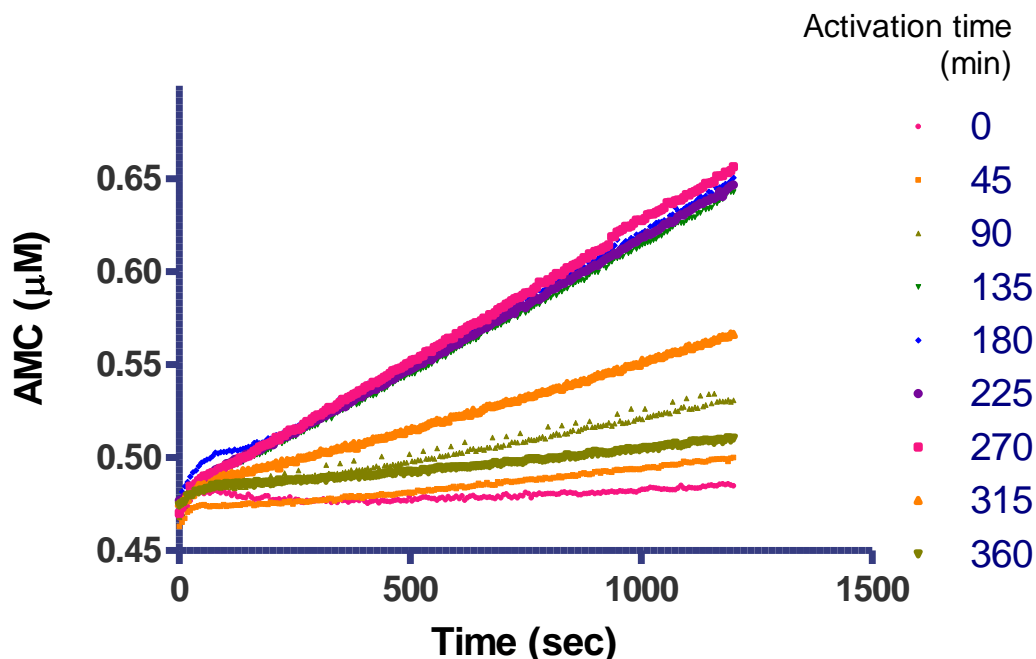


Figure 103. Inhibition of the Activation of procathepsin K by **1** under acidic conditions at 3 °C

Cathepsin K activity was also estimated in nM AMC/s rates. Maximum activities for untreated and treated samples were 0.77 and 0.13 nM AMC/s respectively. Table 116 and figure 90 show the complete evaluation of cathepsin K activity. This represents an 80% inhibition compared to the untreated sample.

Activation of untreated samples occurs rapidly within the first 45 minutes, possibly due to the abundance of natural substrate (procathepsin K) present in solution. The activity increases up to 4 times in that period of time (0.11 vs 0.44 nM AMC/s). Maximum activity was stable between 180 and 225 minutes of activation time, which corresponds with visual observations for the western blotting experimental results. Activity of the active enzyme slowly decreases after 270 minutes. The activity after 6 hours of activation time (3 hours after maximum activity) is 80% with respect to the activated enzyme at 180 minutes.

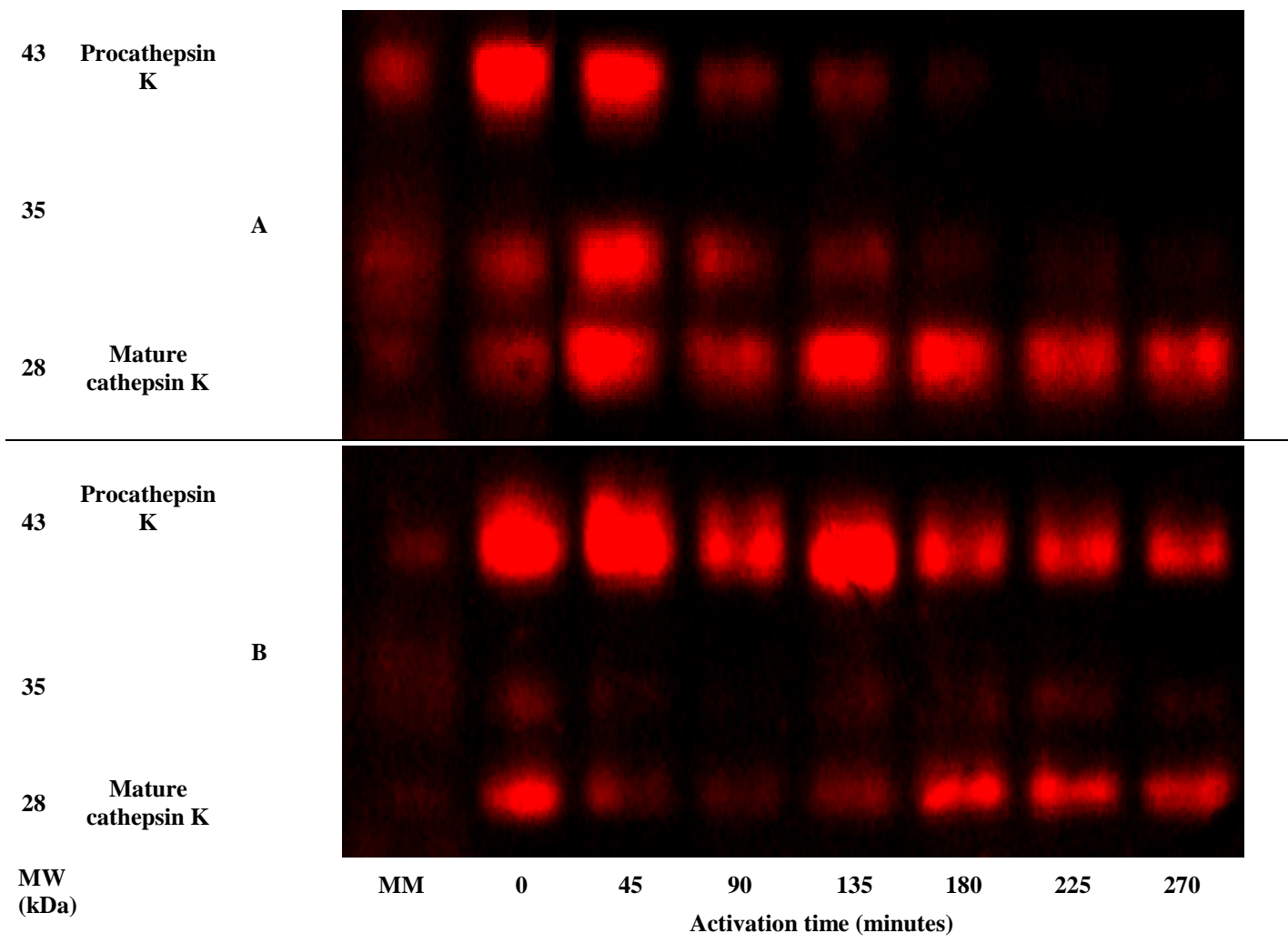


Figure 104. Western Blotting of the Inhibition of Procathepsin K Activation by 1. A. Untreated Sample. B. Treated Sample with 1

On the other hand, activation of treated samples occurs very slowly. Overall activities of the reactions for the first 90 minutes showed no catalytic activity in any sample. The inhibitory potency of **1** with regard to the activation of procathepsin K is clearly seen in the activity of the rest of the samples. Activity of cathepsin K was stable for more almost two hours and maximum activity was observed at 270 minutes, which is 1.5 hours after maximum activity for untreated samples was completed. Samples with inhibitor showed little or no activity after 5 hours of activation time.

Closer examination of the samples reveals that samples showed catalytic activity, probably due to the presence of activated cathepsin K that was not inhibited. This activity is quickly inhibited by analog one within the first 35 seconds and inhibition is observed for the rest of the reaction See Figure 105.

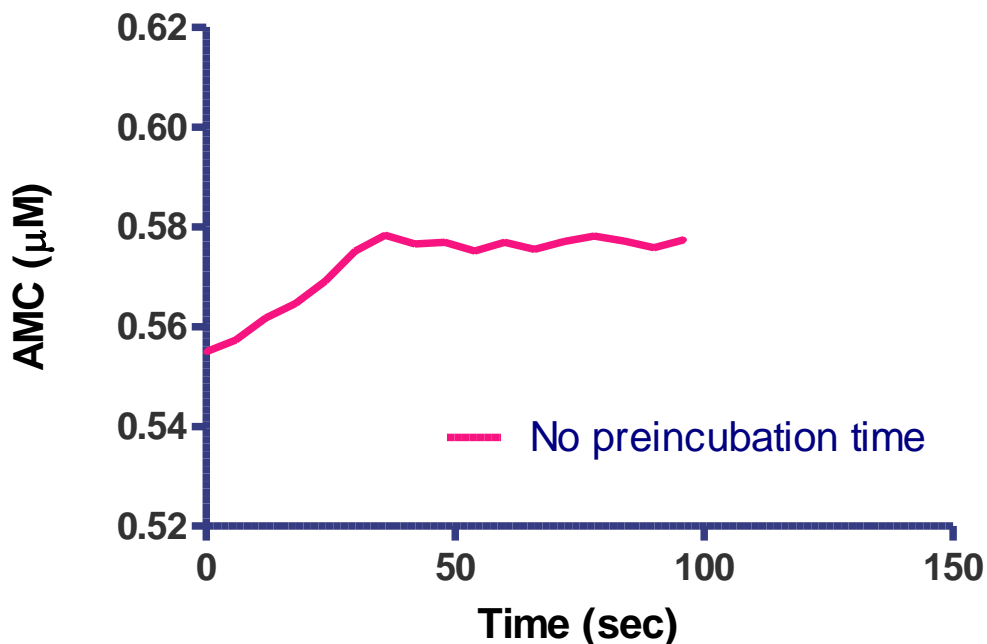


Figure 105. Inactivation of the Activation of Procathepsin K by **1** Under Acidic Conditions at 3 °C (No Preincubation Time).

Table 133. Catalytic Activity of Activated Cathepsin K Measure in nM AMC/s

Activation time (min)	Untreated samples (nM AMC/s)	Treated samples (nM AMC/s)
0	0.11	0.00
45	0.44	0.00
90	0.46	0.00
135	0.65	0.13
180	0.77	0.13
225	0.78	0.13
270	0.73	0.14
315	0.44	0.07
360	0.60	0.00

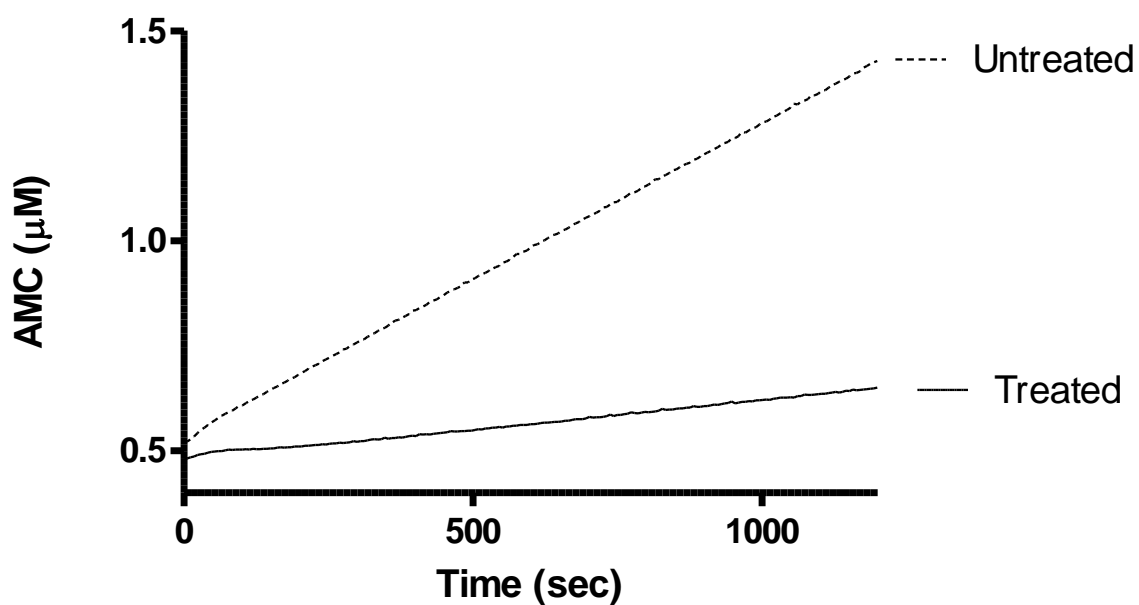


Figure 106. Untreated and Treated Cathepsin K (1.5 nM) with **1** (30 nM) at 3 Hours After Reactions of Activation Process Started

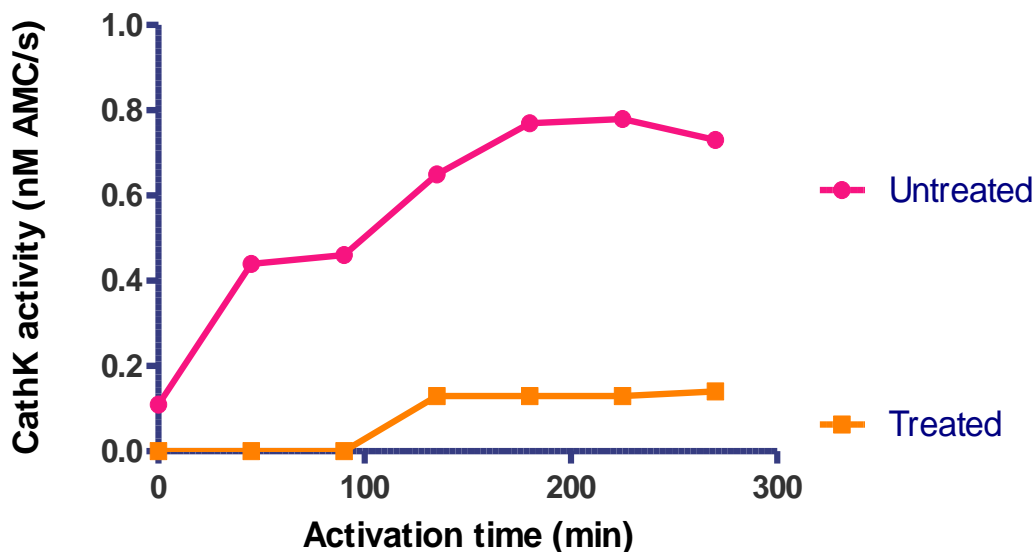


Figure 107. Catalytic activity of Activated Cathepsin K Measure in nM AMC/s

Inhibition of Cathepsin K Proteolytic Activity by a Thiosemicarbazone Derivative

Cathepsin K is the main protease found in osteoclasts, bone cells responsible for bone resorption processes. Research has shown that cathepsin K is capable of degrading high molecular weight proteins found in the extracellular matrix such as type I and type IV collagens.³⁷² Type I collagen is a major component of the ECM, while type IV is the main macromolecule found in the basal membrane. Thus, we explored the catalytic activity of cathepsin K with type IV collagen from human placenta under physiological-like conditions. Additionally, calf brain tubulin (MW: 55 kDa) was used as a potential cathepsin K substrate for comparison. To date, there is no report of explaining the proteolytic ability of cathepsin K to degrade tubulin. However, cathepsin D, an aspartyl protease, was able to degrade microtubulin-associated protein 2 (MAP-2) and tubulin.⁴²² We also tested the ability of one of the lead inhibitors (**1**, IC_{50} : 35.2 nM), to inhibit the type IV collagenase catalytic activity of cathepsin K. Untreated and treated samples were incubated at 37 °C under acidic conditions (pH 5.5). Six sets of samples were prepared

for this experiment. Three sets (one per substrate) were not treated with **1** (DMSO was used as control vehicle). The fourth (type I collagen), fifth (type IV collagen), and sixth (calf brain tubulin) sets were treated with 20 μ M of **1** in 4% DMSO. Activated cathepsin K was added to the samples containing 0.15% chondroitin 4-sulfate with one substrate. The inhibitor **1** was not preincubated with activated cathepsin K in any of these experiments. The role of C4-S was previously established by Li in recent reports. The collagenase activity of cathepsin K is enhanced by the presence of C4-S while cathepsin L can be inhibited by this glycosaminoglycan. Additionally, C4-S increases the stability of the enzyme when C4-S is added to samples containing cathepsin K.^{371,423}

Inhibition of Cathepsin K Type IV Collagenase Activity by Thiosemicarbazone Derivative 1. (Preincubation Time: 0 Hours)

Figure 108 shows a gel image of inhibition of the collagenase activity of cathepsin K with **1** (no preincubation time). The figure shows the results of degradation of type IV collagen from human placenta using activated recombinant human cathepsin K soluble in acidic solutions.³³⁰ The first sample is untreated type IV collagen that was not treated with cathepsin K, inactivated at the beginning of the experiment in the vehicle control (4% DMSO). Reported literature establishes that collagen is large protein that is divided into three defined chains: α , β , and γ heavy chains and is crosslinked and polymerizes into fibrils. The molecular weights of the heavy chains are approximately 92, 169 and 207 kDa α , β , and chains.⁴²⁴ Results show the presence of three major bands with high molecular weights that are consistent with the reported values for α , β , and γ heavy chains in the literature. Approximate molecular weights are: $\alpha \approx 100$ kDa, $\beta \approx 120$ -170 kDa and $\gamma \approx 270$ kDa (lane 1). The second and third bands of the Figure 108 represent untreated and treated samples that were stopped after 30 minutes of the natural

substrate. Untreated samples contained type IV collagen, C4-S and cathepsin K in DMSO, while treated samples consisted of type IV collagen, cathepsin K, C4-S and compound **1** (20 μ M) in DMSO. In the untreated sample ([**1**]: 0 μ M) the γ chain was degraded in as little as 30 minutes. The α , and β , bands look significantly less intense than their respective control at 30 minutes. The γ chain is completely degraded, while the α , and β bands are very significantly degraded. There is evidence of degradation products when compared to untreated collagen at time 0 (bands with smaller molecular weight, data not shown). Similar results can be seen after three and 4.5 hours of reaction time. However, that is not the case for the treated samples with 20 μ M of cathepsin inhibitor **1**. These results indicate that the dibromobenzophenone thiosemicarbazone (**1**) is capable of inhibiting cathepsin K type IV collagenase activity more than fifty percent. A comparison when using human cathepsin L as the target protease also demonstrates the differences of their collagenase activities. Cathepsin K (70 nM) was able to degrade 3 micrograms of type IV collagen, while 270 nM of cathepsin L (4-fold) were needed to degrade the same amount of protein in the same time.

Additionally, Figure 109 shows the results of samples containing cathepsin K with **1** (20 μ M) and tubulin from calf brain that was immediately added without preincubation. Cathepsin K was able to degrade the protein in a lesser extent. Untreated tubulin shows one defined band with a molecular weight of 55 kDa. Significant degradation of the untreated samples was observed after two hours of reactions. Dibromobenzophenone thiosemicarbazone was able to inhibit the proteolytic activity of cathepsin K up to three hours when using tubulin as a substrate. To date, there is no report showing similar experiments.

Additionally, parallel experiments were conducted to verify the stability of cathepsin K when using C4-S. The stability of the enzyme was dramatically increased by the addition of the glycosaminoglycan. Similar reports were found and the activity of cathepsin K is not affected by C4-S.³⁷¹ Eighty percent of the activity remained after 4.5 hours of reaction when compared to the activity at 0 hours. Furthermore, the activity of the enzyme was constant for the first three hours. Previous experiments showed that cathepsin K is stable for approximately 45 minutes when C4-S is not present in solution (See *Effect of Preincubation Studies on Cathepsin K Inhibition Assays using 1* for a complete list of experimental conditions).

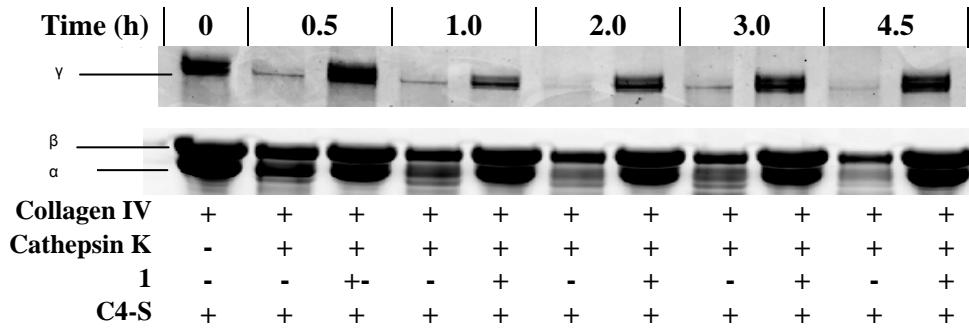


Figure 108. Inhibition of Collagenase Activity of Cathepsin K by **1**, Preincubation Time: 0 hours

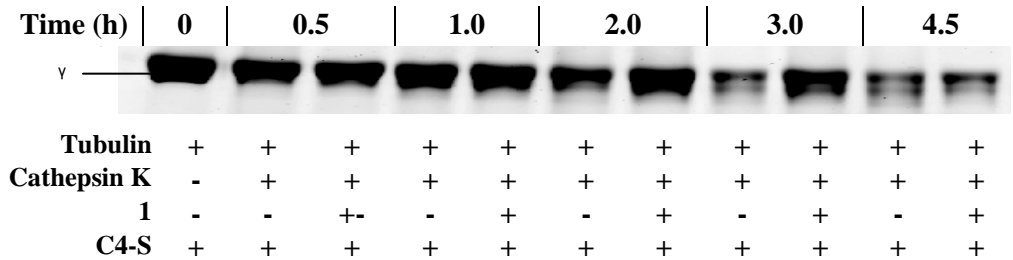


Figure 109. Inhibition of the Proteolytic Activity of Cathepsin K by **1**, No Preincubation Time

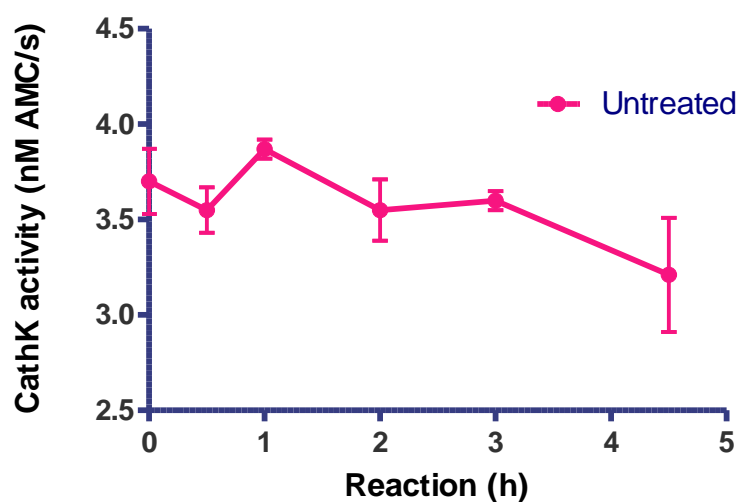
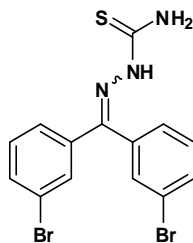


Figure 110. Effect of Chondroitin 4-Sulfate on Cathepsin K Activity

Effect of Chondroitin 4-Sulfate on Cathepsin K Inhibition Assay

Finally we decided to explore the effect of chondroitin 4-sulfate in the inhibitory activity of **1**. The standard IC_{50} assay were conducted using two sets of the thiosemicarbazone analog with the presence or absence of C4-S. Preincubation time was set to 5 minutes. Final conditions were: 150 mM NaOAc pH 5.5, 2.5 mM EDTA, 2.5 mM DTT, 4% DMSO, 50 μ M Z-FR-AMC, 1.5 nM cathepsin K and 0.15% C4-S. A control IC_{50} (i.e. no C4-S) was also tested for comparison purposes. Results can be seen in Figure 111.

Results clearly showed that C4-S did not interfere in the inhibitory activity of the dibromobenzophenone thiosemicarbazone (**1**). Previous reports done by Li and coworkers established the catalytic activity of cathepsin K is not affected when the protease is bound to C4-S.³⁷³ They also found C4-S interacts with the R-domain of the cysteine protease, quite far from the active site of cathepsin K. Our studies also confirmed these statements due to the independence of IC_{50} with respect to C4-S.³⁷³



Without Chondroitin 4-Sulfate
 IC_{50} : 31.9 ± 2.9 nM

With Chondroitin 4-Sulfate (0.15%)
 IC_{50} : 29.9 ± 3.6 nM

Figure 111. Effect of Chondroitin 4-Sulfate on Cathepsin K Inhibition Assay using **1**

Conclusions

More than one hundred TSC analogs were evaluated as inhibitors of recombinant, activated cathepsin K from human liver. The library contained a select group of potent, cathepsin K inhibitors. Figure 112 shows the Structure-Activity Relationship of thiosemicarbazones based on their chemical structures.

Advanced kinetics were used to investigate the mechanism of **1**. The compound was determined to be a time-dependent, slow, reversible, competitive inhibitor of the fluorogenic inhibitor Z-FR-AMC. The increase in IC_{50} values as a function of Z-FR-AMC (substrate) concentration indicates that **1** competes with the substrate for binding to the enzyme active site. Fluorescent western blotting demonstrated that **1** was able to inhibit the cleavage (substrate) and activation of human procathepsin K under acidic conditions and low temperatures.

Finally, the proteolytic activity of mature cathepsin K was examined by using a series of natural substrates type IV collagen from human placenta. The collagenase activity of cathepsin K was inhibited by 3,3'-dibromobenzophenone thiosemicarbazone. Additionally, tubulin from calf brain was used as a potential substrate for the enzyme.

Results indicate tubulin underwent proteolytic degradation by cathepsin K. Analog **1** was also able to delay this activity.

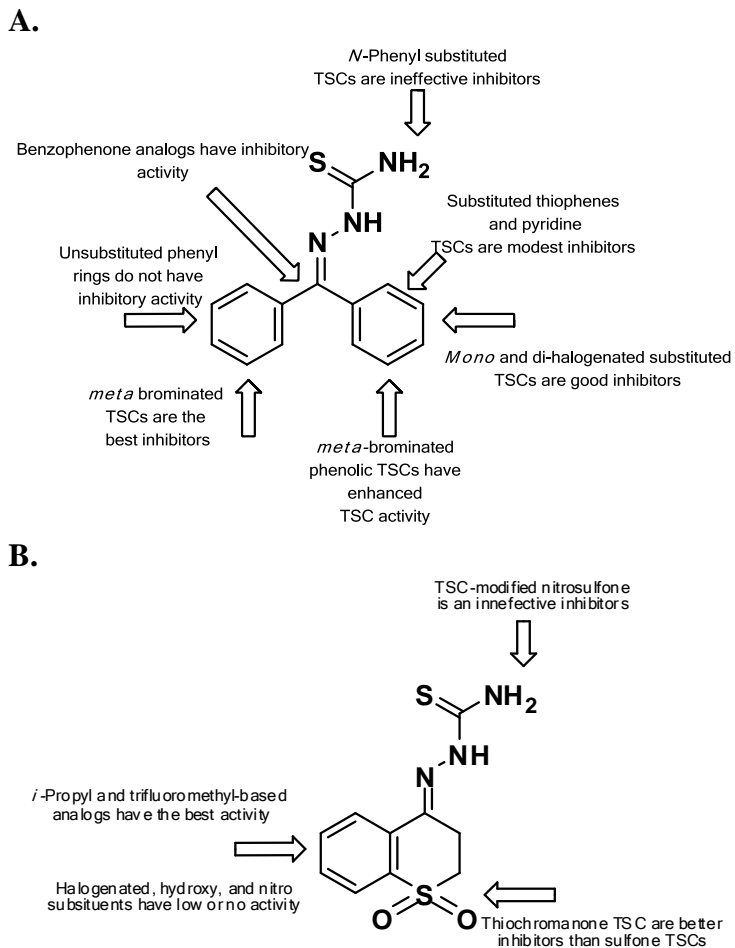
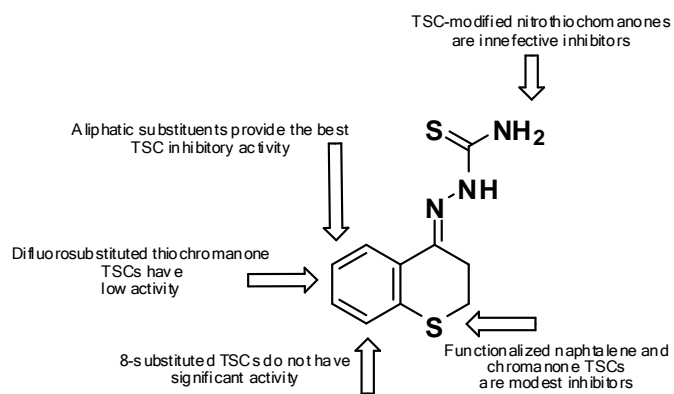


Figure 112. Structure-Activity Relationship for TSCs as Cathepsin K Inhibitors. **A.** Benzophenone TSCs. **B.** Sulfone TSCs

A.



B.

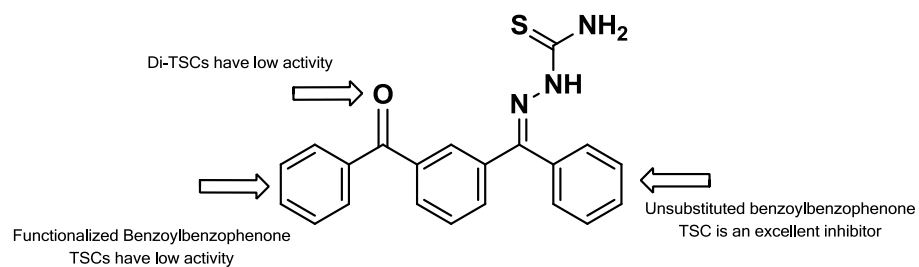


Figure 113. Structure-Activity Relationship for TSCs as Cathepsin K Inhibitors. **A.** Thiochromanone TSCs. **B.** Benzoylbenzophenone TSCs

CHAPTER FOUR

Evaluation of Thiosemicarbazones as Cruzain Inhibitors

Role of Proteases in Parasitic Diseases

Pathological diseases in mammals are often caused by parasites, organisms that get their nutrients at the expense of their hosts. Parasites can be classified as endoparasites or ectoparasites, depending if they live inside or outside the body of the hosts respectively. There are numerous transmission vectors; however, contaminated water, food, insect bites and direct contact between healthy and infected organisms are the most common transmission methods. The list of parasitic diseases includes amoebiasis, enterobiasis, giardiasis, scabies, sleeping sickness, toxoplasmosis among others.⁴²⁵

Parasitic diseases are extremely difficult to understand due to the complexity in the pathology and biology of individual diseases. Some parasites can inhabit their host organisms for years, even decades, being asymptomatic. The lack of effective treatments makes the eradication of these diseases a challenging task. Malaria, as an example, is considered an endemic disease that affects more than 500 million people worldwide.⁴²⁶

One of the main functions of proteases in parasitic diseases is parasite invasion. Parasitic migration and invasion are intrinsically related. Parasites, usually in their larval stage, migrate through organs or tissues. This phenomenon is due to the action of powerful proteases, usually metalloproteases, cysteine, and serine proteases. Research has proven that inhibition of some of these targets might delay or arrest parasitic invasion

in human skin. Recently, Lopez-Quezada was able to arrest skin invasion by *Schistosoma mansoni* by using a serine protease inhibitor.⁴²⁷

General Considerations of Chagas' Disease

History, Statistics, and Geographical Distribution of Chagas' Disease

Carlos Chagas', a Brazilian physician discovered this tropical disease more than one hundred years ago.⁴²⁸ Chagas' disease is also known as mal de Chagas' (in Spanish and Portuguese) and *American trypanosomiasis*.⁴²⁹ Recent demographics showed that more than 8 million people are infected with the parasite and one quarter of the total population in Latin America are at potential risk to be affected.⁴³⁰

Chagas' disease is widely spread in several regions of Latin America, with endemic proportions in more than 18 countries. However, an increasing number of cases have been detected in non-endemic areas, especially in those countries with higher number of immigrants carrying the disease. Patients with Chagas' disease have been reported in Netherlands, Spain, France, Portugal, Switzerland, Germany and United Kingdom.⁴³¹ More than 3 million immigrants live in these countries and the expected number of infected patients could be as high as one hundred thousand cases. This problem has also been extended in the United States and other parts of the world.⁴³² Eleven possible vectors of *Trypanosoma cruzi* have been discovered in characterized in southern states of the nation. Higher prevalence of the vectors (insects infected with the protozoa) can be found in Texas, California and Arizona. A lower number of vectors is also found in Florida, Georgia, Tennessee, Alabama and Louisiana.⁴³³



Figure 114. Estimated Number of Immigrants with *Trypanosoma cruzi* Infection Living in Non-Endemic Countries. (Reproduced from Rassi, page 1391⁴³²)

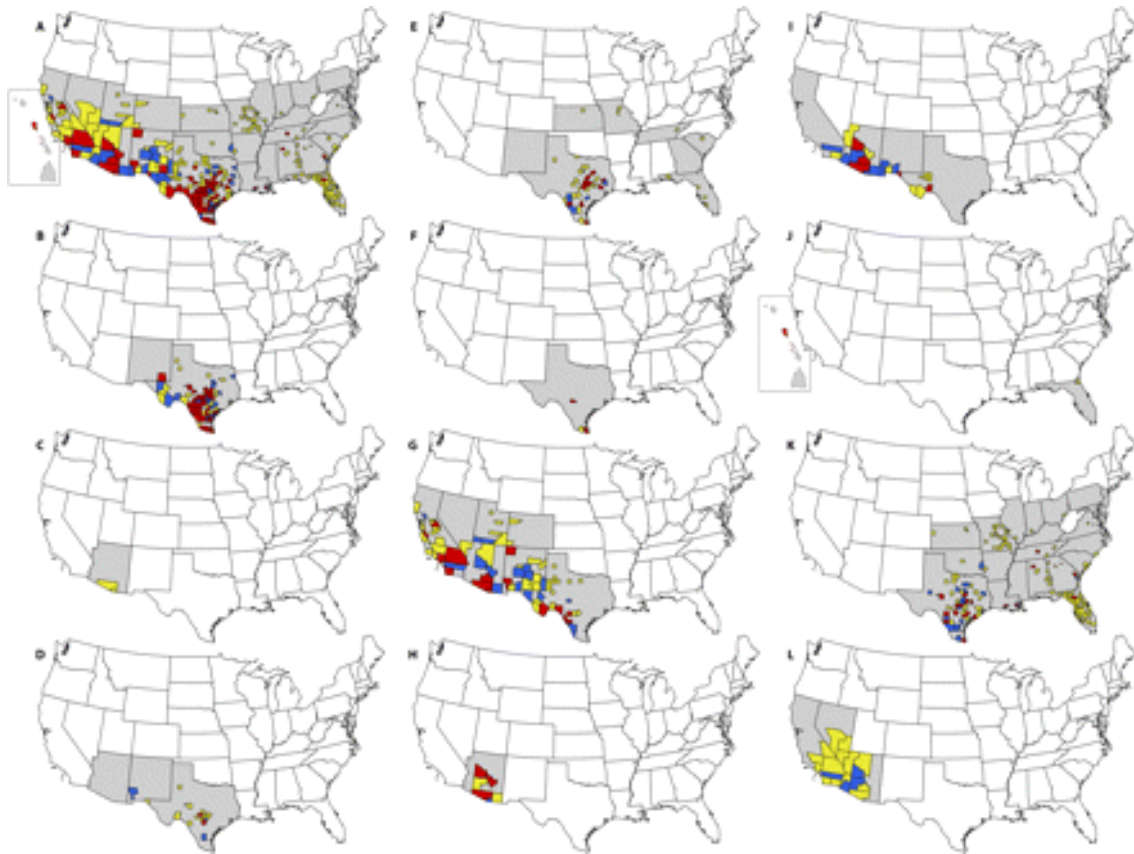


Figure 115. Triatomine Species Geographic Distribution by State (gray areas) and County and *Trypanosoma cruzi* Infection Status by County in the Continental United States and Hawaii. (A) All species; (B) *Triatoma gerstaeckeri*; (C) *T. incrassata*; (D) *T. indictiva*; (E) *T. lecticularia*; (F) *T. neotomae*; (G) *T. protracta*; (H) *T. recurva*; (I) *T. rubida*; (J) *T. rubrofasciata*; (K) *T. sanguisuga*; (L) *Paratriatoma hirsuta*. Red, *T. cruzi*-positive specimens; blue, negative specimens; yellow, no testing reported. (Reproduced from Bern, page 660⁴³³)

Etiology of Trypanosoma cruzi and its Vectors

The protozoan *T. cruzi* belongs to the *Trypanosomatidae* family, or trypanosomes, and can be divided into two major groups: *Stercoraria* and the *T. cruzi* clade.⁴³⁴ Some of the transmission vectors of the trypanosomes found in South America are: *Triatoma infestans*, *Rhodnius prolixus*, *Triatoma dimidiata*, *Mepraia spinolai* and *Mepraia gajardoi*.⁴³⁵ Other names given to the insects include: “kissing bugs” and vinchuca. The former name is given to these insects due to their preference to feed on people’s faces.

Mode of Transmission and Life Cycle of T. cruzi

Several mammals, *Homo sapiens* included, are considered common hosts for Chagas' disease. Mammalian blood is part of the triatomid's dietary habits. Usually, insects that are infected with *T. cruzi* transmit the infection when they bite healthy mammals. The spread of the protozoan occurs when the transmission vectors defecate contaminated feces over the skin of the mammal. The trypomastigotes (first morphological stage of *T. cruzi*) enters in the blood stream by physical and biochemical processes (self scratching and proteolytic degradation by enzymes found in vector's saliva respectively). Then, trypomastigotes differentiate into epimastigotes, the second biological stage of the protozoan via a phagocytosis process by macrophages and leucocytes. The last biological stage of *T. cruzi*, amastigote, multiplies inside of their host cells and differentiates into new trypomastigotes. The new generation of trypomastigotes is transported by the blood system and invades other target organs such as heart, muscles, placenta, glial cells among others.⁴³⁶⁻⁴³⁸ Figure 95 depicts the major stages of the life cycle of *Trypanosoma cruzi*.

Non-traditional modes of transmission have also been detected. Literature reports cases of infected people with Chagas' disease by blood transfusion, organ transplants, contaminated food and placental infection to newborns.⁴³⁹⁻⁴⁴²

Clinical Stages and Symptoms

American trypanosomiasis is divided into three clinical stages: acute, asymptomatic and chronic stages.^{429,443} However, the incubation stage is remarkably important because it happens prior to the acute stage. The mode of transmission determines the duration of that stage. The incubation period varies between five to

fourteen days, if the patient was infected via triatomid feces, and can last up to forty days if the transmission happened due to blood transfusion.

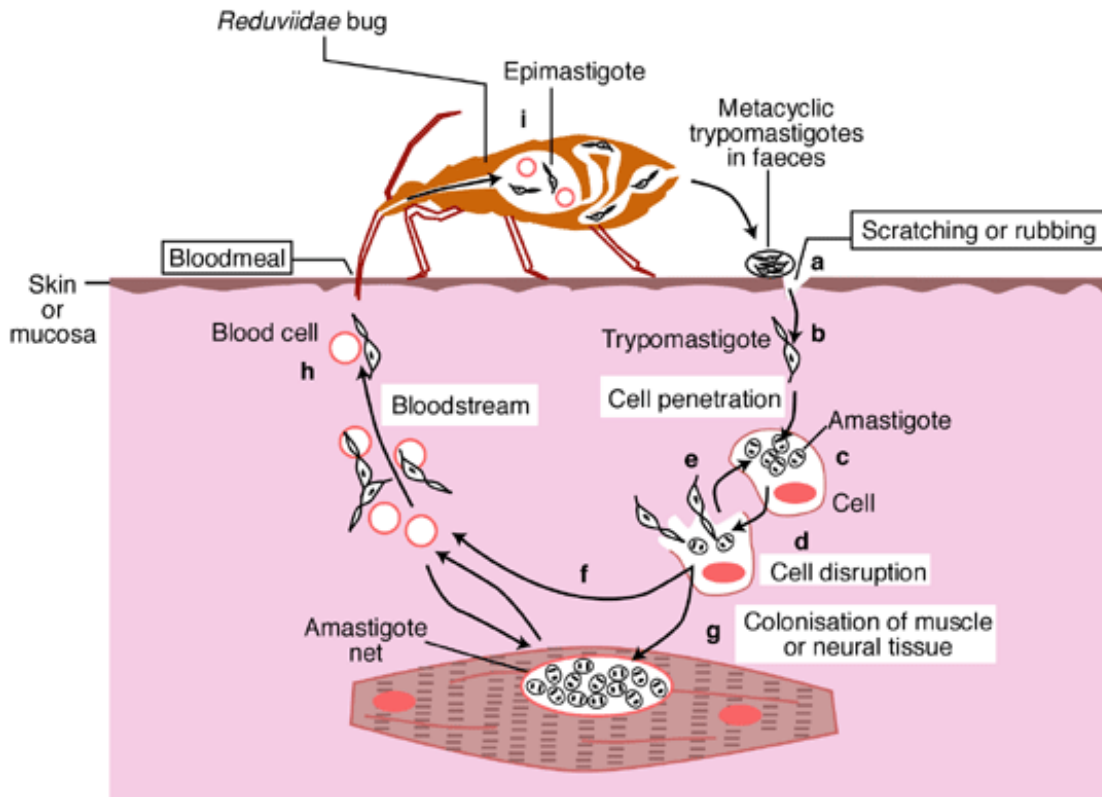


Figure 116. Schematic Representation of the Life Cycle of the Flagellate Protozoan *Trypanosoma cruzi* (reproduced from Macedo, page 3⁴³⁸)

The main characteristic that defines the acute phase is the ability to find the parasites within the blood stream. Patients can develop several symptoms that include: fever, anorexia, myalgia, joint pain, nausea, vomiting, and an unilateral edema, also known as Romana's sign. The latter symptom is particularly helpful for the detection of the infection during this stage.

The asymptomatic stage is the second stage of the disease.⁴⁴⁴ It is also called indeterminate stage because patients do not show any symptoms. In this stage, parasites

practically disappear from the blood stream and they are more difficult to detect. The duration of the stage can vary from a few months to decades.

The chronic stage occurs when some organs start failing. Cardiac and digestive systems are usually the most vulnerable organs to this infection. Patients in a chronic stage develop arrhythmias, cardiac failure, embolic disease, and death in many cases. Digestive anomalies include problems when swallowing, chest pain, and excessive production of saliva among others. Severe constipation can last for months.

Pregnant women and patients with AIDS or weak immune systems are very vulnerable to Chagas' disease.⁴⁴⁵ Newborn infants can get the infection from their mother and develop meningoencephalitis, retina changes and cardiac insufficiency.⁴⁴⁶

Methods of Detection

There are several methods to detect Chagas' disease. The most important are: microscopic techniques, parasite isolation, serology, molecular biology techniques and clinical examination.

- A. Clinical diagnosis at a late stage is usually by electrocardiograms. These examinations look for abnormal arrhythmias, tachycardia, heart block and ventricular aneurysms.⁴⁴⁷
- B. Microscopic techniques are able to detect *T. cruzi* using staining protocols such as Giemsa or Wright tests. The protozoan can be detected in several human fluids and tissues (blood, and cerebrospinal fluid).⁴⁴⁸
- C. Parasite isolation can be lengthy, and expensive. Standard experiments require the inoculation of healthy mammals with blood from infected patients. Other protocols include cell culture, and xenodiagnosis.⁴⁴⁹

D. Serology is one of the most used and precise techniques that are available for the diagnosis and detection of Chagas' disease. Indirect Immunofluorescent Assay (IFA), Enzyme-linked immunosorbent assay (ELISA), hemagglutination, radioimmunoprecipitation are commonly used with high sensitivity and specificity rates.⁴⁵⁰ However, false positives may happen with other parasites from the Leishmania family.⁴⁵¹

E. Molecular biology techniques are quite useful due to the detection of the parasite using polymerase chain reaction (PCR) and immunoblotting protocols.⁴⁵²

Prevention

Preventive techniques are usually suggested in countries when the disease is considered endemic. Some people use permethrin, an insecticide, in their beds to repel the transmission vectors.⁴⁵³

Treatment

Currently, there is no effective treatment to either prevent or cure Chagas' disease. Thus, mortality rates are between 5% and 8% in some regions. Unfortunately, more than twenty percent of infected patients develop the chronic stage. Currently, nifurtimox (Nx, a nitrofurantoin derivative) and benznidazole (Bz, a nitroimidazole derivative) are currently used for the treatment of patients diagnosed with Chagas' disease in its acute stage as effective chemotherapeutic agents.^{454,455} However, treatments with nifurtimox are usually painful, toxic, lengthy and show poor efficacy.⁴⁵⁶ Side effects of these treatments include stomach pain and rash in the case of Nx, while Bz may cause digestive problems in some patients.

A

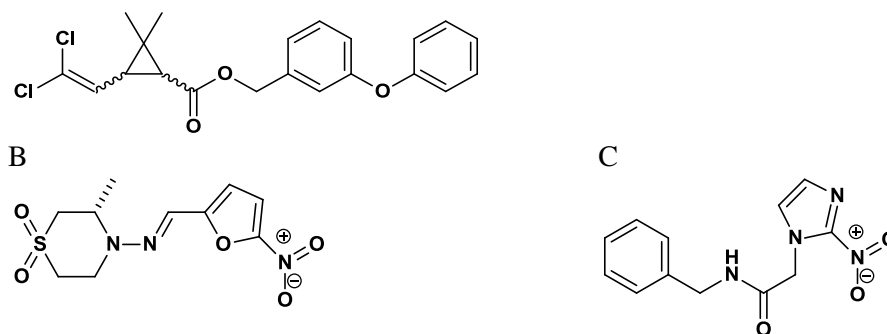


Figure 117. Chemical Structures of A. Permethrin. B. Nifurtimox. C Benznidazole

Mechanisms of Action of Nifurtimox and Benznidazole

Maya and coworkers showed that Nx and Bz act against the formation of electrophilic metabolites within the parasite. They have done an extensive review about the mechanism of action of these anti-parasitic agents. The presence of a nitro group allows nifurtimox and benznidazole to be potent agents. NADPH dependent oxidoreductases, form very reactive radicals and molecules(R-NO). Molecular oxygen facilitates the recovery of the compound via reduction of the former intermediates. Superoxide dismutase utilizes these radicals, in order to form molecular oxygen and hydrogen peroxide. Hydroxyl free radicals bind to critical cell components such as lipids and DNA producing irreversible changes to these cellular components.⁴⁵⁷

Introduction to Cruzain and its Importance in Chagas' disease

Nomenclature, Classification and Historical Background

Cruzain, the recombinant form of cruzipain, is a parasitic hydrolase found in *Trypanosoma cruzi* with an EC number of 3.4.22.51.⁴⁵⁸ Both names are used in literature. Cruzain is also considered a cathepsin-L like enzyme due to their structural similarities and is also considered as an endopeptidase.

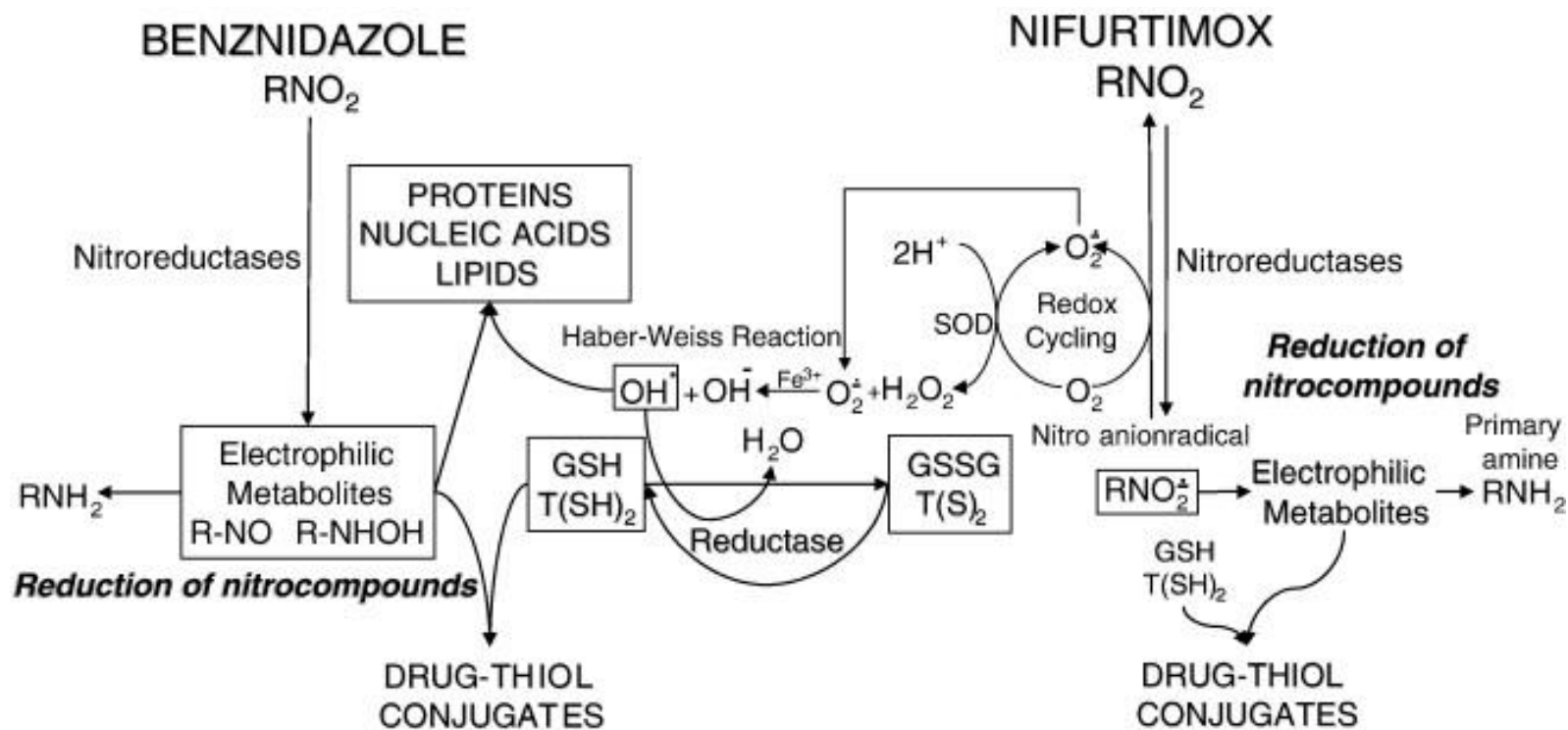


Figure 118. Role of Glutathione and Trypanothione in the Action and Metabolism of the Antichagasic Drugs Nifurtimox and Benznidazole (Reproduced from Maya, page 603⁴⁵⁷)

The official name of the protein is cruzipain. Other names for this protease are cruzain, and cruzaine among others. First reports of cruzain activity were by Itow and coworkers in 1977.⁴⁵⁹ Furthermore, Rangel reported a purified form of cruzain few years later. Cruzain is considered the most important and powerful protease found in *Trypanosoma cruzi*.⁴⁶⁰

Localization of Cruzain

Cruzain is mainly found in parasitic lysosomes, similar to cathepsin L, which is expressed in mammalian organisms. An in-depth investigation confirmed cruzain trafficking is extremely complex. Cruzain is transported to the epimastigote lysosome via endoplasmic reticulum and Golgi apparatus.^{461–463} Immunoblotting data indicate cruzain is also expressed in every stage of the protozoan life cycle. Specifically, cruzain was found at the surfaces of epimastigotes and amastigotes, as well in transitional forms of the trypomastigote-amastigote pseudo stage.^{464–466}

Biological Roles of Cruzain

Cruzain is considered the most important protease due to its high catalytic activity found in *T. cruzi*. Reports showed cruzipain is mainly involved in cell invasion, immune evasion and disease pathogenesis.

Cruzain plays a key role in the parasite's ability to invade host cells. Studies showed cruzain mediates cell invasion by two independent pathways. In the first one, cruzain stimulates the activity of B2 kinin receptor, while the second one is completely kinin independent but helps in cell invasion processes.⁴⁶⁷

Cruzain is also involved in immunogenic processes. Investigations reported that mice immunized with cruzain showed strong responses against muscle and/or cardiac

myosin. The studies also demonstrated that mice immunized with cruzain showed a clear enlargement of their spleens.^{468,469}

Parasite life cycles are well regulated by cytokine activated macrophages.⁴⁷⁰ This is one of the key roles in the pathogenesis of Chagas' disease. However, recent theory showed cruzain is also important in the regulation of macrophages. For example, macrophages were activated when they were cultured with purified cruzipain.⁴⁷¹ Evident signs of this activation include L-arginase activity, and expression of IL-20 and TGF β .⁴⁷² Nevertheless, these results have not been proven in parasite infection in *in vivo* studies.

Cruzain's kininogenase activity releases Lys-bradykinin, an inflammatory peptide and prekallikrein which also generates Lys-bradykinin.⁴⁷³ These two proteins are responsible for increasing parasitic uptake in host cells.⁴²⁶

Other roles of cruzain include: cell remodeling in the epimastigote, activation of bradykinin receptors, and mediation of apoptotic processes.⁴⁷⁴⁻⁴⁷⁶ Researchers have also worked with genetically-modified knockout models of *Trypanosoma cruzi*. Unsuccessful living models indicate that cruzain plays important roles in the life cycle of the parasite.⁴⁷⁷ Engel and coworkers were able to eradicate *in vitro* infections with cysteine proteases, supporting the previous statement.⁴⁶²

Catalytic Activity of Cruzain and Substrate Specificity

Cruzain is a cathepsin L-like cysteine protease. Its amino acid sequences is 45% identical compared to mouse cathepsin L and its substrate specificity is very similar when their catalytic activities are compared.⁴⁷⁸ Several studies have been carried out to understand cruzain's substrate specificity compared to other cysteine proteases members such as cathepsin B, cathepsin L and cathepsin F, which are identical as much as 50%.

Gillmor explored the ability of this parasitic protease to cleave synthetic fluorogenic substrates.⁴⁷⁹ Essentially, they explored five different fluorogenic substrates with the form Z-XR-AMC where X were hydrophilic or hydrophobic residues at the P2 position. Their studies were extended to test cruzain's enzymatic activity over a broad pH range (3 to 8). In summary, cruzain prefers substrates with hydrophobic residues at the P2 position (X= Phe, Ile and Tyr) over hydrophilic ones (Arg, Gln). Nevertheless, the protease is able to cleave hydrophilic residues to a lesser extent. Cruzain's activity was well defined over a broad pH range when hydrophobic residues were tested reaching 50% activity under acidic conditions (pH 5.5). Studies showed the phenolic ring of phenylalanine can be easily accommodated in a hydrophobic pocket formed by Ala133, Leu157, Gly160 and Glu205 and Met68.

Natural Substrates of Cruzain

Cruzain's enzymatic activity has been well studied due to its major roles in cell invasion and immunogenic procedures. The enzyme is able to cleave proteins found in the extracellular matrix such as collagen, and gelatin, among others.⁴⁸⁰ Overall, *Trypanosoma cruzi* is capable of degrading selective members of the extracellular matrix, including collagen, laminin, and heparin sulfate.⁴⁸¹ Interestingly, *in vitro* experiments have shown cruzain can degrade fibronectin, which is one of the few macromolecules that is not affected during parasitic cell invasion. Cruzain also possesses caseinase properties and can also cleave hemoglobin, bovine serum albumin and immunoglobulin.

Structure of Cruzain

Cruzain is expressed in *Trypanosoma cruzi* as a preproenzyme of 467 amino acids residues containing four very well defined portions: a signal peptide (19 residues),

the propeptide (104 residues), the mature portion (215 residues) and a carboxy-terminal domain (129 residues).⁴⁶⁰ The signal and prodomain, which account for 123 amino acids, are cleaved under acidic conditions. The function domain is to protect unnecessary hydrolytic activity of the enzyme inside of the eukaryotic cells of the parasite, similarly to other cysteine proteases found in mammals. Mature cruzain is a monomer containing 215 residues with a molecular weight of 23 kDa; however, glycosylated cruzain molecular weight is 51 kDa. Finally, the 129-residue carboxyl terminal domain is an unusual feature that is a characteristic of cruzain. To date, there are no other known enzymes sharing this portion. Researchers have unsuccessfully tried to crystallize cruzain with its C-terminal. The challenges in the elucidation of the biological functions of the C-terminal have increased due to the lack of validated crystal structures and the existence of other modifications (i.e. isotypes) are also known, yet difficult to understand their implications in *in vivo* studies. Nevertheless, it has been concluded cruzain's catalytic activity is C-terminal independent.⁴⁸²

The amino acid sequence of mature cruzain is shown in Figure 120.¹³¹ The amino acid sequence of both cysteine cathepsins (cathepsin L and cruzain) have been identified and confirmed. Literature reports identities between 45% and 50%. The figure shows the amino acid sequence of mature cruzain containing 215 residues.

The three most abundant residues are alanine, glycine, asparagine and valine (25, 23 and 21 residues respectively). On the other hand, the least abundant residues are histidine (4 residues), arginine and phenylalanine (three residues each). Hydrophobic residues (A, F, I, L, M, P, V and W) account for forty percent of cruzain's composition. Also, 38 percent of the composition is entirely made of polar, uncharged residues.

The secondary structure also reveals that seven alpha helices account for 25% of the structure (55 residues). However, 24% of the sequence forms beta sheets distributed in seventeen strands (52 residues).

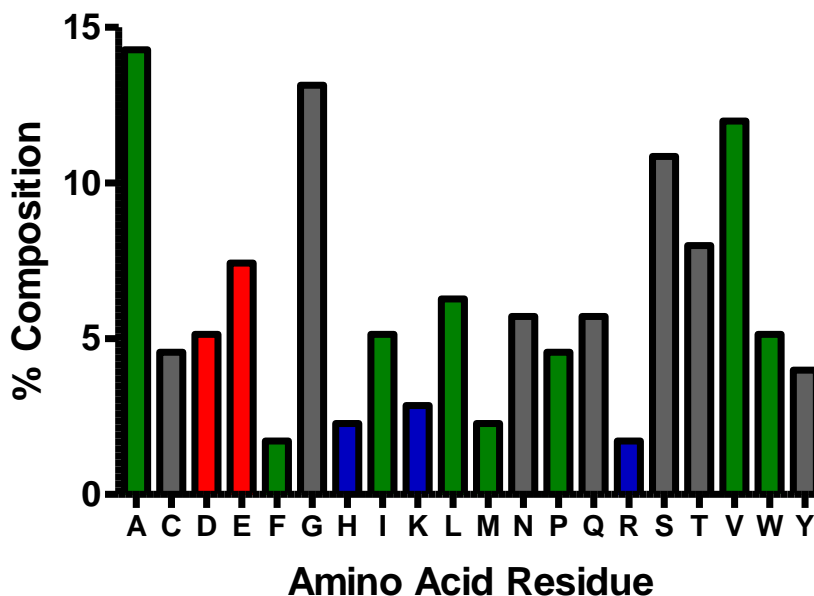


Figure 119. Composition of Mature Cruzain. Legend= blue: basic residues; red: acidic residues; green: hydrophobic (non polar) residues; gray: neutral, polar residues

Table 134. Amino Acid Composition of Mature Cruzain

AA	#	%	AA	#	%	AA	#	%
Ala	25	11.6	Ile	9	4.2	Arg	3	1.4
Cys	8	3.7	Lys	5	2.3	Ser	19	8.8
Asp	9	4.2	Leu	11	5.1	Thr	14	6.5
Glu	13	6.1	Met	4	1.9	Val	21	9.8
Phe	3	1.4	Asn	10	4.7	Trp	9	4.2
Gly	23	10.7	Pro	8	3.7	Tyr	7	3.3
His	4	1.9	Gln	10	4.7	Total	215	100

The complete amino acid sequence of precruzain is shown in Figure 121. The macromolecule is made of 467 residues and divided in four different regions: The prerregion, the propeptide, the mature enzyme and the carboxy-terminal. Eakin and

coworkers offers a complete description of the molecule. Nevertheless, its crystal structure has not been elucidated.⁴⁶⁰

```
1   APAAVDWRARGAVTAVKDQGQCGSCWAFSAIGNVECQWFLAGHPLTNLSE
51  QMLVSCDKTDSGCSGGLMNAFEWIVQENNGAVYTEDSYPYASGEGISPP
101 CTTSGHTVGATITGHVELPQDEAQIAAWLAVNGPVAVAVDASSWMTYTGG
151 VMTSCVSEQLDHGVLLVGYNDSAAVPYWIKNWTTQWGEEGYIRIAKGS
201 NQCLVKEEASSAVVG
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Figure 120. Amino Acid Sequence of Cruzain (PDB: 1ME3¹³¹)

```
1   MSGWARALLLA AVLVMACLVPAATASLHAEETLTSQFAEFKQKHGRVYESAAEEAFRLS
61  VFRENLFARLHAAANPHATFGVTPFSDLTREEFRSRYHNGAAHFAAAQERARVPVKVEV
121 V GAPA AVDWRARGAVTAVKDQGQCGSCWAFSAIGNVECQWFLAGHPLTNLSEQMLVSCDK
241 PQDEAQIAAWLAVNGPVAVAVDASSWMTYTGGVMTSCVSEQLDHGVLLVGYNDSAAVPYW
301 I IKNWTTQWGEEGYIRIAKGSNQCLVKEEASSAVVGGPGPTPEPTTTTTTTSAPGPSY
361 FVQMSCTDAACIVGCENVTLPTGQCLLTTSGVSAIVTCGAETLTEEVFLTSTHCSGPSVR
421 SSVPLNKCNRLLRGSVEFFCGSSSGRLADVDRQRHQPYHSRHRRL
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Figure 121. Amino Acid Sequence of Preprocruzain. Legend: Green: Signal Peptide; Blue: Propeptide; Red: Mature Cruzain; Purple: C-Terminal.⁴⁶⁰

Crystal Structure of Cruzain

The first crystal structure of cruzain was reported McGrath in 1995, when they solved the structure at a 2.35 Å.⁴⁸³ The monomer is folded into two well defined domains: the L-domain which is mainly α -helical and the R-domain with extended antiparallel β -sheet interactions. Similar to cathepsin L, the catalytic triad is composed of Cys25, His159 and Asn175 (papain numbering). The active site is located at the inner face of the two domains. Cruzain's structure and domains are very similar when compared to papain, the major cysteine protease. Nevertheless, cruzain and papain differ in their loops and turns.

The amino acid sequence shows cruzain possesses eight cysteine residues. For example, Cys153 and Cys200 form a disulfide bond and are located in close proximity to

the active site (less than 2.0 Å). Interestingly, four residues (148, 159, 190 and 202) are isostructural in both cysteine proteases (papain and cruzain).

One of the unique features of cruzain is the presence of one cysteine residue at position 36. (Cys36). To date, there are no other cysteine proteases with this residue located at that position. The thiol group was found to have weak interactions with two carbonyl oxygen atoms (Ala12, Gly32) and one water molecule.

A comparison between papain and cruzain molecules inhibited by synthetic inhibitors reveals similarities between both proteases. For example, McGrath proposed that inactivation occurs by alkylation of the active residue Cysteine 25.

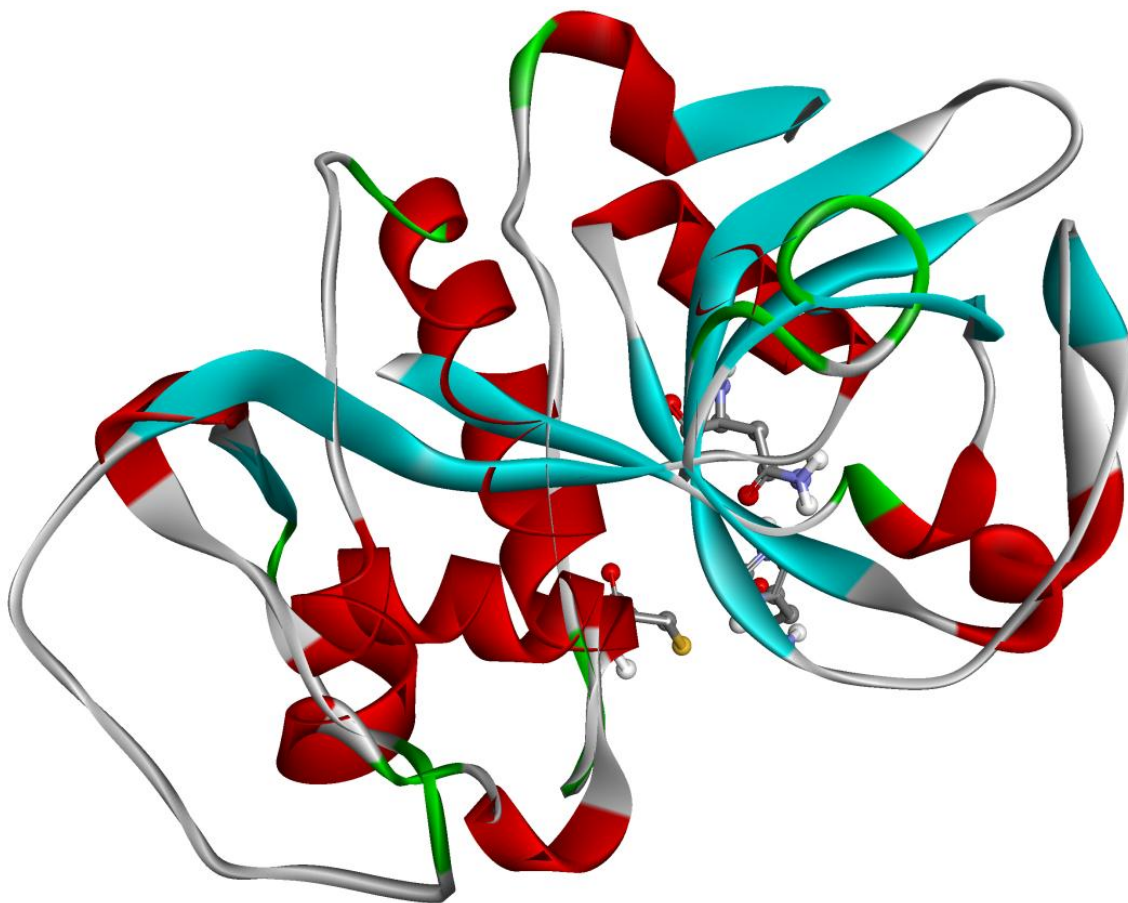


Figure 122. Crystal Structure of Cruzain PDB: 1AIM⁴⁷⁹

Inhibitors of Cruzain

Extensive research related to Chagas' disease is found in recent literature due to the significant number of cases. Thus, a number of synthetic inhibitors have been developed to elucidate possible chemotherapeutic agents against the disease. In this case, a review of cruzain inhibitors will be discussed due to the proteolytic importance of the enzyme in the parasitic disease. Nevertheless, natural inhibitors are also known to be good and potent inhibitors of cruzipain.

Natural Inhibitors of Cruzain

Cruzain is expressed in parasitic organisms as procrucain, similarly to cathepsins L and K, which also belong to the subclan of cysteine proteases. One of the major functions of the propeptide is the prevention of enzymatic activity of cruzain within the parasite. Thus, activation occurs within the parasitic lysosome where cruzain activity is mainly found. Cleavage and activation of the enzyme occur under acidic conditions. Studies showed that this proregion, (103 residues) with a molecular size of 14 kDa, is a potent inhibitor of mature cruzain. Dissociation constant was calculated as low as 18 pM in *in vitro* studies.⁴⁸⁴ Similar studies also confirmed that regions of the propeptides can also inhibit mature cruzain as well.⁴⁸⁵

Other natural inhibitors of cruzain can be found within the protozoan. Cystatins, are small-size macromolecules with regulatory functions. For example chagasin is a potent and reversible inhibitor of cruzain.⁴⁸⁶ These proteins are excellent cruzain inhibitors with dissociation constants in the picomolar range.

The p41 fragment of the MHC class-II associated molecule is a tight-binding, competitive, reversible inhibitor of cruzain. The affinity for cruzain is clearly seen in its dissociation constant ranging in the picomolar values as well (K_i : 58 pM).⁴⁸⁷

Synthetic Inhibitors of Cruzain

A selected group of synthetic inhibitors that have been designed for the inhibition of cruzain can be found in Figure 124.

The discovery of nonpeptidic thiosemicarbazones as cysteine protease inhibitors was reported by Du and coworkers in 2002. They found 3'-bromopropiophenone thiosemicarbazone was able to inhibit cruzain.⁸ Similar findings have been also reported by Chiyanzu et al. and Siles and coworkers in independent studies.^{9,488} The first collaborative work between Dr. Mary Lynn Trawick and Dr. Kevin G. Pinney reported the synthesis of novel functionalized thiosemicarbazones. The list included naphthalenes, benzophenone, propiophenones, among others. They were the first to report the potency of 3,3'-dibromobenzophenone (**1**) as a cysteine protease inhibitor. They tested it against recombinant cruzain obtaining an IC_{50} value of 24 nM.^{3-5,9}

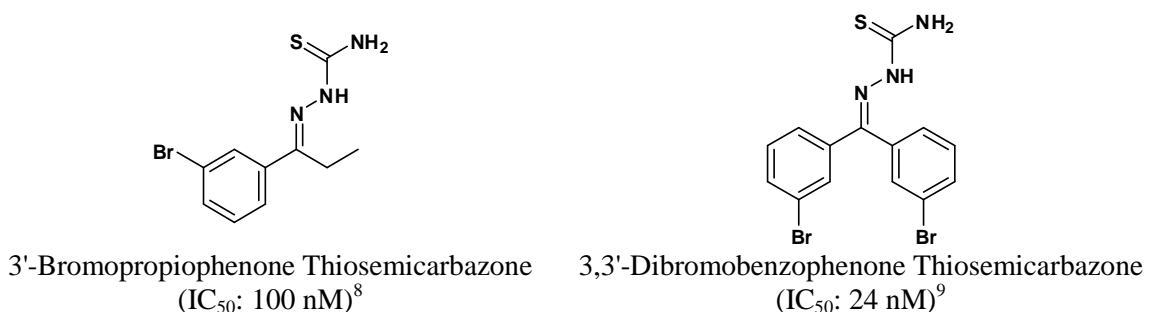


Figure 123. Nonpeptidic Thiosemicarbazone Inhibitors of Cruzain

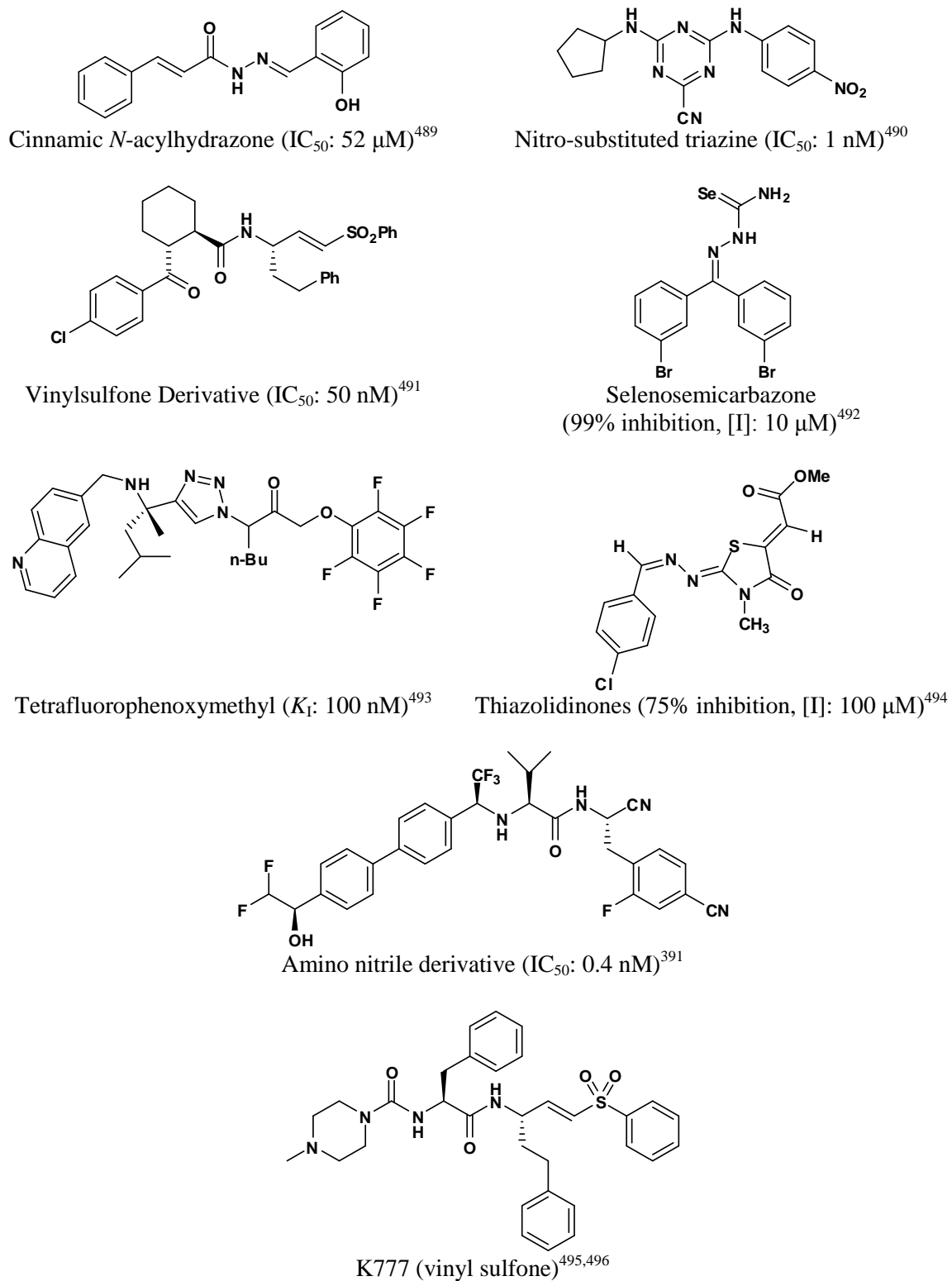


Figure 124. Selected Cruzain Inhibitors Reported Between 2009 and 2012

Material and Methods for the Biological Evaluation of Thiosemicarbazones Derivatives as Inhibitors of Cruzain

Materials

The comprehensive list of materials and equipment is listed in chapter two under the Materials and equipment section. Recombinant cruzain (1.1 μM) was expressed and purified by Dr. Wara M. Arispe and Lauren Adamson under the guidance and supervision of Dr. MaryLynn Trawick.⁵ Cruzain DNA plasmid was kindly donated by Elizabeth Hansell and Dr. James McKerrow from the University of California at San Francisco, CA. Thiosemicarbazone derivatives were synthesized by members of Dr. Kevin G. Pinney's laboratory and have been previously reported.^{3,6,7,9-12}

Experimental Section

Kinetic Cruzain Assay

Kinetic analysis of cruzain was carried out by using a Thermo Fluoroskan microplate reader, 3691 96-well black microplates. Volume of the reaction was 200 μl . Every well contained 100 μl of assay buffer, 10 μl of 35% DMSO solution, 70 μl cruzain stock solution, and 20 μl of Z-FR-AMC stock solution. A mixture containing assay buffer, 35% DMSO, and was preincubated at 25 $^{\circ}\text{C}$ during 5 minutes using 96-black microplates. Triplicate sets of the catalytic activity of cruzain were monitored by adding a concentration of Z-FR-AMC to every reaction. The production of AMC was monitored for 5 minutes at 25 $^{\circ}\text{C}$ using excitation and emission references of 355 and 460 nm respectively. Readings were taken every 10 seconds for five minutes. Reactions were carried out in triplicate. The final concentrations of the kinetic assay are: 100 mM

NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 0.1 nM cruzain. Final concentrations of Z-FR-AMC varied between 0.2 and 15 μ M.

Preliminary Inhibition Studies

Thiosemicarbazone analogs (provided by Dr. Kevin G. Pinney's laboratory) were prescreened to determine if they have inhibitory activity against cruzain.^{3,6,7,9-12}

Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 10 μ l of 35% DMSO or 10 μ l of dilution "1" (final concentration: 10 μ M, See Material and Methods of Chapter Two for complete details), 70 μ l cruzain stock solutions, and 20 μ l of Z-FR-AMC stock solution. The enzyme-inhibitor mixtures (180 μ l assay buffer, 35% DMSO or inhibitor, and cruzain) was preincubated at 25 °C during 5 minutes using 96-black microplates. Reactions were started by adding 20 μ l of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five min. Final reactions were started by adding 20 μ l of Z-FR-AMC. The final concentrations of the preliminary inhibitory reactions were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 0.1 nM cruzain, 10 μ M of the screened inhibitor and 15 μ M of Z-FR-AMC. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Compounds that did not have cruzain inhibitory activity more than 50% (i.e. $v_i/v_o \leq 0.5$) were considered as 'inactive' compounds and a general IC_{50} value greater than 10000 nM was assumed. Compounds that inhibited cruzain inhibitory activity more than 50% were further considered for cruzain inhibitory studies and an exact IC_{50} was determined.

Cruzain Inhibition Assay

Inhibitory analysis of cruzain and its inhibitors was carried out by using a modified protocol of the kinetic cruzain assay, previously described. Total volume of the reaction was 200 μ l. Every well contained 100 μ l of assay buffer, 10 μ l of 35% DMSO or 10 μ l of inhibitor dilutions, 70 μ l cruzain stock solutions, and 20 μ l of Z-FR-AMC stock solution. An 180 μ l mixture containing assay buffer, 35% DMSO or inhibitor, and cruzain was preincubated at 25 °C during 5 minutes using 96-black microplates. Reactions were started by adding 20 μ l of Z-FR-AMC. The release of AMC by inhibited and uninhibited samples were monitored for five minutes. The final concentrations of the inhibitory cruzain assay are: 100 mM NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 0.1 nM cruzain and 15 μ M of Z-FR-AMC. Final concentrations of the inhibitors varied between 10 μ M and 10 pM. Readings were taken every 25 seconds for five minutes and reactions were carried out in triplicate. Table 135 summarizes final volumes and final conditions for the kinetic and cruzain inhibition assays.

Construction of AMC Calibration Curve

The calibration curve obtained for cathepsin L experiments (Chapter 2) was used for similar cruzain experiments. Complete details and results of these experiments can be found in chapter 2 under the subsection “Construction of AMC calibration curve).

Effect of Inhibitor Concentration on Cruzain Progress Curves

Final concentrations, conditions, and volumes are similar to the cruzain inhibition assay. Assay buffer, inhibitors (final concentrations varied between 100 nM and 10 μ M) and Z-FR-AMC (final concentration: 5 μ M) were added to the 96-well black plates

(volume of the substrate-inhibitor mixture: 180 μ l). Then, 20 μ l of cruzain stock solution were added immediately without preincubation time. Readings were taken every 3 seconds for fifty minutes.

Table 135. Preparation Table for Kinetic, Cruzain Inhibition Assays and Construction of AMC Calibration Curves

Reagent	Kinetic assay (μ l)	Inhibitory assay (μ l)	AMC Curve (μ l)
Assay Buffer	100	100	100
Water	0	0	60
Control	10	0	0
Inhibitor	0	10	0
CZ	70	70	20
Z-FR-AMC	20	20	20
Total (μ l)	200	200	200

Effect on Preincubation Studies on Cruzain Inhibition Assays

Preincubation studies with several compounds were carried out by modifying cruzain inhibition assay. Different sets of mixtures containing assay buffer, inhibitor and cruzain were preincubated at various periods of times between 0 and 240 minutes.

Reactions were taken every 25 seconds for five minutes and carried out in triplicate.

Determination of K_i^{app} Using Morrison's Quadratic Equation

Data that were obtained on the effect of preincubation studies for numerous compounds were further analyzed. The possibility of these compounds to be tight binding inhibitors was examined by fitting the data by a nonlinear regression using Morrison's quadratic equation.

Cruzain Reversibility Studies

Three milliliters of assay buffer for reversibility studies were prepared. Twenty five microliters of cruzain (100X: 100 nM) assay buffer for reversibility studies were pre-incubated with an equal amount of a concentrated solution of the inhibitor (100X: 0.1 IC_{50}) at 25 °C between one and four hours. Then, two microliters of the enzyme-inhibitor mixture were rapidly mixed with 198 μ l of ASR in order to start the reaction. Total reaction volume was 200 μ l. Readings were taken every twenty five seconds for four hours. Final concentrations are similar as described previously. Final conditions were: 100 mM NaOAc, pH 5.5, 1 mM EDTA, 2.5 mM EDTA, 0.01% Brij 35, 0.1 nM cruzain and 15 μ M of Z-FR-AMC. Table 136 describes required volumes for this experiment.

Table 136. Preparation Table for Cruzain Reversibility Assay

Reagent	Reversibility studies (μ l)
Assay Buffer	198
Inhibitor/CZ	2
Total	200

Effect of Substrate Concentration (Z-FR-AMC) on IC_{50} Values

The effect of [Z-FR-AMC] was studied with compounds **1** and **17**. Minor modifications of the cruzain inhibition assay were carried out. Different sets of mixtures containing assay, inhibitor and cruzain were preincubated at standard preincubation times (5 minutes). Reactions were initiated by the addition of different concentrations of Z-FR-AMC. Reading were taken every 25 seconds for five minutes and carried out in triplicate.

Effect of Substrate Concentration (Z-FR-AMC) on Cruzain Progress Curves

Final concentrations, conditions, and volumes are similar to those previously described for the cruzain inhibition assay. Assay buffer, one inhibitor concentration (Final concentration for **10**: 100 nM) and Z-FR-AMC (final concentrations varied between 0.5 and 15 μ M) were added to the 96-well black plates (volume of the substrate-inhibitor mixture: 180 μ l). Then, 20 μ l of cruzain stock solution was added immediately without preincubating. Readings were taken every 3 seconds for fifty minutes.

Inhibition of Cruzain Collagenase Activity by Thiosemicarbazone Derivatives

A solution of human type I collagen ([ACI]: 0.4 mg/ml) and cruzain stock solution (CLI) were prepared. The inhibitor stock solution was prepared by mixing 1.04 μ l of 20 mM stock solution of the inhibitor in DMSO, 19.8 μ l of DMSO and 179.2 μ l of water. Conditions of this stock solution are: 10.4% DMSO and 104 μ M of the inhibitor. The experiment was started by preincubating 3 μ l of CZI and 2.5 μ l of the inhibitor stock solution in a microcentrifuge tube at 37 °C for 0.5 hours. Similarly, another group of control samples (without the inhibitor) was also set up. A 1X staining solution was prepared by diluting 10 μ l of 5000X SYPRO® red protein gel staining dye with 49.9 ml of 7.5% acetic acid. Then, reactions were initiated by adding 7.5 μ l of type I collagen and monitored between 0 and 420 minutes at 37 °C. Final conditions of the reactions were: 100 mM NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 20 nM cruzain, 2% DMSO, 20 μ M of the tested inhibitor, and 0.01 mg/ml type I collagen. Reactions were stopped by adding 2 μ l of LDS NuPAGE® sample buffer, heated at 90 degrees for ten minutes and immediately stored at -80°C. Inactivated samples (15 μ l) were loaded onto 4-12 % NuPAGE® Bis-Tris gels. Electrophoresis was carried out by using MOPS running buffer

at 200 V for 50 minutes. Gels were stained in 50 ml of 1X SYPRO® staining solution at room temperature between one and two hours followed by a destaining process using water (1X) and 7.5% acetic acid (1X). Finally, gel imaging was performed with a GE Typhoon 9400 FL with excitation and emission wavelengths of 550 and 630 nm, respectively. Table 137 describes conditions for inhibition of collagenase activity studies.

Table 137. Preparation Table for Cruzain Collagenase Activity Assay

Reagent	Control (µl)	Inhibited (µl)
CZI	3.0	3.0
Control (Water)	2.5	0
Inhibitor	0	2.5
TIC	7.5	7.5
Preincubation time (h)		0.5
Total	13	13

Molecular Modeling Studies

Computational studies were performed using Discovery Studio 3.0. First, a crystal structure was obtained from the PDB database. The chosen molecule was 1ME3.¹³¹

Similar experiments were performed with this molecule by Chen.⁴ His results were used as a comparison point to verify the accuracy of the results.

Preparation of the Protein

The selected crystal structure is related to the analysis of recombinant cruzain bound to a synthetic ketone inhibitor. Therefore, the protein needed to be prepared and validated using a high affinity substrate as reported.¹³¹ Water molecules and other ligands (synthetic ketones) were removed. The structure was then examined in order to

correct possible structural disorders, protein residue connectivity, bond-orders, and missing side-chain or backbone atoms and to correct the predicted pKa values (This option was not performed due to limited options in licensing). The pH of the system was also changed in order to modify the protonation state of cruzain's termini and side chain residues. The pH value was set to 6.8. Finally, the binding site was defined according to default specifications from PDB records. This was done in order to create a spherical binding site object. The binding site appears as a red transparent red sphere.

Preparation of the Ligand (Thiosemicarbazones)

Seven molecules (**1**, **8**, **9**, **10**, **17**, **36** and **58**) were selected to be modeled with cruzain in order to verify their mode of inhibition and major interactions. Analog **1** was used as a comparison because it was previously modeled with cruzain.⁴ Each molecule was drawn in ChemDraw 6.0 and copied into Discovery Studio Client 3.0. Each molecule was prepared as 'ligands'. For each ligand, changes in ionization states, and canonical tautomers were allowed to be carried out. At the end of the run, a list of tautomers was generated for each compound.

Docking Simulations

CDOCKER, a dock ligand protocol was used to docking the selected ligands into the macromolecule. The algorithm allows the simulation of several ligands with a single receptor protein simultaneously. The input receptor was the cruzain molecule (1ME3) and the ligands were the entire list of tautomers for each compound. The total number of top hits was set up to 200. The site sphere coordinates were set: 5.1024, 9.25127, 6.43005, 8.4. These coordinates were set according to the PDB files default conditions

and they established the permitted area where both ligand and receptor (thiosemicarbazones and cruzain respectively) could interact. The number of 'orientations to refine' was set to 20. Similarly, the 'maximum bad orientation' and 'orientation vdW Energy Threshold' numbers were 800 and 800 respectively. The selected forcefield was CHARMM and the ligand partial charge method was changed to MMFF. CHARMM was used due to its extensive use when modeling organic molecules with proven accuracy and works quite well with a variety of solvents used in *in vitro* studies in several receptor-ligand studies where ligands are synthetic molecules. Similarly, MMFF is a type of forcefield derived from experimental calculations. MMFF is extensively used in pharmaceutical industries where for studies involving changes in conformation energies and nonbonded interactions. MMFF also gives accurate results for a wide range of organic molecules but fails when parallel processing is required. Docking simulations took several minutes to be performed. For each compound, the number of conformations varied from a couple of dozen to hundreds to possible conformations. These conformations were arranged based on the interaction energies to localize the conformations with highest interaction energies for every compound. A visible conformation was selected in order to observe interactions between cruzain and its inhibitors. Finally, a selection filtering residues having interactions with every ligand was set. For each modeling, only residues that were located less than 5.0 Å were shown. Distances for selected atoms were selected and calculated. Also, hydrogen bonds in that region were chosen and selected. Discovery Studio 3.0 Client set hydrogen bonds less than 2.5 Å, but this distance was also modified depending on each case.

Results and Discussion

Previous research has indicated that some thiosemicarbazones are lead compounds as cruzain inhibitors.⁹ Thus, new generations of thiosemicarbazones required a complete screening to verify their potency. The Structure-Relationship Activity (SAR) was characterized after completing *in vitro* testing against cruzain (IC₅₀ values). The project was completed in collaboration between Dr Kevin G Pinney and Dr. Mary Lynn Trawick groups. Synthetic nonpeptidic thiosemicarbazones were provided by members of the laboratory of Dr. Kevin G. Pinney at Baylor University.^{3,6,7,9-12}

Fluorometric based assays were utilized to study various assay parameters in inhibitory activities, determination of K_i , and reversibility of thiosemicarbazone inhibitors. To understand the interactions between these inhibitors and cruzain, molecular modeling of thiosemicarbazones was used. These studies will also aid in the understanding of developing inhibitory moieties that could be explored in combination with existing inhibitors. Cruzain studies were carried out using a 96-well microplate fluorometric based assay. Z-FR-AMC is a fluorogenic substrate that has been used with serine and cysteine proteases. However, Salvati and Ascenzi observed substrate inhibition when [Z-FR-AMC] > 15 μ M.^{497,498} Preliminary experiments also confirmed their findings and substrate concentration was set to 15 μ M. Final conditions for the cruzain fluorometric assay are: 100 mM NaOAc pH 5.5, 1 mM EDTA, 2.5 mM DTT, 0.01% Brij 35, 2% DMSO, 0.1 nM cruzain, and 0.5 – 15 μ M Z-FRM-AMC.

Assay Optimization. Effect of DMSO on Cruzain Inhibition Assays

Siles and coworkers reported an IC₅₀ value of 24 nM for analog **1**.⁹ The effect of DMSO during enzymatic catalysis was also evaluated using cruzain as our molecular

target. Experiments were conducted using analog **1**, by increasing the concentration of DMSO in solution. The IC_{50} of compound **1** was revisited and it was found to be between 10 and 12 nM (See Table 138).

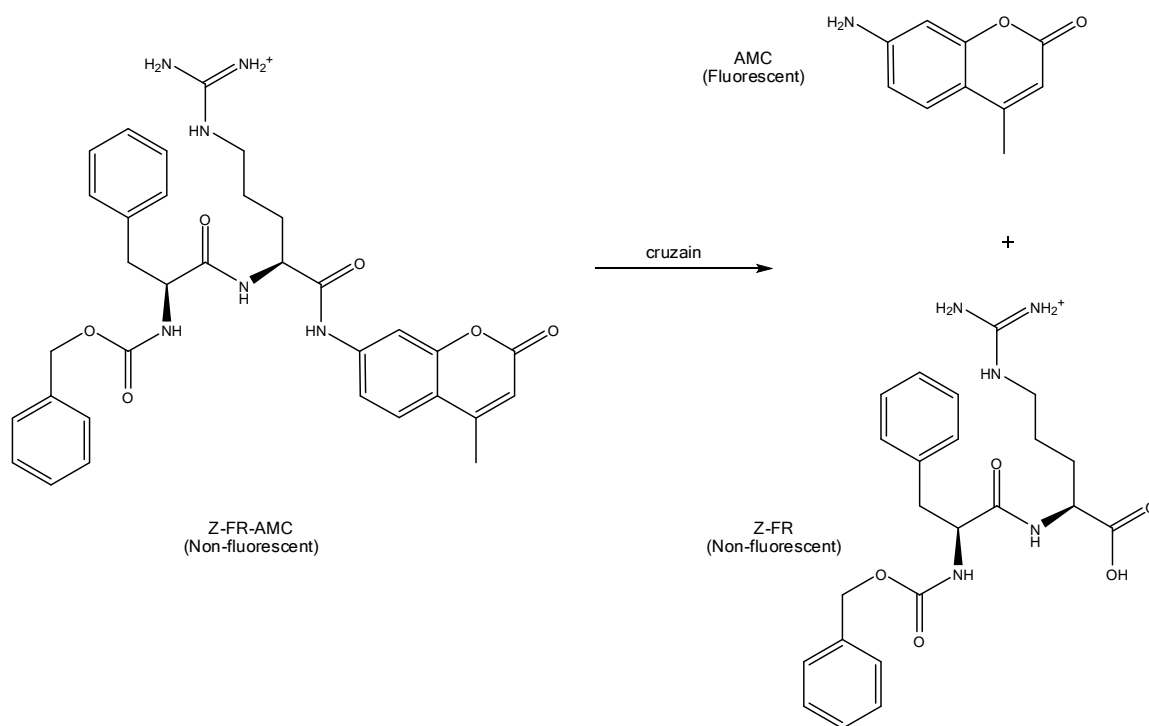


Figure 125. Hydrolysis of the Z-FR-AMC Using cruzain

Table 138. Effect of DMSO Concentration on Cruzain Inhibition Studies

DMSO (%)	1 (IC_{50} , nM)
0.7	24 ⁹
2	12

The obtained results showed the effect of DMSO in the inhibition of cruzain. Solubility of the lead compound (**1**) was increased upon addition of DMSO and their inhibitory activity was enhanced accordingly. An increase in cruzain activity and stability was also observed when DMSO concentration was increased. The efficiency of **1** is 2 times better when comparing its IC_{50} at 0.7 %.

Determination of K_M , V_{MAX} and k_{CAT}

Determination of kinetic constants was based on the analysis assumption that enzymatic activity of cruzain is substrate-concentration dependent. Thus, a high concentration of the substrate (up to 15-fold K_M value) was used for studies related to cruzain. Cruzain catalytic activity showed a linear behavior when 0.5 μ M Z-FR-AMC was used for reaction times that were at least 5 minutes long. The determination of K_M , V_{MAX} and k_{CAT} was possible with experiments that observe the catalytic activity of a fixed concentration of cruzain (0.1 nM) but vary the concentration of Z-FR-AMC (0.2 -15 μ M). Experiments were carried out in triplicate. Catalytic rates were calculated by applying linear regression of the data. ([AMC] is the dependent variable and time (seconds) is the independent variable). Then, a nonlinear regression analysis of the Michaelis-Menten equation (Eq. 1.1) was performed to calculate K_M and V_{MAX} values with the aid of commercially available software (GraphPad 5.0). The k_{CAT} constant value was determined using equation 4.1. The parameter v_o is the initial rate velocity at a specific substrate concentration. The V_{MAX} is the maximum velocity, K_M is the Michaelis-Menten constant, [S] is the substrate concentration, and k_{CAT} is the catalytic rate constant of the reaction.

$$v_o = \frac{V_{MAX}[S]}{K_M + [S]} \quad (1.1)$$

$$k_{CAT} = \frac{V_{MAX}}{[cruzain]} \quad (4.1)$$

K_M , v_{MAX} and k_{CAT} values were found to be $1.01 \pm 0.1 \mu\text{M}$, $1.73 \pm 0.08 \text{ AMC}$ nM/seconds and 17.3 s^{-1} , respectively. These values are similarly to those reported in the literature (K_M : $0.96 \mu\text{M}$).⁴⁹⁹

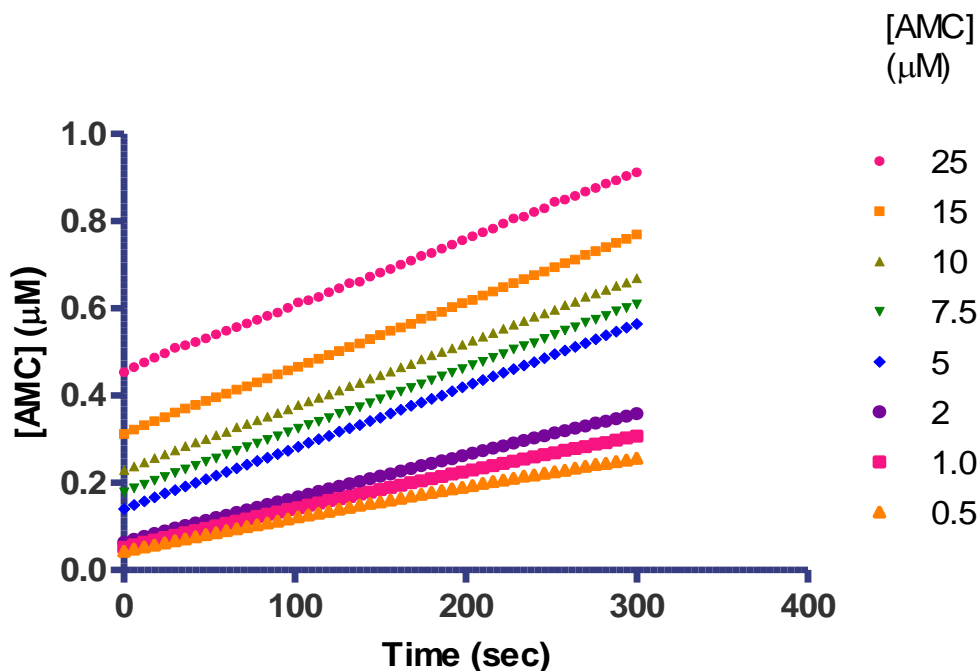


Figure 126. Catalytic Activity of Cruzain Using Z-FR-AMC as the Fluorogenic Substrate

Determination of Inhibitory Efficacy of Thiosemicarbazone Analogs at 10 μM

The first set of experiments that were performed to verify if the individual compound of the library of thiosemicarbazones could be potential cruzain inhibitors. If the catalytic activity of cruzain was inhibited by 50% or more by a fixed concentration (Final concentration: $10 \mu\text{M}$) of the potential inhibitor. Three independent sets of experiments of untreated ($[I]: 0 \mu\text{M}$) and treated samples ($[I]: 10 \mu\text{M}$) were preincubated with 0.1 nM cruzain for 5 minutes at $25 \text{ }^\circ\text{C}$. Inhibitory activities were monitored when

reactions were started by adding 15 μM Z-FR-AMC as a fluorogenic substrate. Reactions demonstrated a linear behavior for at least the first five minutes. Active compounds were further analyzed to determine an exact IC_{50} value. If the ratio $v_i/v_o \leq 0.5$, then the compounds were not considered potential inhibitors and an approximated IC_{50} value ≤ 10000 nM was assigned to them.

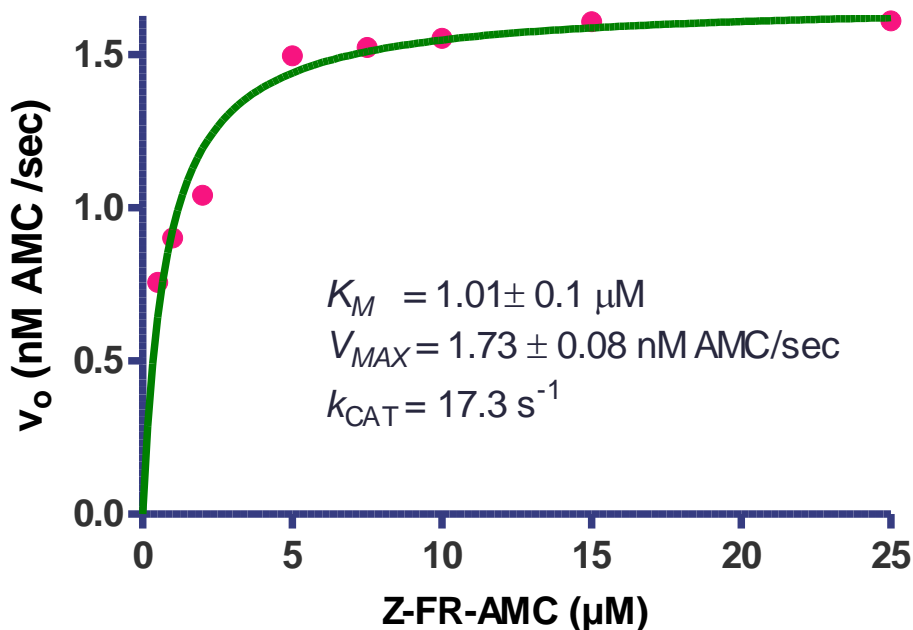


Figure 127. Determination of K_M and V_{MAX} for Cruzain Using Z-FR-AMC as a Fluorogenic Substrate

Determination of IC_{50} values

A section of the library comprising more than sixty compounds was analyzed to verify their efficacy to inhibit the catalytic activity of cruzain. The synthetic compounds were synthesized through a collaborative project under the guidance of Dr. Kevin G. Pinney by several members of his research group.⁹⁻¹² A 96-well microplate fluorometric based assay was utilized to determine the inhibitory activity of each of these inhibitors. Uninhibited cruzain catalytic activity showed linear behavior when 15 μM Z-FR-AMC

was used for reactions times that were 5 minutes long. The determination of the IC₅₀ values was carried out with experiments that observe the inhibitory capacity of the synthetic compounds when a fixed concentration of cruzain (0.1 nM) was preincubated for 5 minutes at 25 °C. The final concentration of each compound varied between 10 pM and 10 μM. Three experiments were carried out for each tested inhibitor. Catalytic rates of uninhibited and inhibited samples were calculated by linear regression of the data. ([AMC]: dependent variable and time (seconds): the independent variable). The data followed a typical sigmoidal dose response, and therefore, a nonlinear regression of the equation 1.2 was performed to calculate IC₅₀ values with the aid of commercially available software (GraphPad 5.0). The value Y represents the inhibited activity (normalized relative to control, X is log([inhibitor]) in M. The velocities v_{\min} and v_{\max} represent when cruzain was preincubated with the highest and lowest inhibitor concentrations respectively (10 μM and 0 pM or control). The Hill slope value is the slope of the sigmoidal curve. IC₅₀ ± S.E. values represent the average and standard errors of at least three independent experiments. A complete structure-activity relationship (SAR) is shown in Table 139 to Table 147 that group these inhibitors by functional groups or families.

$$Y = \frac{v_{MIN} + (v_{MAX} - v_{MIN})}{1 + 10^{(\log(IC_{50}-X)*Hillslope)}} \quad (1.2)$$

Table 139. Inhibition of Cruzain by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12}

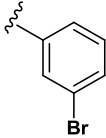
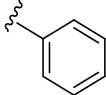

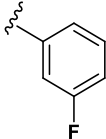
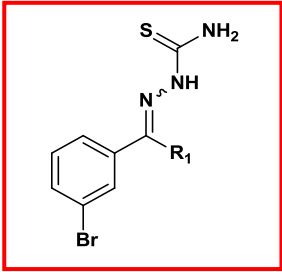
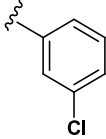
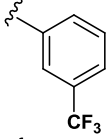
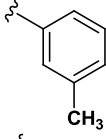
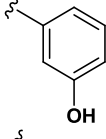
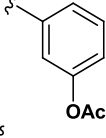
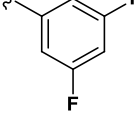
Compound	Structure	R ₁	IC ₅₀ ± S.E. (nM)
1			10.61 ± 0.4
2			73.0 ± 4.7
3 ⁸			221.7 ± 8.98
4			110.8 ± 7.4
5			30.2 ± 2.8
6			64.4 ± 2.9
7			94.4 ± 6.5
8			211.9 ± 16.4
9			12.1 ± 0.2
10			17.6 ± 1.5

Table 139. Inhibition of Cruzain by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

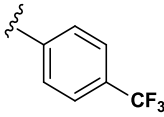
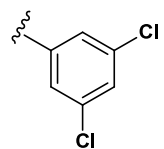
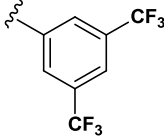
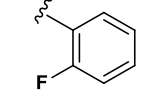
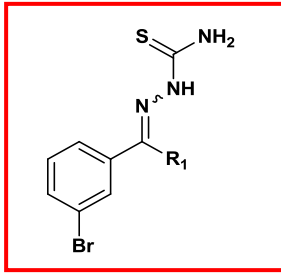
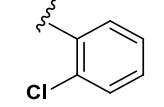
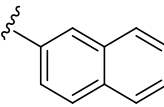
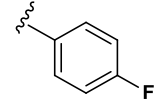
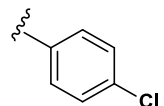
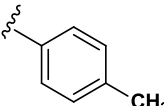
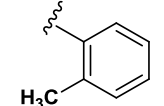
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
11			637.5 ± 47.3
12			97.2 ± 8.9
13			678.2 ± 69.8
14			70.37 ± 6.4
15			182.4 ± 11.7
16			ND
17			14.3 ± 1.5
18			275.3 ± 27.3
19			526.9 ± 33.1
20			770.5 ± 176.8

Table 139. Inhibition of Cruzain by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	Structure	R ₁	IC ₅₀ ± S., (nM)
21			127.0 ± 4.5
22			134.9 ± 5.2
23			83.7 ± 5.6
24			ND
25			266.8 ± 21.7
26			47.7 ± 4.6
27			34.2 ± 2.8
28			113.6 ± 7.2
29			ND
30			64.4 ± 4.1

Table 139. Inhibition of Cruzain by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

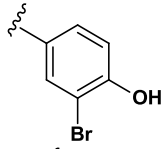
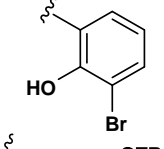
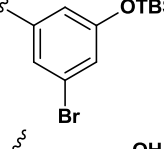
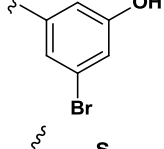
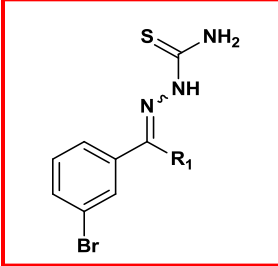
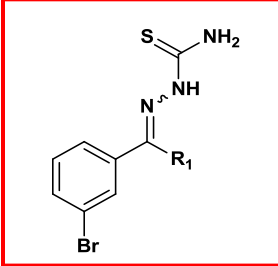
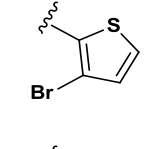
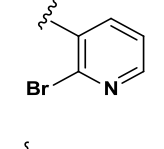
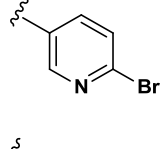
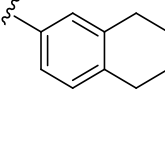
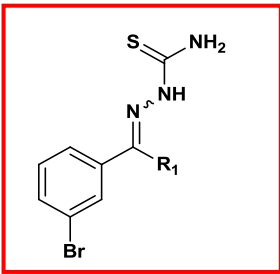
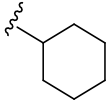
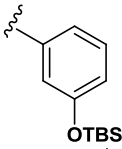
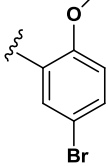
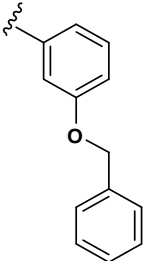
<i>Compound</i>	<i>Structure</i>	<i>R₁</i>	<i>IC₅₀ ± S.E., (nM)</i>
31			ND
32			ND
33			ND
34			97.8 ± 8.3
35			100.1 ± 2.4
36			20.2 ± 0.14
37			31.6 ± 1.6
38			ND
39			≥ 10000
40			ND

Table 139. Inhibition of Cruzain by Thiosemicarbazones Containing a *meta*-Bromophenyl Substituent Group. For Synthesis of Compounds: ^{3,6,7,10-12}(Continued)

Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
41			≥ 10000
42			ND
43			ND
44			46.0 ± 2.1

*ND: Not determined

Table 140. Inhibition of Cruzain by *para*-Bromo Functionalized Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

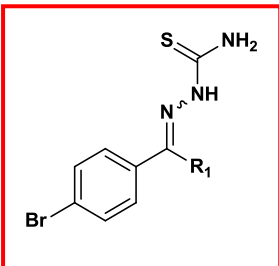
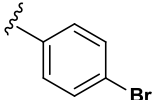
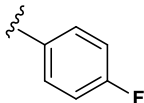
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
47			≥ 10000
48			1453 ± 130.5

Table 141. Inhibition of Cruzain by *para*-Bromo Functionalized Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

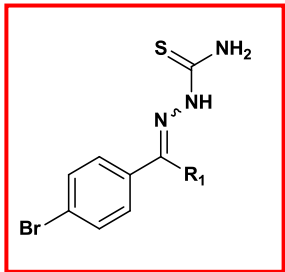
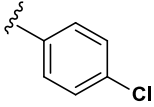
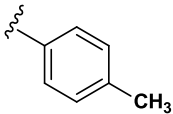
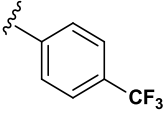
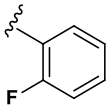
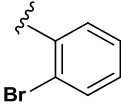
Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
49			≥ 10000
50			1970 ± 151.1
51			2262 ± 247.9
52			466.4 ± 35.8
53			≥ 10000

Table 142. Inhibition of Cruzain by Dihalogen-substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

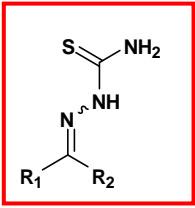
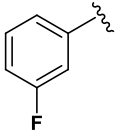
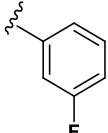
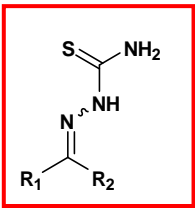
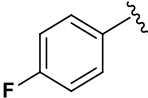
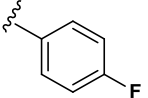
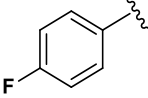
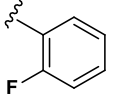
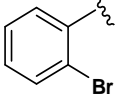
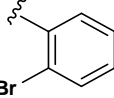
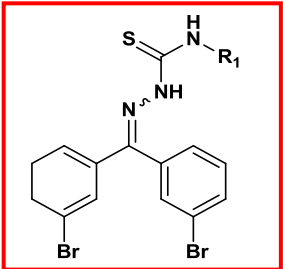
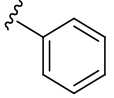
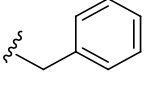

Compound	R ₁	R ₂	IC ₅₀ ± S.E.,(nM)
			
54			484.3 ± 46.5

Table 143. Inhibition of Cruzain by Dihalogen-substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12} (Continued)

Compound	R_1	R_2	$IC_{50} \pm S.E., (nM)$
			
55			≥ 10000
56			494.4 ± 40.9
57			ND

*ND: Not Determined

Table 144. Inhibition of Cruzain by Dibromo-*N*-Substituted Benzophenone Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R_1	$IC_{50} \pm S.E., (nM)$
58			≥ 10000
59			≥ 10000
60			ND

*ND: Not determined

Table 145. Inhibition of Cruzain by Thiosemicarbazones Containing a Phenyl Group.
For Synthesis of Compounds: ^{3,6,7,10-12}

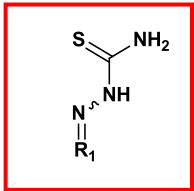
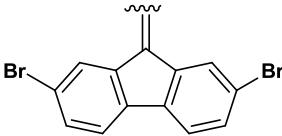
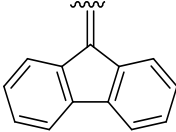
Compound	Structure	R ₁	IC ₅₀ ± S.E., (nM)
61			ND
62			≥ 10000
63			≥ 10000
64			≥ 10000
65			≥ 10000
66			≥ 10000

*ND: Not Determined

Table 146. Inhibition of Cruzain by Substituted Benzophenone Thiosemicarbazones.
For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	R ₁	R ₂	IC ₅₀ ± S.E., (nM)
75			96.0 ± 1.9

Table 147. Inhibition of Cruzain by Functionalized Fluorene Thiosemicarbazones. For Synthesis of Compounds: ^{3,6,7,10-12}

Compound	Structure	R ₁	IC ₅₀ ± S.E.,(nM)
80			≥ 10000
81			≥ 10000

Structure-Activity Relationship (SAR) of Thiosemicarbazones as Cruzain Inhibitors

More than 60 thiosemicarbazone (TSC) synthetic compounds were analyzed during the preliminary screening in order to determinate the potency of these compounds. A considerable number of compounds showed a significant inhibitory potency, and IC₅₀ values were calculated. A selected group of five different compounds were shown to be outstanding cruzain inhibitors with IC₅₀ between 10 and 20 nM. More than half of the screened inhibitors show good inhibitory activities toward cruzain (21 ≤ IC₅₀ ≤ 1000 nM). One fifth of the library (36 analogs) had a moderate activity. Finally, eighteen compounds show low or no cruzain inhibitory activity (IC₅₀ ≥ 1000 nM).

Potent cruzain inhibitors. Five thiosemicarbazones showed very potent inhibitory activities against the cysteine protease. Figure 128 summarizes the structures of the compounds with IC₅₀ values less or equal to 20 nM. Four analogs belong to the subfamily of *metabrominated* benzophenone TSC analogs (**1**, **9**, **17**, **10**). Bromine, acetate and fluorine moieties are the substituents in this subfamily. The fifth analog is a brominated thiophene thiosemicarbazone TSC (**36**).

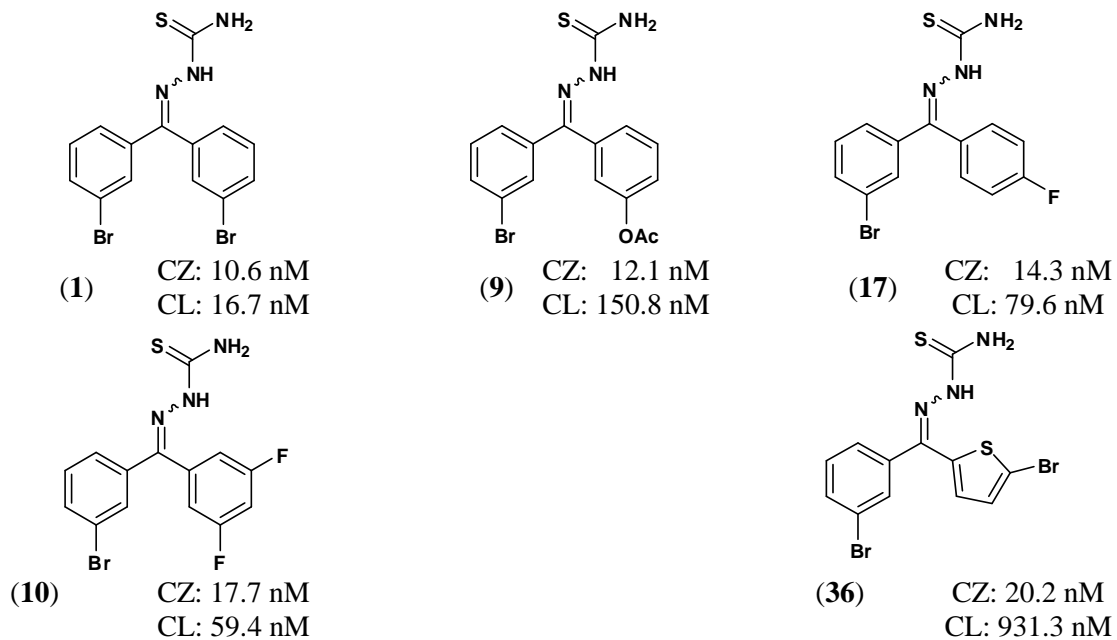


Figure 128. Thiosemicarbazone analogs with Potent Inhibitory Activity against Cruzain⁹⁻¹²

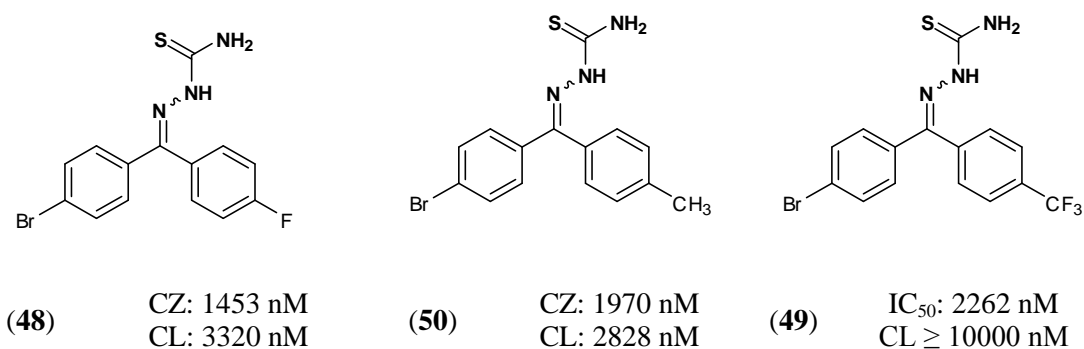


Figure 129. Thiosemicarbazones with Low Activity toward Cruzain

General remarks of the Structure-Activity Relationship

Further analysis of the structure-activity relationship revealed that thiosemicarbazone inhibitory activity can be enhanced or reduced dramatically by moiety substitution. The series of substituted *metabrominated* thiosemicarbazones is the largest subgroup among the compounds that were tested. It consists of forty-four different

analogues with inhibitory activities that varied between 16.7 and ≥ 10000 nM. Halogenated substituents greatly enhanced the activity of the thiosemicarbazones. Compounds **1**, **2**, **4**, **5** (Table 148) are thiosemicarbazones sharing the main moiety (3-bromobenzophenone thiosemicarbazone) but varying the 3'-halogenated substituent. Compound **2** is used as a comparison to complete the series. All of these compounds possessed good inhibitory activity with IC_{50} values less than 150 nM. The brominated analogue is 10-fold more active than its fluorinated analogue. Interestingly, the monobromobenzophenone thiosemicarbazone (**2**) showed good activity against the parasitic enzyme when compared to cathepsin L (See Table 148). Increased solubility and lower enzyme concentration may contribute to its activity. The dependence of IC_{50} as a function of enzyme concentration is also known in literature. Shoited reported that a 10-fold increase in enzyme concentration during *in vitro* assays enhances the potency of tetraiodophenolphthalein as an AmpC inhibitors by a factor of five.⁵⁰⁰

Table 148. Inhibition of Cruzain by 3-Bromo-3'-Halogen Benzophenone Thiosemicarbazones

		R_1					
				<		<	
IC_{50} (nM)		1	5		2		4
CZ		10.2	30.2		73.0		110.8
CL		16.7	131.4		≥ 10000		250.3

Five fluorinated-substituted benzophenone TSC analogs (**4**, **10**, **22**, **23**, and **25**; Table 149) were also compared. Polyfluorination of the benzyl rings proves that cruzain-inhibitory activity of the compounds can be increased with the addition of electronegative substituents. However, a deeper comparison between the two difluoro benzophenone TSCs also reveals that position of the substitutions plays a key role in the activity of these analogs. The *ortho*-difluoro analog is 14 times less active than the *dimeta* counterpart. The effect of other substituents was also evaluated (Table 150). Various groups were substituted at the *meta* position of one of the phenyl rings. The group includes trifluoromethyl, methyl, hydroxyl, and acetate moieties. Two moieties showed good activities compared to the acetate analog, which is a potent inhibitor of cruzain with an IC₅₀ value less than 20 nM (Compounds **6** and **7**).

Table 149. Inhibition of Cruzain by 3-Bromo-poly'-fluoro-Benzophenone Thiosemicarbazones

R_1		>		>	
				>	
IC ₅₀ (nM)	10		23		4
					22
					25
CZ	17.6		22.6		104.2
CL	59.4		63.2		250.3
					134.9
					118.1
					266.8
					609.7

A small trifluoromethyl series of three compounds (**6**, **11**, and **13**; Table 151) differs in their inhibition activity when cruzain and cathepsin L IC₅₀ values are compared.

Only **6**, the *meta*-substituted analog, showed good inhibition values towards cruzain. Its inhibitory activity is 10-fold higher when compared with **13**.

Table 150. Inhibition of Cruzain by 3-Bromo-3'-Heteroatomic Groups Benzophenone Thiosemicarbazones

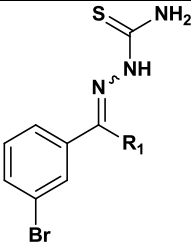
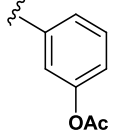
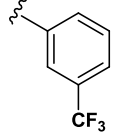
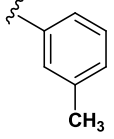
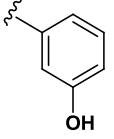
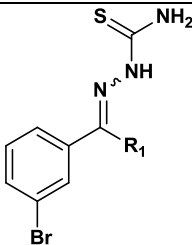
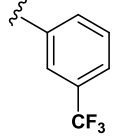
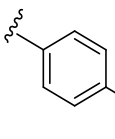
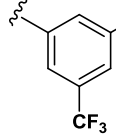
R_1							
		<		<		<	
IC ₅₀ (nM)	9		6		7		8
CZ	12.1		64.4		94.4		211.9
CL	150.8		46.5		224.4		131.4

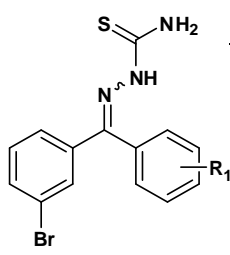
Table 151. Inhibition of Cruzain by 3-Bromo-Trifluoromethyl Benzophenone Thiosemicarbazones

R_1					
		<		<	
IC ₅₀ (nM)	6		11		13
CZ	64.4		637.5		678.2
CL	46.5		520.9		96.0

However, the effect of position of the substituent can be analyzed in depth with the series of twelve 3-bromobenzophenone thiosemicarbazones that are also monosubstituted with halogen and aliphatic groups. (F, Cl, Br, and CH₃). Table 152 compares IC₅₀ values for these compounds. The inhibitory activity of the compounds increases with more electronegative substituents if they are positioned in the *ortho*-position. This trend is not observed for the *meta* and *para* positions on the second phenyl

ring. Furthermore, the *meta* brominated analog is the most potent inhibitor found in this library. In general, *meta*-substituted TSCs analogs are better inhibitors than *ortho* and *para*-substituted TSCs. Interestingly, **17** is the only *para*-halogenated analog with excellent inhibition properties.

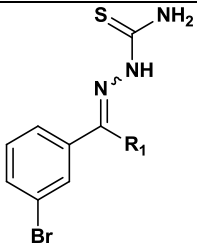
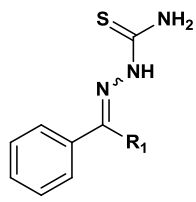
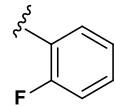
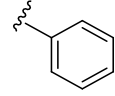
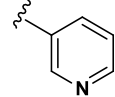
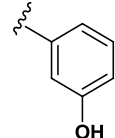
Table 152. Effect of the Position of Substituents in the Inhibitory Activity of 3-Bromo-Benzophenone Thiosemicarbazones

		R_1	IC_{50} (nM)						
			<i>ortho</i>		<i>meta</i>		<i>para</i>		
-F	14	CZ	70.4	4	CZ	110.8	17	CZ	14.3
		CL	23.9		CL	250.3		CL	79.6
-Cl	15	CZ	182.4	5	CZ	30.2	18	CZ	275.3
		CL	1610		CL	131.4		CL	327.1
-Br	24	CZ	ND	1	CZ	10.6	21	CZ	127.0
		CL	2600		CL	16.7		CL	≥ 10000
CH ₃	20	CZ	770.5	7	CZ	94.4	19	CZ	526.9
		CL	≥ 10000		CL	224.4		CL	2156

Compounds **61-65** (Table 153) also demonstrated the importance of the presence of bromine in one of the phenyl rings in order to enhance the potency of thiosemicarbazones. An examination of unsubstituted and bromo-substituted analogs unsubstituted and halogenated benzophenones, phenols, pyridines, and thiophenes; revealed that none of the unsubstituted compounds showed significant inhibitory activity. Bromination in the 3 position of one of the phenyl rings of the benzophenone group is also critical for the activity of TSC as potential cruzain inhibitors. A closer examination of halogenated compounds with similar chemical structures reveals that a different halogen substituent, such as fluorine, greatly reduces the activity of the compounds. The 3,3'-Difluorobenzophenone TSC (**54**) is almost 48 times less active than 3,3'-dibromobenzophenone TSC (**1**). Furthermore, both 4,4'-difluorobenzophenone

TSC (**55**) and 4,4'-dibromobenzophenone TSC (**47**) showed no activity against cruzain, showing that *para*-disubstitution with halogen substituents cannot be used as a possible route for the design of similar compounds.

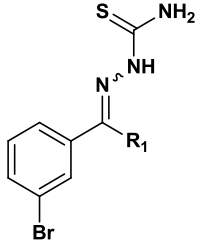
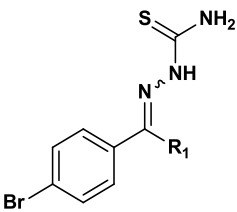
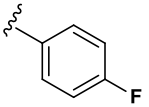
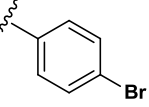
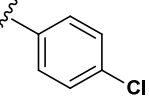
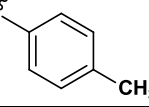
Table 153. Comparison between Brominated and Unbrominated Benzophenone Thiosemicarbazones for Cruzain

R_1						
		IC ₅₀ (nM)				
	14	CZ	70.4	61	CZ	ND
		CL	23.8		CL	≥ 10000
	2	CZ	73.0	62	CZ	≥ 10000
		CL	≥ 10000		CL	≥ 10000
	29	CZ	ND	63	CZ	≥ 10000
		CL	1000		CL	≥ 10000
	8	CZ	211.9	65	CZ	≥ 10000
		CL	188.7		CL	≥ 10000

Compounds **58**, **59**, and **60** (Table 144) demonstrated that modification of the thiosemicarbazone moiety (i.e. substitution on the NH₂ group of the thiosemicarbazone moiety) was detrimental in the activity of the synthetic compounds. None of these compounds showed significant activity towards cruzain.

Finally, Table 147 shows a small series of fluorene thiosemicarbazones. Both of them showed any activity towards cruzain (Analogues **80** and **81**).

Table 154. Comparison of the Inhibitory Activity between *meta* and *para*-Bromination of Benzophenone Thiosemicarbazones

R_1						
	IC ₅₀ (nM)					
	17	CZ	14.3	48	CZ	1453
		CL	79.6		CL	3320
	21	CZ	127.0	47	CZ	≥ 10000
		CL	327.1		CL	≥ 10000
	18	CZ	275.3	49	CZ	≥ 10000
		CL	≥ 10000		CL	≥ 10000
	19	CZ	770.5	50	CZ	2262
		CL	2156		CL	4570

Advanced Kinetic Studies

Four compounds, **1**, **9**, **10** and **17** (Figure 128) were screened against recombinant cruzain and demonstrated to be potent inhibitors of this protease. Their IC₅₀ values varied between 10.6 and 25 nM. Several experiments were carried out to characterize their mode of inhibition against recombinant cruzain. Compounds **1**, **10** and **17** are halogen-substituted bromobenzophenone thiosemicarbazone analogs. Compounds **1**, **10** and **17** are 3,3'-dibromobenzophenone TSC; 3-bromo-3',5'-difluorobenzophenone TSC and 3-bromo-4'-fluorobenzophenone TSC respectively. Finally, analog **9** is 3-bromo-4'-acetobenzophenone TSC.

Kinetic Analysis of 3-Bromo-3'-Bromobenzophenone Thiosemicarbazone (1) as a Cruzain Inhibitor

Effect of Preincubation Studies on Cruzain Inhibition Assays using 1

The effect of preincubation time in the determination of IC₅₀ values was examined using analog **1**. Experimental details are described in Chapter Two. IC₅₀ values were determined at eight different preincubation times ranging between 0 and 240 minutes. Final concentration of the inhibitor varied between 0 and 10 μM. Table 155 and Figure 130 show the results of the studies. Final conditions are: 1 mM EDTA, 2.5 DTT, 0.01% Brij 30, 2% DMSO, 15 μM Z-FR-AMC and 0.1 nM cruzain.

Table 155. Effect of Preincubation Time on IC₅₀ Values of **1** against Cruzain

<i>Pre-incubation times (min)</i>	<i>IC₅₀ ± Standard Error (nM)</i>
240	0.51 ± 0.050
120	0.93 ± 0.08
90	1.22 ± 0.04
60	1.19 ± 0.08
30	2.12 ± 0.20
5	10.6 ± 0.4
1	21.0 ± 1.9
0	207.5 ± 17.6

Preincubation time plays a vital role in the determination for any *in vitro* assay. The effect of preincubation on the potency of **1** as seen in Table 155 is quite evident. The IC₅₀ value is enhanced 10 fold by incubation as little as one minute. (IC_{50(0min)}: 208 and IC_{50(1min)}: 21.0 nM). A similar trend can be observed when preincubation times were increased. The IC₅₀ value when cruzain was preincubated four hours with compound **1** was 0.5 nM. These results confirmed the strong dependence of IC₅₀ value on the preincubation time parameter. Controls (i.e. uninhibited cruzain) were monitored at

every preincubation time. There was no significant loss of catalytic activity at longer preincubation times.

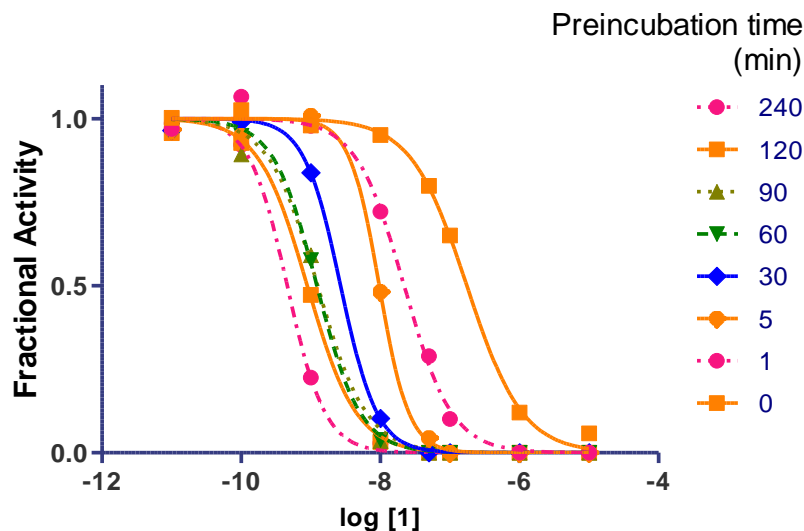


Figure 130. Effect of Preincubation Time on IC₅₀ Values of **1** against Cruzain

*Determination of K_i^{app} using Morrison's Quadratic Equation. Effect of Preincubation Time using **1**.*

The data obtained from the preincubation were further analyzed. The possibility that **1** was a tight-binding inhibitor was analyzed with Morrison's quadratic equation. (See equation 1.11).

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

The rates v_i and v_o are the inhibited and uninhibited cruzain velocities (RFU/s); $[E]_T$ (nM) is, the total concentration of enzyme found in solution (free enzyme and inhibitor-enzyme complex); $[I]_T$ (nM) is the total concentration of inhibitor present in

solution (free inhibitor and inhibitor-enzyme complex); and K_I (nM) is the inhibition constant, often referred as the dissociation constant. The equation may be solved to give two possible answers. However, the equation is written so that there is only one possible answer that fits physiological conditions (i.e. $K_I^{\text{app}} > 0$). GraphPad 5.0 was used to fit the data after data manipulation for every preincubation time point. Inhibitor, substrate, and enzyme concentrations, as well as K_M , K_I^{app} , were all in micromolar units (μM). The residual activity (or v_i/v_0) was normalized to 1 (i.e. $v_0: 1$ and $0 \leq v_i \leq 1$). Normalized residual activity and $[I]$ were defined as the dependent and independent variables respectively. Nonlinear regression was applied using the following conditions: $[S]: 15 \mu\text{M}$, $K_M: 1.01 \mu\text{M}$, $[E]_T: 0.0001 \mu\text{M}$ and $v_0: 1$. Normalization and nonlinear conditions were used to three or four independent experiments per preincubation time. Results and standard errors can be seen in Table 156 and Figure 131.

Table 156. Effect of Preincubation Time in K_I^{app} Values of **1** against Cruzain

<i>Pre-incubation times (min)</i>	$K_I^{\text{app}} \pm \text{Standard Error (nM)}$
0	12.82 ± 1.04
1	1.14 ± 0.16
5	0.59 ± 0.03
30	0.15 ± 0.01
60	0.08 ± 0.01
90	0.08 ± 0.01
120	0.06 ± 0.01

The K_I^{app} values are shown in Table 156. It is apparent that the inhibition constants are also time-dependent. One minute of preincubation shows an excellent inhibition constant with a value of 1.14 nM. Standard preincubation times (5 minutes) enhanced more than 20-fold) when compared the calculated parameter for no preincubation time. The best inhibition constant value was obtained at 2 hours.

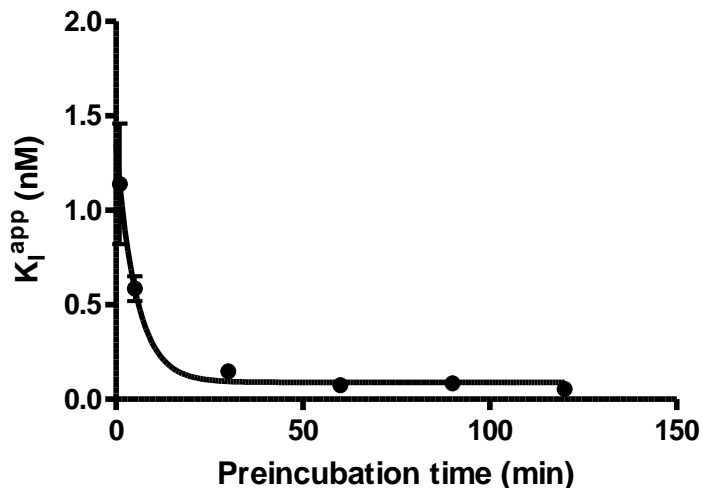


Figure 131. Effect of Preincubation Time in K_I^{app} values of **1** against Cruzain

The apparent K_I was 200-fold better when compared with the corresponding value at no preincubation time ($K_I^{app} = 12.8$ nM). The 240 minutes data did not fit into the equation and K_I^{app} could not be estimated. Inhibition constants were calculated based on the assumption that compound **1** might be a tight binding inhibitor. The obtained results confirmed that dibromobenzophenone thiosemicarbazone is a very tight inhibitor with K_I^{app} as low as 60 pM.

Cruzain Reversibility Studies

Compound **1** was found to be a time-dependent inhibitor in the preincubation studies. Therefore, we decided to explore if this specific compound was a reversible inhibitor of cruzain. A mixture containing 100 X cruzain and 10 X IC_{50} (preincubation time: 5 minutes), which are 10 and 106 nM respectively, were incubated for different preincubation times ranging between 0 and 240 minutes at 25 °C. The inhibition of cruzain by **1** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. Experiment was set up in order to monitor these reactions almost

immediately after adding cruzain to the assay. Final conditions are 0.1 nM cruzain, 1.1 nM of the dibromobenzophenone thiosemicarbazone (**1**), and 15 μ M Z-FR-AMC.

Additionally, a control experiment (cruzain with DMSO as control vehicle) was also carried out for every preincubation time. Figure 132 shows the release of AMC for the first 3500 seconds. Readings were taken every 50 seconds. Uninhibited and inhibited reactions were followed for a total time of four hours. Cruzain was able to recover its catalytic activity after the rapid dilution with assay buffer containing Z-FR-AMC.

Apparent substrate depletion can be observed in the uninhibited reaction after 3500 seconds. Thus, linear regression was applied to the first 3500 seconds of the reactions to determine cruzain activity. Table 157 shows cruzain activities for uninhibited and inhibited reactions. Rates for uninhibited and inhibited (preincubation time: 240 minutes) are 1.89 and 0.62 nM AMC/s respectively (remaining activity: 33%). However, similar experiments reported reversibility studies using a preincubation time of 60 minutes. According to Table 157, cruzain's remaining activity was 40.7% when the mixture cruzain-inhibitor was preincubated for one hour at the specified conditions. Interestingly, the inhibitory activity of **1** was not improved by longer preincubation times suggesting the slow release of the compound (See Figure 132, Figure 133, and Figure 134).

Further observations helped to identify if cruzain was able to recover after 240 minutes of preincubation time with dibromobenzophenone thiosemicarbazone (See Figure 133). Cruzain was able to recover its inhibited activity after 400 seconds of reactions.

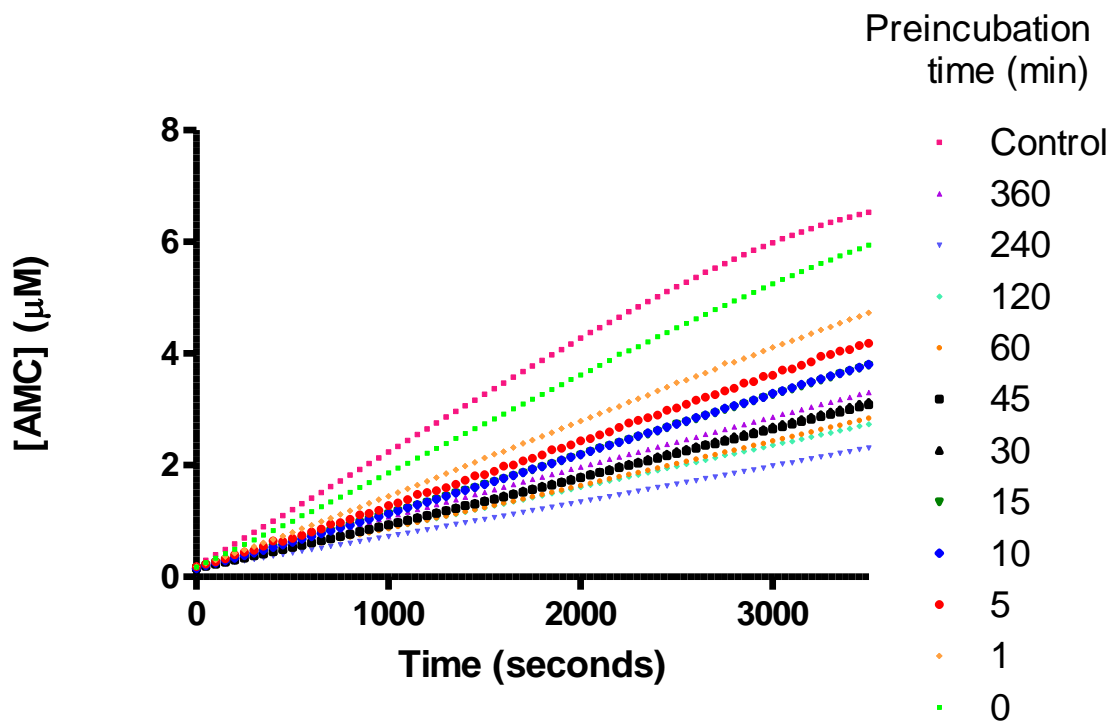


Figure 132. Effect of Preincubation Studies in Cruzain Reversibility Studies with **1** using 15 μM Z-FR-AMC

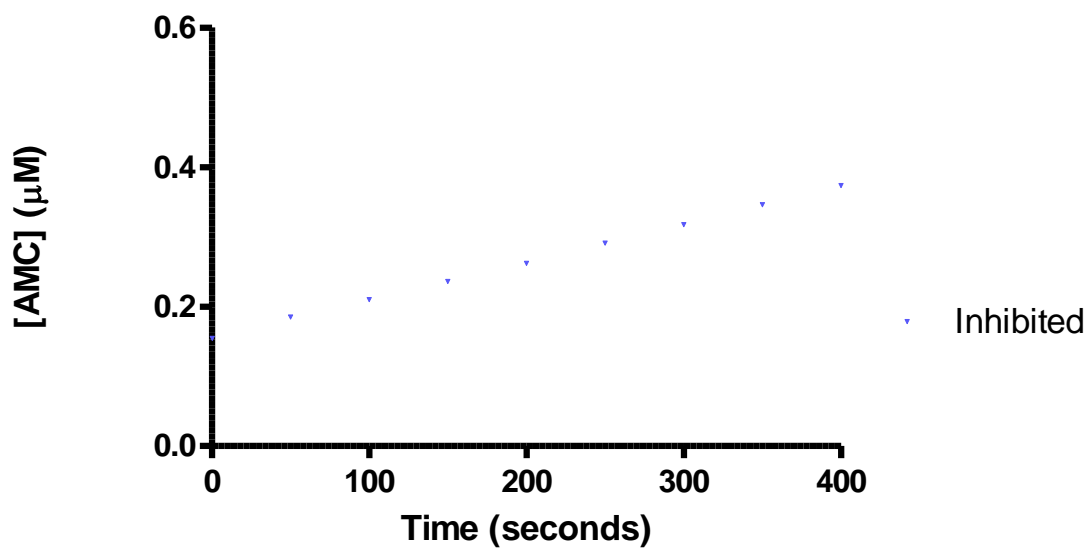


Figure 133. Cruzain Reversibility Studies with **1** Using 15 μM Z-FR-AMC

Table 157. Effect of Preincubation Studies in Cruzain Reversibility Studies with **1**

<i>Pre-incubation times (min)</i>	<i>Inhibited activity (nM AMC/s)</i>	<i>Remaining activity (%)</i>
0 (uninhibited)	1.89	100
0 (inhibited)	1.69	89.4
1	1.32	69.8
5	1.16	61.4
15	1.05	55.6
30	0.85	45
45	0.84	44.4
60	0.77	40.7
120	0.73	38.6
240	0.62	32.8

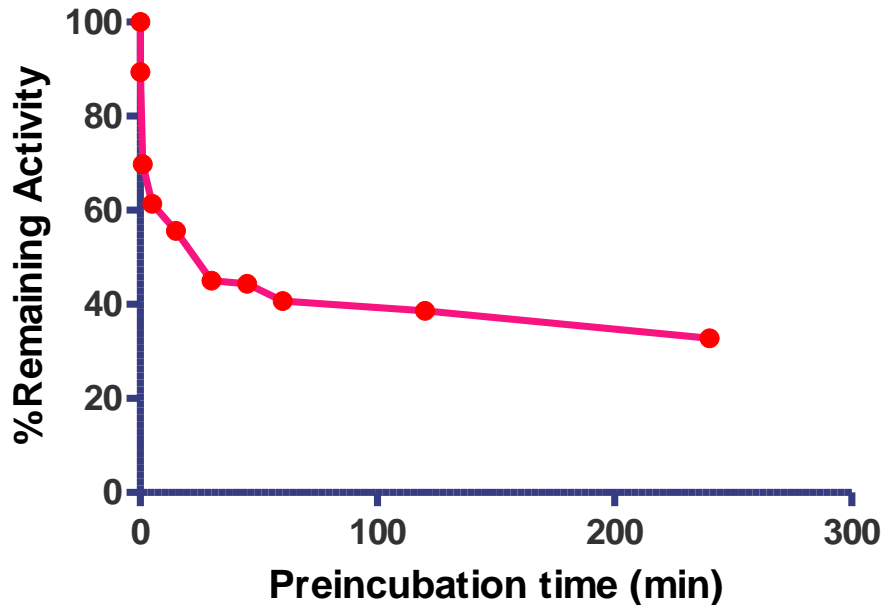


Figure 134. Effect of Preincubation Times on the Activity of Cruzain with **1** Using 15 μ M Z-FR-AMC

Effect of Substrate Concentration (Z-FR-AMC) on IC₅₀ Values

We finally tried to determine the mode of inhibition of thiosemicarbazones as slow-binding inhibitors of cruzain. Therefore, we investigated the effect of substrate concentration on IC₅₀ values.¹⁷⁷ Classical methods for the determination of mode of inhibition (steady-state kinetics) fail when studying a slow, tight-binding inhibitors. In

these cases, Copeland offers a suitable alternative for the investigation. IC_{50} values decrease hyperbolically if the compound is an uncompetitive inhibitor. In the case of competitive inhibitors, IC_{50} values increase in a linear trend with higher substrate concentrations. Six substrate concentrations (Final concentrations: 15, 10, 7.5, 5.0 2.5 and 1.0 μM) were used to determine IC_{50} values for compound **1**. Cruzain and the series of inhibitors were incubated for five minutes. Results can be found in Table 158 and Figure 135.

Table 158. Effect of Substrate Concentration on IC_{50} Values of **1** against Cruzain

<i>[Z-FR-AMC]</i> (μM)	$IC_{50} \pm \text{Standard Error}$ (nM)
15	11.3 ± 1.2
10	10.9 ± 0.2
7.5	9.5 ± 0.7
5.0	7.6 ± 0.7
2.5	6.5 ± 0.1
1.0	4.1 ± 0.2

The effect of substrate concentration was investigated using **1** as a lead compound of the series of thiosemicarbazones. A linear behavior can be observed with the values. Similarly, to cathepsin L case, the inhibitor acts as a competitive inhibitor with respect to the substrate. According to literature, a positive linear behavior (i.e. $IC_{50} \propto [S]$) is an indication the compound acts as competitive inhibitor, that is, both substrate and the compound compete for cruzain active site. Although the compound is a slow-binding inhibitor, the reduction of substrate concentration (15-fold) improves the IC_{50} value by almost 3-fold ($IC_{50}([Z-FR-AMC]: 15 \mu\text{M}): 11.3 \text{ nM}$ vs $IC_{50}([Z-FR-AMC]: 1 \mu\text{M}): 4.1 \text{ nM}$).

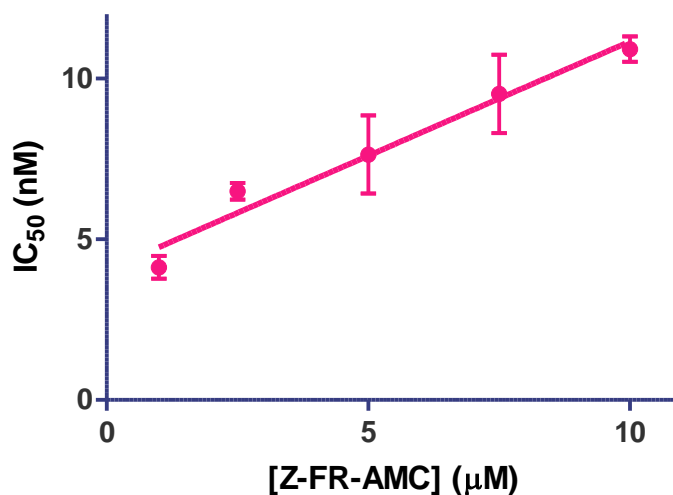


Figure 135. Effect of Substrate Concentration on IC₅₀ Values of **1** against Cruzain

*Kinetic Analysis of 3-Bromo-3',5'-Difluorobenzophenone Thiosemicarbazone (**10**) as a Cruzain Inhibitor*

*Effect of Preincubation Times on Cruzain Inhibition Assays using **10***

The effect of preincubation time in the determination of IC₅₀ values was examined using analog **10**. Experimental details are described in chapter 4 (See Effect of Preincubation Times on Cruzain Inhibition Assays using **1**). Results can be seen in Table 159 and Figure 136.

Table 159. Effect of Preincubation Time on IC₅₀ Values of **10** towards Cruzain

<i>Pre-incubation times (min)</i>	<i>IC₅₀ ± Standard Error (nM)</i>
240	0.8 ± 0.03
120	1.6 ± 0.1
90	2.2 ± 0.2
60	2.3 ± 0.2
30	4.4 ± 0.4
5	17.7 ± 1.5
1	37.3 ± 3.6
0	222.5 ± 6.7

The effect of preincubation times on the potency of **10** can be seen in Table 159 is quite evident. The IC_{50} value is enhanced 6-fold by incubation as little as one minute. ($IC_{50(0min)}$: 222.5 and $IC_{50(1min)}$: 37.3 nM). A similar trend can be observed when preincubation time was increased and the IC_{50} value when cruzain was preincubated four hours with compound **10** was 0.8 nM.

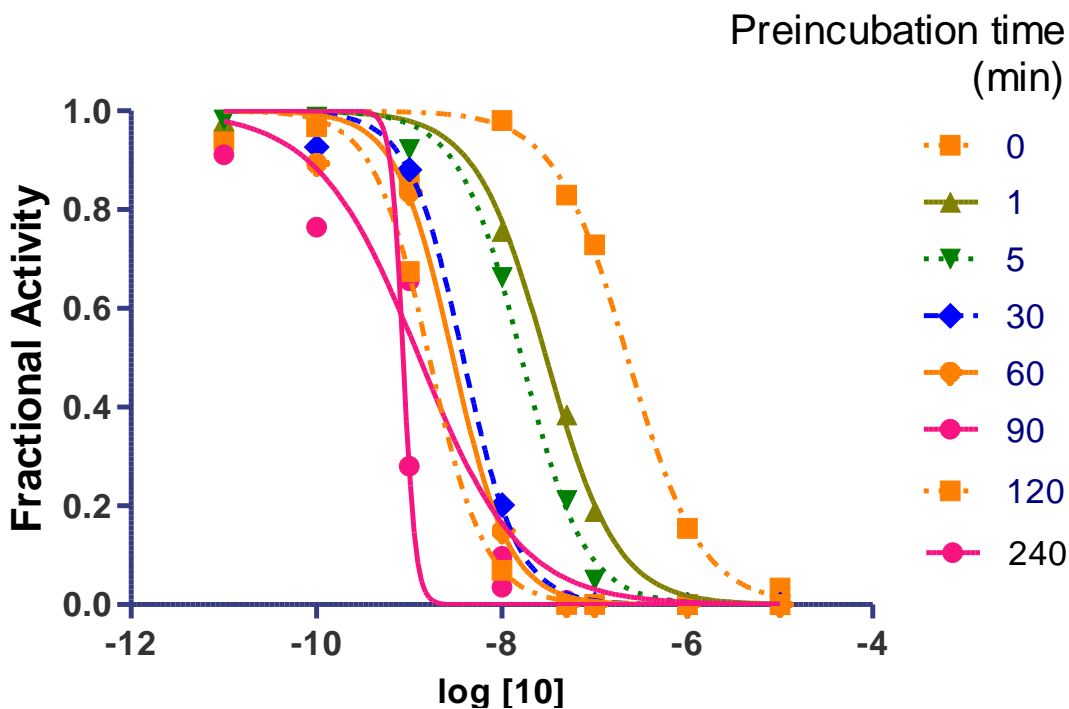


Figure 136. Effect of Preincubation Time on IC_{50} Values of **10** against Cruzain

*Determination of K_i^{app} using Morrison's Quadratic Equation. Effect of Preincubation Time using **10**.*

The data obtained from the preincubation studies were further analyzed. The possibility that **10** was a tight-binding inhibitor, like analog **1**, was analyzed with Morrison's quadratic equation. (See equation 1.11). Results can be found in Table 160 and Figure 137.

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

Table 160. Effect of Preincubation Time on K_I^{app} values of **10** against Cruzain

Pre-incubation times (min)	$K_I^{\text{app}} \pm \text{Standard Error (nM)}$
0	12.23 ± 1.92
1	2.11 ± 0.25
5	1.03 ± 0.09
30	0.23 ± 0.01
60	0.14 ± 0.01
90	0.12 ± 0.01
120	0.10 ± 0.01

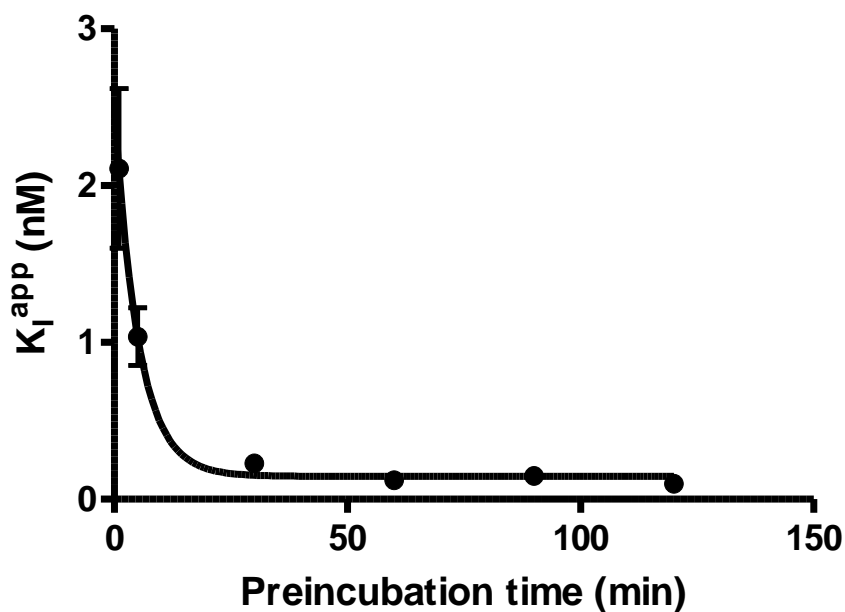


Figure 137. Effect of Preincubation Time in K_I^{app} values of **10** against Cruzain

The K_I^{app} values of **10** are also time-dependent. One minute of preincubation shows an excellent inhibition constant with a value of 2.1 nM. The standard preincubation time (5 minutes) enhanced K_I^{app} more than 12-fold when compared the calculated parameter for no preincubation time (K_I^{app} : 0.10 nM). The best inhibition

constant value was obtained at 2 hours. The apparent K_I was 120-fold better when compared with the corresponding value at no preincubation time ($K_I^{\text{app}} = 12.2$ nM). The 240 minutes data did not fit into the equation and K_I^{app} could not be estimated. The obtained results confirmed that **10**, a polyhalogenated benzophenone thiosemicarbazone is a very tight inhibitor with K_I^{app} as low as 100 pM.

Cruzain Reversibility Studies

Compound **10** was found to be a time-dependent inhibitor in the preincubation time studies. Therefore, we decided to explore if this specific compound was a reversible inhibitor of cruzain. A mixture containing 100 X cruzain and 10 X IC_{50} (preincubation time: 5 minutes), which are 10 and 180 nM respectively, were incubated for different preincubation times ranging between 0 and 240 minutes at 25 °C. The inhibition of cruzain by **10** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. The experiment was set up in order to monitor these reactions almost immediately after adding cruzain to the assay. Final conditions are 0.1 nM cruzain, 1.8 nM of **10**, and 15 μ M. Additionally, a control experiment (cruzain with DMSO as control vehicle) was also carried out for every preincubation time. Figure 138 shows the release of AMC for the first 3500 seconds. Readings were taken every 50 seconds. Uninhibited and inhibited reactions were followed for a total time of four hours. Cruzain was able to recover its catalytic activity after the rapid dilution with assay buffer containing Z-FR-AMC. Apparent substrate depletion can be observed in the uninhibited reaction after 3500 seconds. Thus, linear regression was applied to the first 3500 seconds of the reactions to determine cruzain activity. Figure 138 shows cruzain activities for uninhibited and inhibited reactions. Rates for uninhibited and inhibited (preincubation

time: 240 minutes) are 1.89 and 0.52 nM AMC/s respectively (Remaining activity: 73%). According to Table 161, the polyhalogenated thiosemicarbazone was able to inhibit cruzain activity by 54.0% when the mixture cruzain-inhibitor was preincubated for two hours at the specified conditions. Interestingly, the inhibitory activity of **10** was not improved by longer preincubation times suggesting the slow release of the compound (See Figure 138, Figure 139 and Figure 140). Further observations helped to identify if cruzain was able to recovery after 240 minutes of preincubation time with dibromobenzophenone thiosemicarbazone (See Figure 139, and Figure 140). Cruzain was able to recover its inhibited activity after 400 seconds of reactions.

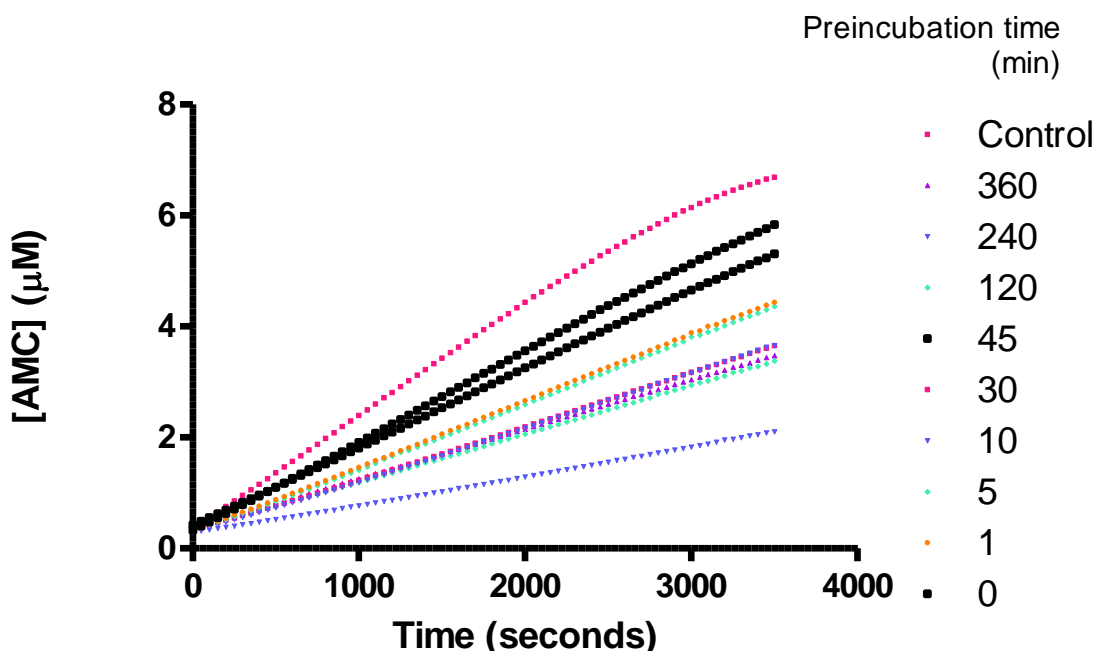


Figure 138. Effect of Preincubation Studies in Cruzain Reversibility Studies with **10** Using 15 µM Z-FR-AMC

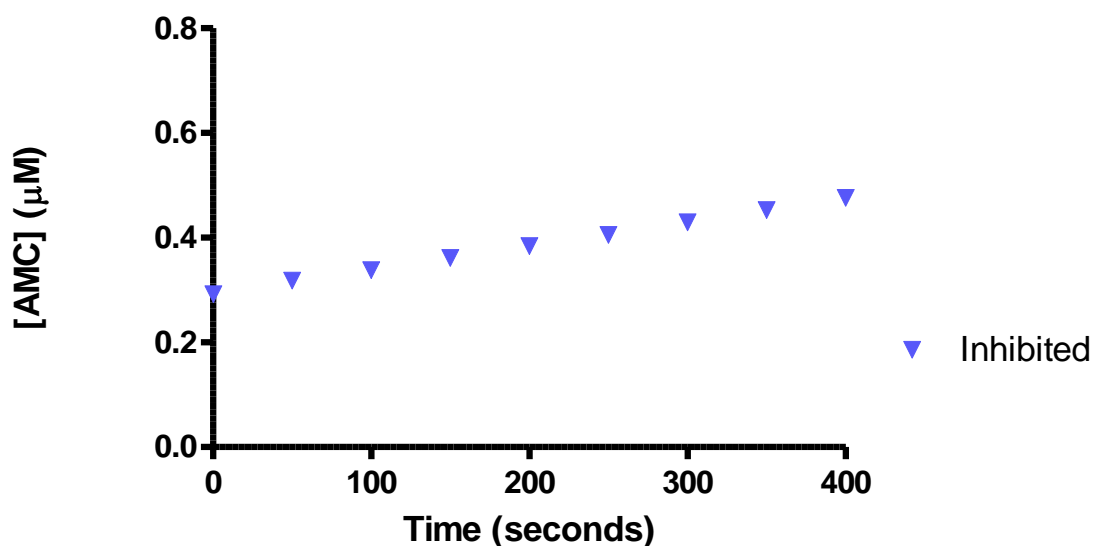


Figure 139. Cruzain Reversibility Studies with **10** Using 15 μM Z-FR-AMC

Table 161. Effect of Preincubation Studies on Cruzain Reversibility Studies with **10**

<i>Pre-incubation times (min)</i>	<i>Inhibited activity (nM AMC/s)</i>	<i>Inhibited activity (%)</i>
0 (uninhibited)	1.89	0
0 (inhibited)	1.60	15.3
1	1.19	37.0
5	1.17	38.1
10	0.97	48.7
30	0.96	49.2
120	0.87	54.0
240	0.52	72.5

Effect of Substrate Concentration (Z-FR-AMC) on IC₅₀ Values

We finally tried to determine the mode of inhibition of **10** as a slow-binding inhibitor of cruzain. Six substrate concentrations (Final concentrations: 15, 10, 7.5, 5.0, 2.5 and 1.0 μM) were used to determine IC₅₀ values for compound **10**. Cruzain and the series of inhibitors were incubated for five minutes. Results can be found in Table 162

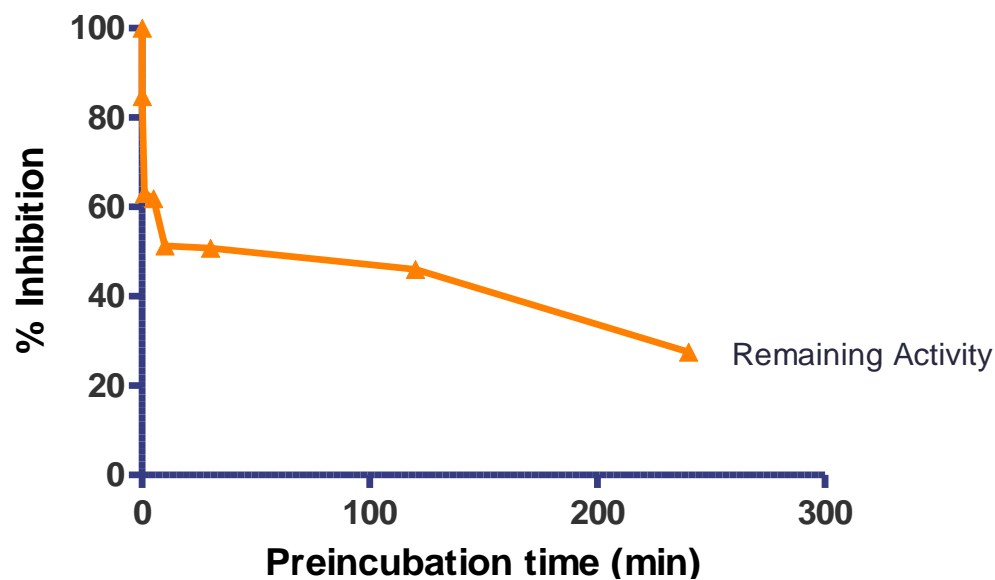


Figure 140. Effect of Preincubation Times on the Activity of Cruzain with **10** using 15 μM Z-FR-AMC

Table 162. Effect of Substrate Concentration on IC_{50} Values of **10** against Cruzain

$[Z\text{-FR-AMC}]$ (μM)	$IC_{50} \pm \text{Standard Error}$ (nM)
15	17.6 ± 0.6
10	17.5 ± 1.15
7.5	14.9 ± 0.8
5.0	13.1 ± 0.4
2.5	13.4 ± 0.6
1.0	4.75 ± 0.3

The effect of substrate concentration was investigated using **10** as a lead compound of the series of thiosemicarbazones. Similar to **1**, **10** acts as a competitive slow-binding inhibitor. The trend shows an increase in IC_{50} value with increased substrate concentration. The reduction of substrate concentration (15-fold) improves the IC_{50} value by almost 4 times. ($IC_{50([Z\text{-FR-AMC}]: 15 \mu\text{M})}$: 17.6 nM vs $IC_{50([Z\text{-FR-AMC}]: 1 \mu\text{M})}$: 4.8 nM).

*Kinetic Analysis of 3-Bromo-3'-Acetobenzophenone Thiosemicarbazone (9)
as a Cruzain Inhibitor*

Effect of Preincubation Times on Cruzain Inhibition Assays Using 9

The effect of preincubation time in the determination of IC₅₀ values was examined using analog **9**. Experimental details are described in chapter 4 (See Effect of Preincubation Times on Cruzain Inhibition Assays using **1**). Preincubation times ranged between 0 and 120 minutes. Results can be seen in Figure 141 and Table 163.

Table 163. Effect of Preincubation Time on IC₅₀ Values of **9** against Cruzain

<i>Pre-incubation times (min)</i>	<i>IC₅₀ ± Standard Error (nM)</i>
120	3.8 ± 0.1
60	4.2 ± 0.4
30	8.8 ± 0.5
5	15.4 ± 1.2
1	132.5 ± 6.2
0	4935 ± 122

The effect of preincubation times on the potency of **9** can be seen in Table 159. The IC₅₀ value is enhanced 37-fold by incubation as little as one minute (IC_{50(0min)}: 4935 and IC_{50(1min)}: 132.5 nM). A similar trend can be observed when preincubation time was increased and the IC₅₀ value when cruzain was preincubated two hours with compound **9** was 3.8 nM.

Determination of K_i^{app} using Morrison's Quadratic Equation. Effect of Preincubation Time using 9.

The data obtained from the preincubation studies were further analyzed. Analogs **1** and **10**, both halogenated thiosemicarbazones inhibitors were tight binding inhibitors of cruzain. The addition of a nonhalogenated group to the main moiety might provide a

different behavior. Thus, the data were analyzed with Morrison's quadratic equation.

(See equation 1.11). Results can be found in Table 164 and Figure 142.

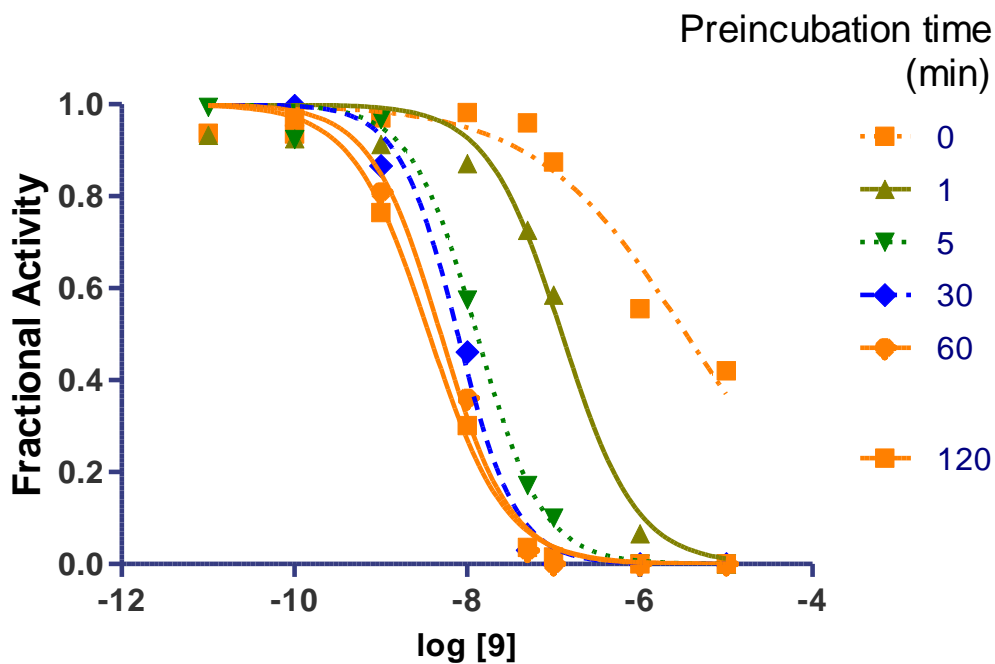


Figure 141. Effect of Preincubation Time on IC_{50} Values of **9** against Cruzain

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

Table 164. Effect of Preincubation Time on K_I^{app} Values of **9** against Cruzain

Pre-incubation times (min)	$K_I^{app} \pm$ Standard Error (nM)
1	11.19 ± 2.21
5	0.92 ± 0.06
30	0.42 ± 0.05
60	0.28 ± 0.03
120	0.27 ± 0.02

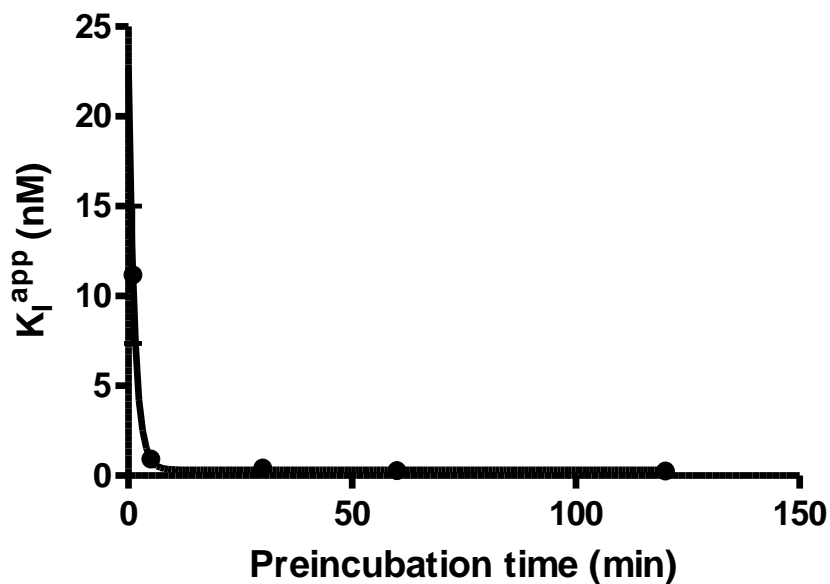


Figure 142. Effect of Preincubation Time in K_I^{app} values of **9** against Cruzain

The K_I^{app} values of **9** are also time-dependent. One minute of preincubation shows a good inhibition constant with a value of 11.2 nM. Standard preincubation times (5 minutes) enhanced the K_I^{app} value more than 12-fold) when compared the calculated parameter for one minute of preincubation time. The dissociation constant for the data set corresponding to 0 minutes preincubation time could not be determined. The best inhibition constant value was obtained at 2 hours. The apparent K_I was 37-fold better when compared with the corresponding value at one minute preincubation time ($K_I^{\text{app}} = 11.2$ nM). The obtained results confirmed that **9**, an 3-bromo-3'-acetobenzophenone thiosemicarbazone is an excellent inhibitor with K_I^{app} as low as 270 pM.

Kinetic Analysis of 3-Bromo-4'-Fluorobenzophenone Thiosemicarbazone (17) as a Cruzain Inhibitor

Effect of Inhibitor Concentration on Cruzain Progress Curves

Compound **17**, is a dihalogenated benzophenone thiosemicarbazone. Its value was determined to be 12.1 nM and is the second most potent inhibitor found in this library. Cruzain (0.1 nM) was added to six different concentration curves ranging from 0 to 10 μM of **17**. Reactions were initiated by the rapid addition of Z-FR-AMC (final concentration: 15 μM). The release of AMC from the nonfluorescent substrate was monitored every 3 seconds for fifty minutes. Figure 143 shows the effect of inhibitor concentration on the cruzain progress curves at a fixed substrate concentration. Uninhibited and inhibited reactions were monitored for 3000 seconds. Visual observations showed analog **17** is a time-dependence inhibitor. Then, the data were fitted to equation 1.7, by nonlinear regression analysis using GraphPad 5.0. P is the concentration of product (μM), v_o and v_s are the initial and steady-state velocities ($\mu\text{M/s}$), t is the time in seconds and k_{obs} the rate constant for conversion of the initial velocity v_o to the steady state velocity v_s . The rate constant (k_{obs}) units are given in s^{-1} . The equation was entered into the computational software, knowing that P and t are the dependent and independent variables, while keeping the velocities and the rate constant as unknowns. For each case, the constraints for their calculation were to give positive values (i.e. $k_{\text{obs}} \geq 0$). It is also worth noting that equation 1.7 is only valid when substrate depletion is insignificant.¹⁸⁸ Therefore, some points were excluded in every case for data fitting. Velocities, rates, r^2 and points analyzed for each substrate concentration are shown in Table 165. Experiments were carried in triplicate.

$$P = v_s t \frac{(v_o - v_s)}{k_{obs}} (1 - e^{-k_{obs}t}) \quad (1.7)$$

Table 165. Calculated Kinetic Parameters from Eq 1.7 for Cruzain Progress Curves with 15 μM Z-FR-AMC

[I] (μM)	1	0.5	0.1	0.05
v_s ($\mu\text{M/s}$)	3.45E-05	7.71E-05	0.000406	0.000731
v_i ($\mu\text{M/s}$)	0.003057	0.002666	0.00211	0.002111
k_{obs} (s^{-1})	0.005461	0.002984	0.00096	0.000843
r^2	≥ 0.94	≥ 0.98	≥ 0.99	≥ 0.99
Points analyzed	1000	1000	1000	1000

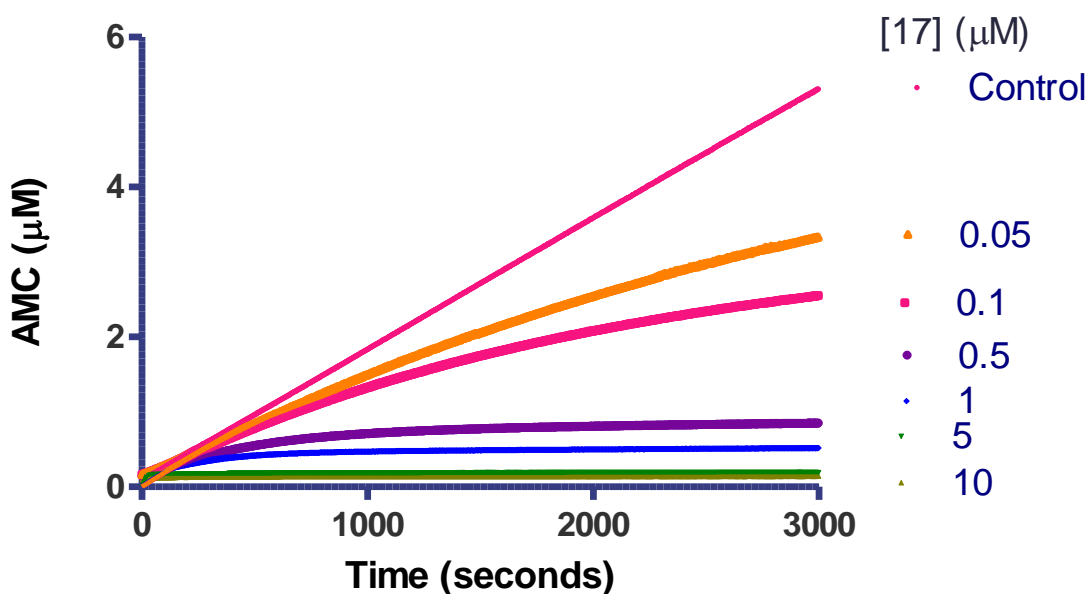


Figure 143. Cruzain Progress Curves with **17** Using 15 μM Z-FR-AMC

Approximate values for the rate constants, and velocities were obtained for every inhibitor concentration ($0.05 \leq [I] \leq 10 \mu\text{M}$). Data were fitted into Equation 1.7, but due to the elevated amount of inhibitor present in solution, compared to the concentration of the cruzain, the results were not reasonable. A better fit was acquired with lower concentrations of **17** (50-1000 nM). We decided to investigate the mechanism of inhibition of **17** as a slow binding inhibitor of cruzain, due to the clear time-dependence

of the progress curves. There are two possible mechanisms that are known for slow binding inhibitors.

Figure 144 shows the two proposed mechanisms that could be observed using the dihalogenated thiosemicarbazone (3-Br-4'-F-TSC) as a slow-binding inhibitor of cruzain. Mechanism A summarizes a simple reversible inhibition with k_{on} and k_{off} values relatively small. The parameters k_{on} and k_{off} are the rate constants for the formation and dissociation of the cruzain-3-Br-4'-F-TSC complex. Mechanism B offers an extra step which is more complicated to monitor or verify. This is a more general approach where it is assumed the cruzain-3-Br-4'-F-TSC complex undergoes an auto-isomerization, or a possible covalent modification of the enzyme due to the presence of the inhibitor.

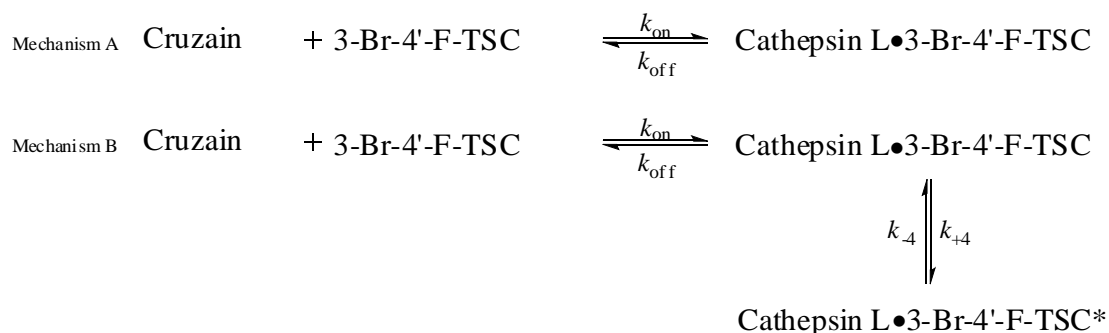


Figure 144. Possible Mechanisms of Inhibition of Cruzain by **17**

The k_{obs} values were plotted versus inhibitor concentration (Figure 145). Mechanism A, (Figure 144) is carried out for slow-binding inhibitors. However, many reversible, yet covalent inhibitors conform to mechanism A, as previously discussed in Chapter Two. The linear dependence of this parameter with respect of the inhibitor concentration is a clear indication the inhibition could follow a simple mechanism. Furthermore, a visual inspection of the progress curves suggests that **17** is a time-

dependent inhibitor of cruzain, but further experiments will need to be performed to validate that mechanism. Other slow-binding inhibitors, covalent and reversible, have also been reported with similar trends but did not follow simple reversible mechanisms.³²⁸ The figure represents the average value of three independent experiments with standards errors.

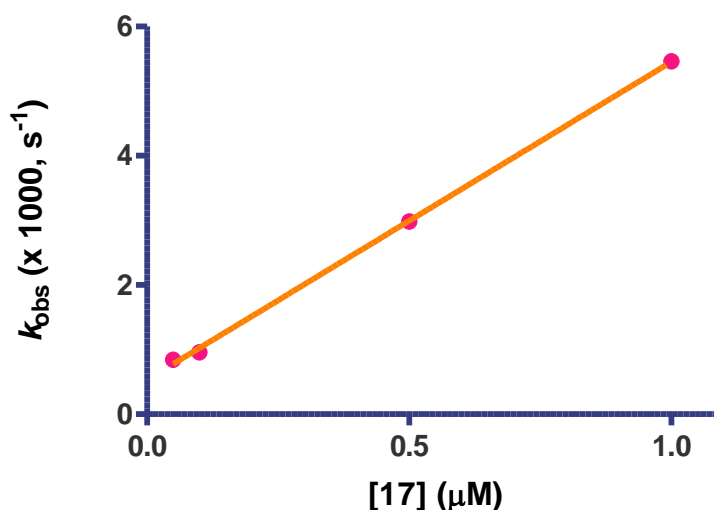


Figure 145. Calculated k_{obs} from Eq. 1.7 Cruzain Progress Curves with **17**

Table 166. Calculated k_{obs} from Eq. 1.7 Cruzain Progress Curves with **17**

[I] (μM)	k_{obs} (s ⁻¹)
1	0.0005852 ± 0.0004082
0.5	0.003082 ± 0.0001027
0.1	0.0009589 ± 4.016e-5
0.05	0.0008181 ± 5.447e-5

Linear regression analysis was performed using GraphPad 5.0 in order to analyze the data (15 μM Z-FR-AMC). Results can be seen in Table 55. The parameters k_{on} and k_{off} can be then calculated using equations 2.1 and 2.2.³²⁸

$$k_{obs} = k_{on}^{app} [I] + k_{off} \quad (2.1)$$

$$k_{obs} = \frac{k_{on}[I]}{\left(1 + \frac{[S]}{K_M}\right)} + k_{off} \quad (2.2)$$

Table 167. Calculated k_{on} , and k_{off} values from Eq. 2.1 Using Cruzain Progress Curves with **17**

[Z-FR-AMC] (μM)	15
$k_{on}^{app} (\text{M}\cdot\text{s})^{-1}$	4922
$k_{on} (\text{M}\cdot\text{s})^{-1}$	307.62
$k_{off} (\text{s}^{-1})$	0.0005318
r^2	0.9994

Effect of Preincubation Studies on Cruzain Inhibition Assays Using 17

The determination of kinetic parameters using cruzain progress curves provided strong evidence that **17** is a slow-binding inhibitor. The effect of preincubation time in the determination of IC_{50} values was examined using analog **17**. Experimental details are described in Chapter 2. Compound **17** IC_{50} values were determined at six different preincubation times ranging between 0 and 120 minutes. Inhibitor final concentration varied between 0 and 10 μM . Figure 146 shows the results of the studies.

Table 168. Effect of Pre-incubation Times on IC_{50} Values of **17** against Cruzain

Pre-incubation times (minutes)	$\text{IC}_{50} \pm \text{Standard Error (nM)}$
120	1.7 ± 0.1
60	2.2 ± 0.3
30	2.7 ± 0.3
5	12.1 ± 1.1
1	91.9 ± 7.4
0	3144 ± 41

The effect of preincubation on the potency of **17** can be seen in Table 168. The 3-bromo-3'-fluorobenzophenone thiosemicarbazone (**17**) showed low inhibitory activity when cruzain was not preincubated with the compound. However, the analog showed good activity with one minute preincubation time. The trend changed when the preincubation time was increased to 5 minutes. The inhibitory activity of **17** increased 32-fold with a remarkable IC_{50} less than 20 nM (12.1 nM). The potency of **17** modestly increased with longer preincubation times. Finally, the best activity was found when **17** was preincubated for two hours with cruzain, the activity of the thiosemicarbazone analog increased 7-fold to give a value of 1.7 nM. These results confirmed the strong dependence of IC_{50} value determination with respect to the preincubation time. Controls (i.e. uninhibited cruzain) were monitored at every preincubation time. There was no significant loss of catalytic activity at longer preincubation times.

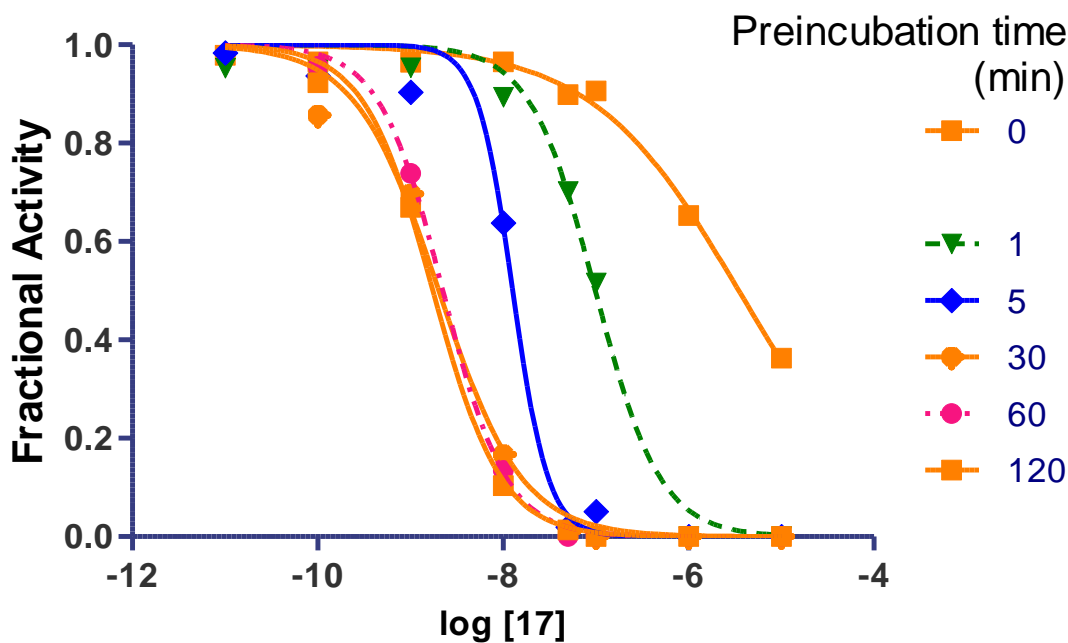


Figure 146. Effect of Preincubation Time on IC_{50} Values of **17** against Cruzain

*Determination of K_i^{app} Using Morrison's Quadratic Equation. Effect of Preincubation Time using **17***

The data obtained from the preincubation was further analyzed. The possibility that **17** was a tight-binding inhibitor was analyzed with Morrison's quadratic equation (See equation 1.11). Detailed analysis was described in chapter two. Results can be seen in Table 169 and Figure 147.

$$\frac{v_i}{v_o} = 1 - \frac{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M}))) - \sqrt{([E]_T + [I]_T + (K_I(1 + \frac{[S]}{K_M})))^2 - 4[E]_T[I]_T}}{2[E]_T} \quad (1.11)$$

Table 169. Effect of Preincubation Time on K_I^{app} Values of **17** against Cruzain

Pre-incubation Times (minutes)	$K_I^{app} \pm \text{Standard Error (nM)}$
0	189.6 ± 1.04
1	5.3 ± 0.6
5	0.8 ± 0.06
30	0.16 ± 0.01
60	0.15 ± 0.02
120	0.11 ± 0.01

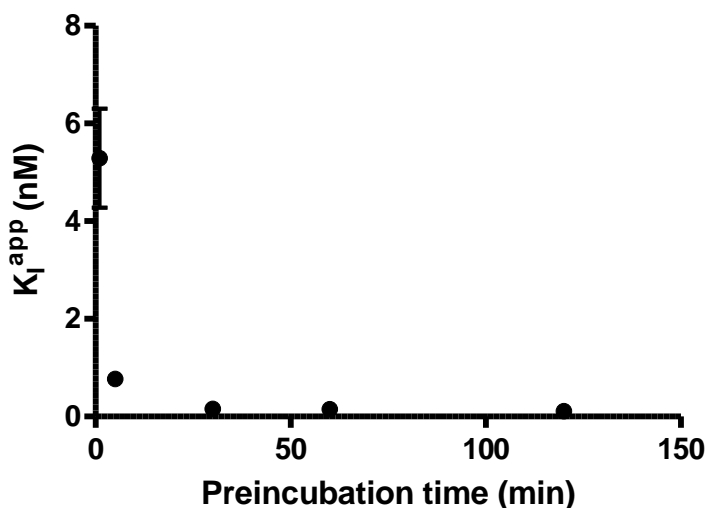


Figure 147. Effect of Preincubation Time on K_I^{app} Values of **17** against Cruzain

The K_I^{app} values are also time-dependent. One minute of preincubation shows an excellent inhibition constant with a value of 5.3 nM. Furthermore, the potency is extremely enhanced (more than 16-fold) for the calculated value at standard conditions (preincubation time: 5 minutes). The best inhibition constant value was obtained at the longest preincubation time (2 hours). The apparent K_I was 48-fold better when compared with the corresponding value at one minute preincubation time ($K_I^{\text{app}} = 0.11$ nM).

Cruzain Reversibility Studies

Compound **17** was found to be a time-dependent inhibitor in the preincubation studies. Therefore, we decided to explore if this specific compound was a reversible inhibitor of cruzain. A mixture containing 100 X cruzain and 10 X IC_{50} (preincubation time: 5 minutes), which are 10 and 120 nM respectively, were incubated at 25 °C for one hour. The inhibition of cruzain by **17** was able to be monitored by the rapid dilution of the mixture (100-fold) with assay buffer containing Z-FR-AMC. The experiment was set up in order to monitor these reactions almost immediately after adding cruzain to the assay. Final conditions were 0.1 nM Cruzain, 1.2 nM of **17**, and 15 μ M. Additionally, a control experiment (cruzain with DMSO as control vehicle) was also carried out. Figure 148 shows the release of AMC for the first 7500 seconds. However, uninhibited and inhibited reactions were followed for a total time of four hours. Cruzain was able to recover its catalytic activity after the rapid dilution with assay buffer containing Z-FR-AMC. The remaining activity was sixty percent, when compared to control, was observed in the recovery curve when cruzain was preincubated for one hour with compound **17**. Apparent substrate depletion can be observed in the uninhibited reaction after 5000 seconds. Thus, linear regression was applied to the first 7000 seconds of the reactions to determine

cruzain activity. Rates for uninhibited and inhibited (preincubation time: 60 minutes) are 1.49 and 0.91 nM AMC/s respectively. Further observations helped to identify if cruzain was able to recovery after 240 seconds of preincubation time with this synthetic compound (See Figure 149)

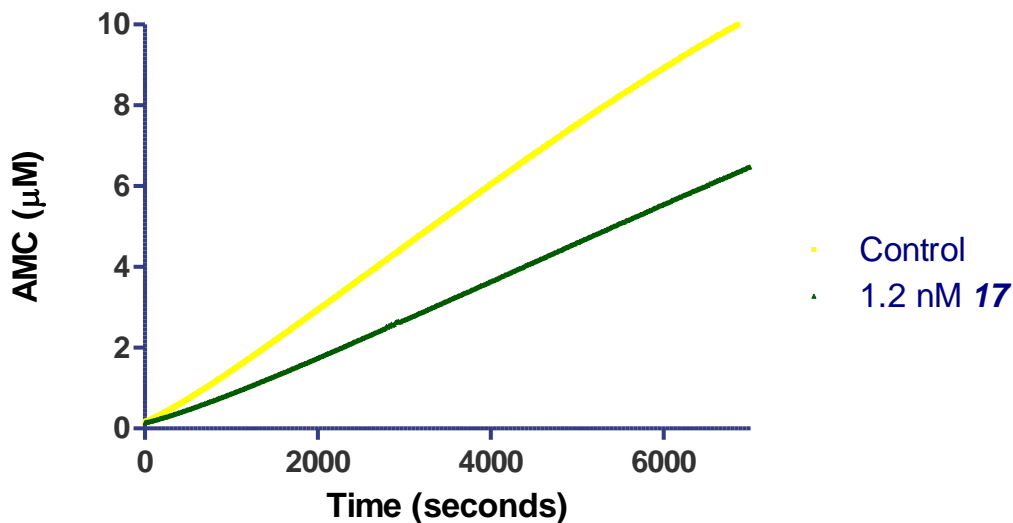


Figure 148. Cruzain Reversibility Studies with **17** Using 15 µM Z-FR-AMC

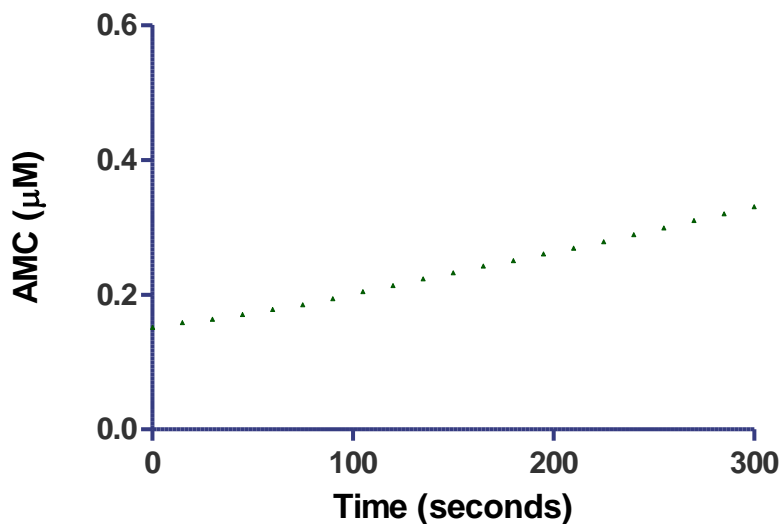


Figure 149. Cruzain Recovery Studies with **17** Using 15 µM Z-FR-AMC

Effect of Substrate Concentration (Z-FR-AMC) on Cruzain Progress Curves

The mode of inhibition of **17** was explored. A fixed concentration of **17** (100 nM) was used to monitor the effect of this compound on cruzain progress curves when using different substrate concentrations. Then, by using non-linear regression analysis (equation 1.7), the parameters k_{obs} , v_o , and v_s were determined. Compound **17** (0.1 μM) and ten Z-FR-AMC concentrations (ranging from 0.5 and 15 μM) were set up in order to verify the effect of substrate on inhibited cruzain progress curves. Reactions were started by the addition of cruzain (final concentration: 0.1 nM). The release of AMC was monitored every five seconds for sixty minutes. Figure 150 shows a typical result of this experiment. The results once again showed the strong time dependence of **17** when inhibiting cruzain. Substrate depletion is negligible for every substrate concentration as seen in the figure. Data were fitted into equation 1.7, (see Effect of inhibitor concentration in Cruzain progress curves. Velocities, rates, r^2 and points analyzed for each substrate concentration are shown in Tables 59, 60, and 61.

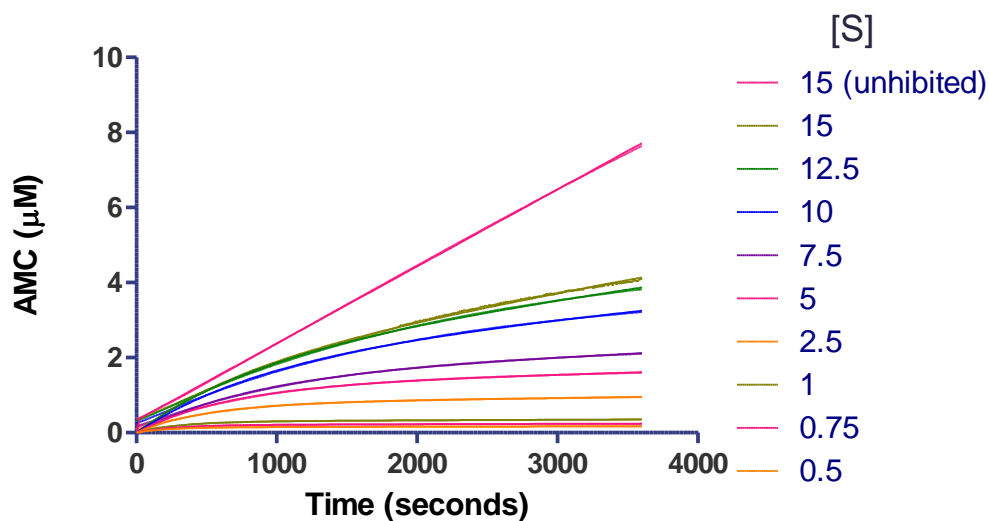


Figure 150. Cruzain Progress Curves with 0.1 μM **17** using Z-FR-AMC

Table 170 shows the average and standard error of calculated k_{obs} as a function of [S]. Data were graphed as a function of the unitless substrate/Michaelis-Menten constant [S]/ K_M , where K_M : 1.01 μM as previously determined. The rate constant k_{obs} follows an inversely hyperbolic trend. The rate constant values increased as the concentration of Z-FR-AMC is decreased in solution. This is a clear indication that **17** is a slow binding reversible competitive inhibitor of cruzain with respect to the fluorogenic substrate.

Table 170. Effect of [Z-FR-AMC] on k_{obs} Values when Using **17** against Cruzain

[Z-FR-AMC] (μM)	$k_{\text{obs}} \pm \text{Standard Error} (x 10^3, \text{s}^{-1})$
15	1.178 ± 0.067
12.5	1.039 ± 0.094
10	1.087 ± 0.011
7.5	1.187 ± 0.006
5	1.483 ± 0.014
2.5	2.101 ± 0.045
1.0	3.673 ± 0.148
0.75	4.172 ± 0.181
0.5	5.069 ± 0.195

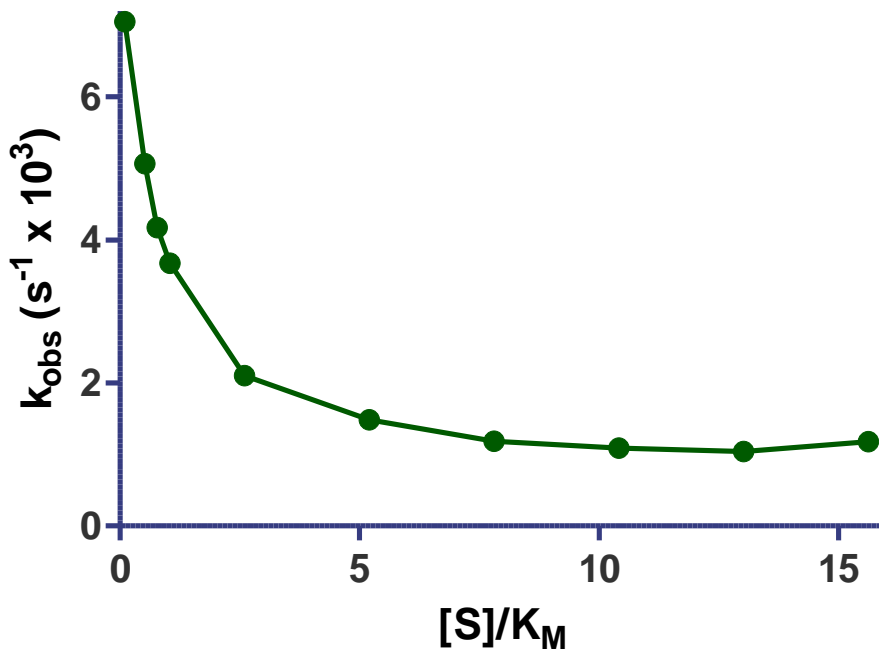


Figure 151. Effect of [Z-FR-AMC] on k_{obs} values when using **17** against Cruzain

Inhibition of Cruzain Collagenase Activity by Thiosemicarbazone Derivatives

Cruzain is the most important protease found in *Trypanosoma cruzi*. It has also been shown that cruzain is also capable of degrading high molecular weight proteins and participates in host invasion by degrading surrounding tissue. Research has shown that cruzain is a validated target in the treatment of Chagas' disease. Thus, we explored the catalytic activity of cruzain by using a natural substrate, type I collagen from human skin. We also tested the ability of one of the lead compounds (**17**, IC₅₀: 12 nM), to inhibit the collagenase catalytic activity of cruzain. Untreated and treated samples were incubated at 37 °C under acidic conditions (pH 5.5). Four sets of samples were prepared for this experiment. Two sets were not treated with **17** (DMSO was used as control vehicle). The third and fourth sets were treated with 20 µM of compound **17** in 2% DMSO. The effect of preincubation activity on the catalytic activity of cruzain and the inhibitory potency of compound **17** was also tested. A preincubation time of 30 minutes was chosen to carry out these experiments.

Inhibition of Cruzain Collagenase Activity by Thiosemicarbazone Derivative 17. (Preincubation Time: 0.5 Hours).

A sample containing three microliters of 1.2 µM Cruzain in CLI and 2.5 µl of 104 µM compound **17** were preincubated as previously described for thirty minutes. Individual samples were prepared to complete a series of six samples. Then, 7.5µl of 0.4 mg/ml type I collagen in acetic acid were added to the cruzain-inhibitor mixture. Every sample was carefully mixed and placed in a 37°C water bath. Reactions were monitored between 0 and 7 hours, stopped with 2.5 µl LDS sample buffer and heated at 90 °C for

every time point. Inactivated samples were immediately stored at -80 °C to preserve them.

Untreated and treated samples were loaded onto 4-12% Bis-Tris SDS gel. A sample of molecular weight standards, one sample with type I collagen only, and one sample of cruzain only were also loaded as reference controls. These last two samples were inactivated after thirty minutes of preincubation time. Electrophoresis was performed at 200 V for 60 minutes. The gel was rinsed with water and placed into a 1X SYPRO® solution, a fluorescent protein staining dye. The gel was stained for a minimum of one hour at room temperature. The destaining process was made by washing the gel once with water and twice with 7.5% acetic acid. Finally, a digital image of the gel was obtained by using a GE Typhoon 9400 fluorescence scanner imager.

Figure 152 shows the results of degradation of type I collagen from human skin using human cruzain soluble in acidic solutions.³³⁰ The activity of the protease was stopped at different time points (0-7 hours). The figure shows the progress of the reaction for the first three hours after collagen I was added to the cruzain-inhibitor mixture that was previously incubated for thirty minutes. The first lane is type I collagen that was not treated with cruzain in the vehicle control (2% DMSO). Reported literature establishes that type I collagen is large protein that is divided into three defined chains: α , β , and γ heavy chains and is crosslinked and polymerizes into fibrils. The molecular weights of the heavy chains are approximately $80 \leq MW_{\alpha} \leq 125$ kDa for α chain ; $160 \leq MW_{\beta} \leq 250$ kDa for β chain; and $240 \leq MW_{\gamma} \leq 375$ kDa for γ chain.^{331,332} Results show the presence of three major bands with high molecular weights that are consistent with the reported values for α , β , and γ heavy chains in the literature. molecular weights are: $\alpha \approx 102$ kDa,

$\beta \approx 150$ kDa and $\gamma \approx 200$ kDa (lane 1). The second and third bands of represent untreated and treated samples that were stopped after 15 minutes of the natural substrate.

Untreated samples contained type I collagen, and cruzain in DMSO, while treated samples consisted of type I collagen, cruzain, and compound **17** in DMSO. The untreated sample showed a slight degradation process even in as little as 15 minutes. The α , β , and γ bands look more diminished with respect to their respective control. The degradation is more evident in the case of the α bands (lane 2). The degradation process is evident after 60 and 180 minutes. All of the bands α , β , and γ , look diminished and there is evidence that heavy bands (compared to untreated collagen at time 0, lane 1) are degraded. A comparison between treated (lanes 3, 5 and 7) and untreated (lanes 4, 6 and 8) samples reveals that compound **17** was able to inhibit the collagenase activity of cruzain. Longer reaction times ($t= 7$ hours) indicate that uninhibited cruzain is able to completely degrade 3 μ g of type I collagen. These results indicate that the **17** is capable of inhibiting cruzain collagenase activity by approximately fifty percent up to 3 hours.

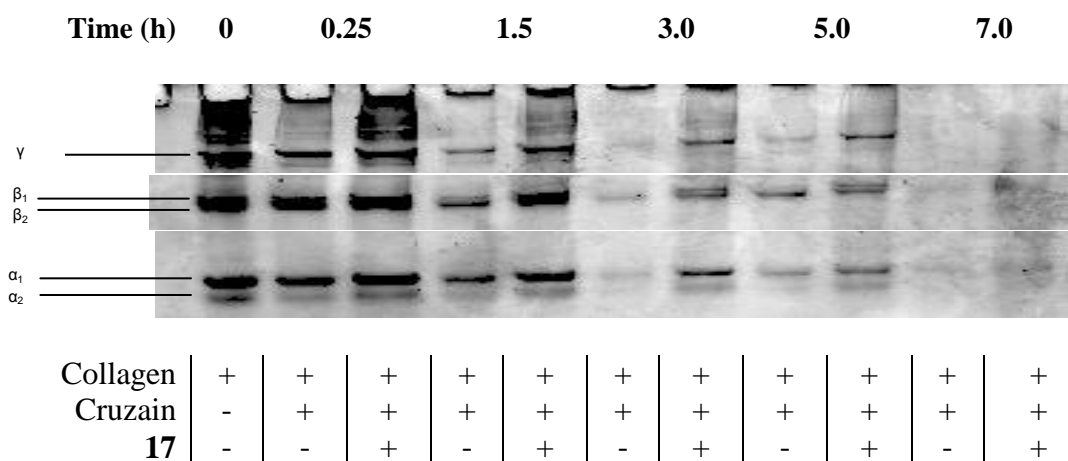


Figure 152. Inhibition of Collagenase Activity of Cruzain by **17**, Preincubation Time: 0.5 hours.

Molecular Docking Studies with Thiosemicarbazones as Inhibitors of Cruzain

An evaluation of the library of thiosemicarbazones showed several compounds are outstanding inhibitors of cruzain with IC_{50} values less than 20 nM. Thus, the most potent five compounds in the library (**1**, **9**, **10**, **17**, and **36**) were docked with cruzain using 1ME3 as a reference structure. Also two compounds with modest (**8**) or no activity (**58**) were modeled for comparison purposes.

The entire library, with the exception of **58** suggested the formation of covalent bonds between the thiocarbonyl carbon and Cys25 (papain numbering) thiolate group of cruzain (C-S⁻), the active residue in the catalytic triad. Calculated distances between these two groups varied between 3.144 (**36**) and 3.467 Å (**9**). The proximity between both groups promotes the formation of a transient covalent bond. Furthermore, the distance for **58** is 4.59 Å, which could have explained the inefficacy of this analog to inhibit cruzain. The modification of the thiosemicarbazone group might be the reason for this behavior. **1** was modeled in previous work (IC_{50} : 12 nM) and served as a positive control for the construction of the screened library. Contacts provided electrostatic interactions with partial positive charges which also stabilize the oxyanion hole and provided evidence for the formation of the transient tetrahedral intermediate formed between the inhibitor and the active site of cruzain. The series of docked molecules with cruzain are shown as follows: cruzain is in the ribbon diagram, (with the exception of the active site residues that were ball and stick format), and thiosemicarbazones are in space-filling format. Atoms were colored by using a modified CPK color code (carbon: green; oxygen: red; sulfur: yellow; bromine: burgundy; nitrogen: blue; hydrogen: white).

Molecular Docking of 9 with Cruzain

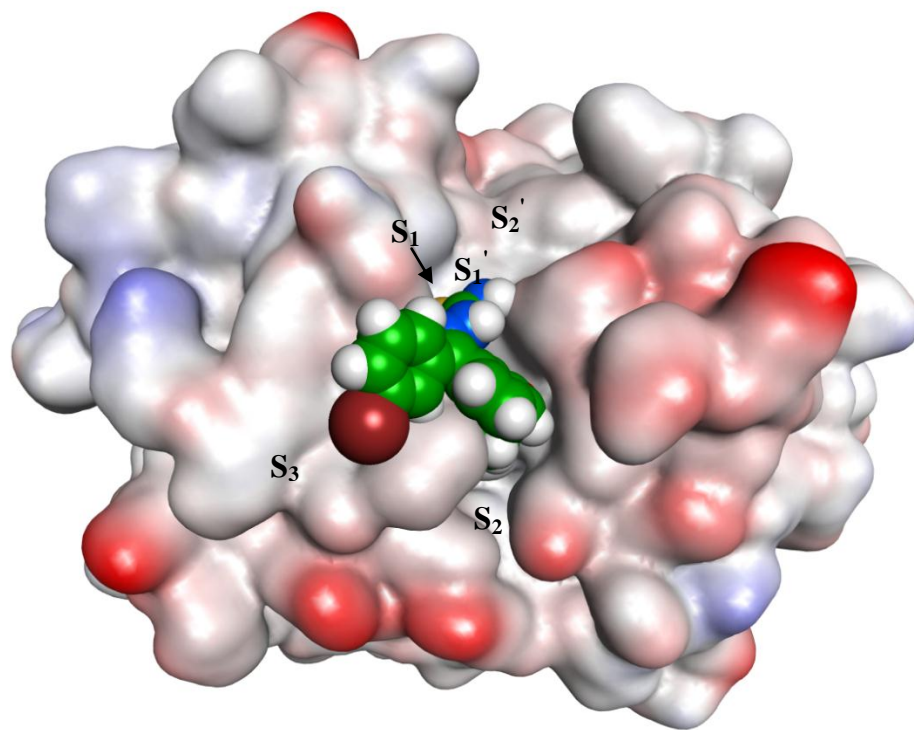
Analog **9**, 3-bromo-3'-acetobenzophenone thiosemicarbazone was the second most potent compound found in the library containing several thiosemicarbazones. Its IC₅₀ value was calculated to be 12.1 nM. The relative interaction energy for the top conformation was calculated to be -62.6 kcal/mol. The top conformation was examined to explore important interactions and possible hydrogen bonds. The thiocarbonyl arm is in the S1 pocket. The functionalized phenyl arm with the acetate group is in the S2 pocket, such that the acetate group is facing down and toward the S2 pocket while the bromophenyl arm is located in the right-hand side of the S3 pocket, so the bromine substituent is pointing out and facing down. Important interactions between **9** and the active site of cruzain are:

1. Hydrogen bond between the backbone carbonyl oxygen of Asp158 with one of the hydrogens of **9** NH₂ group (2.01 Å). This hydrogen bond helps to orient the thiosemicarbazone moiety at the active site.
2. The (C–S) distance is 3.467 Å

Molecular Docking of 10 with Cruzain

Analog **10**, 3-bromo-3',5'-difluorobenzophenone thiosemicarbazone was a potent inhibitor. Its IC₅₀ value was found to be 17.6 nM. The energy interaction for the top conformation was calculated to be -52.8 kcal/mol. The thiocarbonyl arm is in the S1 pocket. The functionalized bromophenyl arm is located deep in the S2 pocket, similar to **9**. Interestingly, the difluorophenyl arm is located along the right-edge of the S3 pocket placing the substituents (fluorines) outwards. Important interactions between **10** and the active site of cruzain are:

A



B

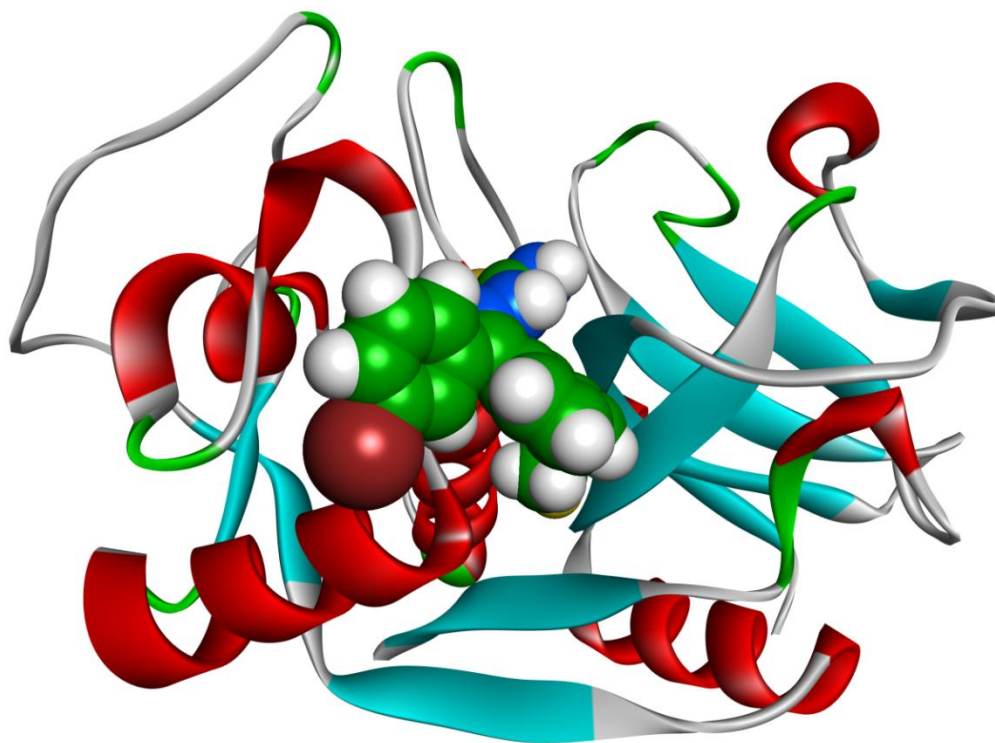


Figure 153. Molecular Docking of **9** with Cruzain. A. Electrostatic Surface of Cruzain Interacting with **9**. B. Ribbon Diagram.

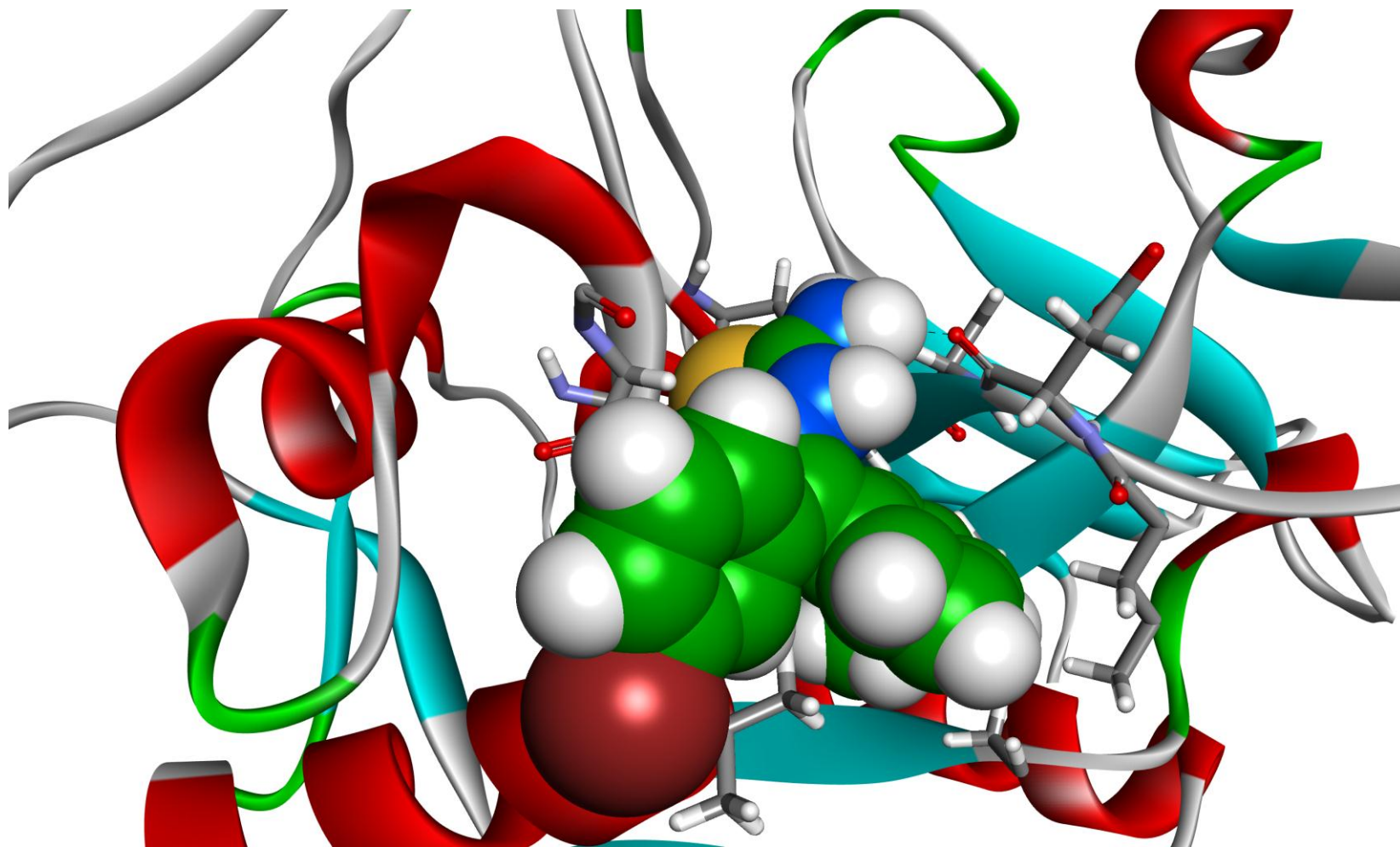


Figure 154. Molecular Docking of Cruzain with 9

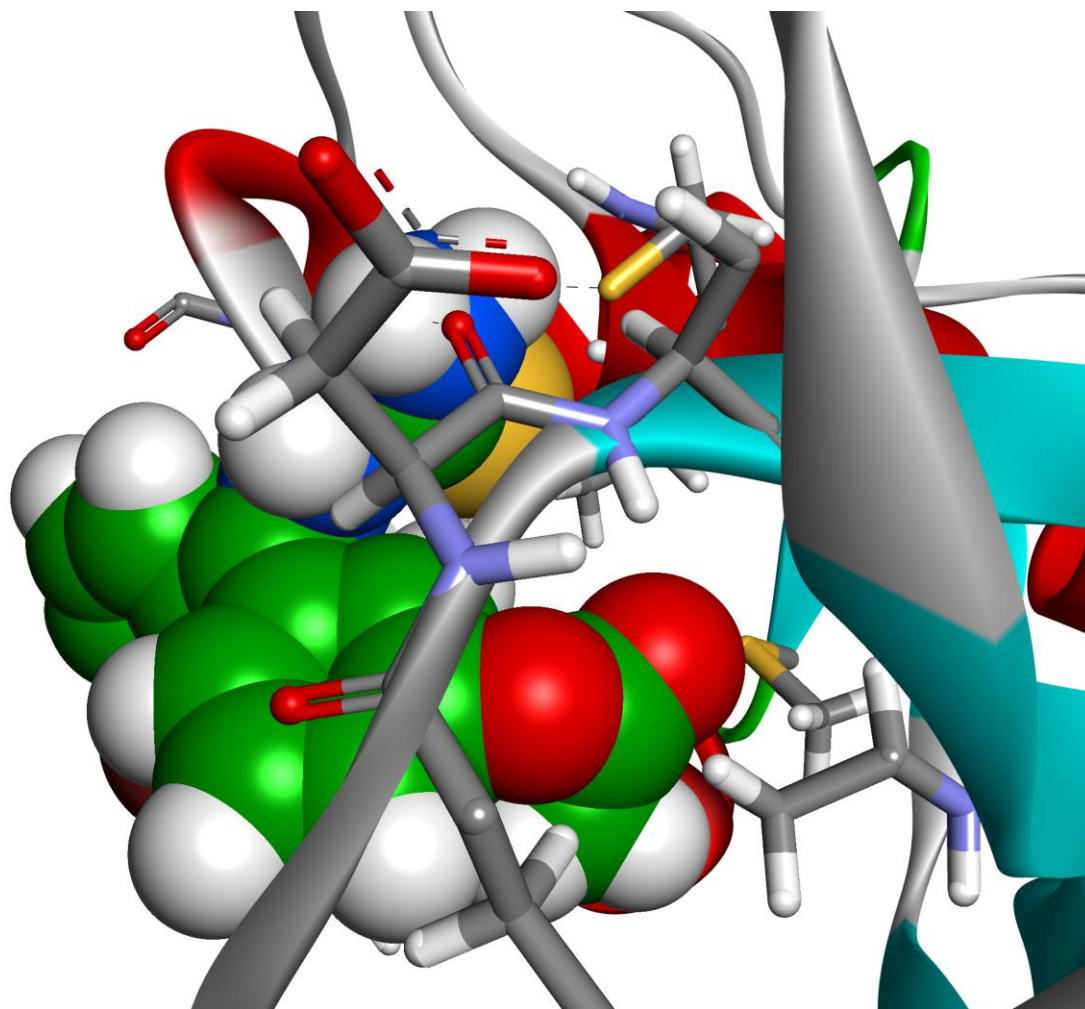


Figure 155. Molecular Docking of **9** with Cruzain. The Thiocarbonyl Carbon and Cys25 (papain numbering) Thiolate Group of Cruzain (C–S) Distance is 3.467 Å

1. The thiocarbonyl sulfur is in close proximity with the side-chain carboxamide hydrogen of Gln19 (2.26 Å)
2. The (C–S) distance is 3.193 Å.

Molecular Docking of 17 with Cruzain

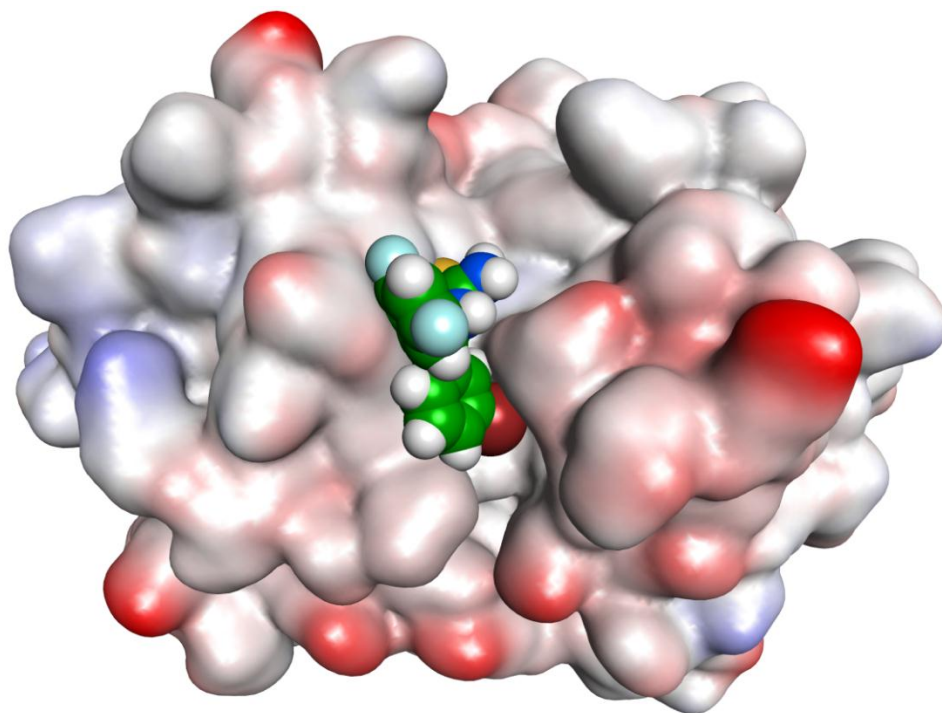
Analog **17**, 3-bromo-4'-fluorobenzophenone thiosemicarbazone was a potent inhibitor with an IC₅₀ value of 14.3 nM. The top energy interaction was -55.0 kcal/mol. The thiocarbonyl arm is in the S1 pocket. The functionalized bromophenyl arm is located deep in the S2 pocket, similar to **9**. Interestingly, the fluorophenyl arm is located along the right-edge of the S3 pocket placing the substituents (fluorines) outwards. Important interactions between **17** and the active site of cruzain are:

1. The thiocarbonyl sulfur is in close proximity with the side-chain carboxamide hydrogen of Gln19 (2.26 Å, see compound **10**)
2. The (C–S) distance is 3.212 Å.

Molecular Docking of 36 with Cruzain

Analog **36**, a dibrominated thiophene thiosemicarbazone had also excellent inhibitory activity towards recombinant cruzain (IC₅₀: 20.2 nM). The top conformation interaction energy was -52.9363kcal/mol. Overall, it was found to have its thiosemicarbazone arm in the right side of the S3 pocket (bromine facing downwards). The thiophene moiety is in the S2 pocket placing the thiophene sulfur outward and the thiophene bromine downwards. The (C–S) distance is 3.144 Å, the shortest distance found in the set of six models. Two hydrogen bonds were found between the backbone carbonyl oxygen of Asp158 and two hydrogens of the thiosemicarbazone moiety of **36**.

A



B

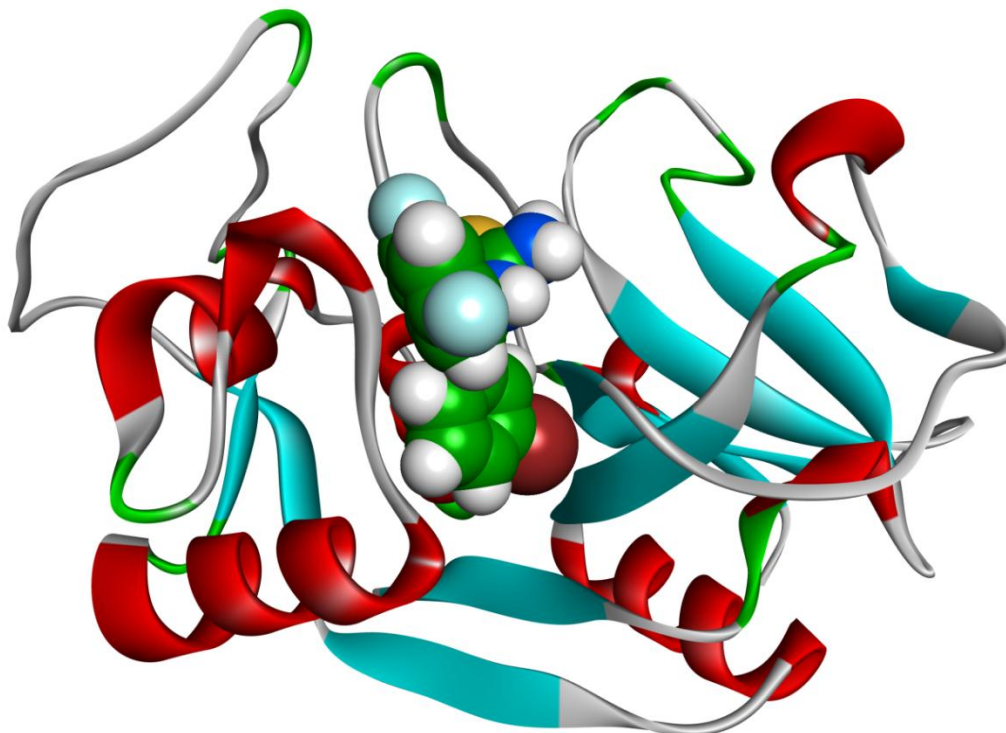


Figure 156. Molecular Docking of **10** with Cruzain. A. Electrostatic Surface of Cruzain Interacting with **10**. B. Ribbon Diagram

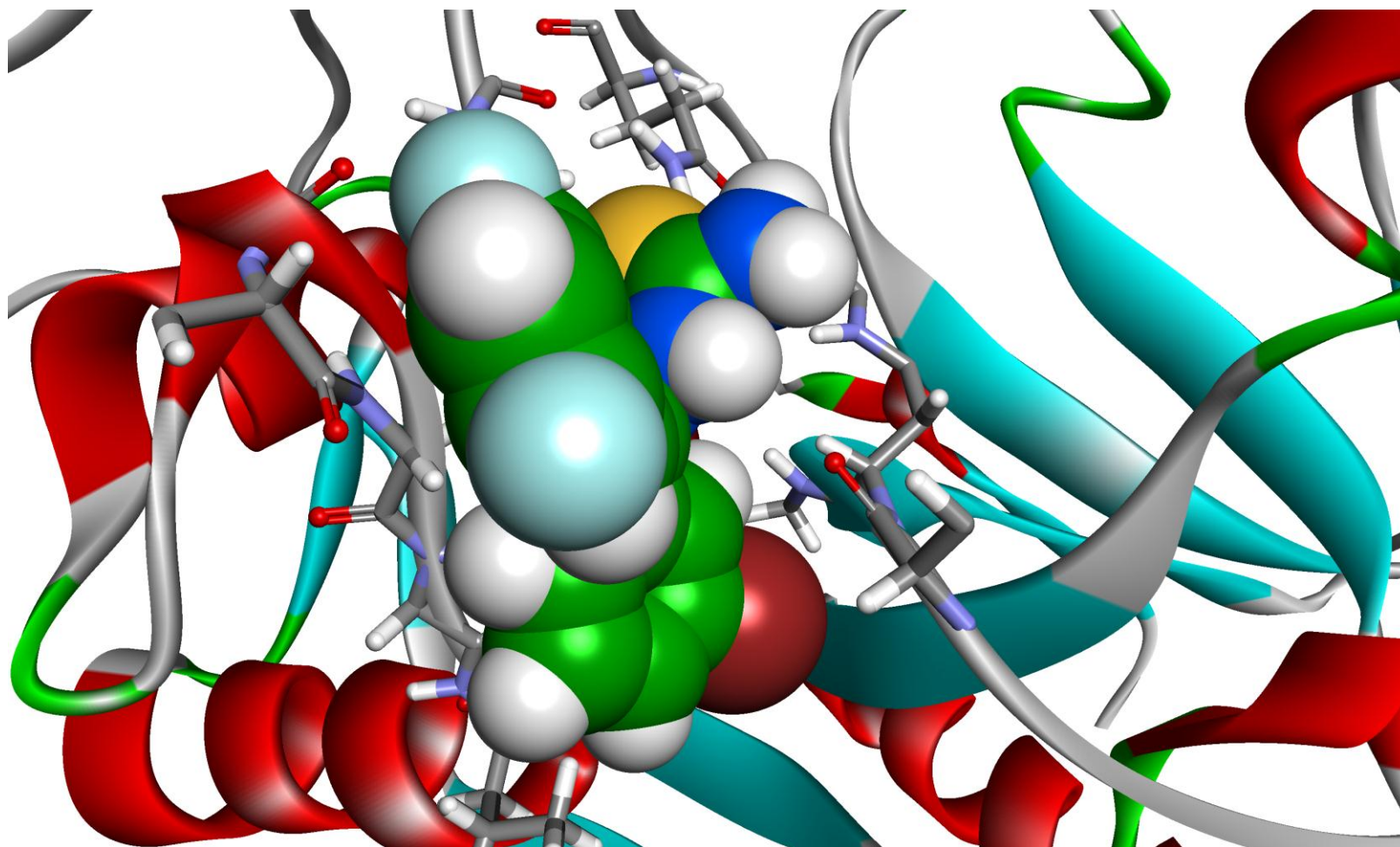


Figure 157. Molecular Docking of Cruzain with **10**

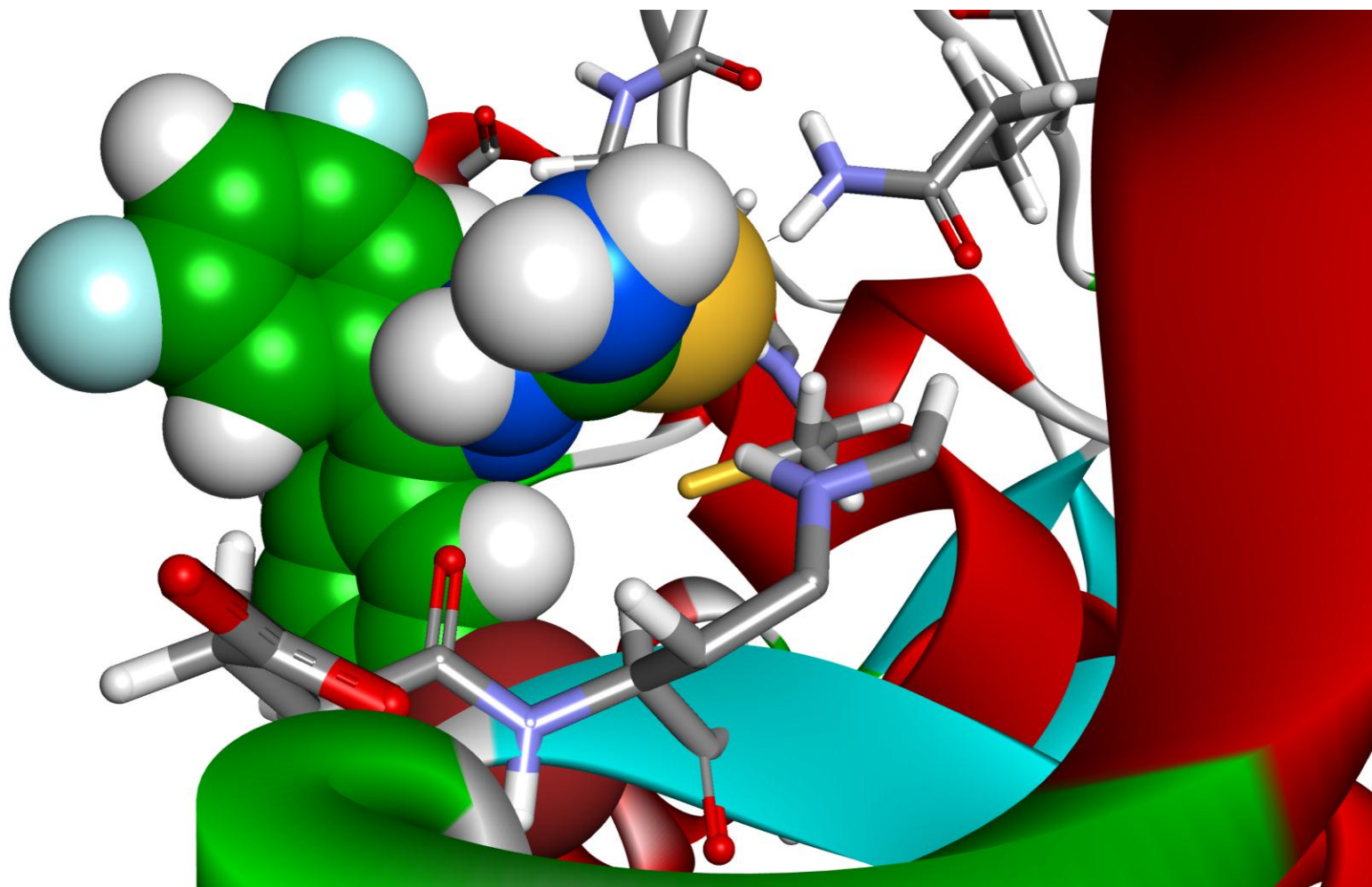
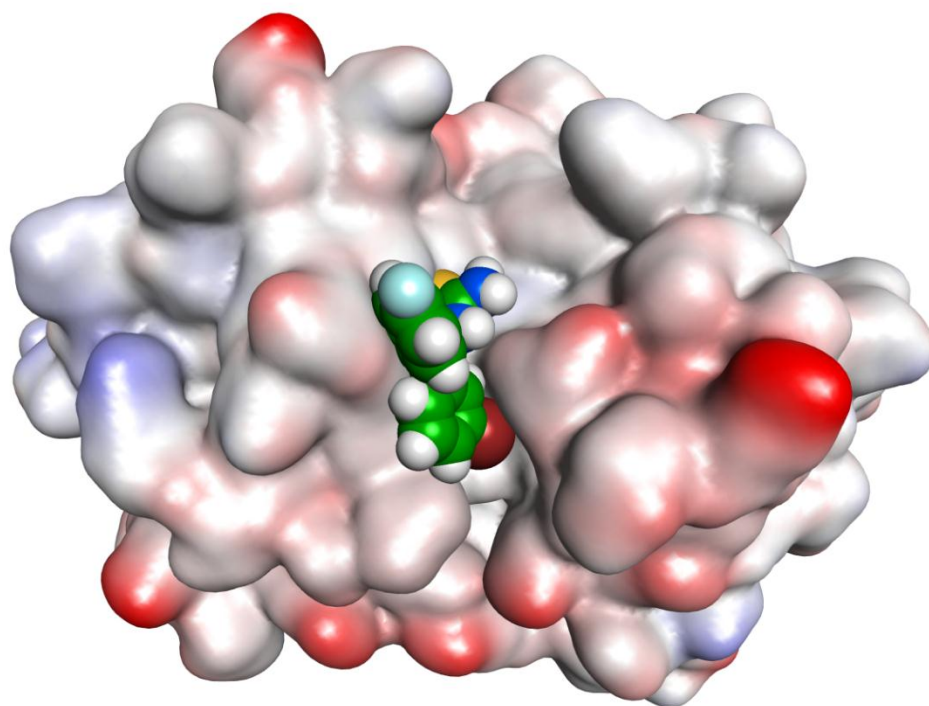


Figure 158. Molecular Docking of **10** with Cruzain. The Thiocarbonyl Carbon and Cys25 (papain numbering) Thiolate Group of Cruzain (C–S) Distance is 3.19 Å

A



B

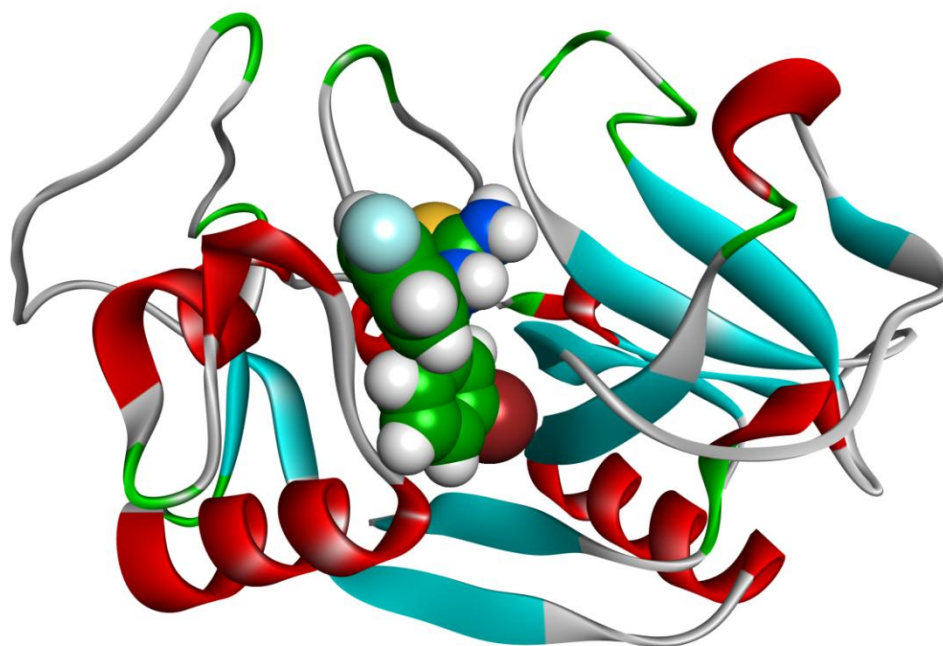


Figure 159. Molecular Docking of **17** with Cruzain. A. Electrostatic Surface of Cruzain Interacting with **17**. B. Ribbon Diagram

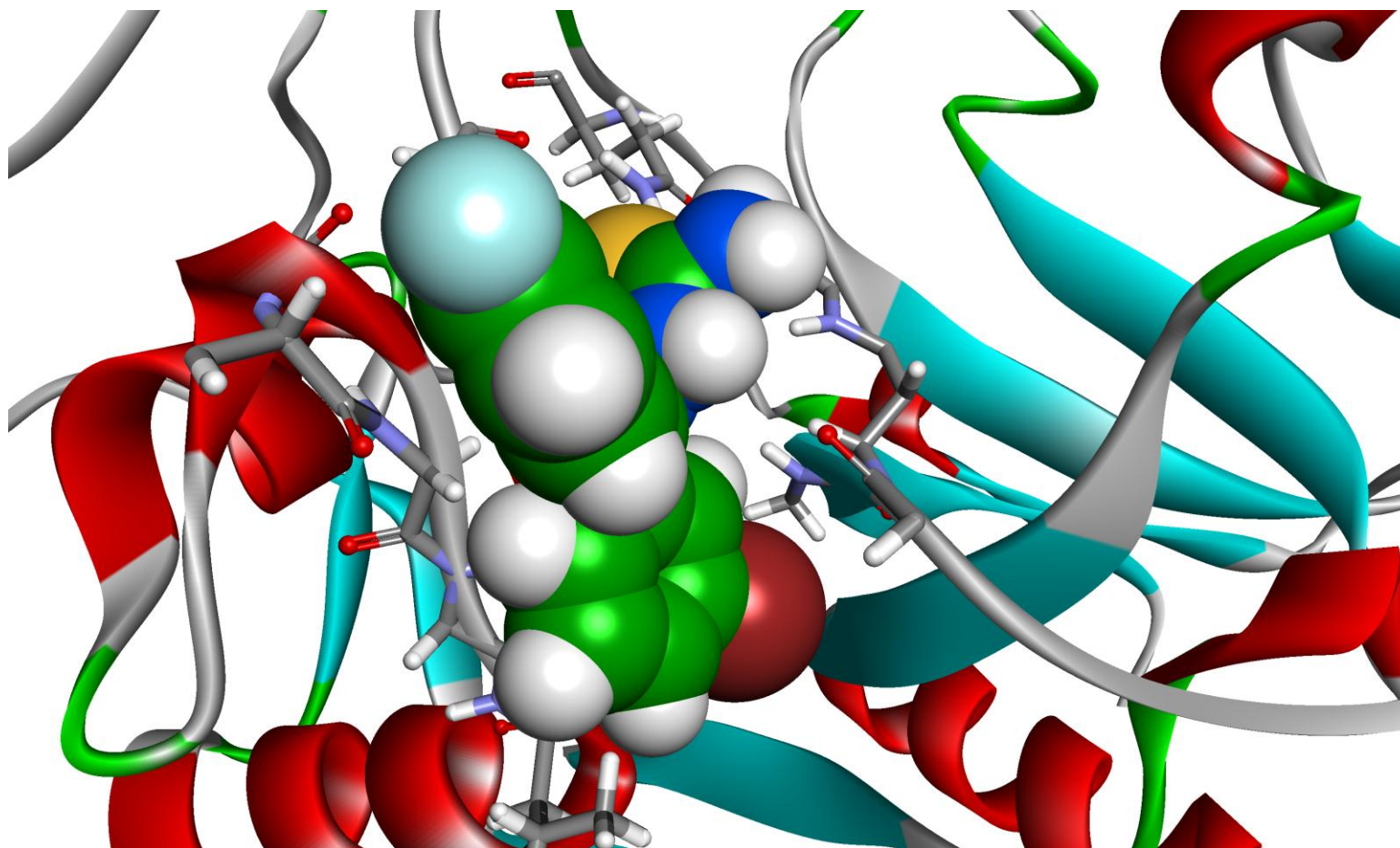


Figure 160. Molecular Docking of Cruzain with **17**

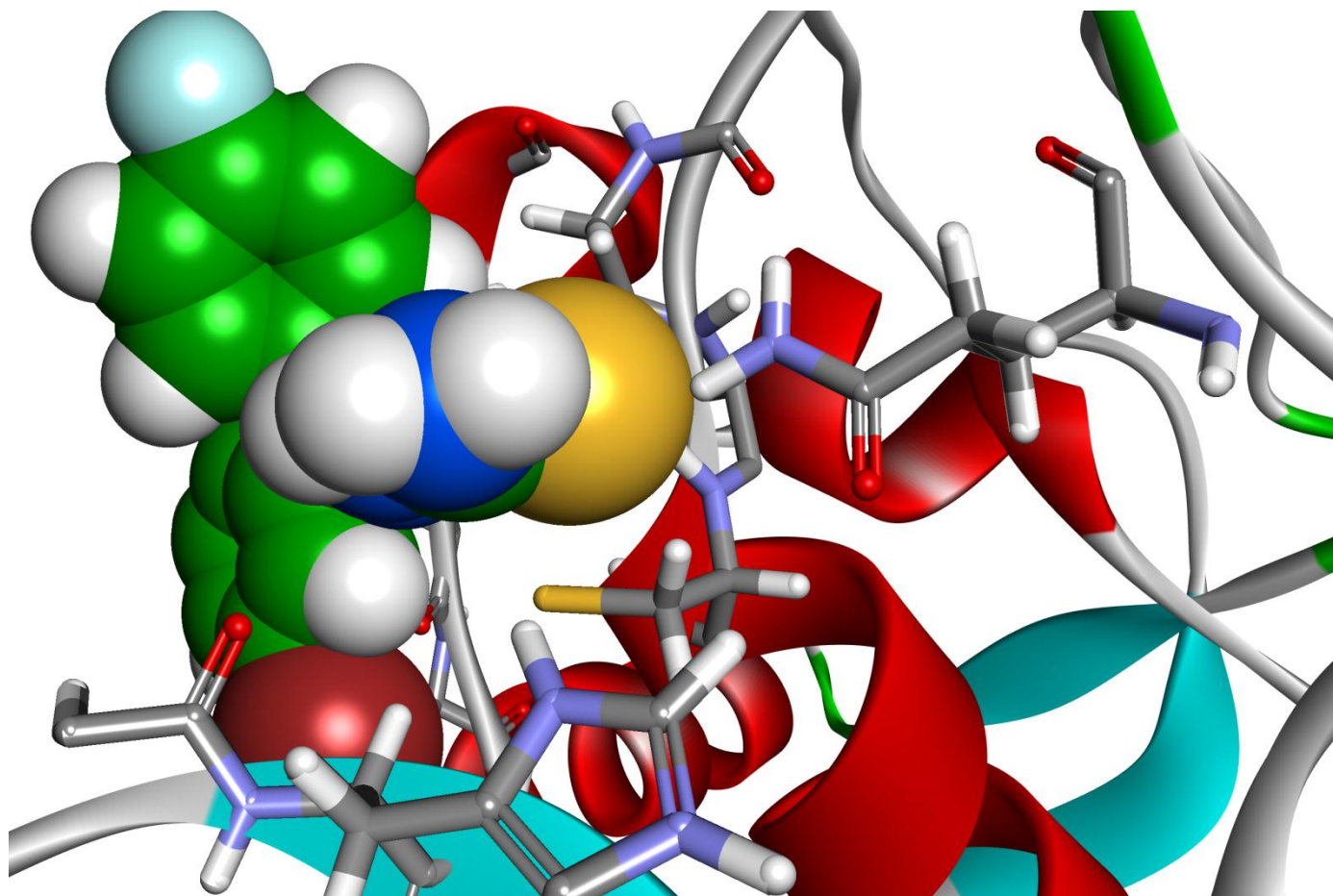


Figure 161. Molecular Docking of **17** with Cruzain. The Thiocarbonyl Carbon and Cys25 (papain numbering) Thiolate Group of Cruzain (C–S) Distance is 3.21 Å

Molecular Docking of 58 with Cruzain

Finally, **58**, an N-phenylated dibromobenzophenone analog with no activity was also modeled. Both bromophenyl arms are located in the S1 and S3 pockets having their substituents facing outward. The phenylated thiosemicarbazone is located in the S2 pocket and in an orientation that does not promote formation of a transient covalent bond. The (C–S) distance is: 4.597, which is considerably much larger than the distance found for the other compounds that were docked with cruzain. The lack of hydrogen bonds in the model, which are very important in orient the compound at the active site, might explain the poor activity of this compound. This model also confirmed the importance of the thiosemicarbazone group in the inhibitory activity of these nonpeptidic compounds.

Conclusions

A series of benzophenone and other functionalized thiosemicarbazones were analyzed as potential inhibitors of cruzain. Five compounds, including a nonbenzophenone thiosemicarbazone showed excellent inhibitory activity toward cruzain with IC₅₀ values less than 20 nM.

Figure 168 shows the Structure-Activity Relationship of these functionalized benzophenone thiosemicarbazones based on their chemical structures. Advanced kinetics were used to investigate the mechanism of four of these compounds (**1**, **9**, **10**, **17**). These compounds were determined to be time dependent, reversible inhibitors of cruzain.

Analog **1** was a competitive inhibitor by examining the effect of substrate concentration on IC₅₀ values. Similarly, **17** was also a competitive inhibitors of the fluorogenic substrate Z-FR-AMC. The decrease in k_{obs} values as a function of Z-FR-

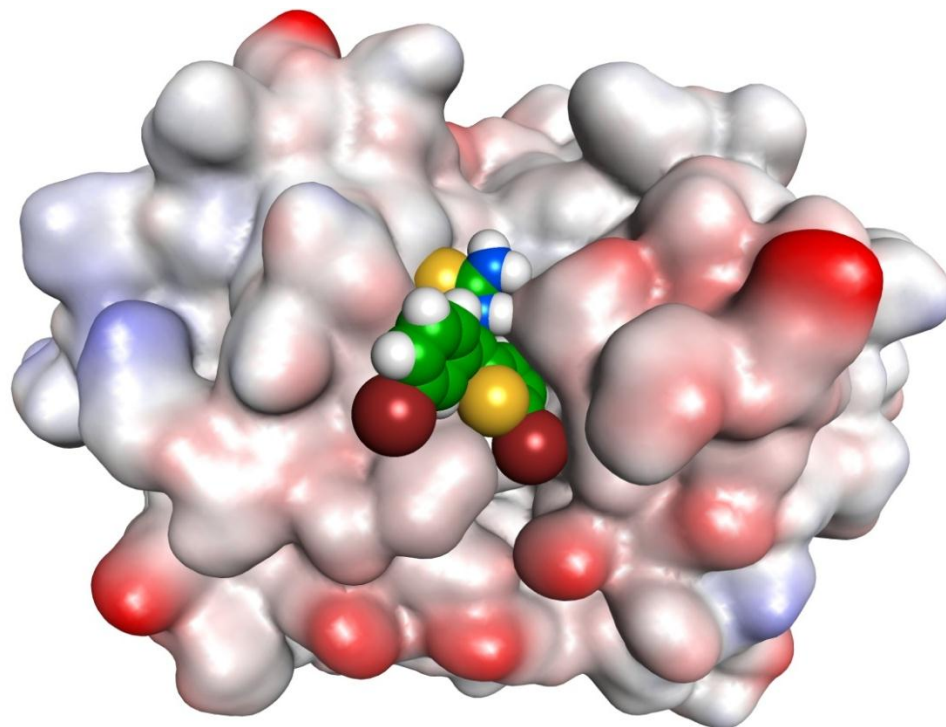
AMC (substrate) concentration in cruzain progress curves indicates that **17** competes with the substrate for binding to the enzyme active site.

The inhibition of type I collagenase activity of recombinant cruzain was examined by using a natural substrate for the enzyme, type I collagen from human skin. The proteolytic activity of cruzain was inhibited by 3-bromo-4'-fluorobenzophenone thiosemicarbazone.

Finally, six of these analogs were docked with cruzain using computational software. Modification of the thiosemicarbazone is detrimental in the activity of these compounds. Molecular docking with analog **58**, N-phenyl-3,3'-dibromobenzophenone thiosemicarbazone ($IC_{50} \geq 10000$ nM), indicated that the bulky phenyl group was located in the S2 pocket and the thiosemicarbazone was not the correct orientation for formation of a covalent bond with the enzyme. The lack of hydrogen bonds may explain the inactivity of these compounds.

For the most potent inhibitors, molecular docking with cruzain showed that thiosemicarbazones are placed at the S1 subsite in close proximity to cruzain Cys25 and in the correct orientation for a chemical reaction.

A.



B.

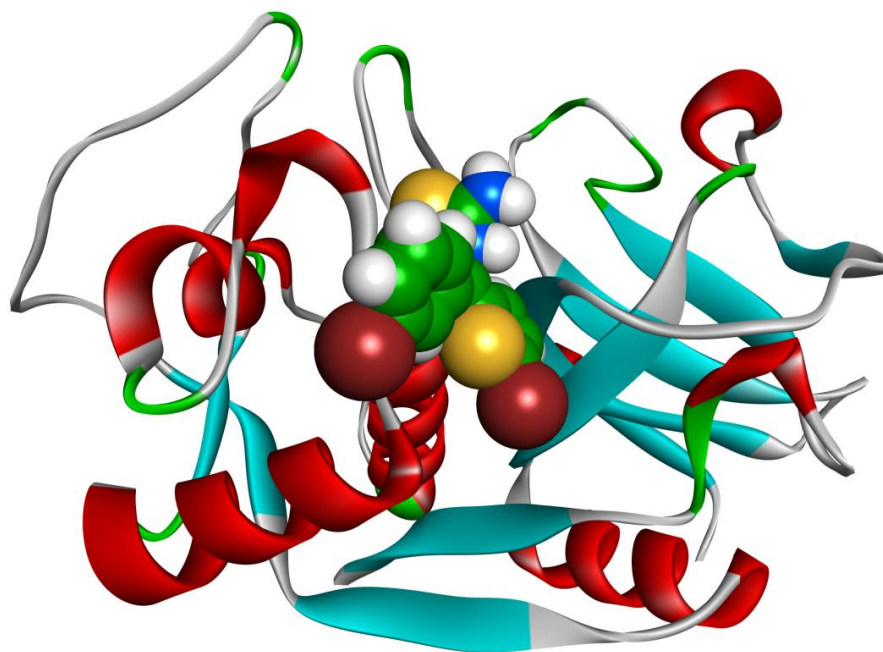


Figure 162. Molecular Docking of **36** with Cruzain. A. Electrostatic Surface of Cruzain Interacting with **36**. B. Ribbon Diagram

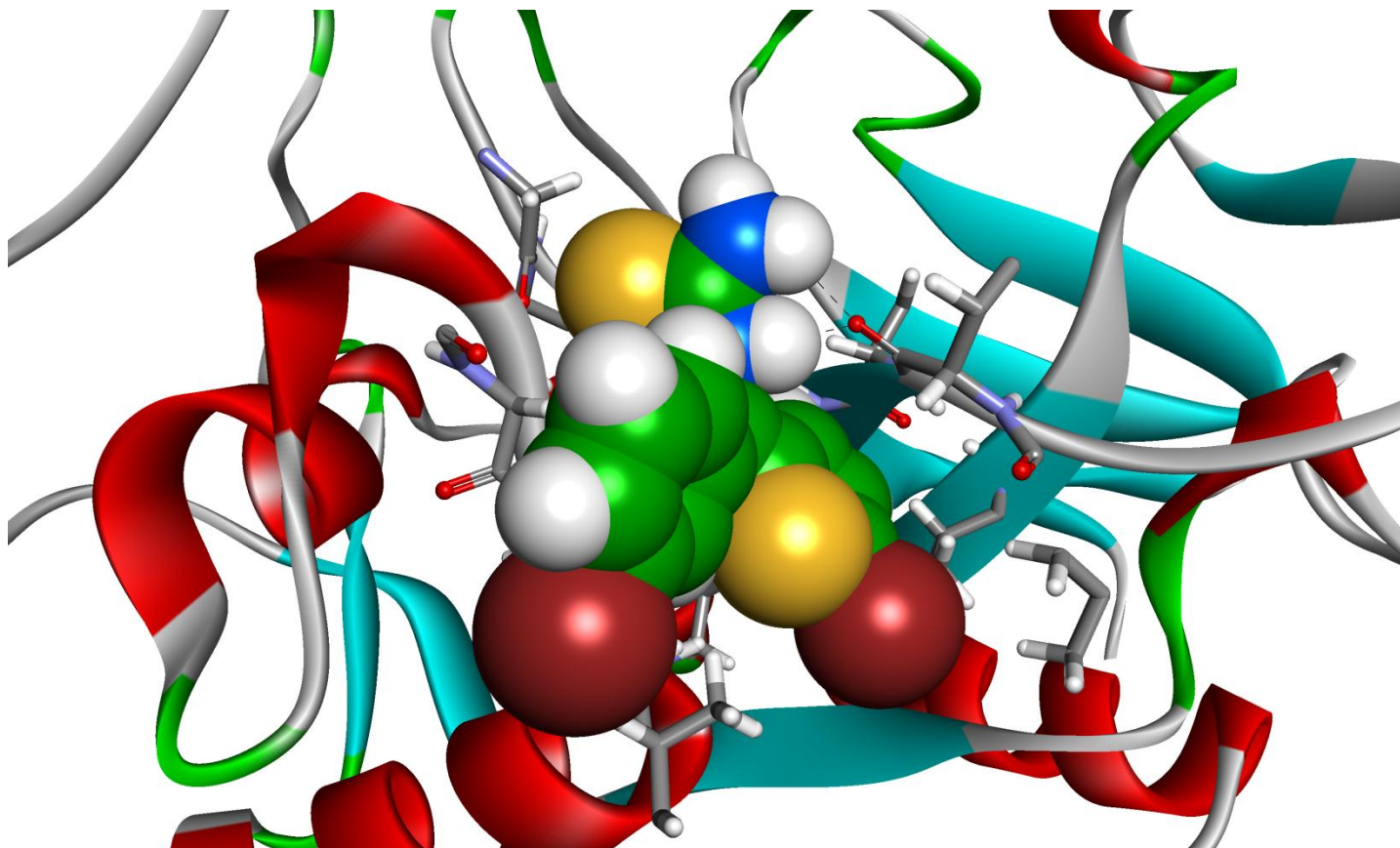


Figure 163. Molecular Docking of Cruzain with **36**

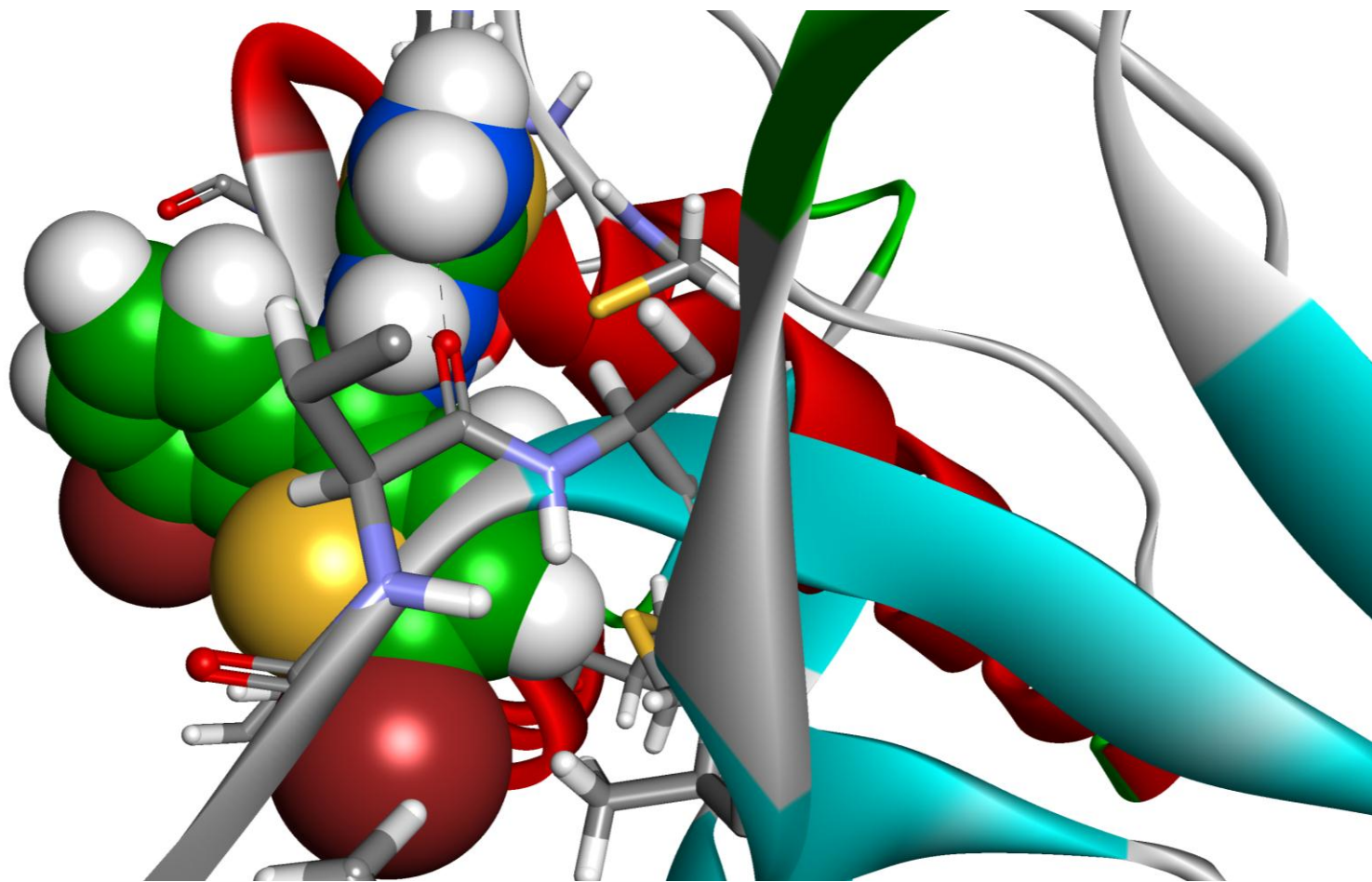
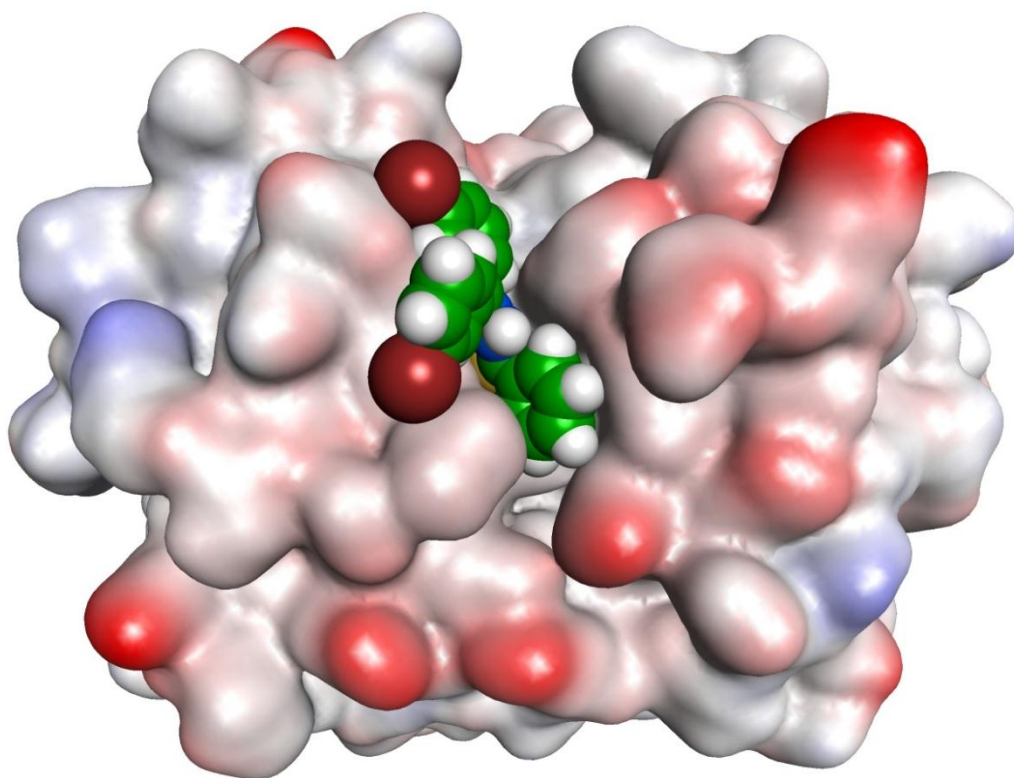


Figure 164. Molecular Docking of **36** with Cruzain. The Thiocarbonyl Carbon and Cys25 (papain numbering) Thiolate Group of Cruzain (C–S) Distance is 3.144 Å

A



B

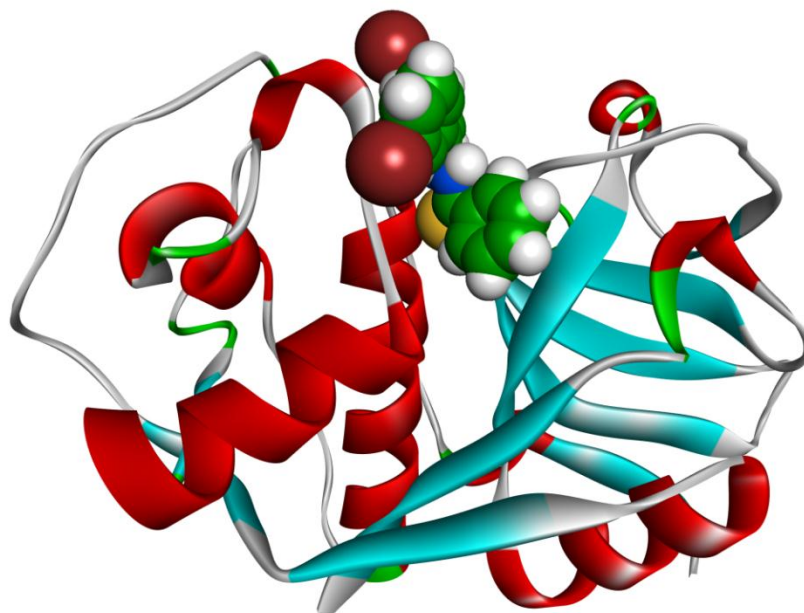


Figure 165. Molecular Docking of **58** with Cruzain. A. Electrostatic Surface of Cruzain Interacting with **58**. B. Ribbon Diagram

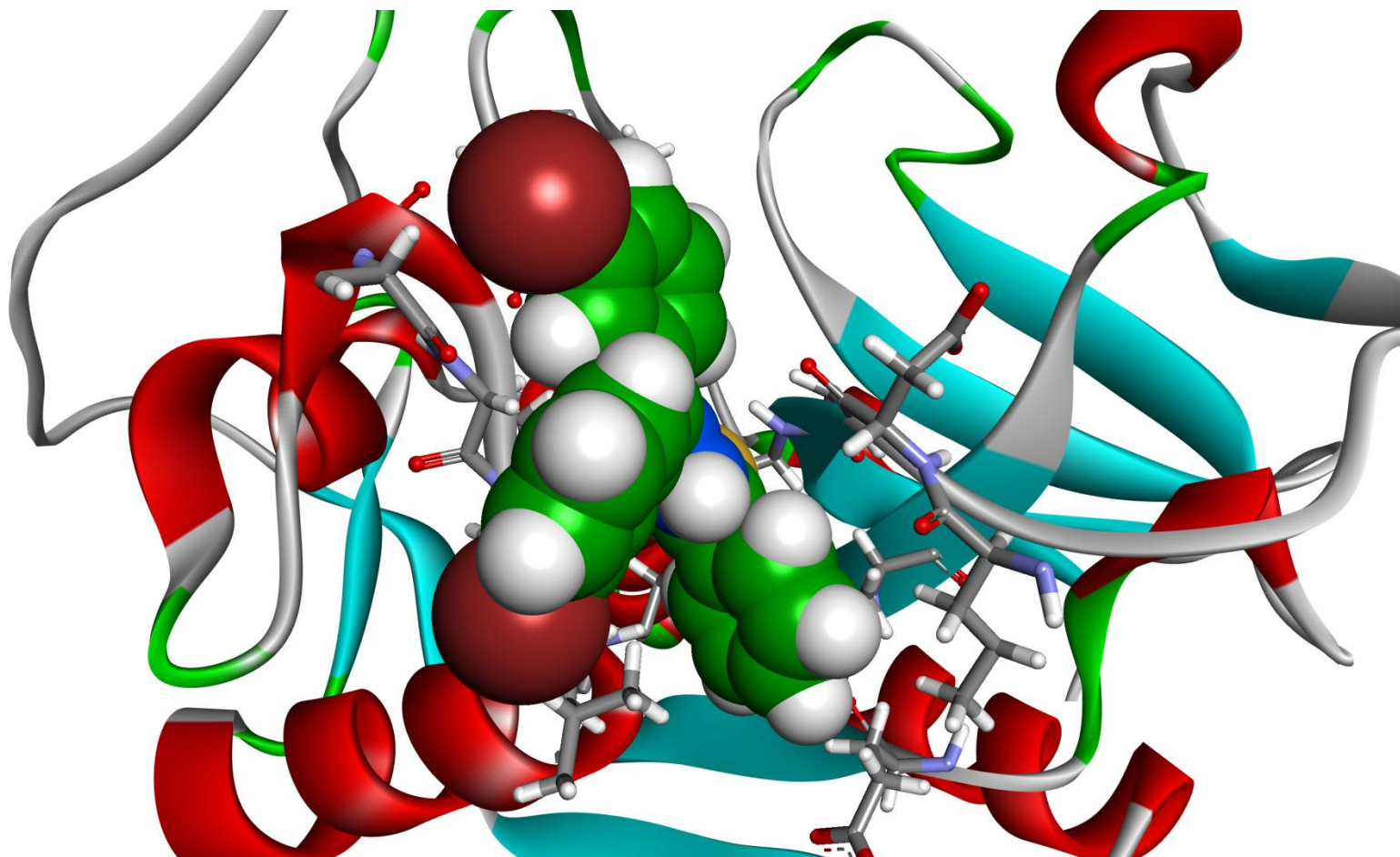


Figure 166. Molecular Docking of Cruzain with **58**. No Hydrogen Bonds were Formed Between **58** and Cruzain Active Site

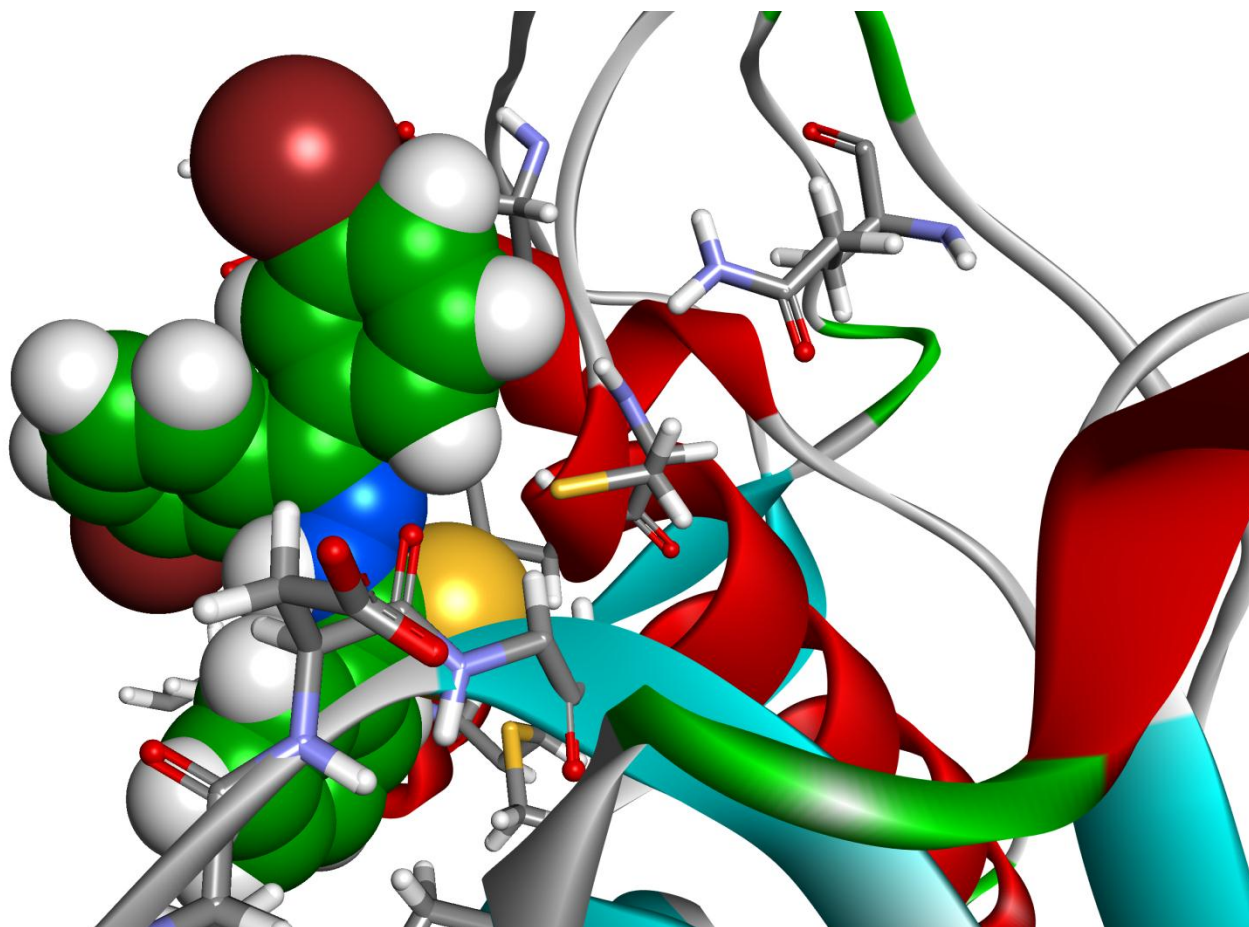


Figure 167. Molecular Docking of **58** with Cruzain. The Thiocarbonyl Carbon and Cys25 (papain numbering) Thiolate Group of Cruzain (C–S) Distance is 4.597Å

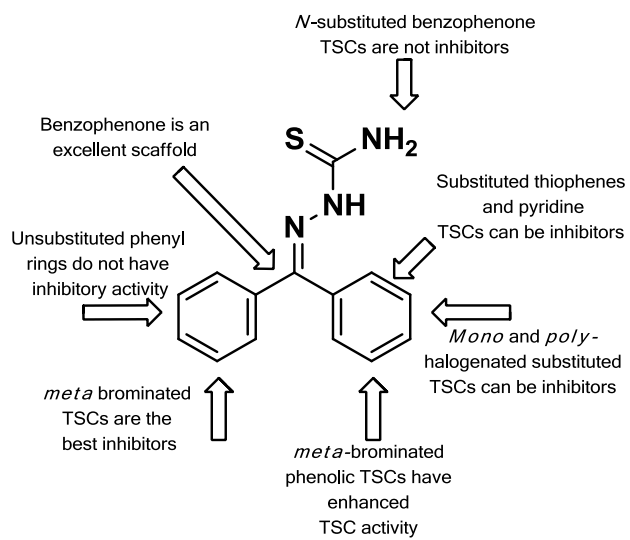


Figure 168. Structure-Activity Relationship for Functionalized Benzophenone TSCs as Cruzain Inhibitors

CHAPTER FIVE

Conclusions and Future Directions

Small, nonpeptidic thiosemicarbazones were tested as inhibitors of three related cysteine proteases, human cathepsins L and K, and cruzain, from *Trypanosoma cruzi*.

Seven compounds were found to be excellent inhibitors of human cathepsin L. Three analogs belong to the subfamily of *metabrominated* benzophenone TSC analogs: 3-bromo-2'-fluorobenzophenone thiosemicarbazone (**14**; IC₅₀: 24 nM), 3-bromo-3'-dibromobenzophenone thiosemicarbazone (**1**; IC₅₀: 17 nM) and 3-bromo-3'-trifluoromethyl benzophenone thiosemicarbazone (**6**; IC₅₀: 47 nM) have all halogenated moieties as substituents in this subfamily. The second subgroup consists of three unsubstituted and substituted benzoyl benzophenone thiosemicarbazones: unsubstituted benzoyl benzophenone thiosemicarbazone (**156**; IC₅₀: 9.9 nM); 5-(4-fluorobenzoyl)-4'-fluorobenzophenone thiosemicarbazone (**157**; IC₅₀: 24.3 nM); and 5-(2-fluorobenzoyl)-3-bromo-2'-fluorobenzophenone thiosemicarbazone (**168**; IC₅₀: 8.1 nM). The best two compounds found in the family of TSC are benzoyl-benzophenone TSC with IC₅₀ values less than 10 nM. Compound **8**, 3-bromo-3'-hydroxybenzophenone thiosemicarbazone (**14**; IC₅₀: 189 nM) was further examined due to the possibility of derivatization and better solubility properties.

Compounds **1** (IC₅₀: 17 nM) and **8** (IC₅₀: 189 nM), functionalized benzophenone thiosemicarbazones were lead inhibitors of cathepsin L, and were evaluated to characterize their mode of action and assess their performance and efficacy in a series of *in vitro* studies. Kinetic analysis was used to investigate the mechanism and mode of

inhibition of these compounds. Compounds **1** and **8** were determined to be slow, time-dependent, reversible, competitive inhibitor of the fluorogenic substrate, Z-FR-AMC. Additionally the mechanism of **8**, 3-bromo-3'-hydroxybenzophenone thiosemicarbazone, was determined to be a slow, simple reversible inhibition but substrate dependent. However, at lower concentrations of the substrate, the kinetics suggested a more complex two-step mechanism of inhibition. Molecular docking of **8** with cathepsin L suggests that it binds in a manner that positions the inhibitor thiocarbonyl moiety for attack by the enzyme Cys25 thiolate to form, at least, a transient tetrahedral intermediate. The kinetic and modeling studies support this proposed mechanism. Compound **8** was also able to significantly inhibit the activity of cathepsin L toward one of its natural substrates, type I collagen from human skin. A selected group of cathepsin L inhibitors (**1**, **8**, **156**, **157**, **168**) displayed potent activity in cell studies by inhibiting invasion and migration of the breast cancer MDA-MB-231 cell line.

Future studies involving these compounds include a complete kinetic characterization of two benzoylbenzophenone thiosemicarbazones (**156**, and **168**) with IC₅₀ values less than 10 nM as cathepsin L inhibitors. The presence of several members of the cathepsin family, including cathepsins K, S, V and F, and cathepsin L-like enzymes in human tumors needs to be considered when testing effective cathepsin L inhibitors in *in vitro* studies. The lack of selective, yet potent inhibitors for each cathepsin and limited *in vitro* cell invasion and cell migration models are the biggest obstacles when confirming effective and selective inhibition of cathepsin L by these novel nonpeptidic thiosemicarbazones. The usage of additional cell lines, like prostate, melanoma and

ovarian cancer cell lines might help to elucidate effective treatment for each one of these diseases.

3-3'-Dibromobenzophenone thiosemicarbazone (**1**) was an excellent inhibitor of human cathepsin K (IC_{50} : 35.2 nM). Advanced kinetics studies were used to investigate the mechanism of **1**. This compound was determined to be a slow, reversible, competitive inhibitor of the protease, similar to cathepsin L. Fluorescent Western blotting demonstrated that **1** was able to inhibit the cleavage and activation of human procathepsin K, and important consideration for use as an antimetastatic agent. The expression of cathepsin K as a proenzyme is well known, yet their mechanism of activation (cleavage of the propeptide) is poorly understood under physiological conditions. The acidic microenvironment of tumors and bone resorption cycles promotes the cleavage of overexpressed procathepsin K, which is vital in those processes. Targeting the inhibition of the activation or the expression of procathepsin K might emerge as a possible alternative for the treatment of osteoarthritis and bone metastasis due to the activity of cathepsin K.

Lastly, **1** was able to inhibit the powerful collagenase and proteolytic activity of cathepsin K when using type IV collagen, a natural substrate of this cysteine protease. Tubulin, from calf brain was susceptible to degradation by cathepsin K and compound **1** impeded the proteolytic degradation of this alternative substrate.

Interestingly, the majority of the potent compounds do not exhibit a clear selectivity toward cathepsin L or K, perhaps due to the similarities in the active sites of both cysteine proteases. Three compounds with a high degree of selectivity were found in the library of thiosemicarbazones and could be used as promising agents of cathepsins

L or K. Nevertheless, subtle differences in cathepsins L and K *in vitro* assays might explain the difference in activities toward both proteases. A more in depth screening using alternative substrates (specific substrates) might help to elucidate their activities.

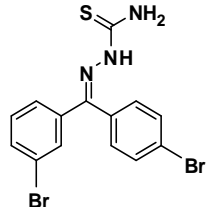
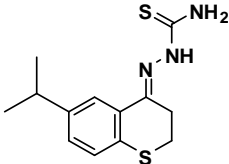
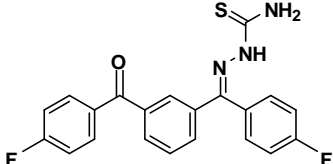
IC ₅₀ (nM)			
	49	123	157
Cathepsin L	≥ 10000	≥ 10000	50
Cathepsin K	49.1	21.3	8500

Figure 169. Chemical Structures of Selective Inhibitors of Human Cathepsins L and K

Four functionalized thiosemicarbazones derivatives (**1**, **9**, **10**, and **17**) were found to be time-dependent inhibitors of recombinant cruzain. Compounds **1**, **9**, and **17** were slow reversible, competitive inhibitors of cruzain. Their proposed mode of inhibition is via transient covalent modification due to the interactions between the thiocarbonyl carbon of the thiosemicarbazone moiety and the thiolate group of cysteine 25, the key residue in the activity of these cysteine proteases. This mechanism is supported by advanced kinetic studies using **17** and molecular docking using the aforementioned inhibitors and cruzain. It was also demonstrated that the presence of the thiosemicarbazone moiety in these molecules is vital for their inhibitory potency. Modification of the functional group impedes both covalent and hydrogen-bond interactions between the inhibitor and cruzain's active site. Lastly, the collagenase activity of cruzain was reduced by the presence of **17**, a dihalogenated benzophenone thiosemicarbazone.

These preliminary studies have contributed to understanding the characteristics of thiosemicarbazones as potential anticancer and/or anti-Chagas' disease agents.

In summary, advanced kinetic studies and molecular modeling provide evidence for the proposal that the best thiosemicarbazone inhibitors of each enzyme form at least a transient covalent bond with the active site Cys25 (Figure 170).

The possibility that thiocarbamoylated cruzain, cathepsin L or cathepsin K is formed followed by slow hydrolysis cannot be ruled out, and future studies using mass spectrometry techniques could be addressed to elucidate this mechanism.

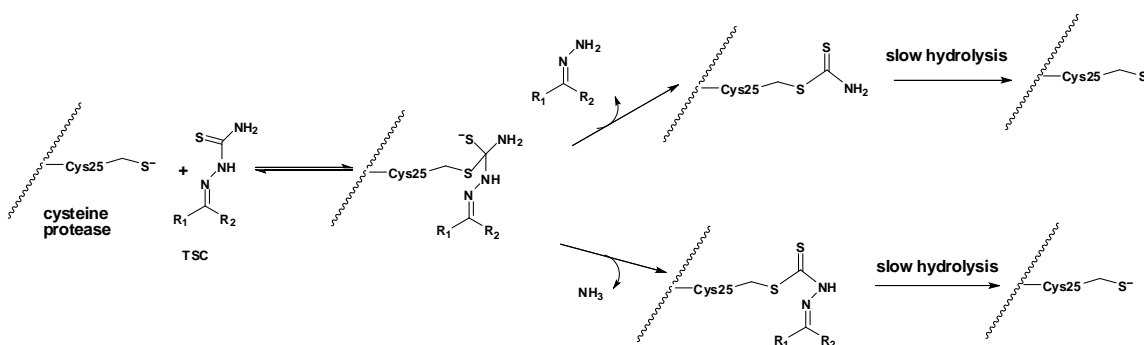
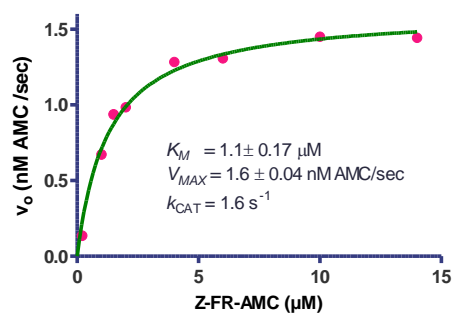


Figure 170. Proposed mechanism of Cysteine Protease Inhibition by Thiosemicarbazone Derivatives

APPENDICES

APPENDIX A

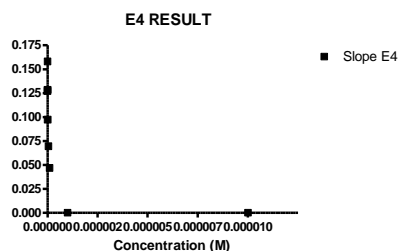
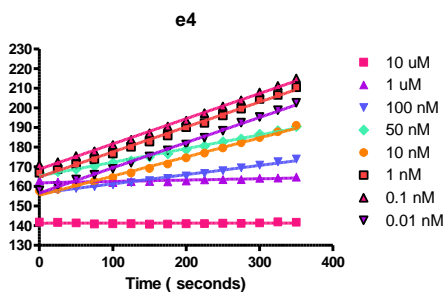
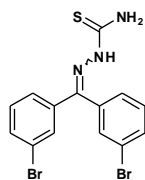
In Vitro Evaluations of Thiosemicarbazones as Inhibitors of Human Cathepsin L



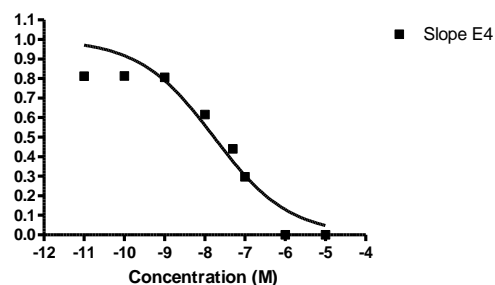
Substrate Concentration (uM)	2 A
14.00000	1.443
10.00000	1.450
6.00000	1.307
4.00000	1.283
2.00000	0.983
1.50000	0.937
1.00000	0.671
0.20000	0.135

	2 A
Michaelis-Menten	
Best-fit values	
Vmax	1.618
Km	1.278
Std. Error	
Vmax	0.05065
Km	0.1511
95% Confidence Intervals	
Vmax	1.494 to 1.742
Km	0.9086 to 1.648
Goodness of Fit	
Degrees of Freedom	6
R square	0.9871
Absolute Sum of Squares	0.01838
Sy.x	0.05535
Constraints	
Km	Km > 0.0
Number of points	
Analyzed	8

A.1 Calculation of Michaelis-Menten Constant for Cathepsin L using Z-FR-AMC as a Substrate



Transform of E4 RESULT



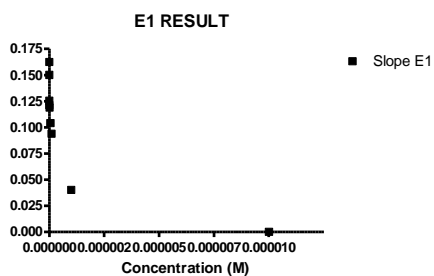
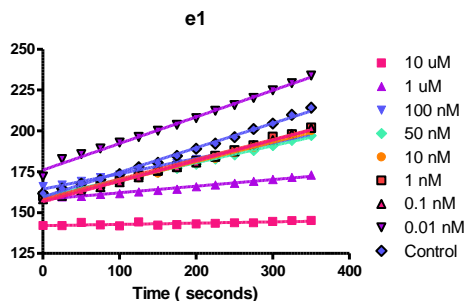
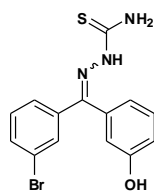
Concentration (M)	Slope E4
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.046930
5.000000e-008	0.069420
1.000000e-008	0.097350
1.000000e-009	0.127300
1.000000e-010	0.128500
1.000000e-011	0.128400

Concentration (M)	Slope E4
-5.000	0.000
-6.000	0.000
-7.000	0.297
-7.301	0.439
-8.000	0.616
-9.000	0.805
-10.000	0.813
-11.000	0.812

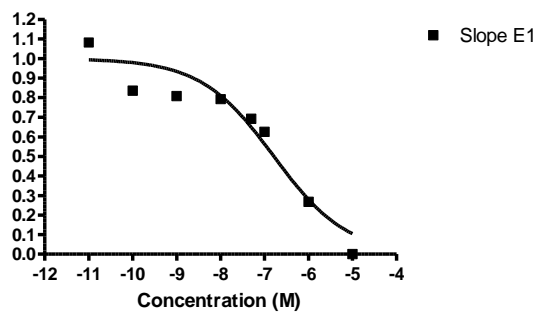
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.771
HillSlope	-0.4669
EC50	1.692e-008
Std. Error	
LogEC50	0.2083
HillSlope	0.1056
95% Confidence Intervals	
LogEC50	-8.281 to -7.262
HillSlope	-0.7253 to -0.2086
EC50	5.234e-009 to 5.473e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9263
Absolute Sum of Squares	0.06201
Sy.x	0.1017
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

A.2 Representative IC₅₀ Calculation Using **1** as a Cathepsin L Inhibitor

GDK-II-98



Transform of E1 RESULT



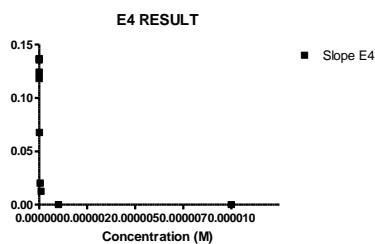
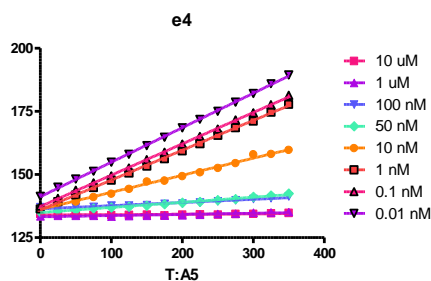
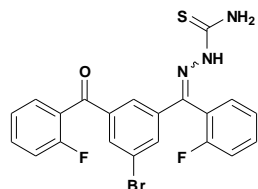
Concentration (M)	Slope E1
1.00e-005	0.000000
1.00e-006	0.040220
1.00e-007	0.093860
5.00e-008	0.104000
1.00e-008	0.118900
1.00e-009	0.121300
1.00e-010	0.125400
1.00e-011	0.162500

Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.268
-7.000	0.625
-7.301	0.693
-8.000	0.792
-9.000	0.808
-10.000	0.835
-11.000	1.083

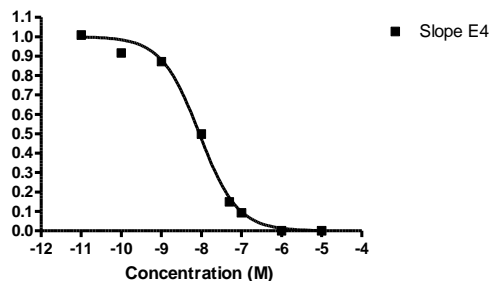
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-6.788
HillSlope	-0.5218
EC50	1.630e-007
Std. Error	
LogEC50	0.1977
HillSlope	0.1258
95% Confidence Intervals	
LogEC50	-7.272 to -6.304
HillSlope	-0.8297 to -0.2139
EC50	5.350e-008 to 4.966e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9262
Absolute Sum of Squares	0.06172
Sy.x	0.1014
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.3 Representative IC₅₀ Calculation using **8** as a Cathepsin L Inhibitor

ENP-I-151



Transform of E4 RESULT

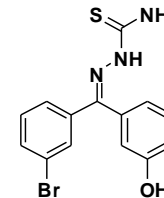
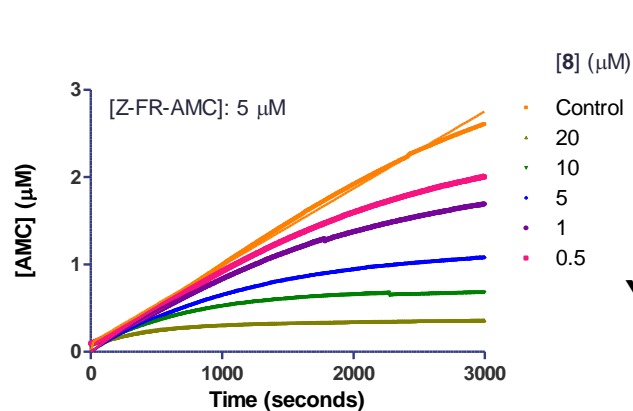


Concentration (M)	Slope E4
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.012590
5.000000e-008	0.020180
1.000000e-008	0.067530
1.000000e-009	0.118100
1.000000e-010	0.124100
1.000000e-011	0.136700

Concentration (M)	Slope E4
-5.000	0.000
-6.000	0.000
-7.000	0.093
-7.301	0.149
-8.000	0.498
-9.000	0.871
-10.000	0.915
-11.000	1.008

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-8.050
HillSlope	-0.9136
EC50	8.915e-009
Std. Error	
LogEC50	0.04992
HillSlope	0.08138
95% Confidence Intervals	
LogEC50	-8.172 to -7.928
HillSlope	-1.113 to -0.7144
EC50	6.729e-009 to 1.181e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9953
Absolute Sum of Squares	0.006209
Sy.x	0.03217
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

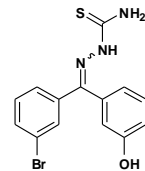
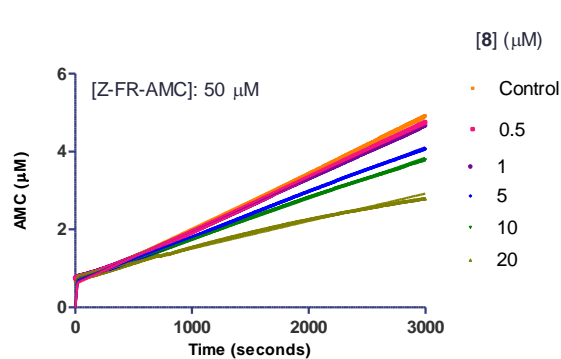
A.4 Representative IC₅₀ Calculation Using **168** as a Cathepsin L Inhibitor



$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	Control	20 uM	10 uM	5 uM	1 uM	0.5 uM
Progress Curve	Ambiguous				Hit constraint	
Best-fit values						
vs	0.0008844	2.732e-005	8.012e-006	4.627e-005	~ 1.888e-016	2.040e-015
vi	~ 0.09269	0.001111	0.001048	0.0009878	0.001053	0.001114
kobs	~ 0.9178	0.003824	0.001566	0.0009347	0.0004653	0.0003643
Std. Error						
vs	1.789e-006	7.660e-007	2.648e-006	5.860e-006		2.403e-006
vi	~ 0.2554	1.200e-005	6.525e-006	4.565e-006	3.956e-006	2.000e-006
kobs	~ 2.555	5.560e-005	2.185e-005	1.764e-005	1.864e-005	1.980e-006
95% Confidence Intervals						
vs	0.0008809 to 0.0008879	2.582e-005 to 2.882e-005	2.821e-006 to 1.320e-005	3.479e-005 to 5.776e-005		0.0 to 4.710e-006
vi	(Very wide)	0.001088 to 0.001135	0.001035 to 0.001061	0.0009789 to 0.0009968	0.001046 to 0.001061	0.001110 to 0.001118
kobs	(Very wide)	0.003715 to 0.003933	0.001523 to 0.001609	0.0009001 to 0.0009693	0.0004288 to 0.0005019	0.0003605 to 0.0003682
Goodness of Fit						
Degrees of Freedom	997	997	997	997	997	997
R square	0.9960	0.9682	0.9909	0.9973	0.9988	0.9989
Absolute Sum of Squares	2.378	0.1373	0.2438	0.2249	0.2790	0.3652
Sy.x	0.04883	0.01173	0.01564	0.01502	0.01673	0.01914
Constraints						
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	1000	1000	1000	1000	1000	1000

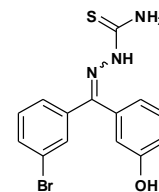
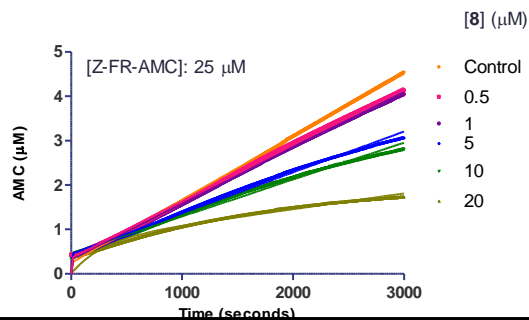
A.5 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 5 µM (Preincubation Time: 0 minutes)



$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	Control	20 µM	10 µM	5 µM	1 µM	0.5 µM
Progress Curve	Ambiguous		Ambiguous	Ambiguous	Ambiguous	Ambiguous
Best-fit values						
vs	0.001423	0.0007067	0.001048	0.001150	0.001340	0.001376
vi	~ 5.266	0.4482	~ 12.38	~ 6.261	~ 5.213	~ 5.182
kobs	~ 8.798	0.5569	~ 17.64	~ 9.394	~ 8.232	~ 8.597
Std. Error						
vs	2.150e-006	2.212e-006	1.169e-006	1.149e-006	1.497e-006	1.668e-006
vi	~ 3.733e+008	0.09774	~ 2.060e+009	~ 4.754e+009	~ 2.460e+008	~ 3.505e+008
kobs	~ 6.239e+008	0.1218	~ 2.936e+009	~ 7.133e+009	~ 3.886e+008	~ 5.817e+008
95% Confidence Intervals						
vs	0.001419 to 0.001428	0.0007024 to 0.0007110	0.001046 to 0.001050	0.001148 to 0.001152	0.001337 to 0.001343	0.001373 to 0.001380
vi	(Very wide)	0.2566 to 0.6398	(Very wide)	(Very wide)	(Very wide)	(Very wide)
kobs	(Very wide)	0.3181 to 0.7956	(Very wide)	(Very wide)	(Very wide)	(Very wide)
Goodness of Fit						
Degrees of Freedom	997	997	997	997	997	997
R square	0.9981	0.9904	0.9988	0.9990	0.9988	0.9987
Absolute Sum of Squares	2.871	3.628	1.018	0.9840	1.672	1.837
Sy.x	0.05366	0.06032	0.03196	0.03142	0.04095	0.04293
Constraints						
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	1000	1000	1000	1000	1000	1000

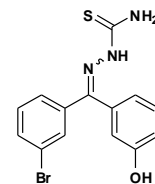
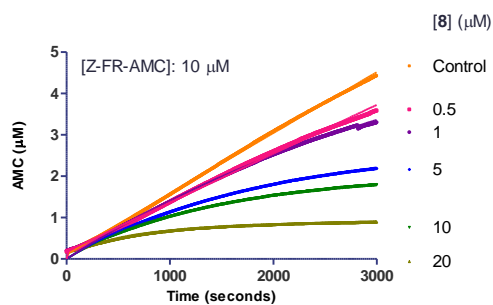
A.6 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 50 µM (Preincubation Time: 0 minutes)



$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	Control	20 uM	10 uM	5 uM	1 uM	0.5 uM
Progress Curve	Ambiguous			Ambiguous	Ambiguous	Ambiguous
Best-fit values						
vs	0.001423	0.0003545	0.0008308	0.0009205	0.001256	0.001283
vi	~ 1.833	0.002994	0.2028	~ 0.2786	~ 2.064	~ 2.577
kobs	~ 7.446	0.003522	0.4363	~ 0.6231	~ 6.293	~ 7.384
Std. Error						
vs	1.634e-006	4.516e-006	2.284e-006	2.013e-006	1.009e-006	1.088e-006
vi	~ 6.361e+007	6.051e-005	0.06619	~ 0.1129	~ 1.372e+006	~ 3.658e+007
kobs	~ 2.586e+008	0.0001091	0.1433	~ 0.2538	~ 4.187e+006	~ 1.049e+008
95% Confidence Intervals						
vs	0.001419 to 0.001426	0.0003456 to 0.0003633	0.0008263 to 0.0008352	0.0009166 to 0.0009244	0.001254 to 0.001258	0.001281 to 0.001285
vi	(Very wide)	0.002876 to 0.003113	0.07309 to 0.3326	(Very wide)	(Very wide)	(Very wide)
kobs	(Very wide)	0.003309 to 0.003736	0.1556 to 0.7171	(Very wide)	(Very wide)	(Very wide)
Goodness of Fit						
Degrees of Freedom	997	997	997	997	997	997
R square	0.9988	0.9734	0.9926	0.9953	0.9994	0.9993
Absolute Sum of Squares	1.845	4.255	3.861	3.006	0.7572	0.8543
Sy.x	0.04302	0.06533	0.06223	0.05491	0.02756	0.02927
Constraints						
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	1000	1000	1000	1000	1000	1000

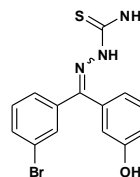
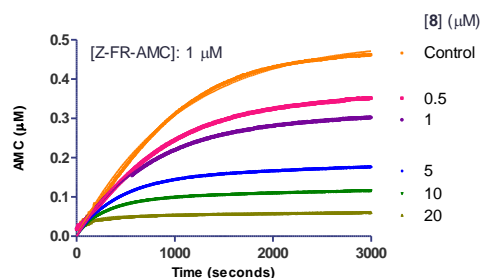
A.7 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 25 μ M (Preincubation Time: 0 minutes)



$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	Control	20 uM	10 uM	5 uM	1 uM	0.5 uM
Progress Curve	Ambiguous					Ambiguous
Best-fit values						
vs	0.001469	8.407e-005	0.0001831	0.0002328	2.674e-014	0.001182
vi	~ 0.8307	0.001703	0.001567	0.001577	0.001591	~ 0.3222
kobs	~ 7.910	0.002457	0.001048	0.0008123	0.0002494	~ 1.800
Std. Error						
vs	1.337e-006	2.671e-006	1.011e-005	1.706e-005	1.705e-005	2.006e-006
vi	~ 1.068e+008	1.739e-005	1.018e-005	9.726e-006	3.808e-006	~ 3.825
kobs	~ 1.019e+009	4.219e-005	2.609e-005	2.747e-005	4.326e-006	~ 21.46
95% Confidence Intervals						
vs	0.001466 to 0.001472	7.884e-005 to 8.931e-005	0.0001633 to 0.0002030	0.0001994 to 0.0002663	0.0 to 3.343e-005	0.001178 to 0.001186
vi	(Very wide)	0.001669 to 0.001737	0.001547 to 0.001587	0.001558 to 0.001596	0.001583 to 0.001598	(Very wide)
kobs	(Very wide)	0.002375 to 0.002540	0.0009972 to 0.001100	0.0007585 to 0.0008662	0.0002409 to 0.0002579	(Very wide)
Goodness of Fit						
Degrees of Freedom	997	997	997	997	997	997
R square	0.9993	0.9793	0.9956	0.9967	0.9982	0.9972
Absolute Sum of Squares	1.063	0.7811	0.9939	1.161	1.663	2.991
Sy.x	0.03265	0.02799	0.03157	0.03412	0.04084	0.05477
Constraints						
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	1000	1000	1000	1000	1000	1000

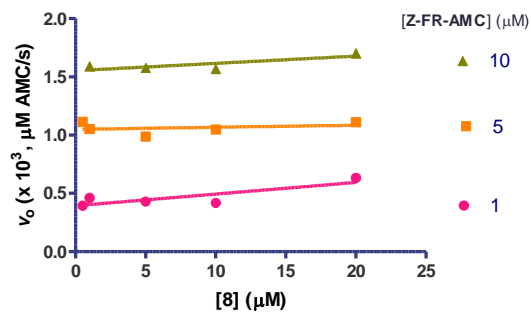
A.8 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 10 μ M (Preincubation Time: 0 minutes)



$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	Control	20 µM	10 µM	5 µM	1 µM	0.5 µM
Progress Curve	Hit constraint					Hit constraint
Best-fit values						
vs	~ 2.162e-016	4.460e-006	7.883e-006	1.078e-005	1.211e-005	~ 1.304e-016
vi	0.0004800	0.0004655	0.0003572	0.0003810	0.0004115	0.0004014
kobs	0.0009615	0.009451	0.003719	0.002538	0.001476	0.001099
Std. Error						
vs		9.250e-008	1.565e-007	2.425e-007	5.294e-007	
vi	1.409e-006	7.037e-006	2.326e-006	1.689e-006	1.144e-006	7.453e-007
kobs	7.352e-006	0.0001589	3.283e-005	1.817e-005	9.933e-006	5.596e-006
95% Confidence Intervals						
vs		4.279e-006 to 4.642e-006	7.576e-006 to 8.190e-006	1.030e-005 to 1.125e-005	1.107e-005 to 1.315e-005	
vi	0.0004772 to 0.0004828	0.0004517 to 0.0004793	0.0003527 to 0.0003618	0.0003776 to 0.0003843	0.0004092 to 0.0004137	0.0003999 to 0.0004028
kobs	0.0009471 to 0.0009759	0.009139 to 0.009762	0.003654 to 0.003783	0.002503 to 0.002574	0.001457 to 0.001496	0.001088 to 0.001110
Goodness of Fit						
Degrees of Freedom	997	997	997	997	997	997
R square	0.9970	0.9333	0.9885	0.9953	0.9986	0.9991
Absolute Sum of Squares	0.05083	0.004174	0.005518	0.006891	0.008179	0.007771
Sy,x	0.007140	0.002046	0.002353	0.002629	0.002864	0.002792
Constraints						
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	1000	1000	1000	1000	1000	1000

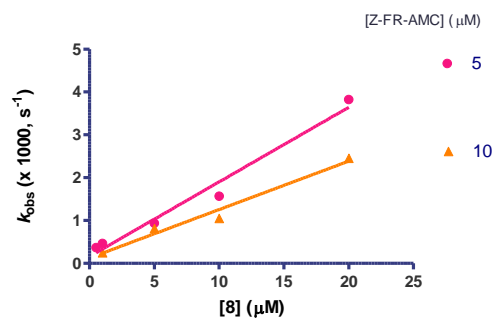
A.9 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 1 µM (Preincubation Time: 0 minutes)



	[I]	1 uM [S]	5 uM [S]	10 uM [I]
	20.000	0.633	1.111	1.703
	10.000	0.418	1.048	1.567
	5.000	0.430	0.988	1.577
	1.000	0.461	1.053	1.591
	0.500	0.394	1.114	

	1 uM [S]	5 uM [S]	10 uM [I]
Best-fit values			
Slope	0.009989 ± 0.003717	0.001762 ± 0.003591	0.006257 ± 0.003161
Y-intercept when X=0.0	0.3943 ± 0.03813	1.050 ± 0.03684	1.553 ± 0.03625
X-intercept when Y=0.0	-39.47	-595.8	-248.2
1/slope	100.1	567.5	159.8
95% Confidence Intervals			
Slope	-0.001837 to 0.02182	-0.009663 to 0.01319	-0.007346 to 0.01986
Y-intercept when X=0.0	0.2729 to 0.5156	0.9327 to 1.167	1.397 to 1.709
X-intercept when Y=0.0	-infinity to -13.46	-infinity to -73.01	-infinity to -71.83
Goodness of Fit			
R square	0.7066	0.07432	0.6620
Sy.x	0.05991	0.05787	0.04493
Is slope significantly non-zero?			
F	7.223	0.2409	3.918
DFn, DFd	1.000, 3.000	1.000, 3.000	1.000, 2.000
P value	0.0746	0.6572	0.1863
Deviation from zero?	Not Significant	Not Significant	Not Significant
Data			
Number of X values	5	5	4
Maximum number of Y replicates	1	1	1
Total number of values	5	5	4
Number of missing values	0	0	1

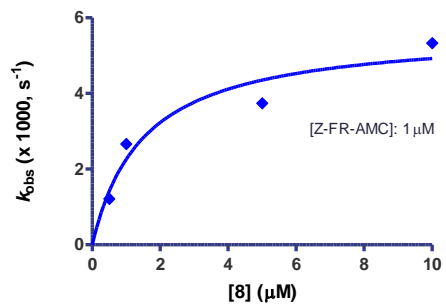
A.10 Plot of v_o , vs [8] Using Linear Regression. (Preincubation Time: 0 minutes)



[I]	5 uM [S]	10 uM [S]
20.000	3.824	2.457
10.000	1.566	1.048
5.000	0.935	0.812
1.000	0.465	0.249
0.500	0.364	

	5 uM [S]	10 uM [S]
Best-fit values		
Slope	0.1741 ± 0.01543	0.1130 ± 0.01250
Y-intercept when X=0.0	0.1598 ± 0.1583	0.1245 ± 0.1433
X-intercept when Y=0.0	-0.9180	-1.101
1/slope	5.743	8.848
95% Confidence Intervals		
Slope	0.1250 to 0.2232	0.05924 to 0.1668
Y-intercept when X=0.0	-0.3438 to 0.6634	-0.4923 to 0.7412
X-intercept when Y=0.0	-4.917 to 1.662	-11.37 to 3.249
Goodness of Fit		
R square	0.9770	0.9761
Sy.x	0.2487	0.1777
Is slope significantly non-zero?		
F	127.4	81.76
DFn, DFd	1.000, 3.000	1.000, 2.000
P value	0.0015	0.0120
Deviation from zero?	Significant	Significant
Data		
Number of X values	5	4
Maximum number of Y replicates	1	1
Total number of values	5	4
Number of missing values	0	1

A.11 Plot of k_{obs} vs [8] Using Linear Regression. ([S]: 5 and 10 μM) (Preincubation Time: 0 minutes)

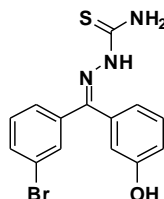


[I]	1 μM [S]
20.000	
10.000	5.331
5.000	3.741
1.000	2.667
0.500	1.218

	1 μM [S]
Michaelis-Menten	
Best-fit values	
Vmax	5.668
Km	1.511
Std. Error	
Vmax	0.8430
Km	0.7593
95% Confidence Intervals	
Vmax	2.040 to 9.295
Km	0.0 to 4.778
Goodness of Fit	
Degrees of Freedom	2
R square	0.9177
Absolute Sum of Squares	0.7442
Sy.x	0.6100
Constraints	
Km	Km > 0.0
Number of points	
Analyzed	4

A.12 Plot of k_{obs} vs [8] Using Non-linear Regression. ([S]: 1 μM) (Preincubation Time: 0 minutes)

RESULTS



GDK-II-98 4a KGP94

% Inhibition
7.900000
8.400000
3.200000

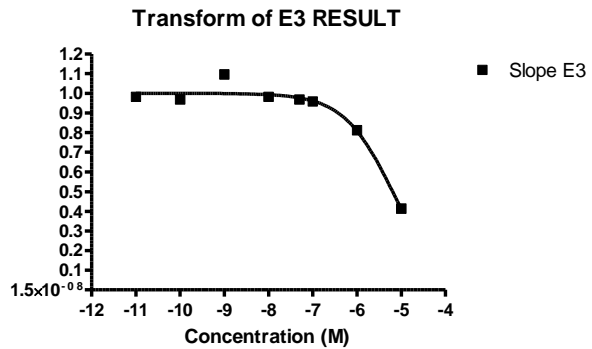
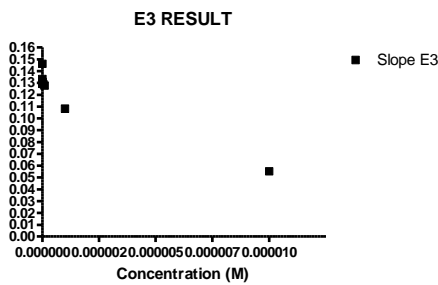
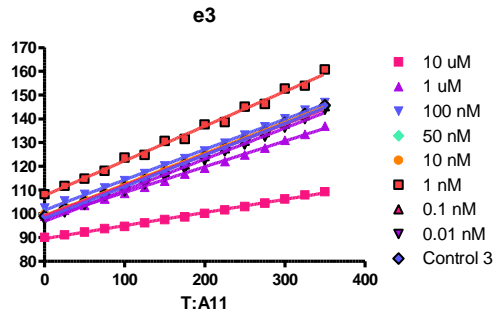
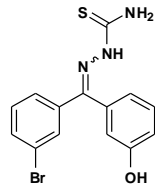
	<u>% Inhibition</u>
Number of values	3
Mean	6.500
Std. Deviation	2.869
Std. Error	1.656
Sum	19.50

	EDTA	DTT	BRU	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	μ M
	1	3	0.01	100	2	1	50

**Pre-incubation time: 0
min**

A.13 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 0 minutes)

GDK-II-98 4a



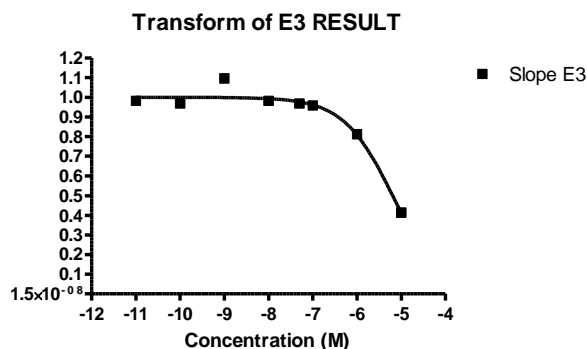
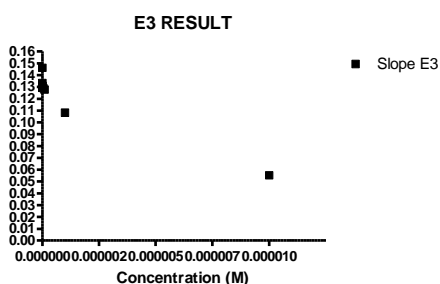
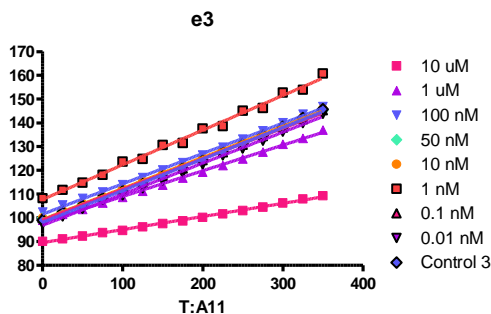
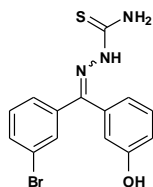
Concentration (M)	Slope E3
0.000010	0.055100
0.000001	0.108100
1.000000e-007	0.127600
5.000000e-008	0.128900
1.000000e-008	0.130800
1.000000e-009	0.146000
1.000000e-010	0.129100
1.000000e-011	0.130800
0.000000	0.133100

Concentration (M)	Slope E3
-5.000	0.414
-6.000	0.812
-7.000	0.959
-7.301	0.968
-8.000	0.983
-9.000	1.097
-10.000	0.970
-11.000	0.983
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-5.192
HillSlope	-0.7760
EC50	6.420e-006
Std. Error	
LogEC50	0.08515
HillSlope	0.1225
95% Confidence Intervals	
LogEC50	-5.401 to -4.984
HillSlope	-1.076 to -0.4764
EC50	3.974e-006 to 1.037e-005
Goodness of Fit	
Degrees of Freedom	6
R square	0.9644
Absolute Sum of Squares	0.01102
Sy.x	0.04285
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.14 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 1 minute)

GDK-II-98 4a



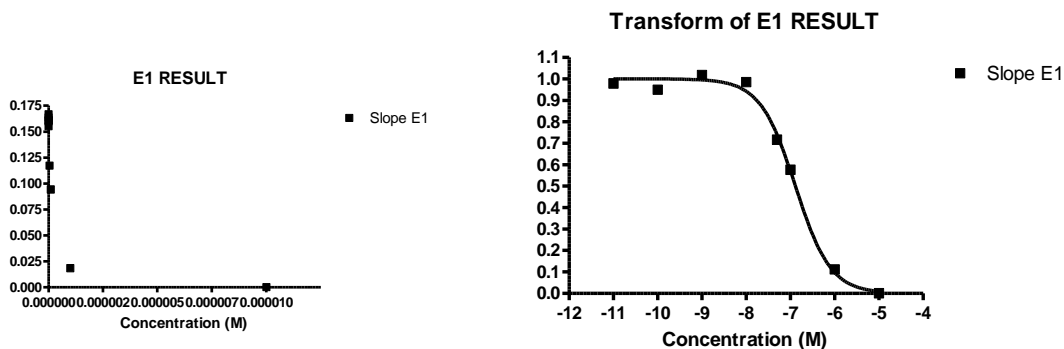
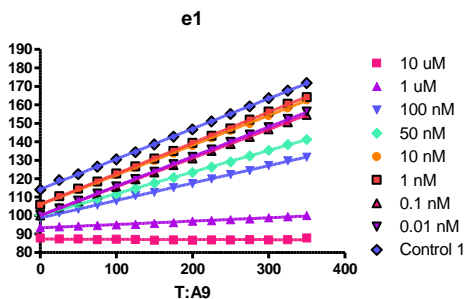
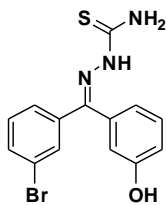
Concentration (M)	Slope E3
0.000010	0.055100
0.000001	0.108100
1.000000e-007	0.127600
5.000000e-008	0.128900
1.000000e-008	0.130800
1.000000e-009	0.146000
1.000000e-010	0.129100
1.000000e-011	0.130800
0.000000	0.133100

Concentration (M)	Slope E3
-5.000	0.414
-6.000	0.812
-7.000	0.959
-7.301	0.968
-8.000	0.983
-9.000	1.097
-10.000	0.970
-11.000	0.983
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-5.192
HillSlope	-0.7760
EC50	6.420e-006
Std. Error	
LogEC50	0.08515
HillSlope	0.1225
95% Confidence Intervals	
LogEC50	-5.401 to -4.984
HillSlope	-1.076 to -0.4764
EC50	3.974e-006 to 1.037e-005
Goodness of Fit	
Degrees of Freedom	6
R square	0.9644
Absolute Sum of Squares	0.01102
Sy.x	0.04285
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.15 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 1 minute)

GDK-II-98 4a



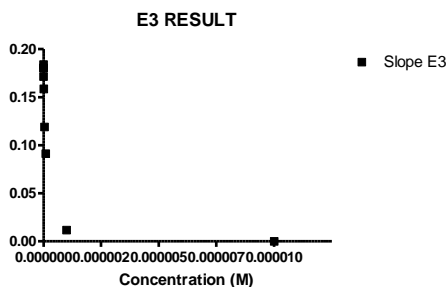
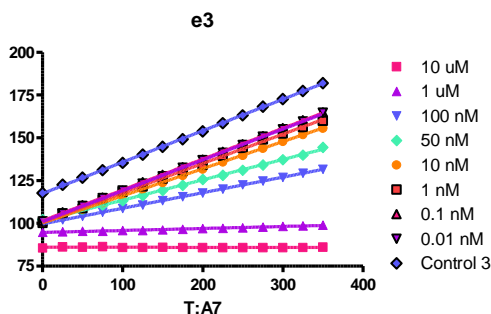
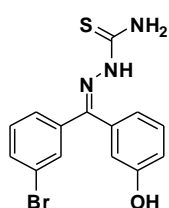
Concentration (M)	Slope E1
0.000010	0.000000
0.000001	0.018250
1.000000e-007	0.094290
5.000000e-008	0.117200
1.000000e-008	0.161100
1.000000e-009	0.166600
1.000000e-010	0.155300
1.000000e-011	0.160100
0.000000	0.163600

Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.112
-7.000	0.576
-7.301	0.716
-8.000	0.985
-9.000	1.018
-10.000	0.949
-11.000	0.979
	1.000

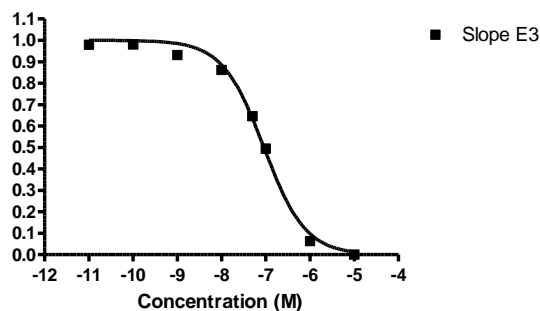
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-6.884
HillSlope	-1.091
EC50	1.305e-007
Std. Error	
LogEC50	0.04297
HillSlope	0.1169
95% Confidence Intervals	
LogEC50	-6.990 to -6.779
HillSlope	-1.377 to -0.8046
EC50	1.024e-007 to 1.662e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9947
Absolute Sum of Squares	0.006132
Sy.x	0.03197
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.16 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 30 minutes)

GDK-II-98 4a



Transform of E3 RESULT



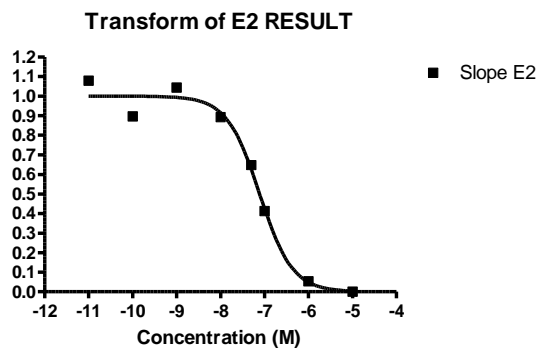
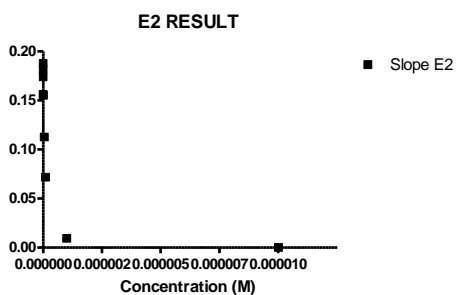
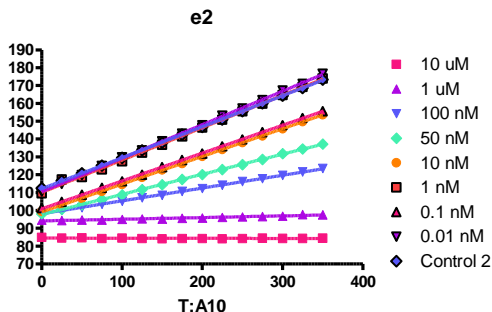
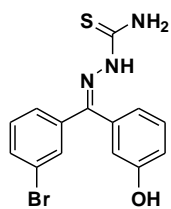
Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.011730
1.000000e-007	0.091020
5.000000e-008	0.118900
1.000000e-008	0.158600
1.000000e-009	0.171500
1.000000e-010	0.180400
1.000000e-011	0.180200
0.000000	0.184000

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.064
-7.000	0.495
-7.301	0.646
-8.000	0.862
-9.000	0.932
-10.000	0.980
-11.000	0.979
	1.000

Slope E3	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.040
HillSlope	-0.9265
EC50	9.120e-008
Std. Error	
LogEC50	0.04095
HillSlope	0.09133
95% Confidence Intervals	
LogEC50	-7.140 to -6.940
HillSlope	-1.150 to -0.7030
EC50	7.240e-008 to 1.149e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9948
Absolute Sum of Squares	0.005802
Sy.x	0.03110
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.17 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 60 minutes)

GDK-II-98 4a



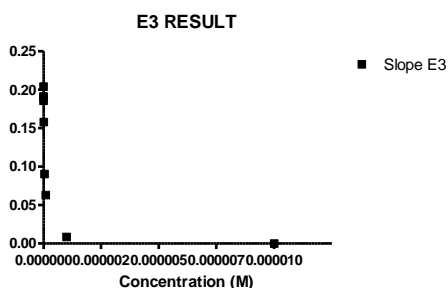
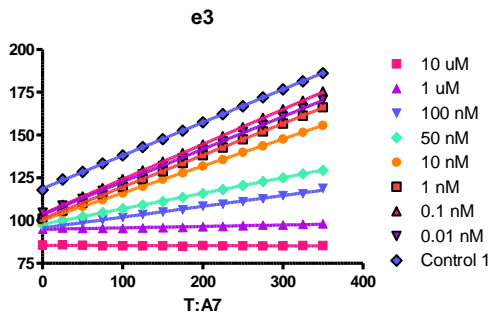
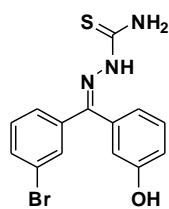
Concentration (M)	Slope E2
0.000010	0.000000
0.000001	0.009365
1.000000e-007	0.071770
5.000000e-008	0.112700
1.000000e-008	0.155400
1.000000e-009	0.181600
1.000000e-010	0.156100
1.000000e-011	0.188000
0.000000	0.174100

Concentration (M)	Slope E2
-5.000	0.000
-6.000	0.054
-7.000	0.412
-7.301	0.647
-8.000	0.893
-9.000	1.043
-10.000	0.897
-11.000	1.080
	1.000

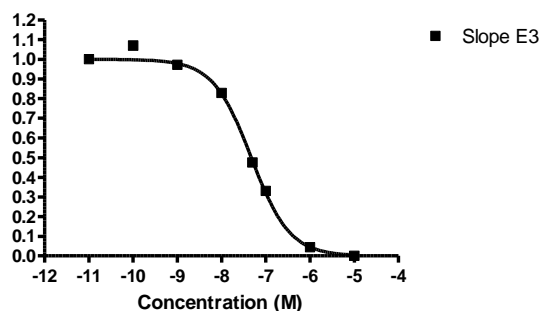
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.109
HillSlope	-1.164
EC50	7.779e-008
Std. Error	
LogEC50	0.06374
HillSlope	0.2564
95% Confidence Intervals	
LogEC50	-7.265 to -6.953
HillSlope	-1.791 to -0.5363
EC50	5.432e-008 to 1.114e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9840
Absolute Sum of Squares	0.02069
Sy.x	0.05872
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.18 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 120 minutes)

GDK-II-98 4a



Transform of E3 RESULT



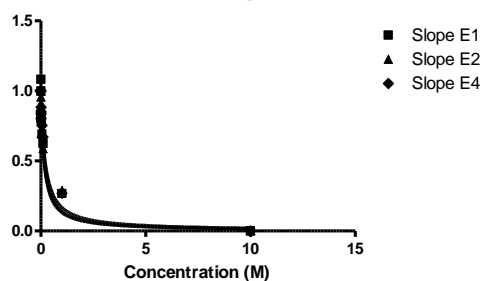
Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.008498
1.000000e-007	0.063020
5.000000e-008	0.090610
1.000000e-008	0.158100
1.000000e-009	0.185500
1.000000e-010	0.204100
1.000000e-011	0.191000
0.000000	0.190700

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.045
-7.000	0.330
-7.301	0.475
-8.000	0.829
-9.000	0.973
-10.000	1.070
-11.000	1.002
	1.000

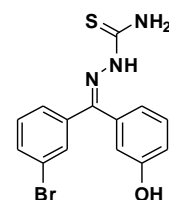
Slope E3	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.326
HillSlope	-1.000
EC50	4.722e-008
Std. Error	
LogEC50	0.03631
HillSlope	0.09587
95% Confidence Intervals	
LogEC50	-7.415 to -7.237
HillSlope	-1.235 to -0.7657
EC50	3.849e-008 to 5.794e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9958
Absolute Sum of Squares	0.005528
Sy.x	0.03035
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

A.19 Representative IC₅₀ Calculation Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 240 minutes)

Transform of Comparison



Ki (nM)
3.254
3.510
4.390



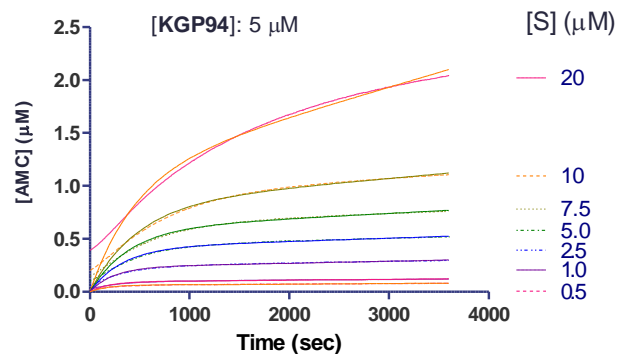
Concentration (M)	Slope E1	Slope E2	Slope E3	Slope E4
10.000	0.000	0.000		0.000
1.000	0.268	0.289		0.269
0.100	0.625	0.588		0.672
0.050	0.693	0.789		0.753
0.010	0.792	0.697		0.837
0.001	0.808	0.911		0.884
1.000e-004	0.835	0.826		0.805
1.000e-005	1.083	0.958		0.851
0.000	1.000	1.000		1.000

	Ki (nM)
Number of values	3
Mean	3.718
Std. Deviation	0.5959
Std. Error	0.3440
Sum	11.15

preincubation: 5 min
[S]: 50 μ M

	Slope E1	Slope E2	Slope E3	Slope E4
Morrison Ki				
Best-fit values				
Vo	= 1.000	= 1.000		= 1.000
Et	= 0.0010	= 0.0010		= 0.0010
Ki	0.003254	0.003519		0.004394
S	= 50.00	= 50.00		= 50.00
Km	= 1.100	= 1.100		= 1.100
Std. Error				
Ki	0.001173	0.001349		0.001583
95% Confidence Intervals				
Ki	0.0005496 to 0.005958	0.0004078 to 0.006629		0.0007435 to 0.008044
Goodness of Fit				
Degrees of Freedom	8	8		8
R square	0.8821	0.8618		0.8836
Absolute Sum of Squares	0.1123	0.1231		0.09893
Sy.x	0.1185	0.1240		0.1112
Constraints				
Vo	Vo = 1.000	Vo = 1.000		Vo = 1.000
Et	Et = 0.0010	Et = 0.0010		Et = 0.0010
S	S = 50.00	S = 50.00		S = 50.00
Km	Km = 1.100	Km = 1.100		Km = 1.100
Number of points				
Analyzed	9	9		9

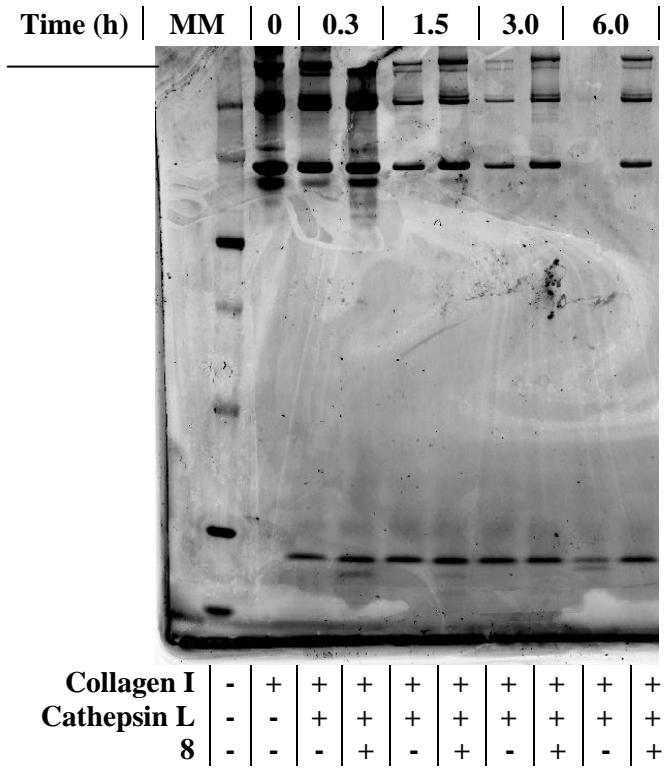
A.20 Calculation of K_i^{app} Using **8** as a Cathepsin L Inhibitor (Preincubation Time: 5 minutes)



$$Y=vs*x+((vi-vs)/kobs)*(1-exp(-kobs*x))$$

	20 uM	10 uM	7.5 uM	5.0 uM	2.5 uM	1.0 uM	0.5 uM
Progress Curve							
Best-fit values							
vs	0.0002770	8.329e-005	4.884e-005	3.055e-005	1.919e-005	7.727e-006	5.059e-006
vi	0.002723	0.001808	0.001560	0.001302	0.001007	0.0005189	0.0003824
kobs	0.002219	0.002096	0.002542	0.003058	0.004260	0.005365	0.005975
Std. Error							
vs	5.683e-006	2.799e-006	1.558e-006	8.708e-007	4.093e-007	1.656e-007	1.192e-007
vi	4.380e-005	1.918e-005	1.579e-005	1.265e-005	1.095e-005	6.628e-006	5.734e-006
kobs	6.071e-005	3.673e-005	3.802e-005	4.051e-005	5.724e-005	8.064e-005	0.0001035
95% Confidence Intervals							
vs	0.0002658 to 0.000288	7.781e-005 to 8.878e-005	4.579e-005 to 5.189e-005	2.884e-005 to 3.225e-005	1.838e-005 to 1.999e-005	7.402e-006 to 8.051e-006	4.825e-006 to 5.292e-006
vi	0.002638 to 0.002809	0.001770 to 0.001845	0.001529 to 0.001590	0.001277 to 0.001327	0.0009856 to 0.001028	0.0005059 to 0.0005319	0.0003712 to 0.0003937
kobs	0.002100 to 0.002338	0.002024 to 0.002168	0.002467 to 0.002616	0.002979 to 0.003138	0.004148 to 0.004372	0.005207 to 0.005523	0.005772 to 0.006177
Goodness of Fit							
Degrees of Freedom	718	718	718	718	718	718	718
R square	0.9801	0.9836	0.9821	0.9816	0.9770	0.9671	0.9534
Absolute Sum of Squares	3.133	0.6766	0.3023	0.1259	0.04074	0.008050	0.004483
Sy.x	0.06605	0.03070	0.02052	0.01324	0.007533	0.003348	0.002499
Constraints							
vs	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points							
Analyzed	721	721	721	721	721	721	721

A.21 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [I]: 5 µM (Preincubation Time: 0 minutes)



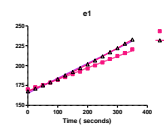
A.22 Inhibition of collagenase activity of cathepsin L by **8** (Preincubation Time: 120 minutes)

	10 μ M	Control
Best-Ft values		
Slope	0.1448 \pm 0.002335	0.1615 \pm 0.002039
Y-intercept when X=0.0	168.2 \pm 0.4803	165.3 \pm 0.5249
X-intercept when Y=0.0	-1162	-811.9
1/slope	6.907	6.234
95% Confidence Intervals		
Slope	0.1387 to 0.1488	0.1539 to 0.1640
Y-intercept when X=0.0	167.2 to 169.3	164.0 to 166.7
X-intercept when Y=0.0	-1210 to -1117	-920.4 to -945.9
Goodness of Fit		
R square	0.9966	0.9966
Sy/x	0.0359	1.271
Is slope significantly non-zero?		
F	3844	3807
DfN, DfD	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data	15	15
Number of X values	1	1
Maximum number of Y replicates	15	15
Total number of values	15	15
Number of missing values	0	0

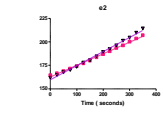
	10 μ M	Control
Best-Ft values		
Slope	0.1235 \pm 0.002290	0.1501 \pm 0.005115
Y-intercept when X=0.0	162.6 \pm 0.4522	159.6 \pm 0.6469
X-intercept when Y=0.0	-1317	-1054
1/slope	8.099	6.662
95% Confidence Intervals		
Slope	0.1187 to 0.1282	0.1434 to 0.1588
Y-intercept when X=0.0	161.7 to 163.6	158.3 to 161.0
X-intercept when Y=0.0	-1377 to -1262	-1122 to -1010
Goodness of Fit		
R square	0.9929	0.9944
Sy/x	1.902	1.034
Is slope significantly non-zero?		
F	3150	2318
DfN, DfD	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data	15	15
Number of X values	1	1
Maximum number of Y replicates	15	15
Total number of values	15	15
Number of missing values	0	0

	10 μ M	Control 3	AMC
Best-Ft values			
Slope	0.1346 \pm 0.001676	0.2257 \pm 0.001938	0.00257 \pm 0.0008649
Y-intercept when X=0.0	156.8 \pm 0.3447	166.7 \pm 0.3686	171.6 \pm 0.1408
X-intercept when Y=0.0	-1258	-747.1	-2859
1/slope	6.025	4.430	173.7
95% Confidence Intervals			
Slope	0.1210 to 0.1282	0.2216 to 0.2299	0.004278 to 0.007237
Y-intercept when X=0.0	156.1 to 157.6	167.8 to 168.5	171.3 to 171.9
X-intercept when Y=0.0	-1307 to -1218	-794.7 to -730.2	-4078 to -2367
Goodness of Fit			
R square	0.9977	0.9990	0.8446
Sy/x	0.7972	0.8110	0.2866
Is slope significantly non-zero?			
F	6527	13659	70.65
DfN, DfD	1,000, 13,000	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data	15	15	15
Number of X values	1	1	1
Maximum number of Y replicates	15	15	15
Total number of values	15	15	15
Number of missing values	0	0	0

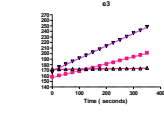
	10 μ M	Control
Best-Ft values		
Slope	0.1242 \pm 0.001880	0.1459 \pm 0.002086
Y-intercept when X=0.0	159.8 \pm 0.3882	164.9 \pm 0.6346
X-intercept when Y=0.0	-1286	-1130
1/slope	8.050	6.865
95% Confidence Intervals		
Slope	0.1202 to 0.1283	0.1362 to 0.1525
Y-intercept when X=0.0	158.8 to 160.6	163.5 to 166.3
X-intercept when Y=0.0	-1326 to -1229	-1163 to -1073
Goodness of Fit		
R square	0.9970	0.9942
Sy/x	1.7907	1.291
Is slope significantly non-zero?		
F	4330	2234
DfN, DfD	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data	15	15
Number of X values	1	1
Maximum number of Y replicates	15	15
Total number of values	15	15
Number of missing values	0	0



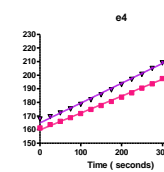
Time (seconds)	10 μ M	Control
0.00	110.234	107.163
25.00	172.792	171.136
50.01	175.431	175.226
75.00	178.748	179.070
100.01	182.095	182.932
125.01	185.481	187.708
150.01	189.086	192.038
175.02	192.426	196.860
200.02	196.326	201.701
225.01	200.219	206.583
250.01	203.963	211.686
275.01	208.287	216.938
300.02	212.060	222.022
325.01	215.807	227.655
350.01	220.487	232.668



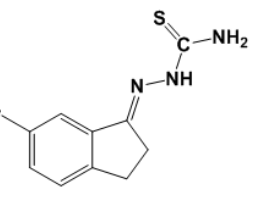
Time (seconds)	10 μ M	Control 3	AMC
0.00	156.248	162.512	172.054
24.97	160.483	175.601	171.714
49.97	163.293	181.067	171.702
74.97	166.161	182.722	172.106
99.97	168.468	191.755	172.016
124.97	171.677	196.864	171.991
149.98	174.696	202.323	172.678
174.97	178.000	207.922	172.718
199.97	181.221	213.440	172.473
224.98	184.366	219.246	172.947
249.97	187.801	224.703	172.975
274.97	191.329	230.391	173.145
299.98	194.974	236.762	173.009
324.98	197.740	241.841	173.311
349.97	201.356	248.124	174.260



Time (seconds)	10 μ M	Control
0.00	167.215	167.801
24.97	163.676	169.518
49.97	166.094	172.247
74.98	168.716	175.253
99.97	171.742	178.653
124.98	174.815	181.935
149.98	177.628	185.022
174.97	180.665	189.325
199.98	183.859	193.160
224.98	187.045	198.344
249.97	190.422	200.635
274.98	193.654	204.998
299.98	197.476	209.069
324.97	201.072	213.539
349.97	204.330	217.859



	% Inhibition
Number of values	3
Mean	18.47
Std. Deviation	4.005
Std. Error	2.313
Sum	55.40



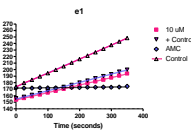
ALJ-I-15
KGP242

	E1	E2	E3	E4
Control	0.000	0.000		0.000
10 μ M	0.228	0.177		0.149

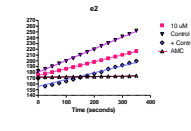
	EDTA	DTT	BRJ	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	μ M
	1	3	0.01	100	2	1	50

A.23 Representative IC₅₀ Calculation Using **83** as a Cathepsin L Inhibitor (Preincubation

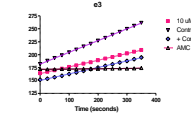
	10 uM	Control	+ Control
BestFit values			
Slope	0.1146 ± 0.0007484	0.2129 ± 0.0008208	0.1273 ± 0.001252
Y-intercept when X=0.0	153.3 ± 0.1539	173.8 ± 0.1688	154.9 ± 0.2074
X-intercept when Y=0.0	-1338	-915.1	-1217
1/slope	8.723	4.697	7.867
95% Confidence Intervals			
Slope	0.1150 to 0.1162	0.2112 to 0.2147	0.1268 to 0.1300
Y-intercept when X=0.0	153.0 to 153.7	173.2 to 173.9	154.3 to 155.4
X-intercept when Y=0.0	-1359 to -1317	-923.5 to -906.9	-1247 to -1198
Goodness of Fit			
R square	0.9994	0.9987	0.9987
Sy/x	0.3321	0.3434	0.3237
Is slope significantly non-zero?			
F	23463	6739	10308
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0



	10 uM	Control	+ Control
BestFit values			
Slope	0.1198 ± 0.001127	0.2024 ± 0.002186	0.1182 ± 0.01085
Y-intercept when X=0.0	174.8 ± 0.2217	180.3 ± 0.4499	155.7 ± 2.231
X-intercept when Y=0.0	-4463	-402.6	-1316
1/slope	8.414	4.940	8.483
95% Confidence Intervals			
Slope	0.1184 to 0.1213	0.1977 to 0.2072	0.09473 to 0.1416
Y-intercept when X=0.0	174.1 to 175.1	179.4 to 181.3	150.9 to 160.5
X-intercept when Y=0.0	-4503 to -4406	-414.4 to -406.3	-1589 to -1070
Goodness of Fit			
R square	0.9985	0.9985	0.9912
Sy/x	0.4714	0.5153	0.4338
Is slope significantly non-zero?			
F	11121	8961	118.6
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

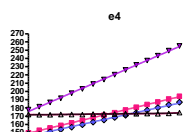


	10 uM	Control	+ Control
BestFit values			
Slope	0.1288 ± 0.0008498	0.2274 ± 0.0007680	0.1525 ± 0.001141
Y-intercept when X=0.0	163.2 ± 0.1747	181.8 ± 0.1579	150.1 ± 0.2346
X-intercept when Y=0.0	-1557	-799.3	-1163
1/slope	7.791	4.388	7.347
95% Confidence Intervals			
Slope	0.1280 to 0.1317	0.2257 to 0.2291	0.1524 to 0.1583
Y-intercept when X=0.0	162.8 to 163.8	181.4 to 182.1	148.6 to 150.6
X-intercept when Y=0.0	-1577 to -1523	-809.5 to -792.2	-1220 to -1166
Goodness of Fit			
R square	0.9994	0.9999	0.9999
Sy/x	0.3666	0.3213	0.4772
Is slope significantly non-zero?			
F	23345	67669	12166
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

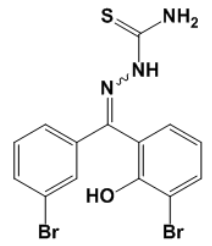


	10 uM	Control	+ Control
BestFit values			
Slope	0.1288 ± 0.0008498	0.2274 ± 0.0007680	0.1525 ± 0.001141
Y-intercept when X=0.0	163.2 ± 0.1747	181.8 ± 0.1579	150.1 ± 0.2346
X-intercept when Y=0.0	-1557	-799.3	-1163
1/slope	7.791	4.388	7.347
95% Confidence Intervals			
Slope	0.1280 to 0.1317	0.2257 to 0.2291	0.1524 to 0.1583
Y-intercept when X=0.0	162.8 to 163.8	181.4 to 182.1	148.6 to 150.6
X-intercept when Y=0.0	-1577 to -1523	-809.5 to -792.2	-1220 to -1166
Goodness of Fit			
R square	0.9994	0.9999	0.9999
Sy/x	0.3666	0.3213	0.4772
Is slope significantly non-zero?			
F	23345	67669	12166
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

	10 uM	Control	+ Control	AMC
BestFit values				
Slope	0.1146 ± 0.0007484	0.2129 ± 0.0008208	0.1273 ± 0.001252	0.092757 ± 0.0008648
Y-intercept when X=0.0	153.3 ± 0.1539	173.8 ± 0.1688	154.9 ± 0.2074	171.8 ± 0.1408
X-intercept when Y=0.0	-1338	-915.1	-1217	-2895
1/slope	8.723	4.697	7.867	10.805
95% Confidence Intervals				
Slope	0.1150 to 0.1162	0.2112 to 0.2147	0.1268 to 0.1300	0.094278 to 0.070237
Y-intercept when X=0.0	153.0 to 153.7	173.2 to 173.9	154.3 to 155.4	171.3 to 171.9
X-intercept when Y=0.0	-1359 to -1317	-923.5 to -906.9	-1247 to -1198	-4073 to -2877
Goodness of Fit				
R square	0.9994	0.9987	0.9987	0.8446
Sy/x	0.3321	0.3434	0.3237	0.2865
Is slope significantly non-zero?				
F	23463	6739	10308	70.86
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant	Significant
Data				
Number of X values	15	15	15	15
Maximum number of Y replicates	1	1	1	1
Total number of values	15	15	15	15
Number of missing values	0	0	0	0



	Number of values	% Inhibition
Mean	4	43.55
Std. Deviation		2.047
Std. Error		1.023
Sum	174.2	



	10 uM	Control	+ Control	AMC
BestFit values				
Slope	0.1288 ± 0.0008498	0.2274 ± 0.0007680	0.1525 ± 0.001141	0.092757 ± 0.0008648
Y-intercept when X=0.0	163.2 ± 0.1747	181.8 ± 0.1579	150.1 ± 0.2346	171.8 ± 0.1408
X-intercept when Y=0.0	-1557	-799.3	-1163	-2895
1/slope	7.791	4.388	7.347	10.805
95% Confidence Intervals				
Slope	0.1280 to 0.1317	0.2257 to 0.2291	0.1524 to 0.1583	0.094278 to 0.070237
Y-intercept when X=0.0	162.8 to 163.8	181.4 to 182.1	148.6 to 150.6	171.3 to 171.9
X-intercept when Y=0.0	-1577 to -1523	-809.5 to -792.2	-1220 to -1166	-4073 to -2877
Goodness of Fit				
R square	0.9994	0.9999	0.9999	0.8446
Sy/x	0.3666	0.3213	0.4772	0.2865
Is slope significantly non-zero?				
F	23345	67669	12166	70.86
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant	Significant
Data				
Number of X values	15	15	15	15
Maximum number of Y replicates	1	1	1	1
Total number of values	15	15	15	15
Number of missing values	0	0	0	0

	10 uM	Control	+ Control	AMC
BestFit values				
Slope	0.1285 ± 0.0007710	0.2260 ± 0.001901	0.1142 ± 0.002460	0.092757 ± 0.0008648
Y-intercept when X=0.0	149.4 ± 0.1461	175.7 ± 0.3909	145.4 ± 0.5049	171.8 ± 0.1408
X-intercept when Y=0.0	-1187	-789.7	-1278	-2895
1/slope	7.796	4.444	8.769	10.805
95% Confidence Intervals				
Slope	0.1249 to 0.1283	0.2209 to 0.2292	0.1098 to 0.1196	0.094278 to 0.070237
Y-intercept when X=0.0	149.0 to 149.7	174.9 to 176.5	144.4 to 147.0	171.3 to 171.9
X-intercept when Y=0.0	-1198 to -1165	-798.0 to -763.5	-1349 to -1214	-4072 to -2876
Goodness of Fit				
R square	0.9996	0.9991	0.9940	0.8446
Sy/x	0.2973	0.7962	1.027	0.2865
Is slope significantly non-zero?				
F	31667	14014	2161	70.87
DFx, DFd	1,000, 13.00	1,000, 13.00	1,000, 13.00	1,000, 13.00
P value	< 0.0001	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant	Significant
Data				
Number of X values	15	15	15	15
Maximum number of Y replicates	1	1	1	1
Total number of values	15	15	15	15
Number of missing values	0	0	0	0

	E1	E2	E3	E4
Control	0.000	0.000	0.000	0.000
10 uM	0.462	0.413	0.429	0.438
+ Control (50 nM)	0.402	0.416	0.447	0.492

**GDK-III-35 (1 mg)
KGP224**

	EDTA	DTT	BRUJ	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	µM
	1	3	0.01	100	2	1	50

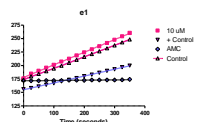
A.23 Representative IC₅₀ Calculation Using **32** as a Cathepsin L Inhibitor

	10 uM	Control	+ Control
Basic values			
Slope	0.2344 ± 0.00180	0.2129 ± 0.006208	0.1273 ± 0.001252
Y-intercept when X=0.0	173.9 ± 0.3701	173.8 ± 0.1688	154.3 ± 0.2374
X-intercept when Y=0.0	-758.3	-815.1	-1217
Yslope	4.266	4.697	7.857
95% Confidence Intervals			
Slope	0.2295 to 0.2393	0.2112 to 0.2147	0.1248 to 0.1300
Y-intercept when X=0.0	172.0 to 178.8	172.3 to 173.0	154.3 to 165.4
X-intercept when Y=0.0	-774.2 to -743.0	-823.5 to -806.9	-1247 to -1188
Yslope	4.092	4.998	6.987
Goodness of Fit	0.7931	0.8434	0.5237
R square	0.9992	0.9986	0.9987
Chi square	1.000	1.000	1.000
DFs, DfG	18951	8789	10336
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	15	15	15
Total number of values	15	15	15
Number of missing values	0	0	0

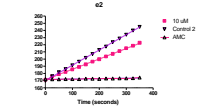
	10 uM	Control 2	AMC
Basic values			
Slope	0.1453 ± 0.001050	0.2133 ± 0.001033	0.05757 ± 0.0008849
Y-intercept when X=0.0	171.2 ± 0.2248	170.1 ± 0.2124	171.8 ± 0.1438
X-intercept when Y=0.0	-1178	-715.5	-2865
Yslope	6.882	4.688	173.7
95% Confidence Intervals			
Slope	0.1430 to 0.1477	0.2111 to 0.2155	0.054278 to 0.057237
Y-intercept when X=0.0	170.7 to 171.7	169.7 to 170.8	171.3 to 171.9
X-intercept when Y=0.0	-1201 to -1157	-807.8 to -706.3	-4073.6 to -2967.7
Yslope	6.920	4.987	0.8446
Goodness of Fit	0.4470	0.4322	0.2885
R square	0.9997	0.9997	0.8446
Chi square	1	1	1
DFs, DfG	18950	13500	70.85
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	15	15	1
Total number of values	15	15	15
Number of missing values	0	0	0

	10 uM	Control 3	AMC
Basic values			
Slope	0.1841 ± 0.001633	0.2257 ± 0.001940	0.002757 ± 0.0008849
Y-intercept when X=0.0	195.3 ± 0.3358	188.7 ± 0.3988	171.8 ± 0.1438
X-intercept when Y=0.0	-1194	-747.1	-2865
Yslope	6.093	4.430	173.7
95% Confidence Intervals			
Slope	0.1808 to 0.1876	0.2218 to 0.2299	0.004278 to 0.007237
Y-intercept when X=0.0	193.2 to 198.7	187.8 to 193.5	171.3 to 171.9
X-intercept when Y=0.0	-1224.9 to -1165	-764.7 to -730.2	-4073.6 to -2967.6
Yslope	6.097	4.990	0.8446
Goodness of Fit	0.6832	0.8114	0.2885
R square	0.9997	0.9990	0.8446
Chi square	1	1	1
DFs, DfG	18950	13544	70.86
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	15	15	1
Total number of values	15	15	15
Number of missing values	0	0	0

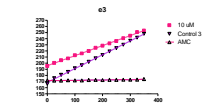
	10 uM	Control 4	AMC
Basic values			
Slope	0.1383 ± 0.001325	0.2087 ± 0.001368	0.059757 ± 0.0008849
Y-intercept when X=0.0	188.8 ± 0.2724	188.8 ± 0.2812	171.8 ± 0.1438
X-intercept when Y=0.0	-1387	-808	-2865
Yslope	7.233	4.730	173.7
95% Confidence Intervals			
Slope	0.1354 to 0.1411	0.2058 to 0.2117	0.054278 to 0.057237
Y-intercept when X=0.0	188.4 to 189.4	188.2 to 189.4	171.3 to 171.9
X-intercept when Y=0.0	-1400 to -1336	-823 to -795.0	-4073.6 to -2967.6
Yslope	6.988	4.994	0.8446
Goodness of Fit	0.5542	0.722	0.2885
R square	0.9998	0.9994	0.8446
Chi square	1	1	1
DFs, DfG	18950	23202	70.87
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	15	15	1
Total number of values	15	15	15
Number of missing values	0	0	0



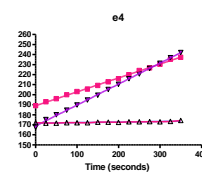
Time (seconds)	10 uM	Control	+ Control	Time (seconds)	10 uM	Control 2	AMC
0.00	175.970	173.572	156.187	0.00	171.548	169.766	172.054
24.96	184.909	178.026	158.332	24.97	178.834	176.059	171.714
49.95	190.199	184.411	161.127	49.97	178.886	181.630	171.702
74.94	195.946	189.555	164.147	74.97	182.014	185.975	172.106
99.94	201.231	194.905	167.119	99.97	185.655	191.341	172.016
124.94	207.169	199.587	170.616	124.98	188.899	196.345	171.991
149.94	212.695	205.078	173.471	149.97	192.532	201.651	172.678
174.94	218.721	210.603	176.667	174.97	196.353	207.242	172.718
199.94	224.257	216.091	180.029	199.97	199.760	212.745	172.473
224.94	230.105	221.252	183.042	224.97	203.221	217.588	172.987
249.94	235.713	226.610	186.720	249.97	207.275	223.537	172.975
274.94	241.764	231.998	190.111	274.98	211.130	228.878	173.145
299.94	247.945	237.525	193.066	299.97	215.204	233.997	173.009
324.94	254.298	242.949	196.794	324.97	218.518	240.072	173.311
349.95	260.466	248.526	199.959	349.97	222.758	244.950	174.260



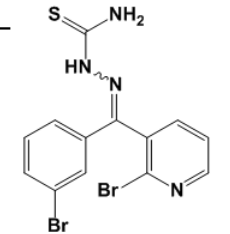
Time (seconds)	10 uM	Control 3	AMC	Time (seconds)	10 uM	Control 4	AMC	
0.00	24.96	200.464	175.601	171.714	24.94	192.902	175.232	171.714
24.96	204.920	181.067	171.702	49.94	196.068	180.001	171.702	
49.96	207.820	185.722	172.106	74.94	199.887	184.662	172.106	
74.96	212.929	191.755	172.016	99.94	203.099	189.728	172.016	
99.96	215.998	196.864	171.991	124.94	205.601	194.787	171.991	
124.96	219.923	202.323	172.878	149.94	207.821	199.887	172.878	
149.96	225.175	207.922	172.718	174.94	212.880	205.131	172.718	
169.96	228.661	213.440	172.473	199.94	215.982	210.431	172.473	
189.96	232.496	218.946	172.947	224.96	218.899	215.820	172.947	
209.96	236.037	224.703	172.975	249.94	224.061	220.648	172.975	
229.96	240.791	230.391	173.145	274.94	226.752	226.260	173.145	
249.96	244.852	236.762	173.009	299.94	230.295	231.390	173.009	
269.96	250.873	241.841	173.311	324.94	235.211	236.986	173.311	
349.96	253.246	248.124	174.260	349.94	237.226	242.101	174.260	



Time (seconds)	10 uM	Control 4	AMC
0.00	186.577	186.572	172.054
24.96	196.464	181.067	171.702
49.96	204.920	185.722	172.106
74.96	212.929	191.755	172.016
99.96	215.998	196.864	171.991
124.96	219.923	202.323	172.878
149.96	225.175	207.922	172.718
169.96	228.661	213.440	172.473
189.96	232.496	218.946	172.947
209.96	236.037	224.703	172.975
229.96	240.791	230.391	173.145
249.96	244.852	236.762	173.009
269.96	250.873	241.841	173.311
349.96	253.246	248.124	174.260



	Number of values	% Inhibition
Mean	3	30.97
Std. Deviation		3.301
Std. Error		1.906
Sum		92.90



**GDK-III-52 (1.5 mg)
KGP233**

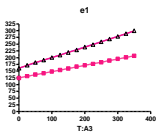
	E2	E3	E4
Control	0.000	0.000	0.000
10 uM	0.319	0.273	0.337

	EDTA	DTT	BRUJ	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	μM
	1	3	0.01	100	2	1	50

A.23 Representative IC₅₀ Calculation Using **38** as a Cathepsin L Inhibitor

	10 uM	Control
Best Fit values		
Slope	0.2333 ± 0.001955	0.3927 ± 0.001924
Y-intercept when X=0.0	124.8 ± 0.2251	161.9 ± 0.3955
X-intercept when Y=0.0	-535.1	-409.9
1/slope	4.287	2.546
95% Confidence Intervals		
Slope	0.2309 to 0.2357	0.3885 to 0.3969
Y-intercept when X=0.0	124.4 to 125.3	160.1 to 161.8
X-intercept when Y=0.0	-532.5 to -527.9	-416.7 to -403.6
Goodness of Fit		
R square	0.9997	0.9997
Sy	0.6560	-0.8487
Is slope significantly non-zero?		
F	45387	41857
Df(n), Df(d)	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data		
Number of X values	15	15
Maximum number of Y replicates	1	1
Total number of values	15	15
Number of missing values	0	0

E1



T.A3	10 uM	Control
0.00	124.407	160.577
24.94	131.353	171.757
49.84	136.827	181.583
74.74	141.905	190.668
99.94	148.506	200.202
124.94	153.616	209.539
149.91	159.428	219.230
174.94	166.433	228.573
199.93	171.366	238.893
224.93	176.835	248.022
249.93	182.591	258.410
274.93	188.888	269.054
299.93	195.100	278.818
324.93	200.735	289.130
349.93	206.783	299.724

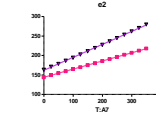
E1

T.A7	MC7	MHS
0.00	144.293	163.562
24.93	148.530	171.093
49.93	154.826	178.690
74.92	159.739	186.654
99.93	165.358	194.372
124.91	169.882	202.754
149.91	175.291	210.988
174.92	180.258	219.381
199.91	185.456	227.515
224.89	190.965	236.274
249.89	196.197	244.773
274.89	201.924	253.403
299.89	206.964	261.969
324.89	212.387	270.994
349.88	217.927	279.549

E2

	MC7	MHS
Best Fit values		
Slope	0.2096 ± 0.0008027	0.3328 ± 0.002048
Y-intercept when X=0.0	144.1 ± 0.1650	161.9 ± 0.4211
X-intercept when Y=0.0	-687.4	-486.4
1/slope	4.771	3.004
95% Confidence Intervals		
Slope	0.2079 to 0.2113	0.3284 to 0.3373
Y-intercept when X=0.0	143.7 to 144.4	161.0 to 162.8
X-intercept when Y=0.0	-694.6 to -680.2	-495.5 to -477.6
Goodness of Fit		
R square	0.9998	0.9995
Sy	0.3357	0.8567
Is slope significantly non-zero?		
F	68184	26406
Df(n), Df(d)	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data		
Number of X values	15	15
Maximum number of Y replicates	1	1
Total number of values	15	15
Number of missing values	0	0

E2



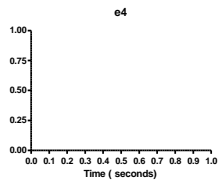
T.A11	MC11	Control
0.00	132.898	164.913
24.92	133.039	172.469
49.92	136.596	179.750
74.92	140.518	186.872
99.92	144.192	194.147
124.91	148.467	201.709
149.90	152.331	210.322
174.92	156.580	218.039
199.91	160.538	226.805
224.90	164.584	235.001
249.89	168.706	243.521
274.89	172.857	251.944
299.87	177.162	260.516
324.87	181.662	269.109
349.86	186.139	277.618

E3

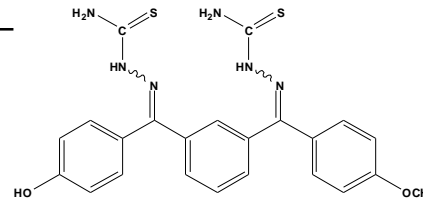
E4

	MC11	Control
Best Fit values		
Slope	0.1987 ± 0.002885	0.3229 ± 0.002882
Y-intercept when X=0.0	129.3 ± 0.5931	162.9 ± 0.5923
X-intercept when Y=0.0	-614.9	-503.6
1/slope	6.301	3.091
95% Confidence Intervals		
Slope	0.1925 to 0.1649	0.3173 to 0.3297
Y-intercept when X=0.0	128.0 to 130.6	161.7 to 164.2
X-intercept when Y=0.0	-656.6 to -777.3	-517.1 to -490.6
Goodness of Fit		
R square	0.9967	0.9990
Sy	1.207	1.205
Is slope significantly non-zero?		
F	3025	12605
Df(n), Df(d)	1,000, 13,000	1,000, 13,000
P value	< 0.0001	< 0.0001
Deviation from zero?	Significant	Significant
Data		
Number of X values	15	15
Maximum number of Y replicates	1	1
Total number of values	15	15
Number of missing values	0	0

E3



	% Inhibition
Number of values	3
Mean	42.83
Std. Deviation	7.214
Std. Error	4.165
Sum	128.5



	E1	E2	E3
Control	0.000	0.000	0.000
10 uM	0.406	0.370	0.509

SJL-III-124-3
KGP271

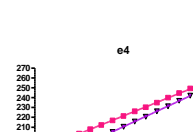
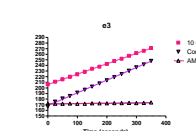
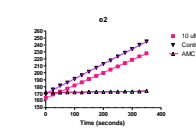
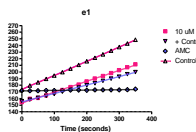
E4

	EDTA	DTT	BRUJ	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	μM
	1	3	0.01	100	2	1	50

A.23 Representative IC₅₀ Calculation Using **166** as a Cathepsin L Inhibitor

	10 uM	Control	+ Control
Best fit values			
Slope	0.1636 ± 0.002270	0.2129 ± 0.0008208	0.1273 ± 0.001252
Y-intercept when X=0.0	153.3 ± 0.4666	173.6 ± 0.1688	154.3 ± 0.2574
X-intercept when Y=0.0	-937.3	-815.1	-1217
1/slope	6.114	4.697	7.857
95% Confidence Intervals			
Slope	0.1587 to 0.1685	0.2112 to 0.2147	0.1246 to 0.1300
Y-intercept when X=0.0	152.3 to 154.3	173.2 to 173.9	154.3 to 155.4
X-intercept when Y=0.0	-971.8 to -904.7	-823.5 to -856.9	-1247 to -1188
Goodness of Fit			
R square	0.9975	0.9999	0.9987
Sy.x	0.9494	0.3434	0.5237
F			
F	5194	67289	10336
DfN, DfD	1.000, 13.000	1.000, 13.000	1.000, 13.000
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?			
Significant	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

E1



	10 uM	Control 2	AMC
Best fit values			
Slope	0.1624 ± 0.0007915	0.2133 ± 0.001033	0.009757 ± 0.0006949
Y-intercept when X=0.0	153.7 ± 0.1607	170.1 ± 0.2124	171.6 ± 0.1468
X-intercept when Y=0.0	-895.5	-797.5	-2906
1/slope	5.476	4.688	173.7
95% Confidence Intervals			
Slope	0.1589 to 0.1643	0.2111 to 0.2155	0.004279 to 0.007237
Y-intercept when X=0.0	153.4 to 164.1	169.7 to 170.8	171.3 to 171.9
X-intercept when Y=0.0	-905.5 to -886.6	-807.8 to -787.3	-4017.5 to -2367.7
Goodness of Fit			
R square	0.9998	0.9997	0.9446
Sy.x	0.3269	0.4322	0.2865
F			
F	54608	42634	70.66
DfN, DfD	1.000, 13.000	1.000, 13.000	1.000, 13.000
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?			
Significant	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

E2

	10 uM	Control 3	AMC
Best fit values			
Slope	0.1893 ± 0.0007596	0.2207 ± 0.001040	0.009757 ± 0.0006949
Y-intercept when X=0.0	206.0 ± 0.1642	168.7 ± 0.2988	171.6 ± 0.1429
X-intercept when Y=0.0	-1113	-747.1	-2905
1/slope	5.403	4.480	173.7
95% Confidence Intervals			
Slope	0.1833 to 0.1968	0.2216 to 0.2299	0.004279 to 0.007237
Y-intercept when X=0.0	205.8 to 206.3	167.3 to 169.4	171.3 to 171.9
X-intercept when Y=0.0	-1125 to -1101	-764.7 to -730.2	-4017.4 to -2367.6
Goodness of Fit			
R square	0.9998	0.9990	0.9446
Sy.x	0.3341	0.8114	0.2865
F			
F	53701	13544	70.66
DfN, DfD	1.000, 13.000	1.000, 13.000	1.000, 13.000
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?			
Significant	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

E3

	10 uM	Control 4	AMC
Best fit values			
Slope	0.1621 ± 0.0007426	0.2087 ± 0.001368	0.009757 ± 0.0006949
Y-intercept when X=0.0	186.2 ± 0.0207	168.8 ± 0.2812	171.6 ± 0.1468
X-intercept when Y=0.0	-1017	-808.8	-2905
1/slope	5.497	4.790	173.7
95% Confidence Intervals			
Slope	0.1565 to 0.1637	0.2098 to 0.2117	0.004279 to 0.007237
Y-intercept when X=0.0	184.8 to 185.5	169.2 to 169.4	171.3 to 171.9
X-intercept when Y=0.0	-1027 to -1006	-823.0 to -795.0	-4017.2 to -2367.9
Goodness of Fit			
R square	0.9998	0.9994	0.9446
Sy.x	0.3107	0.5722	0.2865
F			
F	60144	23292	70.67
DfN, DfD	1.000, 13.000	1.000, 13.000	1.000, 13.000
P value	< 0.0001	< 0.0001	< 0.0001
Deviation from zero?			
Significant	Significant	Significant	Significant
Data			
Number of X values	15	15	15
Maximum number of Y replicates	1	1	1
Total number of values	15	15	15
Number of missing values	0	0	0

E4

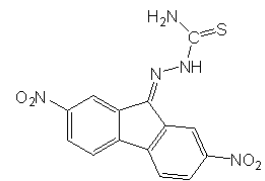
Time (seconds)	10 uM	Control	+ Control	Time (seconds)	10 uM	Control 2	AMC
0.00	154.585	173.572	156.187	0.00	164.068	169.766	172.054
24.96	159.546	178.626	158.332	24.97	168.760	176.659	171.714
49.95	160.940	184.411	161.127	49.97	172.968	181.630	171.702
74.94	164.510	189.555	164.147	74.97	177.286	185.975	172.106
99.94	168.915	194.505	167.119	99.97	181.783	191.341	172.016
124.94	172.896	199.587	170.616	124.96	186.256	196.345	171.991
149.94	177.112	205.078	173.471	149.97	190.995	201.651	172.678
174.94	181.061	210.693	176.667	174.97	195.519	207.242	172.718
199.94	185.744	216.091	180.029	199.97	199.852	212.745	172.473
224.94	189.965	221.252	183.042	224.97	204.600	217.588	172.947
249.94	193.931	226.610	186.720	249.97	208.946	223.537	172.975
274.94	198.277	231.998	190.111	274.96	213.781	228.676	173.145
299.94	202.622	237.525	193.056	299.97	218.747	233.997	173.009
324.94	207.084	242.949	196.794	324.97	223.571	240.072	173.311
349.95	211.487	248.526	199.959	349.97	228.006	244.950	174.260

E1

Time (seconds)	10 uM	Control 3	AMC	Time (seconds)	10 uM	Control 4	AMC
0.00	206.471	166.572	172.054	0.00	185.592	167.585	172.054
24.96	210.960	175.601	171.714	24.94	189.928	175.232	171.714
49.95	215.312	181.067	171.702	49.94	194.513	180.001	171.702
74.96	219.712	185.722	172.106	74.94	198.314	184.662	172.106
99.96	224.064	191.735	172.016	99.94	203.402	189.726	172.016
124.96	228.697	196.864	171.991	124.94	207.915	194.787	171.991
149.96	233.900	202.323	172.678	149.94	212.229	199.569	172.678
174.96	237.830	207.322	172.718	174.94	216.950	205.331	172.718
199.96	242.687	213.440	172.473	199.94	221.301	210.431	172.473
224.97	247.799	219.246	172.947	224.95	226.122	215.820	172.947
249.96	252.071	224.703	172.975	249.94	230.288	220.648	172.975
274.96	257.015	230.391	173.145	274.94	234.071	226.260	173.145
299.96	261.885	236.762	173.009	299.94	238.026	231.390	173.009
324.96	266.135	241.841	173.311	324.94	244.628	236.986	173.311
349.96	270.960	248.124	174.260	349.94	249.348	242.101	174.260

E3

	% Inhibition
Number of values	3
Mean	15.03
Std. Deviation	2.706
Std. Error	1.562
Sum	45.10



Sample Reference: SJL-II-138
KGP236

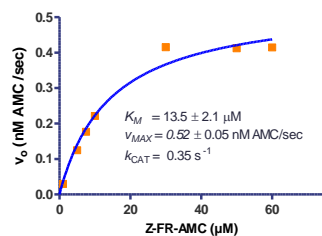
Stock sln: bright orange
Precipitated at 35% DMSO inhibitor dilution

	EDTA	DTT	BRJ	NaOAc	DMSO	CAT L	Z-FR-AMC
	mM	mM	%	mm	%	nM	µM
	1	3	0.01	100	2	1	50

A.23 Representative IC₅₀ Calculation Using **82** as a Cathepsin L Inhibitor

APPENDIX B

In Vitro Evaluations of Thiosemicarbazones as Inhibitors of Human Cathepsin K

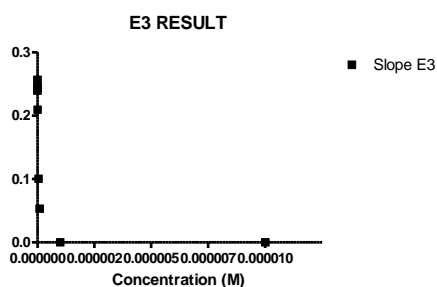
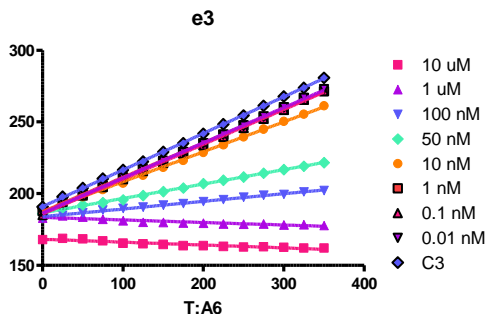
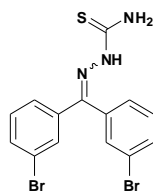


Substrate Concentration (μM)	
1.00000	0.030
5.00000	0.125
7.50000	0.177
10.00000	0.221
30.00000	0.416
50.00000	0.413
60.00000	0.415
	0.372

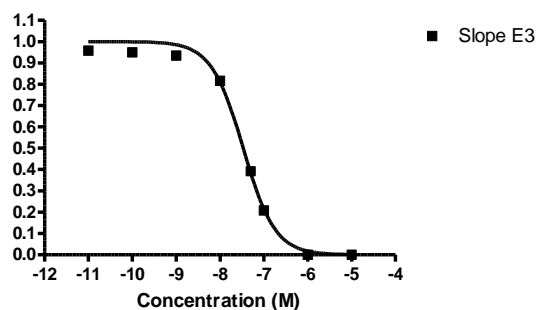
Michaelis-Menten	
Best-fit values	
Vmax	0.5422
Km	14.34
Std. Error	
Vmax	0.03844
Km	2.914
95% Confidence Intervals	
Vmax	0.4433 to 0.6410
Km	6.845 to 21.83
Goodness of Fit	
Degrees of Freedom	5
R square	0.9779
Absolute Sum of Squares	0.003341
Sy.x	0.02585
Constraints	
Km	Km > 0.0
Number of points	
Analyzed	7

B.1 Calculation of Michaelis-Menten Constant for Cathepsin K using Z-FR-AMC as a Substrate

LJ-I-006



Transform of E3 RESULT



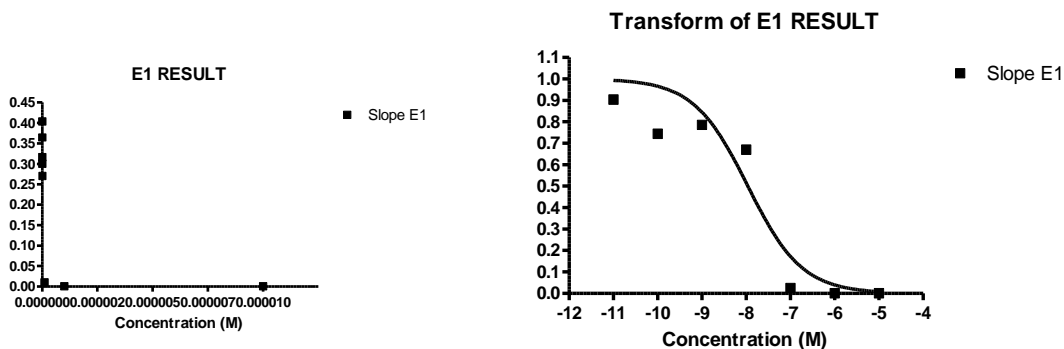
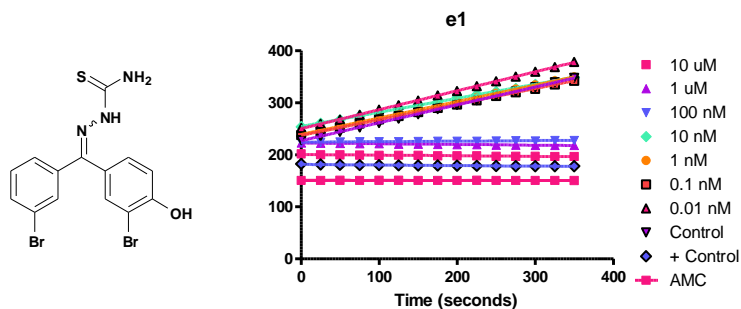
Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.053130
5.000000e-008	0.100400
1.000000e-008	0.209100
1.000000e-009	0.239400
1.000000e-010	0.243100
1.000000e-011	0.245300
0.000000	0.256200

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.207
-7.301	0.392
-8.000	0.816
-9.000	0.934
-10.000	0.949
-11.000	0.957
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.471
HillSlope	-1.193
EC50	3.381e-008
Std. Error	
LogEC50	0.03995
HillSlope	0.1255
95% Confidence Intervals	
LogEC50	-7.569 to -7.373
HillSlope	-1.500 to -0.8858
EC50	2.699e-008 to 4.234e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9943
Absolute Sum of Squares	0.007339
Sy.x	0.03497
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

B.2 Representative IC₅₀ Calculation Using 1 as a Cathepsin K Inhibitor

GDK-III-29



Concentration (M)	Slope E1
1.00e-005	0.000000
1.00e-006	0.000000
1.00e-007	0.009867
1.00e-008	0.270200
1.00e-009	0.316500
1.00e-010	0.300100
1.00e-011	0.364400
0.00	0.403300

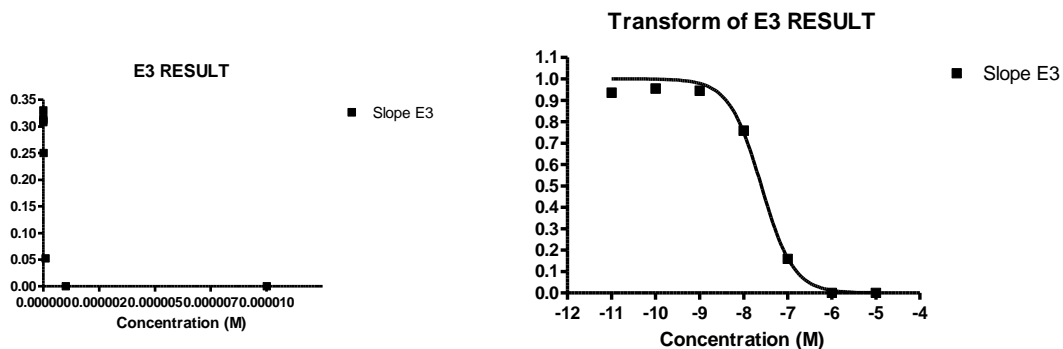
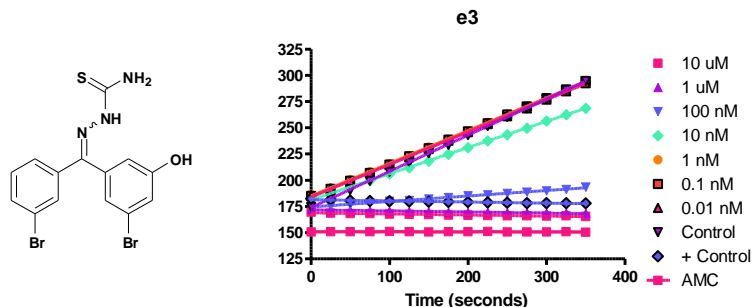
Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.000
-7.000	0.024
-8.000	0.670
-9.000	0.785
-10.000	0.744
-11.000	0.904
	1.000

	E1
Control	0.000
+ Control (100 nM)	0.882

	Slope E1
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.960
HillSlope	-0.7084
EC50	1.095e-008
Std. Error	
LogEC50	0.2811
HillSlope	0.2893
95% Confidence Intervals	
LogEC50	-8.683 to -7.238
HillSlope	-1.452 to 0.03548
EC50	2.075e-009 to 5.784e-008
Goodness of Fit	
Degrees of Freedom	5
R square	0.8962
Absolute Sum of Squares	0.1078
Sy.x	0.1468
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	7

B.3 Representative IC₅₀ Calculation using **31** as a Cathepsin K Inhibitor

GDK-III-35



Concentration (M)	Slope E3
1.00e-005	0.000000
1.00e-006	0.000000
1.00e-007	0.052280
1.00e-008	0.250000
1.00e-009	0.311200
1.00e-010	0.314700
1.00e-011	0.308200
0.00	0.329600

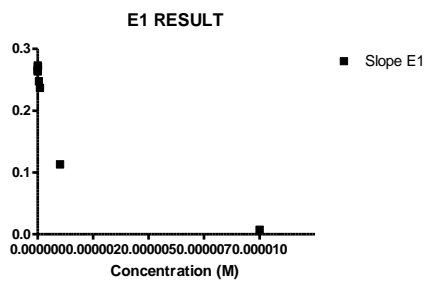
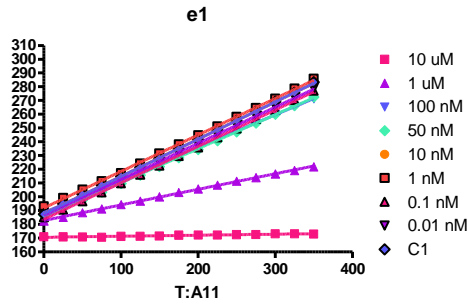
Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.159
-8.000	0.758
-9.000	0.944
-10.000	0.955
-11.000	0.935
	1.000

	E3
Control	0.000
+ Control (100 nM)	0.881

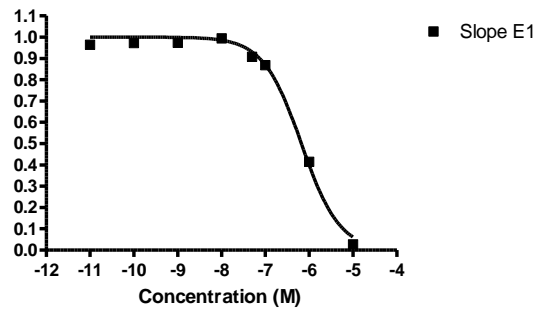
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.598
HillSlope	-1.190
EC50	2.524e-008
Std. Error	
LogEC50	0.06148
HillSlope	0.1464
95% Confidence Intervals	
LogEC50	-7.756 to -7.440
HillSlope	-1.567 to -0.8136
EC50	1.754e-008 to 3.632e-008
Goodness of Fit	
Degrees of Freedom	5
R square	0.9940
Absolute Sum of Squares	0.007570
Sy.x	0.03891
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	7

B.4 Representative IC₅₀ Calculation Using **34** as a Cathepsin K Inhibitor

GDK-II-63



Transform of E1 RESULT

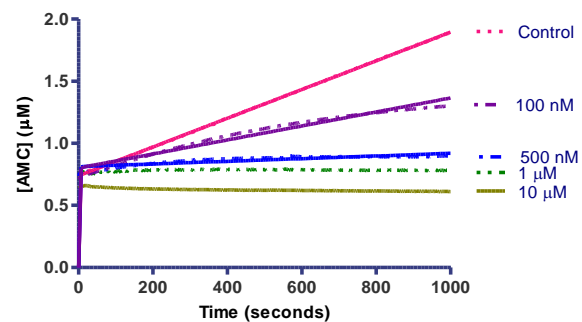


Concentration (M)	Slope E1
0.000010	0.007452
0.000001	0.113100
1.000000e-007	0.236700
5.000000e-008	0.247500
1.000000e-008	0.270900
1.000000e-009	0.265500
1.000000e-010	0.264900
1.000000e-011	0.263000
0.000000	0.272800

Concentration (M)	Slope E1
-5.000	0.027
-6.000	0.415
-7.000	0.868
-7.301	0.907
-8.000	0.993
-9.000	0.973
-10.000	0.971
-11.000	0.964
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-6.177
HillSlope	-1.001
EC50	6.650e-007
Std. Error	
LogEC50	0.04391
HillSlope	0.08421
95% Confidence Intervals	
LogEC50	-6.285 to -6.070
HillSlope	-1.207 to -0.7945
EC50	5.193e-007 to 8.517e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9945
Absolute Sum of Squares	0.004800
Sy.x	0.02828
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

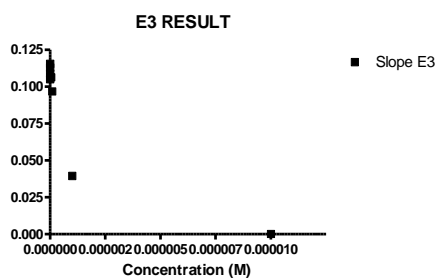
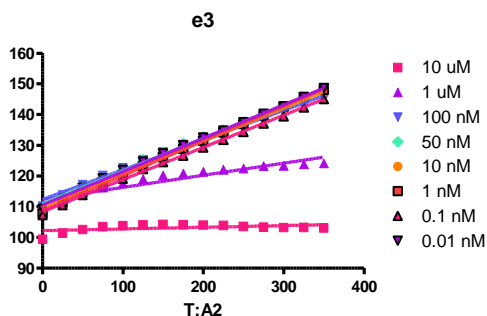
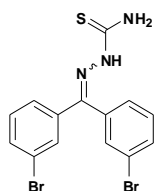
B.5 Representative IC₅₀ Calculation Using **18** as a Cathepsin K Inhibitor



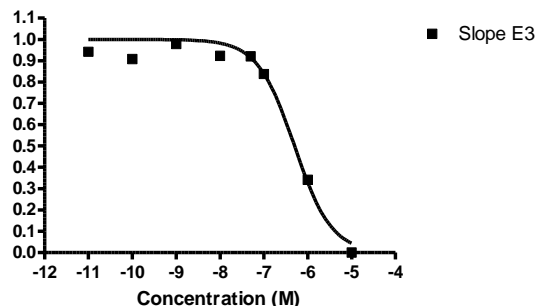
	Control	Control	10 μM	1 μM	500 nM	100 nM
Progress Curve	Ambiguous		Not converged	Not converged		
Best-fit values						
vs	0.001157				0.0001072	0.0005641
vi	~ 1.386				0.7186	0.5464
kobs	~ 1.873				0.8844	0.6809
Std. Error						
vs	7.686e-006				8.494e-006	9.655e-006
vi	~ 3.500				0.2047	0.1242
kobs	~ 4.736				0.2525	0.1554
95% Confidence Intervals						
vs	0.001142 to 0.001172				9.059e-005 to 0.0001239	0.0005452 to 0.0005830
vi	(Very wide)				0.3174 to 1.120	0.3030 to 0.7898
kobs	(Very wide)				0.3895 to 1.379	0.3762 to 0.9855
Goodness of Fit						
Degrees of Freedom	331				331	331
R square	0.9857				-0.4317	0.9101
Absolute Sum of Squares	0.5367				0.6537	0.8423
Sy.x	0.04027				0.04444	0.05044
Constraints						
vs	vs > 0.0		vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0		vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0		kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points						
Analyzed	334		334	334	334	334

B.6 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 50 μM (Preincubation Time: 0 minutes)

LJ-I-006



Transform of E3 RESULT



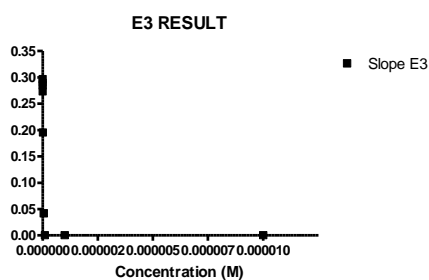
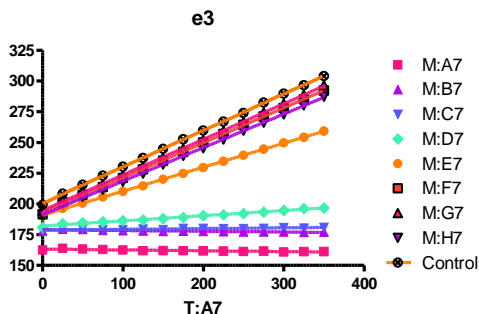
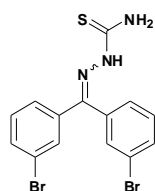
Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.039420
1.000000e-007	0.096690
5.000000e-008	0.106300
1.000000e-008	0.106600
1.000000e-009	0.113000
1.000000e-010	0.104900
1.000000e-011	0.108800
0.000000	0.115500

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.341
-7.000	0.837
-7.301	0.920
-8.000	0.923
-9.000	0.978
-10.000	0.908
-11.000	0.942
	1.000

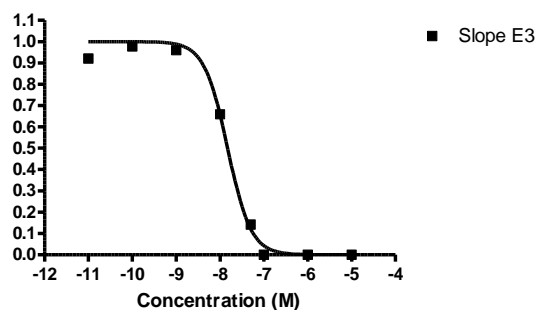
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-6.301
HillSlope	-1.027
EC50	4.996e-007
Std. Error	
LogEC50	0.08417
HillSlope	0.1550
95% Confidence Intervals	
LogEC50	-6.507 to -6.095
HillSlope	-1.406 to -0.6477
EC50	3.109e-007 to 8.027e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9804
Absolute Sum of Squares	0.01781
Sy,x	0.05449
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

B.7 Representative IC₅₀ Calculation Using **1** as a Cathepsin K Inhibitor (Preincubation Time: 0 minutes)

LJ-I-006



Transform of E3 RESULT



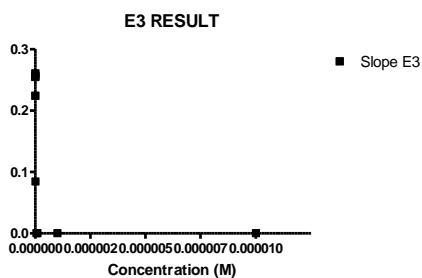
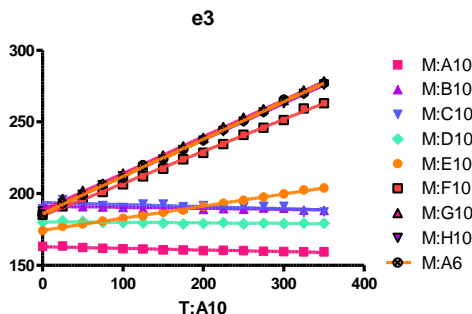
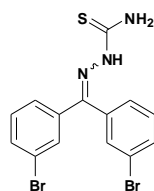
Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.041930
1.000000e-008	0.195400
1.000000e-009	0.284700
1.000000e-010	0.289800
1.000000e-011	0.273300
0.000000	0.296800

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.141
-8.000	0.658
-9.000	0.959
-10.000	0.976
-11.000	0.921
	1.000

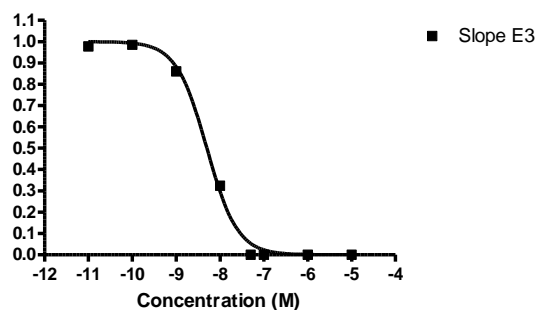
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.826
HillSlope	-1.631
EC50	1.494e-008
Std. Error	
LogEC50	0.04278
HillSlope	0.2266
95% Confidence Intervals	
LogEC50	-7.930 to -7.721
HillSlope	-2.185 to -1.076
EC50	1.174e-008 to 1.901e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9934
Absolute Sum of Squares	0.009853
Sy.x	0.04052
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

B.8 Representative IC₅₀ Calculation Using 1 as a Cathepsin K Inhibitor (Preincubation Time: 15 minutes)

LJ-I-006



Transform of E3 RESULT

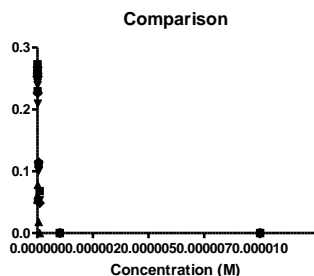


Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.000000
1.000000e-008	0.084340
1.000000e-009	0.224000
1.000000e-010	0.256500
1.000000e-011	0.254500
0.000000	0.260300

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.000
-8.000	0.324
-9.000	0.861
-10.000	0.985
-11.000	0.978
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-8.299
HillSlope	-1.250
EC50	5.024e-009
Std. Error	
LogEC50	0.04477
HillSlope	0.1188
95% Confidence Intervals	
LogEC50	-8.408 to -8.189
HillSlope	-1.541 to -0.9595
EC50	3.904e-009 to 6.466e-009
Goodness of Fit	
Degrees of Freedom	6
R square	0.9966
Absolute Sum of Squares	0.005143
Sy.x	0.02928
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

B.9 Representative IC₅₀ Calculation Using 1 as a Cathepsin K Inhibitor (Preincubation Time: 45 minute)

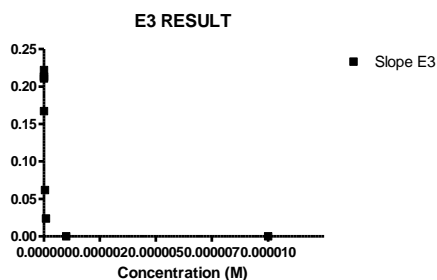
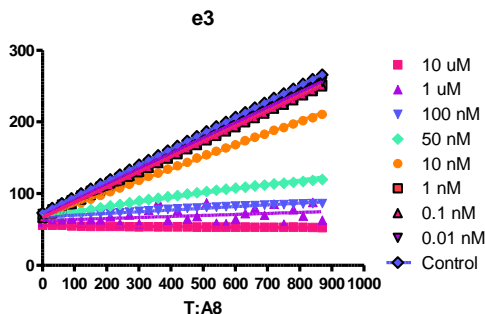
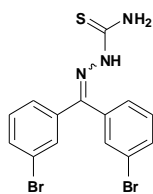


	Concentration (M)	Slope E1	Slope E2	Slope E3	Slope E4
	10.000	0.000	0.000	0.000	0.000
	1.000	0.000	0.000	0.000	0.000
	0.100	0.248	0.000	0.207	0.182
	0.050	0.409	0.295	0.392	0.428
	0.010	0.841	1.272	0.816	0.846
	0.001	0.946	0.921	0.934	0.944
	1.000e-004	0.948	0.891	0.949	0.948
	1.000e-005	0.949	0.876	0.957	0.964
	0.000	1.000	1.000	1.000	1.000

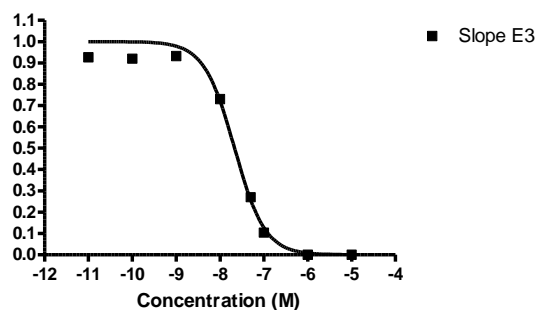
	Slope E1	Slope E2	Slope E3	Slope E4
Morrison Ki				
Best-fit values				
Vo	= 1.000	= 1.000	= 1.000	= 1.000
Et	= 0.0015	= 0.0015	= 0.0015	= 0.0015
Ki	0.006857	0.006632	0.006050	0.006490
S	= 50.00	= 50.00	= 50.00	= 50.00
Km	= 11.80	= 11.80	= 11.80	= 11.80
Std. Error				
Ki	0.0006982	0.003949	0.0006160	0.0008231
95% Confidence Intervals				
Ki	0.005247 to 0.008467	-0.002474 to 0.01574	0.004630 to 0.007471	0.004592 to 0.008388
Goodness of Fit				
Degrees of Freedom	8	8	8	8
R square	0.9929	0.8281	0.9932	0.9895
Absolute Sum of Squares	0.01036	0.3519	0.01008	0.01589
Sy.x	0.03599	0.2097	0.03550	0.04457
Constraints				
Vo	Vo = 1.000	Vo = 1.000	Vo = 1.000	Vo = 1.000
Et	Et = 0.0015	Et = 0.0015	Et = 0.0015	Et = 0.0015
S	S = 50.00	S = 50.00	S = 50.00	S = 50.00
Km	Km = 11.80	Km = 11.80	Km = 11.80	Km = 11.80
Number of points				
Analyzed	9	9	9	9

B.10 Calculation of K_I^{app} Using **1** as a Cathepsin K Inhibitor (Preincubation Time: 5 minutes)

LJ-I-006



Transform of E3 RESULT

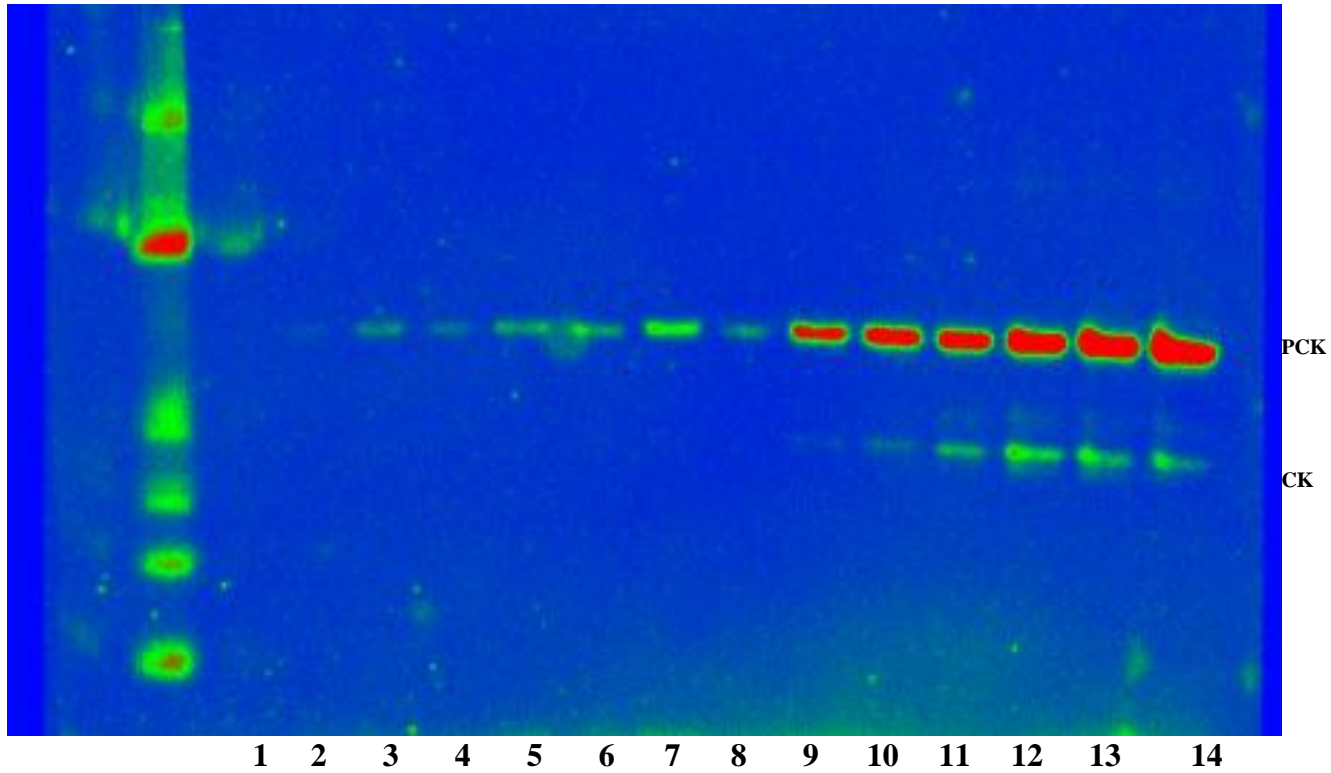


Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.023880
5.000000e-008	0.061840
1.000000e-008	0.167200
1.000000e-009	0.213400
1.000000e-010	0.210800
1.000000e-011	0.212300
0.000000	0.221900

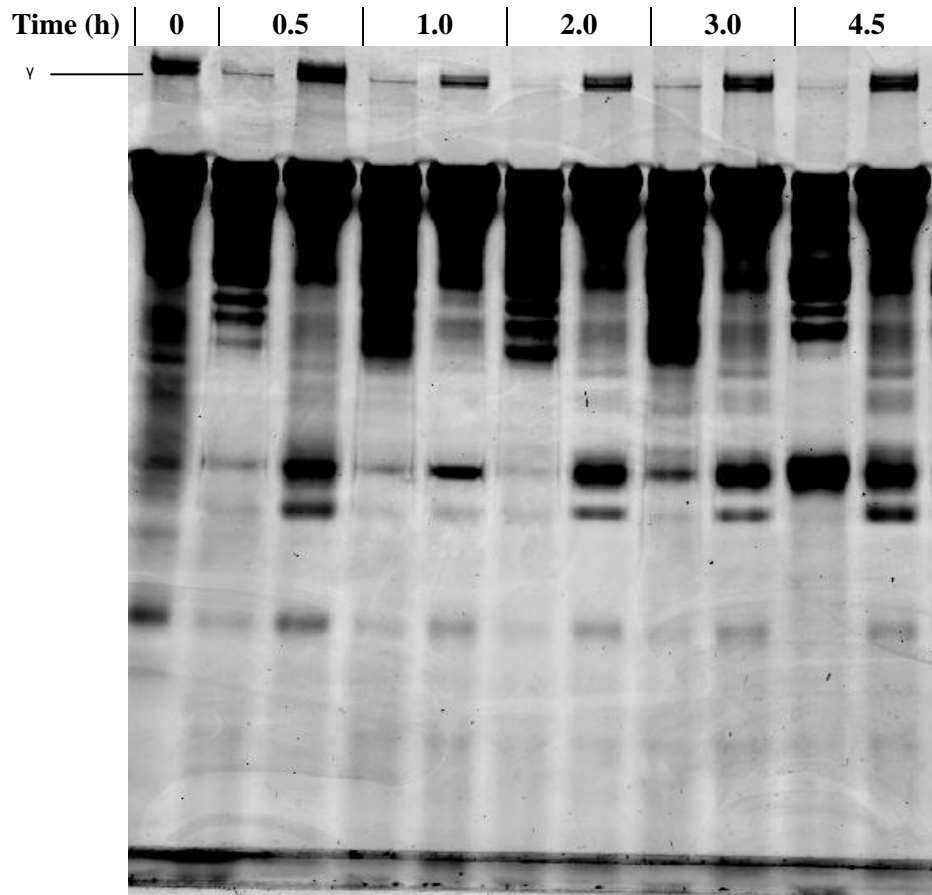
Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.104
-7.301	0.270
-8.000	0.730
-9.000	0.931
-10.000	0.920
-11.000	0.927
	0.969

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.668
HillSlope	-1.249
EC50	2.148e-008
Std. Error	
LogEC50	0.05853
HillSlope	0.1739
95% Confidence Intervals	
LogEC50	-7.811 to -7.525
HillSlope	-1.675 to -0.8235
EC50	1.545e-008 to 2.988e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9888
Absolute Sum of Squares	0.01461
Sy.x	0.04934
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

B.11 Representative IC₅₀ Calculation Using 1 as a Cathepsin K Inhibitor (Preincubation Time: 5 minutes, [S]: 10 μM)

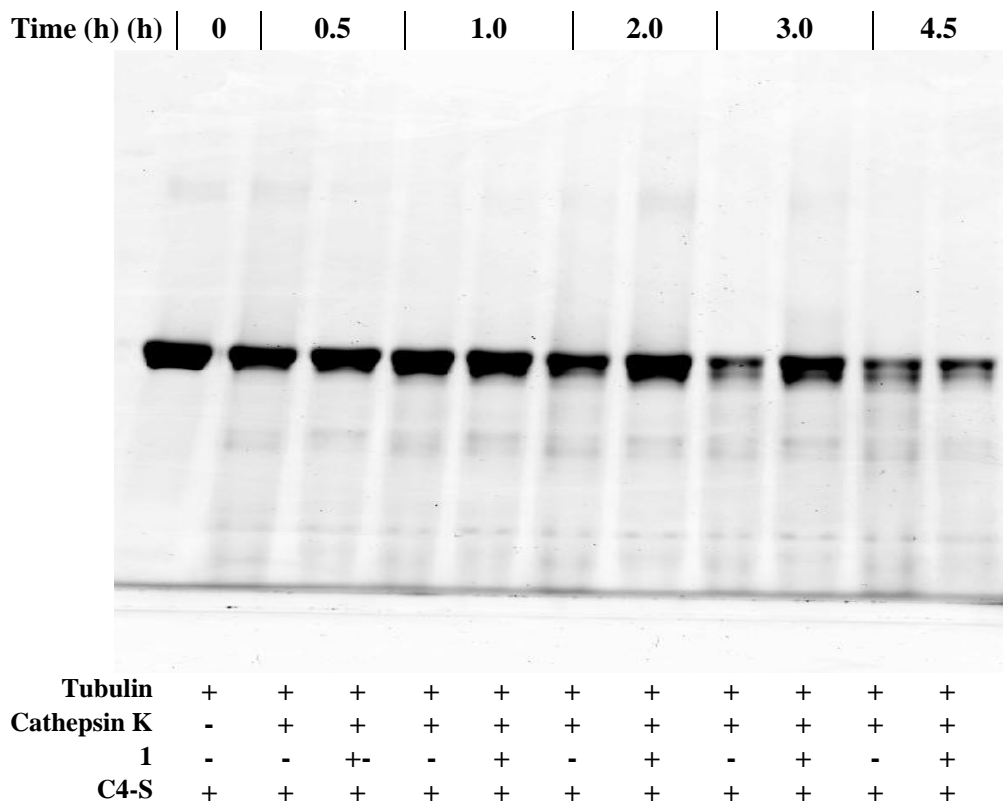


B.12 Western Blotting of the Detection of Procathepsin K. Legend: **1:** 5 ng (not detected); **2:** 10 ng; **3:** 15 ng; **4:** 20 ng; **5:** 25 ng; **6:** 30 ng; **7:** 35 ng; **8:** 40 ng; **9:** 45 ng; **10:** 55 ng; **11:** 65 ng; **12:** 100 ng; **13:** 150 ng; **14:** 200 ng; PCK: procathepsin K (mW~ 43 kDa); CK: active cathepsin K (MW ~29 kDa)



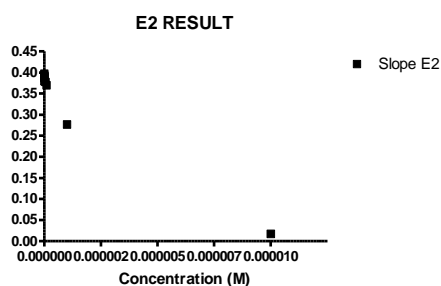
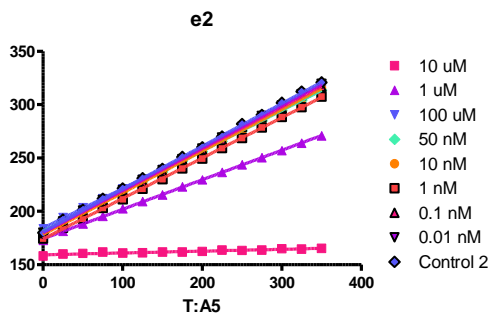
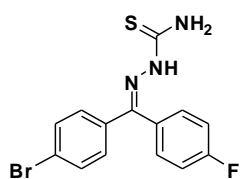
Collagen IV	+	+	+	+	+	+	+	+	+	+
Cathepsin K	-	+	+	+	+	+	+	+	+	+
1	-	-	+/-	-	+	-	+	-	+	-
C4-S	+	+	+	+	+	+	+	+	+	+

B13. Inhibition of collagenase IV activity of cathepsin K by **1**, preincubation time: 0 hours.

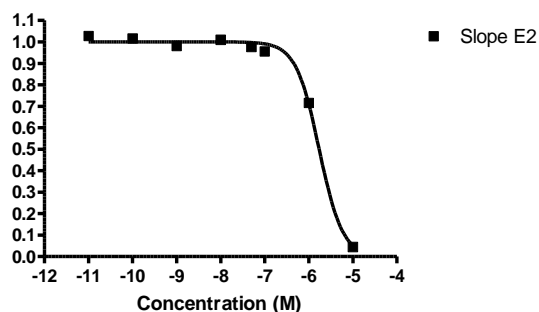


B14. Inhibition of proteolytic activity of cathepsin K by **1**, preincubation time: 0 hours

GDK-II-33



Transform of E2 RESULT



Concentration (M)	Slope E2
0.000010	0.017250
0.000001	0.276700
1.000000e-007	0.369400
5.000000e-008	0.377600
1.000000e-008	0.390500
1.000000e-009	0.379400
1.000000e-010	0.392800
1.000000e-011	0.397200
0.000000	0.386700

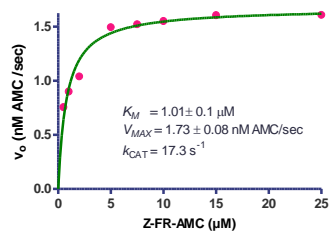
Concentration (M)	Slope E2
-5.000	0.045
-6.000	0.716
-7.000	0.955
-7.301	0.976
-8.000	1.010
-9.000	0.981
-10.000	1.016
-11.000	1.027
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-5.762
HillSlope	-1.649
EC50	1.730e-006
Std. Error	
LogEC50	0.03681
HillSlope	0.2037
95% Confidence Intervals	
LogEC50	-5.852 to -5.672
HillSlope	-2.148 to -1.151
EC50	1.406e-006 to 2.129e-006
Goodness of Fit	
Degrees of Freedom	6
R square	0.9959
Absolute Sum of Squares	0.003222
Sy.x	0.02317
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

B.15 Representative IC₅₀ Calculation Using **48** as a Cathepsin K Inhibitor

APPENDIX C

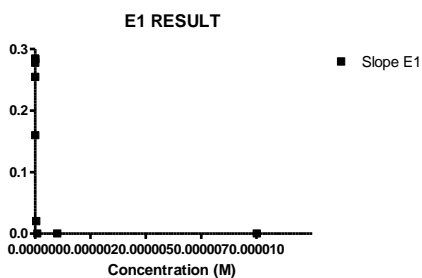
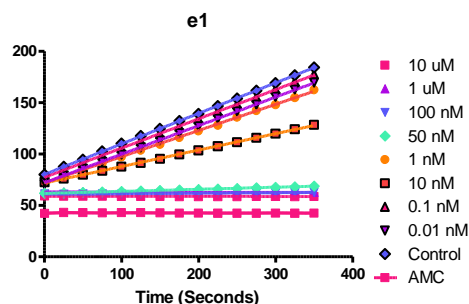
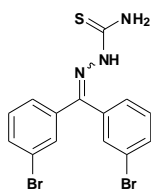
In Vitro Evaluations of Thiosemicarbazones as Inhibitors of Cruzain



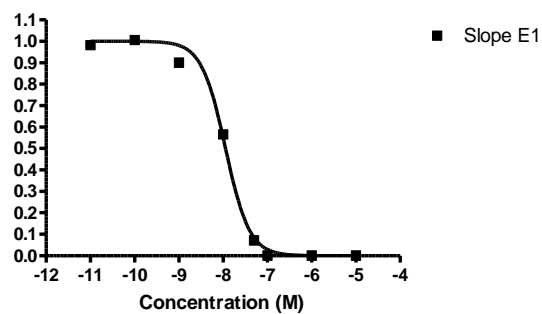
Substrate Concentration (uM)	2 A
0.50000	0.756
1.00000	0.902
2.00000	1.041
5.00000	1.496
7.50000	1.523
10.00000	1.553
15.00000	1.606
25.00000	1.610

	2 A
Michaelis-Menten	
Best-fit values	
Vmax	1.672
Km	0.8077
Std. Error	
Vmax	0.05042
Km	0.1279
95% Confidence Intervals	
Vmax	1.549 to 1.795
Km	0.4947 to 1.121
Goodness of Fit	
Degrees of Freedom	6
R square	0.9530
Absolute Sum of Squares	0.04052
Sy.x	0.08218
Constraints	
Km	Km > 0.0
Number of points	
Analyzed	8

C.1 Calculation of Michaelis-Menten Constant for Cruzain using Z-FR-AMC as a Substrate



Transform of E1 RESULT

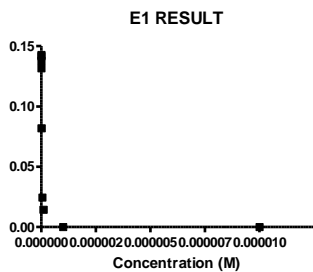
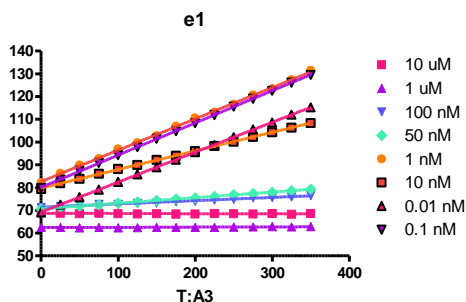
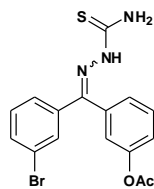


0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.020070
1.000000e-008	0.160000
1.000000e-009	0.254600
1.000000e-010	0.284600
1.000000e-011	0.277700
0.000000	0.283100

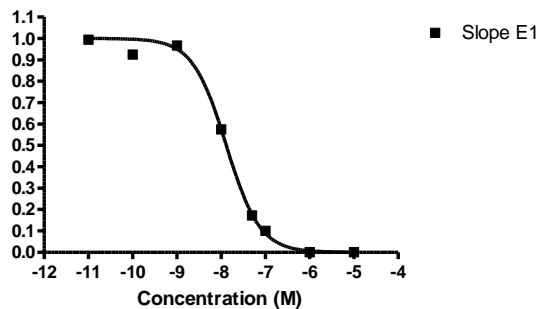
Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.071
-8.000	0.565
-9.000	0.899
-10.000	1.005
-11.000	0.981
	1.000

Slope E1	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.944
HillSlope	-1.611
EC50	1.137e-008
Std. Error	
LogEC50	0.03819
HillSlope	0.2613
95% Confidence Intervals	
LogEC50	-8.038 to -7.851
HillSlope	-2.250 to -0.9711
EC50	9.165e-009 to 1.409e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9947
Absolute Sum of Squares	0.008201
Sy.x	0.03697
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.2 Representative IC₅₀ Calculation Using 1 as a Cruzain Inhibitor



Transform of E1 RESULT

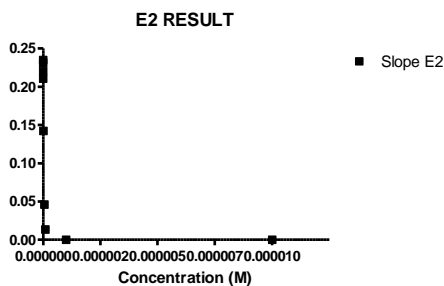
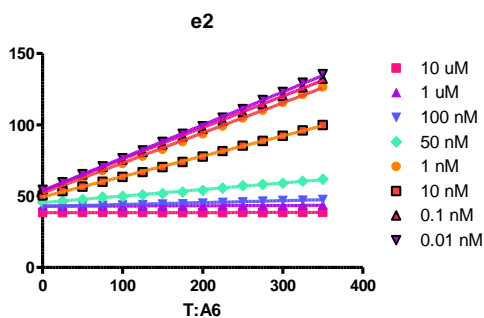
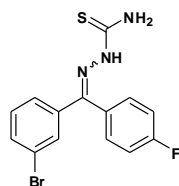


0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.014260
5.000000e-008	0.024420
1.000000e-008	0.081800
1.000000e-009	0.137700
1.000000e-010	0.131600
1.000000e-011	0.141500
0.000000	0.142400

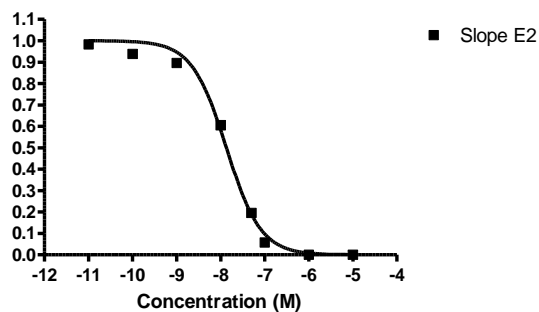
Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.000
-7.000	0.100
-7.301	0.171
-8.000	0.574
-9.000	0.967
-10.000	0.924
-11.000	0.994
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.883
HillSlope	-1.146
EC50	1.310e-008
Std. Error	
LogEC50	0.04018
HillSlope	0.1086
95% Confidence Intervals	
LogEC50	-7.981 to -7.784
HillSlope	-1.412 to -0.8806
EC50	1.045e-008 to 1.643e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9959
Absolute Sum of Squares	0.005736
Sy.x	0.03092
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.3 Representative IC₅₀ Calculation using 9 as a Cruzain Inhibitor



Transform of E2 RESULT

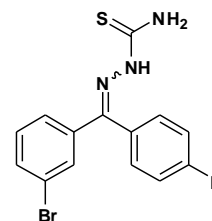
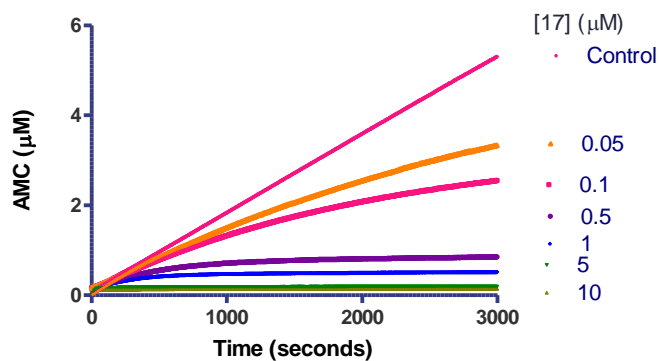


Concentration (M)	Slope E2
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.013450
5.000000e-008	0.046040
1.000000e-008	0.142400
1.000000e-009	0.210800
1.000000e-010	0.220600
1.000000e-011	0.231300
0.000000	0.235200

Concentration (M)	Slope E2
-5.000	0.000
-6.000	0.000
-7.000	0.057
-7.301	0.196
-8.000	0.605
-9.000	0.896
-10.000	0.938
-11.000	0.983
	1.000

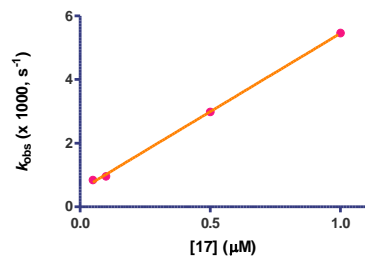
Slope E2	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.869
HillSlope	-1.110
EC50	1.351e-008
Std. Error	
LogEC50	0.04964
HillSlope	0.1253
95% Confidence Intervals	
LogEC50	-7.991 to -7.748
HillSlope	-1.417 to -0.8037
EC50	1.022e-008 to 1.787e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9938
Absolute Sum of Squares	0.008485
Sy.x	0.03761
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.4 Representative IC₅₀ Calculation Using 17 as a Cruzain Inhibitor



	Control	Control	10 µM	5 µM	1 µM	0.5 µM	0.1 µM	0.05 µM
Progress Curve	Ambiguous							
Best-fit values								
vs	0.001741		4.533e-006	7.658e-006	2.832e-005	5.803e-005	0.0003832	0.0007004
vi	~ 0.7972		0.04930	0.01915	0.002535	0.002149	0.001920	0.001950
kobs	~ 7.790		0.3660	0.1090	0.005666	0.003036	0.001029	0.0009268
Std. Error								
vs	4.190e-007		1.773e-007	2.679e-007	9.089e-007	1.643e-006	9.676e-006	1.208e-005
vi	~ 4.648e+007		0.003839	0.0009045	2.909e-005	1.646e-005	9.352e-006	9.232e-006
kobs	~ 4.552e+008		0.02857	0.005192	7.804e-005	3.429e-005	2.166e-005	2.695e-005
95% Confidence Intervals								
vs	0.001740 to 0.001742		4.185e-006 to 4.880e-006	7.133e-006 to 8.183e-006	2.654e-005 to 3.010e-005	5.481e-005 to 6.125e-005	0.0003642 to 0.0004021	0.0006768 to 0.0007241
vi	(Very wide)		0.04178 to 0.05683	0.01738 to 0.02092	0.002478 to 0.002592	0.002117 to 0.002181	0.001902 to 0.001938	0.001932 to 0.001968
kobs	(Very wide)		0.3100 to 0.4220	0.09886 to 0.1192	0.005513 to 0.005819	0.002968 to 0.003103	0.0009870 to 0.001072	0.0008740 to 0.0009791
Goodness of Fit								
Degrees of Freedom	997		997	997	997	997	997	997
R square	0.9999		0.2366	0.4979	0.9524	0.9848	0.9982	0.9989
Absolute Sum of Squares	0.1309		0.02324	0.05176	0.2932	0.4443	0.8554	0.9273
Sy,x	0.01146		0.004828	0.007205	0.01715	0.02111	0.02929	0.03050
Constraints								
vs	vs > 0.0		vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0	vs > 0.0
vi	vi > 0.0		vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0	vi > 0.0
kobs	kobs > 0.0		kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0	kobs > 0.0
Number of points								
Analyzed	1000		1000	1000	1000	1000	1000	1000

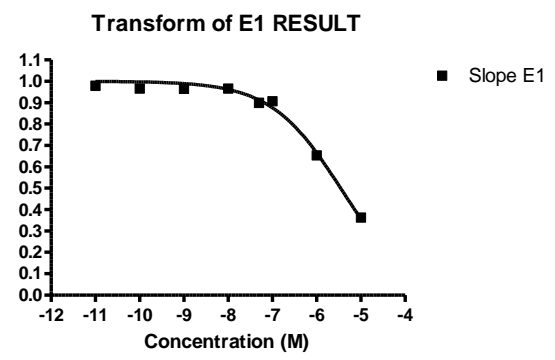
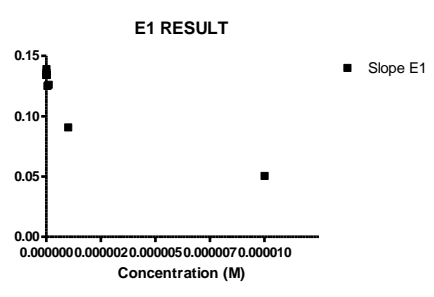
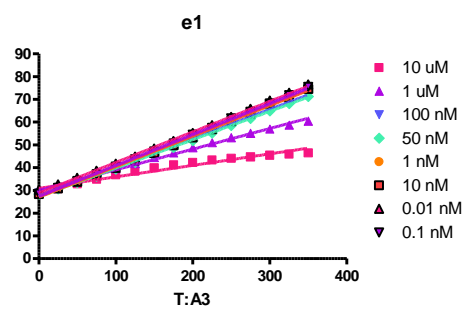
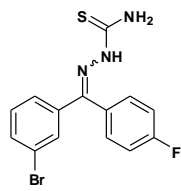
C.5 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [S]: 15 µM (Preincubation Time: 0 minutes)



[I]	kobs		
	Mean	SEM	N
1.000	5.461	0.408	3
0.500	2.984	0.103	3
0.100	0.960	0.040	3
0.050	0.843	0.054	3

kobs	
Best-fit values	
Slope	4.922 ± 0.08528
Y-intercept when X=0.0	0.5318 ± 0.04791
X-intercept when Y=0.0	-0.1080
1/slope	0.2032
95% Confidence Intervals	
Slope	4.555 to 5.289
Y-intercept when X=0.0	0.3256 to 0.7379
X-intercept when Y=0.0	-0.1593 to -0.06262
Goodness of Fit	
R square	0.9994
Sy.x	0.06505
Is slope significantly non-zero?	
F	3331
DFn, DFd	1.000, 2.000
P value	0.0003
Deviation from zero?	
Significant	
Data	
Number of X values	4
Maximum number of Y replicates	1
Total number of values	4
Number of missing values	0

C.6 Plot of k_{obs} vs [17] Using Linear Regression. (Preincubation Time: 0 minutes)

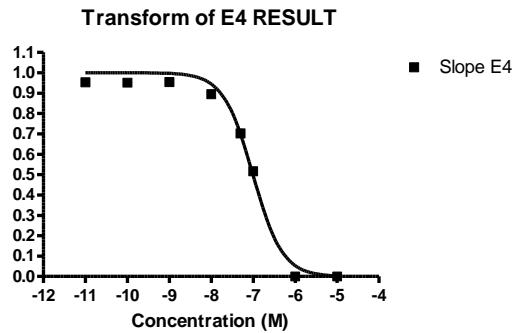
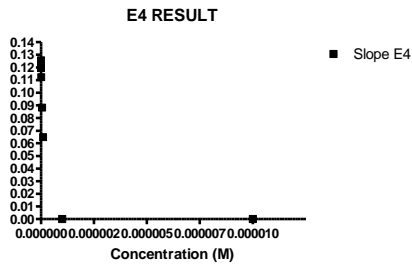
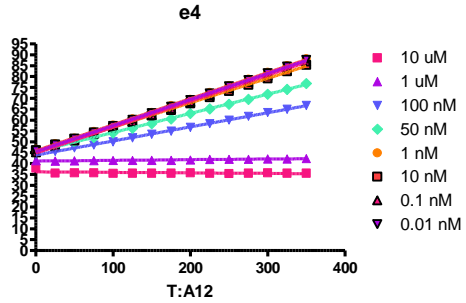
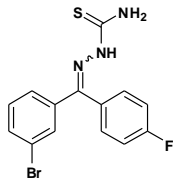


0.000010	0.050430
0.000001	0.090750
1.000000e-007	0.126000
5.000000e-008	0.124900
1.000000e-008	0.134200
1.000000e-009	0.134100
1.000000e-010	0.134200
1.000000e-011	0.136000
0.000000	0.139000

Concentration (M)	Slope E1
-5.000	0.363
-6.000	0.653
-7.000	0.906
-7.301	0.899
-8.000	0.965
-9.000	0.965
-10.000	0.965
-11.000	0.978
	1.000

Slope E1	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-5.457
HillSlope	-0.5508
EC50	3.490e-006
Std. Error	
LogEC50	0.05655
HillSlope	0.03711
95% Confidence Intervals	
LogEC50	-5.595 to -5.319
HillSlope	-0.6416 to -0.4600
EC50	2.538e-006 to 4.800e-006
Goodness of Fit	
Degrees of Freedom	6
R square	0.9902
Absolute Sum of Squares	0.003289
Sy.x	0.02341
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.7 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 0 minutes)

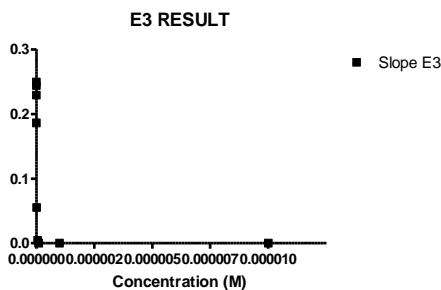
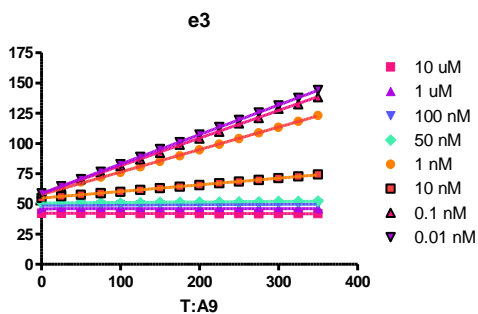
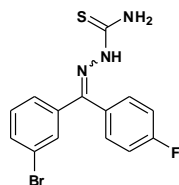


Concentration (M)	Slope E4
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.064850
5.000000e-008	0.088190
1.000000e-008	0.112300
1.000000e-009	0.119800
1.000000e-010	0.119500
1.000000e-011	0.119700
0.000000	0.125600

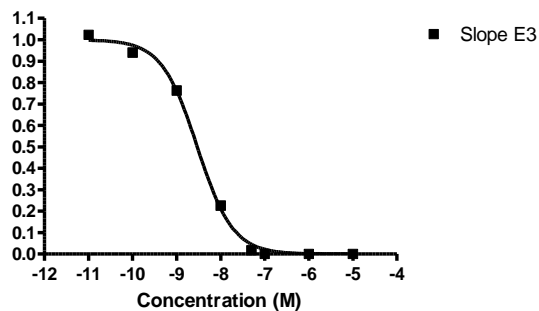
Concentration (M)	Slope E4
-5.000	0.000
-6.000	0.000
-7.000	0.516
-7.301	0.702
-8.000	0.894
-9.000	0.954
-10.000	0.951
-11.000	0.953
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.005
HillSlope	-1.238
EC50	9.887e-008
Std. Error	
LogEC50	0.05165
HillSlope	0.2217
95% Confidence Intervals	
LogEC50	-7.131 to -6.879
HillSlope	-1.780 to -0.6950
EC50	7.390e-008 to 1.323e-007
Goodness of Fit	
Degrees of Freedom	6
R square	0.9897
Absolute Sum of Squares	0.01225
Sy.x	0.04518
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.8 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 1 minute)



Transform of E3 RESULT

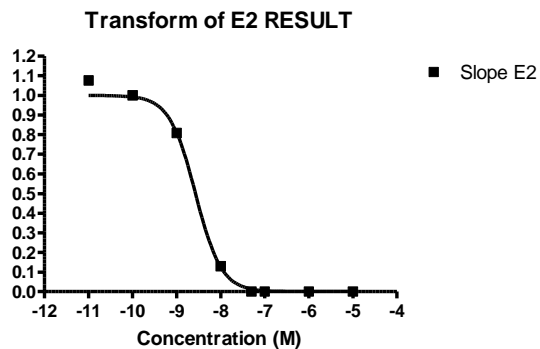
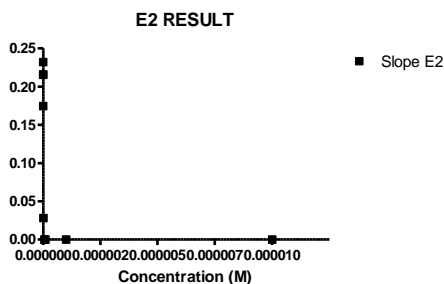
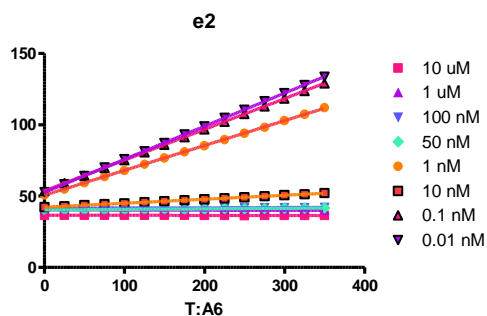
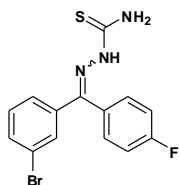


Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.004460
1.000000e-008	0.055080
1.000000e-009	0.186100
1.000000e-010	0.229200
1.000000e-011	0.249400
0.000000	0.243800

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.018
-8.000	0.226
-9.000	0.763
-10.000	0.940
-11.000	1.023
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-8.530
HillSlope	-1.075
EC50	2.955e-009
Std. Error	
LogEC50	0.03718
HillSlope	0.07053
95% Confidence Intervals	
LogEC50	-8.620 to -8.439
HillSlope	-1.247 to -0.9021
EC50	2.396e-009 to 3.643e-009
Goodness of Fit	
Degrees of Freedom	6
R square	0.9978
Absolute Sum of Squares	0.003238
Sy.x	0.02323
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.9 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 30 minutes)

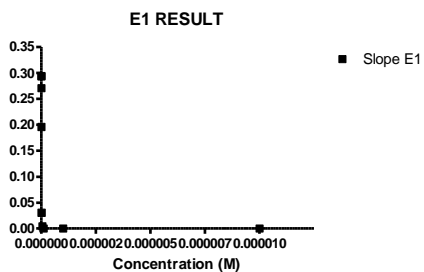
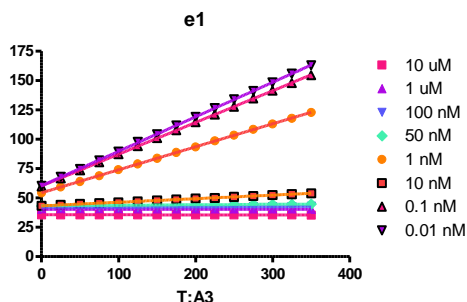
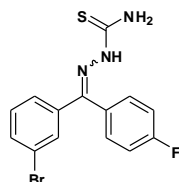


Concentration (M)	Slope E2
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.000000
1.000000e-008	0.028020
1.000000e-009	0.174400
1.000000e-010	0.215700
1.000000e-011	0.232200
0.000000	0.215600

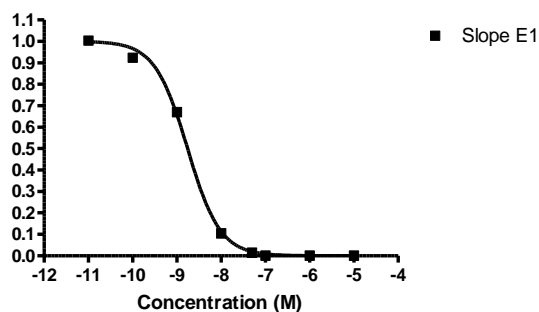
Concentration (M)	Slope E2
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.000
-8.000	0.130
-9.000	0.809
-10.000	1.000
-11.000	1.077
	1.000

Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-8.571
HillSlope	-1.474
EC50	2.687e-009
Std. Error	
LogEC50	0.05098
HillSlope	0.1523
95% Confidence Intervals	
LogEC50	-8.696 to -8.446
HillSlope	-1.847 to -1.101
EC50	2.016e-009 to 3.581e-009
Goodness of Fit	
Degrees of Freedom	6
R square	0.9963
Absolute Sum of Squares	0.006256
Sy.x	0.03229
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.10 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 60 minutes)



Transform of E1 RESULT



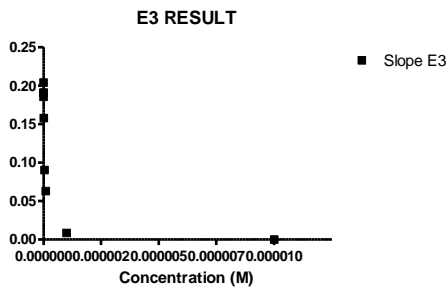
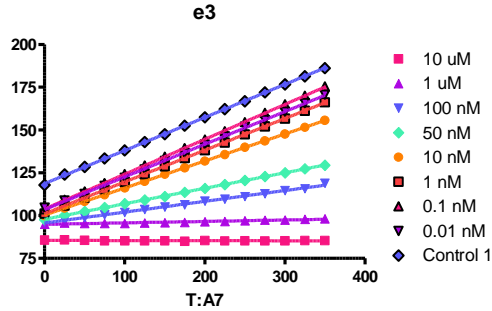
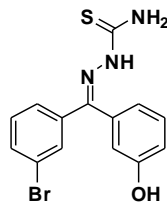
0.000010	0.000000
0.000001	0.000000
1.000000e-007	0.000000
5.000000e-008	0.004140
1.000000e-008	0.030410
1.000000e-009	0.196000
1.000000e-010	0.270400
1.000000e-011	0.293800
0.000000	0.292900

Concentration (M)	Slope E1
-5.000	0.000
-6.000	0.000
-7.000	0.000
-7.301	0.014
-8.000	0.104
-9.000	0.669
-10.000	0.923
-11.000	1.003
	1.000

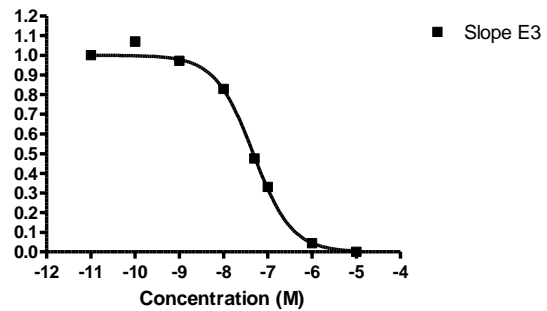
Slope E1	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-8.758
HillSlope	-1.172
EC50	1.744e-009
Std. Error	
LogEC50	0.02908
HillSlope	0.07936
95% Confidence Intervals	
LogEC50	-8.830 to -8.687
HillSlope	-1.366 to -0.9779
EC50	1.481e-009 to 2.055e-009
Goodness of Fit	
Degrees of Freedom	6
R square	0.9984
Absolute Sum of Squares	0.002223
Sy.x	0.01925
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points	
Analyzed	8

C.11 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 120 minutes)

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Transform of E3 RESULT

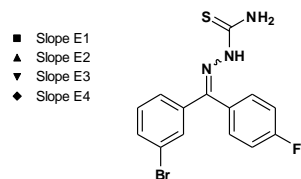
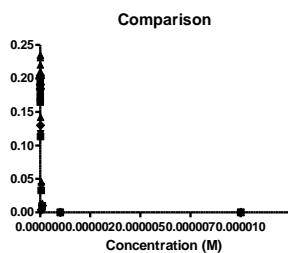


Concentration (M)	Slope E3
0.000010	0.000000
0.000001	0.008498
1.000000e-007	0.063020
5.000000e-008	0.090610
1.000000e-008	0.158100
1.000000e-009	0.185500
1.000000e-010	0.204100
1.000000e-011	0.191000
0.000000	0.190700

Concentration (M)	Slope E3
-5.000	0.000
-6.000	0.045
-7.000	0.330
-7.301	0.475
-8.000	0.829
-9.000	0.973
-10.000	1.070
-11.000	1.002
	1.000

Slope E3	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-7.326
HillSlope	-1.000
EC50	4.722e-008
Std. Error	
LogEC50	0.03631
HillSlope	0.09587
95% Confidence Intervals	
LogEC50	-7.415 to -7.237
HillSlope	-1.235 to -0.7657
EC50	3.849e-008 to 5.794e-008
Goodness of Fit	
Degrees of Freedom	6
R square	0.9958
Absolute Sum of Squares	0.005528
Sy.x	0.03035
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

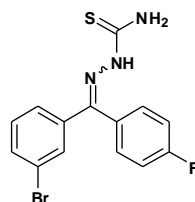
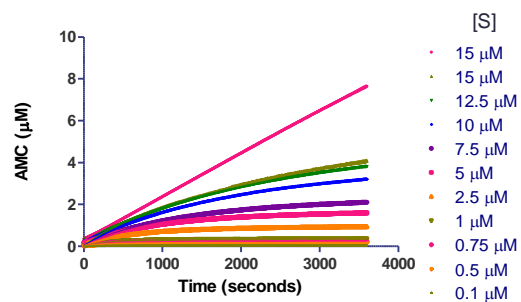
C.12 Representative IC₅₀ Calculation Using **17** as a Cruzain Inhibitor (Preincubation Time: 240 minutes)



Concentration (M)	Slope E1	Slope E2	Slope E3	Slope E4
10.000	0.000	0.000	0.000	0.000
1.000	0.000	0.000	0.000	0.000
0.100	0.048	0.057	0.052	0.051
0.050	0.166	0.196	0.207	0.020
0.010	0.572	0.605	0.634	0.638
0.001	0.832	0.896	0.962	0.904
1.000e-004	0.873	0.938	0.938	0.937
1.000e-005	0.932	0.983	0.993	0.983
0.000	1.000	1.000	1.000	1.000

	Slope E1	Slope E2	Slope E3	Slope E4
Morrison Ki				
Best-fit values				
Vo	= 1.000	= 1.000	= 1.000	= 1.000
Et	= 1.000e-004	= 1.000e-004	= 1.000e-004	= 1.000e-004
Ki	0.0006741	0.0008138	0.0009111	0.0006836
S	= 15.00	= 15.00	= 15.00	= 15.00
Km	= 1.010	= 1.010	= 1.010	= 1.010
Std. Error				
Ki	0.0001346	8.783e-005	0.0001069	0.0001633
95% Confidence Intervals				
Ki	0.0003637 to 0.0009846	0.0006113 to 0.001016	0.0006646 to 0.001158	0.0003070 to 0.001060
Goodness of Fit				
Degrees of Freedom	8	8	8	8
R square	0.9795	0.9942	0.9932	0.9752
Absolute Sum of Squares	0.03090	0.009465	0.01149	0.04438
Sy.x	0.06215	0.03440	0.03790	0.07449
Constraints				
Vo	Vo = 1.000	Vo = 1.000	Vo = 1.000	Vo = 1.000
Et	Et = 1.000e-004	Et = 1.000e-004	Et = 1.000e-004	Et = 1.000e-004
S	S = 15.00	S = 15.00	S = 15.00	S = 15.00
Km	Km = 1.010	Km = 1.010	Km = 1.010	Km = 1.010
Number of points				
Analyzed	9	9	9	9

C.13 Calculation of K_I^{app} Using **17** as a Cruzain Inhibitor (Preincubation Time: 5 minutes)

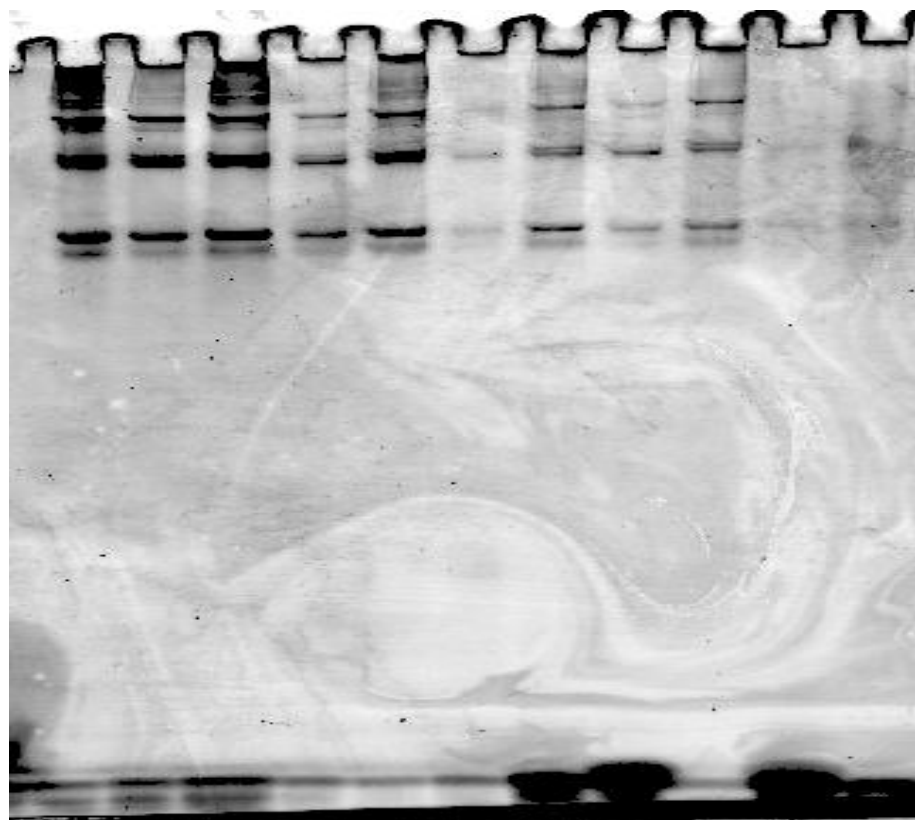


$$Y = v_s * x + ((v_i - v_s) / k_{obs}) * (1 - \exp(-k_{obs} * x))$$

	15 µM	15 µM	15 µM	12.5 µM	10 µM	7.5 µM	5 µM	2.5 µM	1 µM	0.75 µM	0.5 µM	0.1 µM
Progress Curve	Ambiguous											
Best-Fit values												
v_s	0.002052	0.0006798	0.0005365	0.0003764	0.0001708	0.0001053	0.0001053	5.128e-005	1.554e-005	1.083e-005	8.102e-006	2.082e-006
v_i	-3.240	0.002798	0.002738	0.002475	0.001988	0.001356	0.001607	0.001607	0.001021	0.0007539	0.0005661	0.0002876
k_{obs}	-9.855	0.001235	0.001114	0.001085	0.001190	0.001484	0.002021	0.002021	0.003379	0.003689	0.004712	0.006389
Std. Error												
v_s	1.823e-006	1.223e-005	1.243e-005	1.032e-005	5.766e-006	3.248e-006	1.281e-006	2.292e-007	3.312e-007	2.292e-007	1.637e-007	5.943e-008
v_i	-7.775e+008	2.676e-005	1.690e-005	1.690e-005	1.160e-005	1.081e-005	8.138e-006	5.808e-006	5.132e-006	5.232e-006	3.202e-006	3.202e-006
k_{obs}	-2.366e+009	3.650e-005	2.825e-005	2.320e-005	1.839e-005	1.728e-005	1.701e-005	2.516e-005	3.141e-005	4.316e-005	8.701e-005	8.701e-005
95% Confidence Intervals												
v_s	0.002049 to 0.002056	0.000559 to 0.0007038	0.0005121 to 0.0005608	0.0003562 to 0.0003966	0.0001595 to 0.0001821	9.891e-005 to 0.0001116	4.877e-005 to 5.379e-005	1.489e-005 to 1.619e-005	1.038e-005 to 1.127e-005	7.781e-006 to 8.423e-006	1.966e-006 to 2.199e-006	1.966e-006 to 2.199e-006
v_i	(Very wide)	0.002744 to 0.002849	0.002696 to 0.002780	0.002442 to 0.002508	0.001966 to 0.002011	0.001834 to 0.001977	0.001591 to 0.001623	0.001009 to 0.001032	0.0007838 to 0.0008039	0.0006758 to 0.0006963	0.0002813 to 0.0002738	0.0002813 to 0.0002738
k_{obs}	(Very wide)	0.001164 to 0.001307	0.001059 to 0.001169	0.001039 to 0.001130	0.001154 to 0.001226	0.001460 to 0.001528	0.001988 to 0.002054	0.003330 to 0.003429	0.003788 to 0.003921	0.004628 to 0.004797	0.006219 to 0.006560	0.006219 to 0.006560
Goodness of Fit												
Degrees of Freedom	718	718	718	718	718	718	718	718	718	718	718	718
R square	0.9998	0.9960	0.9966	0.9969	0.9968	0.9962	0.9958	0.9930	0.9906	0.9867	0.9825	0.9825
Absolute Sum of Squares	0.5637	3.326	2.499	1.585	0.6594	0.4037	0.1312	0.02077	0.01158	0.007120	0.001161	0.001161
Sy.x	0.02802	0.05806	0.05899	0.04698	0.03031	0.02371	0.01352	0.005378	0.004016	0.003149	0.001271	0.001271
Constraints												
v_s	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$	$v_s > 0.0$
v_i	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$	$v_i > 0.0$
k_{obs}	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$	$k_{obs} > 0.0$
Number of points												
Analyzed	721	721	721	721	721	721	721	721	721	721	721	721

C.14 Calculation of k_{obs} , v_s , and v_o Using Nonlinear Regression. [I]: 100 nM (Preincubation Time: 0 minutes)

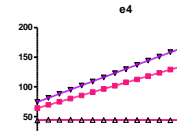
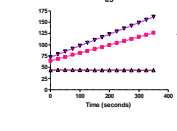
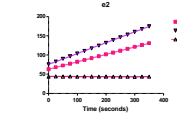
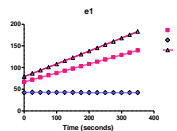
Time (h) 0 0.25 1.5 3.0 5.0 7.0



Collagen	+	+	+	+	+	+	+	+	+	+	+
Cruzain	-	+	+	+	+	+	+	+	+	+	+
17	-	-	+	-	+	-	+	-	+	-	+

C15. Inhibition of collagenase activity of Cruzain by **17**, preincubation time: 0.5 hours.

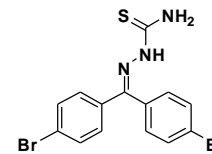
Best fit values	10 uM		Control	AMC
	Slope	0.2077 ± 0.0004500	0.2070 ± 0.0007883	-0.0009435 ± 0.0005039
Y-intercept when X=0.0	66.89 ± 0.00252	78.82 ± 0.1621	42.82 ± 0.1036	
X-intercept when Y=0.0	-322.0	-265.8	45380	
1/slope	4.814	3.367	-1060	
95% Confidence Intervals				
Slope	0.2067 to 0.2087	0.2953 to 0.2887	-0.002032 to 0.0001448	
Y-intercept when X=0.0	66.68 to 67.07	78.57 to 79.27	42.59 to 43.04	
X-intercept when Y=0.0	-324.3 to -318.6	-288.4 to -263.2	21156 to infinity	
Goodness of Fit				
R square	0.9999	0.9999	0.0214	
Sy.x	0.1882	0.3268	0.2108	
Is slope significantly non-zero?				
F	213079	141014	3.506	
DFs, DFd	1.000, 13.000	1.000, 13.000	1.000, 13.000	
P value	< 0.0001	< 0.0001	0.0838	
Deviation from zero?	Significant	Significant	Not Significant	
Data				
Number of X values	15	15	15	
Maximum number of Y replicates	1	1	1	
Total number of values	15	15	15	
Number of missing values	0	0	0	



Time (seconds)	10 uM			Control			AMC		
	0.00	67.148	78.385	42.290	0.00	62.879	75.813	44.003	
24.94	72.179	80.785	43.866	24.97	68.869	83.028	44.322		
49.95	77.404	83.897	42.870	49.97	72.526	90.086	44.131		
74.94	82.522	101.075	42.813	74.97	77.311	96.568	44.207		
99.94	87.339	108.271	42.846	99.97	82.031	103.687	43.806		
124.94	92.704	115.845	42.993	124.98	86.922	110.455	44.205		
149.94	97.830	123.057	42.695	149.97	91.766	117.451	43.903		
174.94	102.992	130.659	42.654	174.97	96.667	124.559	43.786		
199.94	108.242	138.004	42.506	199.97	101.556	131.262	43.321		
224.94	113.593	145.477	42.477	224.97	106.233	138.039	43.719		
249.94	118.726	153.105	42.443	249.97	111.217	146.029	43.642		
274.94	123.992	160.480	42.444	274.98	116.081	153.131	43.525		
299.94	129.292	167.936	42.625	299.97	121.193	160.275	43.477		
324.94	134.348	175.734	42.403	324.97	125.969	167.700	43.509		
349.95	139.895	183.446	42.648	349.97	130.790	174.794	43.759		

Time (seconds)	10 uM			Control			AMC		
	0.00	64.085	72.448	43.441	0.00	64.229	74.689	43.551	
24.96	68.370	78.932	44.274	24.94	69.735	81.651	44.617		
49.96	72.703	85.136	43.941	49.94	74.941	88.439	44.540		
74.96	78.006	91.252	43.835	74.94	80.168	95.303	44.544		
99.96	81.557	97.668	43.893	99.94	85.611	102.335	44.276		
124.96	85.631	104.227	43.823	124.94	91.071	108.879	44.219		
149.96	90.485	110.409	43.256	149.94	96.383	116.130	44.182		
174.96	94.781	116.803	43.309	174.94	101.817	123.147	44.093		
199.96	99.373	123.122	43.538	199.94	107.515	129.794	44.314		
224.97	103.871	129.885	43.388	224.95	112.441	137.017	44.218		
249.96	108.220	136.091	43.512	249.94	118.034	144.185	44.090		
274.96	112.764	142.338	43.348	274.94	123.547	151.134	44.600		
299.96	117.488	148.954	43.140	299.94	129.087	158.532	43.908		
324.96	121.797	155.504	43.222	324.94	134.587	165.233	44.111		
349.96	126.452	162.084	43.408	349.94	140.438	172.561	44.076		

	% Inhibition
Number of values	3
Mean	30.63
Std. Deviation	0.6807
Std. Error	0.3930
Sum	91.90



Best fit values	10 uM		Control	AMC
	Slope	0.1939 ± 0.0004041	0.2824 ± 0.0007757	-0.002004 ± 0.0003735
Y-intercept when X=0.0	63.32 ± 0.08308	75.55 ± 0.1595	44.21 ± 0.07881	
X-intercept when Y=0.0	-324.0	-267.5	20668	
1/slope	5.158	3.541	-488.9	
95% Confidence Intervals				
Slope	0.1930 to 0.1948	0.2807 to 0.2841	-0.002811 to -0.001198	
Y-intercept when X=0.0	62.84 to 63.00	75.21 to 75.90	44.05 to 44.38	
X-intercept when Y=0.0	-328.3 to -321.7	-270.3 to -264.9	15777 to 38801	
Goodness of Fit				
R square	0.9999	0.9999	0.6889	
Sy.x	0.1900	0.3345	0.1963	
Is slope significantly non-zero?				
F	220251	132032	28.79	
DFs, DFd	1.000, 13.000	1.000, 13.000	1.000, 13.000	
P value	< 0.0001	< 0.0001	0.0001	
Deviation from zero?	Significant	Significant	Significant	
Data				
Number of X values	15	15	15	
Maximum number of Y replicates	1	1	1	
Total number of values	15	15	15	
Number of missing values	0	0	0	

Best fit values	10 uM		Control	AMC
	Slope	0.1778 ± 0.0006521	0.2556 ± 0.0004939	-0.002033 ± 0.0005734
Y-intercept when X=0.0	63.95 ± 0.1341	72.25 ± 0.1015	43.91 ± 0.1179	
X-intercept when Y=0.0	-382.7	-282.7	21603	
1/slope	5.624	3.913	-482.0	
95% Confidence Intervals				
Slope	0.1764 to 0.1792	0.2545 to 0.2567	-0.002771 to -0.0007940	
Y-intercept when X=0.0	63.66 to 64.24	72.03 to 72.47	43.66 to 44.17	
X-intercept when Y=0.0	-364.0 to -355.4	-284.7 to -280.7	13490 to 55028	
Goodness of Fit				
R square	0.9998	1.000	0.4915	
Sy.x	0.2728	0.2066	0.2399	
Is slope significantly non-zero?				
F	74337	26782	12.57	
DFs, DFd	1.000, 13.000	1.000, 13.000	1.000, 13.000	
P value	< 0.0001	< 0.0001	0.0036	
Deviation from zero?	Significant	Significant	Significant	
Data				
Number of X values	15	15	15	
Maximum number of Y replicates	1	1	1	
Total number of values	15	15	15	
Number of missing values	0	0	0	

Best fit values	10 uM		Control	AMC
	Slope	0.2169 ± 0.0002517	0.2794 ± 0.0006306	-0.0009839 ± 0.0005337
Y-intercept when X=0.0	64.03 ± 0.1134	74.38 ± 0.1064	44.91 ± 0.1303	
X-intercept when Y=0.0	-295.2	-266.2	63857	
1/slope	4.611	3.579	-1441	
95% Confidence Intervals				
Slope	0.2157 to 0.2181	0.2780 to 0.2809	-0.002063 to 0.0008750	
Y-intercept when X=0.0	63.79 to 64.28	74.09 to 74.68	44.03 to 44.59	
X-intercept when Y=0.0	-297.9 to -292.6	-268.5 to -263.9	21598 to infinity	
Goodness of Fit				
R square	0.9999	0.9999	0.0843	
Sy.x	0.2398	0.2776	0.2651	
Is slope significantly non-zero?				
F	154540	177304	1.199	
DFs, DFd	1.000, 13.000	1.000, 13.000	1.000, 13.000	
P value	< 0.0001	< 0.0001	0.2994	
Deviation from zero?	Significant	Significant	Not Significant	
Data				
Number of X values	15	15	15	
Maximum number of Y replicates	1	1	1	
Total number of values	15	15	15	
Number of missing values	0	0	0	

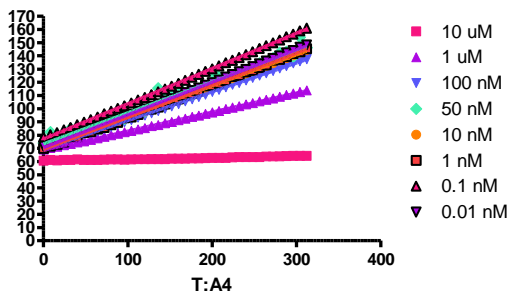
	E1	E2	E3	E4
Control	0.000	0.000	0.000	0.000
10 uM	0.301	0.314	0.304	0.224

FINAL CONDITIONS								
	EDTA	DTT	BRUJ	NaOAc	DMSO	Cruzain	SUBS	
	mM	mM	%	mm	%	nM	uM	
FC	1	2.5	0.01	100		2	0.1001	15

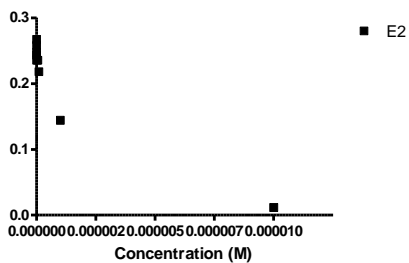
C.16. Representative Analysis of 47 as an Inactive Cruzain Inhibitor.

GDK-II-33

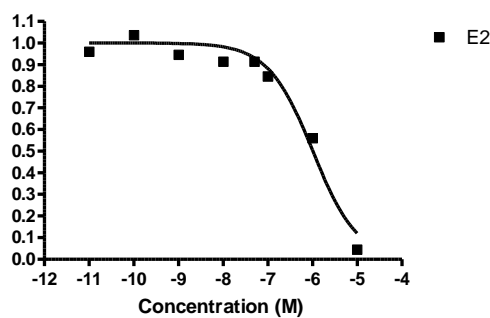
e2



E2 RESULT



Transform of E2 RESULT

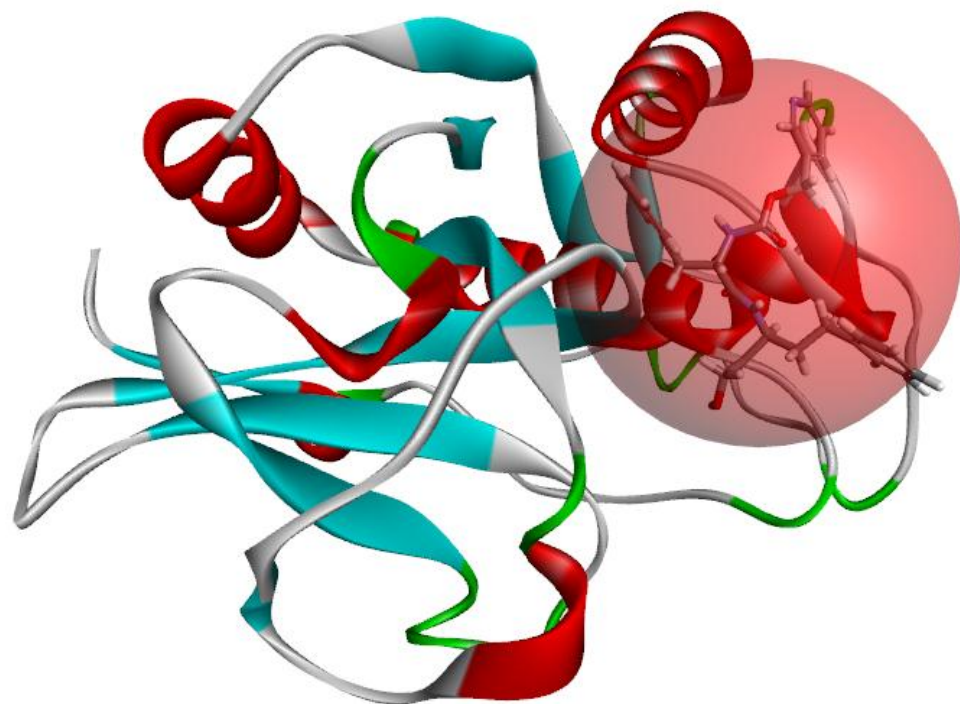


Concentration (M)	E2
0.000010	0.011490
0.000001	0.144200
1.000000e-007	0.217800
5.000000e-008	0.235400
1.000000e-008	0.235600
1.000000e-009	0.243500
1.000000e-010	0.267000
1.000000e-011	0.247300

Concentration (M)	E2
-5.000	0.045
-6.000	0.559
-7.000	0.845
-7.301	0.913
-8.000	0.914
-9.000	0.945
-10.000	1.036
-11.000	0.959

E2	
Sigmoidal dose-response (variable slope)	
Best-fit values	
Bottom	= 0.0
Top	= 1.000
LogEC50	-5.996
HillSlope	-0.8665
EC50	1.009e-006
Std. Error	
LogEC50	0.1002
HillSlope	0.1484
95% Confidence Intervals	
LogEC50	-6.241 to -5.751
HillSlope	-1.230 to -0.5033
EC50	5.739e-007 to 1.776e-006
Goodness of Fit	
Degrees of Freedom	6
R square	0.9720
Absolute Sum of Squares	0.02110
Sy,x	0.05930
Constraints	
Bottom	Bottom = 0.0
Top	Top = 1.000
Number of points Analyzed	8

C.17. Representative Analysis of **48** as an Inactive Cruzain Inhibitor



	-CDOCKER_ENERGY	-CDOCKER_INTERACTION_ENERGY	POSE_NUMBER	NumberOfTautomers	StereoisomerIndex	TautomerIndex	Ionization pH	Number
	71.3987	69.0827	1	1	4	1	7.5	4
	71.3014	69.3615	2	1	4	1	7.5	4
	71.0952	69.3003	3	1	4	1	7.5	4
	70.986	68.9851	4	1	4	1	7.5	4
	70.6787	68.4982	5	1	4	1	7.5	4
	70.0659	68.6286	6	1	4	1	7.5	4
	69.5473	68.1114	7	1	4	1	7.5	4
	69.1792	68.6784	8	1	4	1	7.5	4

C18. Validation of Cruzain Crystal Structure docked an Irreversible Inhibitor Using Discovery Studio 3.0

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